

Yuan Xian-Zhu, Zhang Ding-Wa, Yi Xiu-Guang*, Zeng Hui-Lei and Yi Zhi-Qiang*

Synthesis and crystal structure of methyl 2-[[4-(4-cyclopropyl-1-naphthyl)-4H-1,2,4-triazole-3-yl]thio} acetate, C₁₈H₁₇N₃O₂S

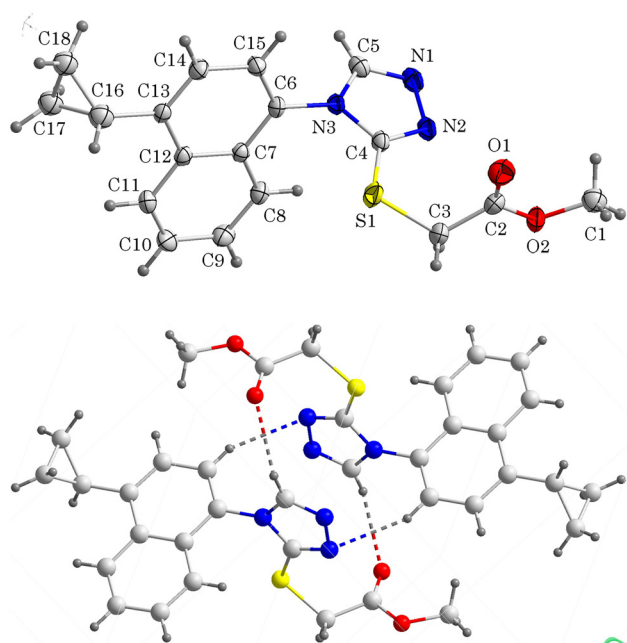


Table 1: Data collection and handling.

Crystal:	Yellow block
Size:	0.23 × 0.17 × 0.14 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	0.21 mm ⁻¹
Diffractometer, scan mode:	D8/APEX2, φ and ω
$\theta_{\max}$, completeness:	25.0°, >99 %
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	9313, 2906, 0.019
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$, 2430
$N(\text{param})_{\text{refined}}$:	226
Programs:	Bruker [1], SHELX [2, 3], Diamond [4]

Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

1 Source of materials

Intermediate 4-(4-cyclopropylnaphthalen-1-yl)-2,4-dihydro-3H-1,2,4-triazole-3-thione (45.0 g, 0.169 mol) and triethylamine (18.7 g, 0.183 mol) of were added into absolute ethanol (400 mL) at room temperature. Cooling to 0 °C, dropping 150 mL of absolute ethanol solution of ethyl bromoacetate (29.6 g, 0.18 mol), and controlling the temperature to 0 °C for 1 h after dropping. Most of the solvent was evaporated under reduced pressure, and the reaction solution was poured into 250 mL of water, extracted with dichloromethane (100 mL × 3), and the organic layers were combined. Wash with 200 mL water and 200 mL saturated sodium chloride solution, and dry the organic layer with anhydrous sodium sulfate. 52.4 g yellow oil product was obtained with a yield of 88.2 %.

2 Experimental details

All H atoms were included in calculated positions and refined as riding atoms, with C–H = 0.90–0.97 Å with $U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{C})$ for methyl H atoms and 1.2 $U_{\text{eq}}(\text{C})$ for all other H atoms.

<https://doi.org/10.1515/ncrs-2024-0056>

Received February 6, 2024; accepted March 20, 2024;

published online April 4, 2024

Abstract

C₁₈H₁₇N₃O₂S, monoclinic, $P2_1/c$ (no. 14), $a = 10.9713(5)$ Å, $b = 15.4445(8)$ Å, $c = 9.8221(4)$ Å, $\beta = 98.275(4)^\circ$, $V = 1646.99(13)$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.0401$, $wR_{\text{ref}}(F^2) = 0.0990$, $T = 296(2)$ K.

CCDC no.: 2341009

***Corresponding authors:** Yi Xiu-Guang, School of Chemistry and Chemical Engineering, Jinggangshan Univeristy, Ji'an, Jiangxi, 343709, P.R. China, E-mail: jayxgggchem@163.com; and Yi Zhi-Qiang, Ji'an Central People's Hospital, Ji'an, Jiangxi, 343700, P.R. China, E-mail: yzqqq2024@163.com. <https://orcid.org/0009-0006-8649-9616> (Y. Zhi-Qiang)

Yuan Xian-Zhu, Taihe County Maternal and Child Health Hospital in Ji'an City, Ji'an, Jiangxi, 343700, P.R. China

Zhang Ding-Wa, School of Chemistry and Chemical Engineering, Jinggangshan Univeristy, Ji'an, Jiangxi, 343709, P.R. China

