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The crystal structure of 1-(3-carboxypropyl)-1*H*-imidazole-3-oxide, C₇H₁₀N₂O₃

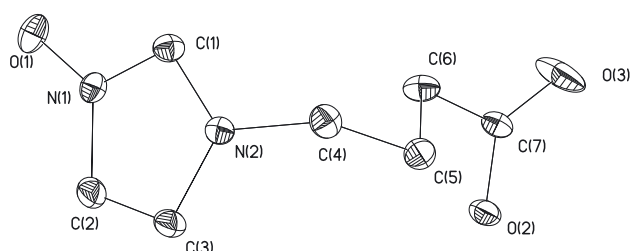


Table 1: Data collection and handling.

Crystal:	Colourless block
Size:	0.13 × 0.07 × 0.04 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	0.11 mm ⁻¹
Diffraction, scan mode:	Bruker D8 VENTURE, φ and ω
$\theta_{\max}$, completeness:	26.4°, 99 %
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	2664, 1316, 0.058
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$, 1235
$N(\text{param})_{\text{refined}}$:	114
Programs:	Olex2 [1], SHELX [2, 3], Bruker [4]

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Abstract

C₇H₁₀N₂O₃, monoclinic, *Cc* (no. 9), $a = 8.8611(4)$ Å, $b = 13.8991(7)$ Å, $c = 7.7194(5)$ Å, $\beta = 124.527(1)^\circ$, $V = 783.27(7)$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.0493$, $wR_{\text{ref}}(F^2) = 0.1257$, $T = 170$ K.

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The molecular structure is shown in the figure (hydrogen atoms are omitted for clarity). Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

1 Source of material

Add formaldehyde, glyoxal, butyric acid, and hydroxylamine to the reactor. The temperature of the reactor is 298 K. Turn on the mixer at a speed of 600 rpm. The raw materials

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.9469 (3)	0.6337 (2)	0.7812 (4)	0.0338 (7)
O2	0.1219 (4)	0.62114 (17)	0.6234 (4)	0.0339 (7)
N1	0.7800 (4)	0.5947 (2)	0.6450 (4)	0.0239 (6)
N2	0.4960 (3)	0.5796 (2)	0.3886 (4)	0.0218 (6)
O3	0.0543 (6)	0.7741 (2)	0.5381 (6)	0.0643 (11)
C2	0.7301 (5)	0.5017 (3)	0.6492 (6)	0.0311 (8)
H2A	0.806522	0.453029	0.746315	0.037*
C1	0.6374 (5)	0.6412 (2)	0.4857 (5)	0.0237 (7)
H1	0.635804	0.706355	0.447420	0.028*
C3	0.5515 (5)	0.4923 (3)	0.4891 (6)	0.0279 (8)
H3	0.478486	0.436109	0.453218	0.033*
C7	0.1333 (5)	0.7015 (2)	0.5454 (5)	0.0271 (8)
C6	0.2514 (5)	0.6957 (3)	0.4628 (5)	0.0300 (8)
H6A	0.378841	0.680930	0.580463	0.036*
H6B	0.251528	0.758970	0.404084	0.036*
C4	0.3087 (4)	0.6063 (2)	0.2164 (5)	0.0252 (7)
H4A	0.257573	0.555836	0.106612	0.030*
H4B	0.312133	0.667260	0.152359	0.030*
C5	0.1845 (5)	0.6186 (2)	0.2923 (5)	0.0260 (7)
H5A	0.175625	0.556382	0.348249	0.031*
H5B	0.060129	0.635977	0.170940	0.031*
H2	0.062 (6)	0.625 (4)	0.685 (7)	0.049 (15)*

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undergo condensation reaction in the reactor. After the reaction, a white solid is obtained. Dissolve the white solid in deionized water and pour it into an evaporating dish. Place the dish at room temperature to evaporate deionized water to obtain colorless block crystals.

2 Experimental details

Hydrogen atoms were placed in their geometrically idealized positions and constrained to ride on their parent atoms.

3 Comment

Imidazole compounds have always been a hot research topic in the field of organic chemistry [5–8]. The title compound is obtained by condensation reaction of formaldehyde, glyoxal, butyric acid, and hydroxylamine. The title compound is an imidazole oxide and has high research value as an organic intermediate [9–10].

The asymmetric unit of the title compound is 1-(3-carboxypropyl)-1H-imidazole-3-oxide molecule. The bond length and angle are within the expected range [11]. Almost all non hydrogen atoms in a molecule are in three planes. The seven atoms of O1, N1, C1, C2, N2, C3, and C4 are in the first plane. The four atoms C4, C5, C6, and C7 are in the second plane. The four atoms C6, C7, O2, and O3 are in the third plane. N1, C1, N2, C3, and C2 form imidazole structures similar to those reported in references.

Conflict of interest statement: The authors declare no conflicts of interest regarding this article.

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