Zeng Hui-Lei, Ji'an Central People's Hospital, Ji'an, Jiangxi, 343700, P.R. China

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	x	y	z	<i>U</i> _{iso} */ <i>U</i> _{eq}
C1	1.1379 (2)	0.1369 (2)	1.5252 (3)	0.0835 (9)
H1A	1.1233	0.1180	1.6145	0.125*
H1B	1.1824	0.0930	1.4838	0.125*
H1C	1.1853	0.1894	1.5342	0.125*
C2	1.0259 (2)	0.17954 (14)	1.3117 (2)	0.0481 (5)
C3	0.8980 (2)	0.2046 (2)	1.2460 (2)	0.0571 (6)
C4	0.85857 (17)	0.07197 (13)	1.06705 (18)	0.0389 (4)
C5	0.83748 (17)	−0.05848 (13)	0.9907 (2)	0.0430 (5)
H5	0.8198	−0.1059	0.9329	0.052*
C6	0.75820 (16)	0.05789 (12)	0.82003 (18)	0.0349 (4)
C7	0.63682 (15)	0.09122 (11)	0.81232 (17)	0.0322 (4)
C8	0.56984 (17)	0.08977 (12)	0.92520 (19)	0.0396 (5)
H8	0.6058	0.0666	1.0088	0.048*
C9	0.45342 (18)	0.12197 (14)	0.9127 (2)	0.0470 (5)
H9	0.4103	0.1200	0.9876	0.056*
C10	0.39791 (18)	0.15789 (14)	0.7888 (2)	0.0500 (5)
H10	0.3188	0.1807	0.7822	0.060*
C11	0.45914 (17)	0.15966 (13)	0.6774 (2)	0.0432 (5)
H11	0.4212	0.1838	0.5954	0.052*
C12	0.57983 (16)	0.12522 (11)	0.68472 (18)	0.0342 (4)
C13	0.64333 (16)	0.12171 (11)	0.56667 (18)	0.0349 (4)
C14	0.75834 (17)	0.08656 (13)	0.58105 (19)	0.0417 (5)
H14	0.7990	0.0834	0.5044	0.050*
C15	0.81785 (17)	0.05486 (13)	0.70790 (19)	0.0414 (5)
H15	0.8970	0.0321	0.7149	0.050*
C16	0.58272 (18)	0.15701 (13)	0.43202 (18)	0.0414 (5)
H16	0.5770	0.2203	0.4283	0.050*
C17	0.4766 (2)	0.11063 (16)	0.3497 (2)	0.0597 (6)
H17A	0.4519	0.0554	0.3840	0.072*
H17B	0.4098	0.1454	0.3029	0.072*
C18	0.5974 (2)	0.11506 (18)	0.2987 (2)	0.0682 (7)
H18A	0.6040	0.1524	0.2207	0.082*
H18B	0.6461	0.0625	0.3019	0.082*
H3A	0.834 (2)	0.1770 (16)	1.292 (2)	0.071 (8)*
H3B	0.894 (2)	0.2632 (18)	1.252 (3)	0.075 (8)*
N1	0.88388 (15)	−0.06394 (12)	1.11983 (18)	0.0506 (5)
N2	0.89793 (15)	0.02069 (12)	1.16988 (16)	0.0467 (4)
N3	0.81796 (13)	0.02530 (10)	0.95072 (15)	0.0373 (4)
O1	1.11751 (16)	0.18538 (12)	1.26099 (17)	0.0684 (5)
O2	1.02109 (14)	0.15273 (11)	1.43938 (14)	0.0606 (4)
S1	0.86185 (6)	0.18445 (4)	1.06464 (5)	0.0578 (2)

3 Comment

Lesinurad sodium is the first inhibitor of uric acid selective reabsorption transporter-1 (URAT1), which was first developed by Aread Biosciences and then continued to be developed by Astra Zeneca [5]. The chemical name of lesinurad sodium is 2-[[5-bromo-4-(4-cyclopropylnaphthalene-1-yl)-4H-1,2,4-triazol-3-yl]thio} acetic acid, which is mainly used for treating

hyperuricemia related to gout in combination with xanthine oxidase inhibitors (such as allopurinol and febuxostat) [6–8]. Optimizing the synthesis process of lesinurad sodium and its derivatives has attracted the attention of researchers [9–11]. Recently, we have been focusing on the synthesis of esinurad sodium and optimizing its synthesis process. Here we reveal the synthesis and crystal structure of methyl 2-[[4-(4-cyclopropyl-1-naphthyl)-4H-1,2,4-triazole-3-yl]thio} acetate, the key intermediate of lesinurad sodium.

In the molecules of the title structure bond lengths and angles are very similar to those given in the literature [12–15]. In the title structure, the dihedral angle formed by the naphthalene ring plane, the 1,2,4-triazole-3-thione plane and the cyclopropyl group plane are 54.6°, 71.5(73)° and 86.9°, respectively. These parameters are different from the data reported before [13]. The dihedral angle between the 1,2,4-triazole-3-thione plane and the ester plane to which it is connected is 89.0(213)°. The torsion angles of N2–C4–S1–C3 and C4–S1–C3–C2 are −12.798(207)° and 69.286(186)°. Inter-molecular C–H...O hydrogen bonds and C–H...N hydrogen bonds connect two adjacent compounds into a dimer.

Acknowledgment: X-ray data were collected at Instrumental Analysis Center Nanchang Hangkong University, Nanchang, 330063, People's Republic of China.

Author contribution: All the authors have accepted responsibility for the entire content of this submitted manuscript and approved submission.

Research funding: This research was supported by the National Natural Science Foundation of China (22168018), the National Natural Science Foundation of Jiangxi, China (20232BAB203048; 20202BABL206111), The Science and Technology Plan Project Fund of Jiangxi Provincial Health Planning Commission (202411004), the Science and Technology Plan Project of Ji'an (20233–043398, 20233–043995).

Conflict of interest: The authors declare no conflicts of interest regarding this article.

References

1. Bruker. *APEX2, SAINT and SADABS*; Bruker AXS Inc.: Madison, Wisconsin, USA, 2009.
2. Sheldrick G. M. A short history of *SHELX*. *Acta Crystallogr.* 2008, *A64*, 112–122.
3. Sheldrick G. M. Crystal structure refinement with *SHELXL*. *Acta Crystallogr.* 2015, *C71*, 3–8.
4. Brandenburg K. *DIAMOND. Visual Crystal Structure Information System. Ver. 4.0*; Crystal Impact: Bonn, Germany, 2015.
5. Hoy S. M. Lesinurad: first global approval. *Drugs* 2016, *76*, 509–516.

6. Sattui S. E., Gaffo A. L. Treatment of hyperuricemia in gout: current therapeutic options, latest developments and clinical implications. *Ther. Adv. Musculoskel. Dis.* 2016, 8, 145–159.
7. Shen Z., Yeh L. T., Wallach K., Zhu N., Kerr B., Gillen M. *In vitro* and *in vivo* interaction studies between lesinurad, a selective urate reabsorption inhibitor, and major liver or kidney transporters. *Clin. Drug Invest.* 2016, 36, 443–452.
8. Miner J. N., Tan P. K., Hyndman D., Liu S., Iverson C., Nanavati P., Hagerty D. T., Manhard K., Shen Z., Girardet J. L., Yeh L. T., Terkeltaub R., Quart B. Lesinurad, a novel, oral compound for gout, acts to decrease serum uric acid through inhibition of urate transporters in the kidney. *Arthritis Res. Ther.* 2016, 18, 214–236.
9. Zou L., Liu Y., Yao K., Li J. Q., Zhang Z. X. Synthesis of lesinurad. *Chin. J. Pharm.* 2017, 48, 488–491.
10. Pan W., Wan T., Jiang N., Gong P., Zhai X. Improved synthetic process of lesinurad sodium. *Chin. J. Med. Chem.* 2019, 29, 363–367.
11. Chen J., Su W. Q., Chen N. G., Wang Y. M. Study on synthesis process of lesinurad. *Chin. J. Guangzhou Chem. Ind.* 2023, 51, 102–104.
12. Drumright R. E., Mas R. H., Merola J. S., Tanko J. M. Interplay between conjugative and steric effects in cyclopropylarenes. *J. Org. Chem.* 1990, 55, 4098–4102.
13. Zeng H. L., He Y., Zhang D. W., Yi X. G., Yi Z. Q. Synthesis and crystal structure of 4-(4-cyclopropyl-naphthalen-1-yl)-2,4-dihydro-3H-1,2,4-triazole-3-thione, $C_{15}H_{13}N_3S$. *Z. Kristallogr. N. Cryst. Struct.* 2024, 239, 415–417.
14. Palanisamy V., Sanphui P., Prakash M., Chernyshev V. Multicomponent solid forms of the uric acid reabsorption inhibitor lesinurad and cocrystal polymorphs with urea: DFT simulation and solubility study. *Acta Crystallogr.* 2019, C75, 1102–1117.
15. Terruzzi S., Bellomi S., Marras G., Barreca G., Ventimiglia G., Cervellino A., Masciocchi N. Disclosing the rich crystal chemistry of lesinurad by ab initio laboratory X-ray powder diffraction methods. *Cryst. Growth Des.* 2018, 18, 6863–6872.