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The crystal structure of poly[di(μ_2 -aqua)-diaqua-bis(3-aminopyridine-4-carboxylate- $\kappa^2O: O'$)-tetra(μ_2 -3-aminopyridine-4-carboxylate- $\kappa^2O: O'$)-dineodymium(III), $[\text{Nd}_2(\text{C}_6\text{H}_5\text{N}_2\text{O}_2)_6(\text{H}_2\text{O})_4]_n$

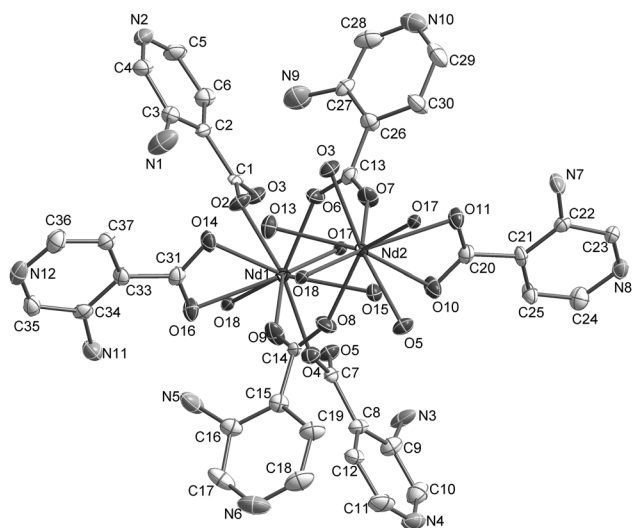


Table 1: Data collection and handling.

Crystal:	Colorless block
Size:	0.20 × 0.18 × 0.10 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	2.66 mm ⁻¹
Diffractometer, scan mode:	Bruker APEX-II, φ and ω
θ_{max} , completeness:	27.6°, >99 %
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	24,277, 9122, 0.063
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 6092
$N(\text{param})_{\text{refined}}$:	605
Programs:	Bruker [1], Diamond [2], Olex2 [3], SHELX [4, 5]

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Abstract

$[\text{Nd}_2(\text{C}_6\text{H}_5\text{N}_2\text{O}_2)_6(\text{H}_2\text{O})_4]_n$ monoclinic, $P2_1/c$ (no. 14), $a = 9.383(3)$ Å, $b = 22.741(6)$ Å, $c = 18.775(5)$ Å, $\beta = 90.231(5)^\circ$, $V = 4006(2)$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.0427$, $wR_{\text{ref}}(F^2) = 0.0862$, $T = 296$ K.

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The molecular structure is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

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1 Source of materials

3-Aminoisonicotinic acid (0.6 mmol 0.083 g), Nd_2O_3 (0.2 mmol 0.067 g), and 0.4 mol L^{-1} NaOH solution (1 mL), were dissolved in deionized water (7 mL). The mixture was sealed in a 25-mL Teflon-lined steel autoclave, and heated at 453 K for 8 days, then slowly cooled to room temperature. Colorless crystals of title complex could be found in the product.

2 Experimental details

All hydrogen atoms were placed in calculated positions and refined as riding atoms. The values were set to be $1.2U_{\text{eq}}$ or $1.5U_{\text{eq}}$ of the parent atoms.

3 Comment

Lanthanide complexes of 3-aminoisonicotinic acid have attracted some attentions during recent decades [6–11]. However, transition metal complexes for this ligand are relatively rare. A complex $[\text{Nd}_2(\text{C}_6\text{H}_5\text{N}_2\text{O}_2)_6(\text{H}_2\text{O})_4]_n$ of 3-aminoisonicotinic acid has been prepared under solvothermal reaction conditions. The asymmetric unit of title complex contains one Nd^{1+} ion, one Nd^{2+} ion, six L^- ($\text{HL} = 3\text{-aminoisonicotinic acid}$) ligands and four coordinated aqua molecules. Each Nd^{1+} ion is nine-coordinated by

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$).

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Nd1	0.49211 (3)	0.35460 (2)	0.25256 (2)	0.01435 (8)
Nd2	1.04449 (3)	0.39830 (2)	0.26271 (2)	0.01375 (8)
O17	0.3219 (3)	0.43644 (14)	0.28824 (19)	0.0190 (8)
H17A	0.332005	0.444117	0.332611	0.028*
H17B	0.336265	0.465857	0.263161	0.028*
O2	0.3641 (4)	0.32326 (16)	0.3573 (2)	0.0227 (9)
O3	0.1471 (4)	0.36036 (16)	0.3721 (2)	0.0223 (9)
O4	0.3895 (4)	0.38804 (16)	0.14489 (19)	0.0219 (9)
O5	0.1729 (4)	0.42775 (16)	0.1527 (2)	0.0240 (9)
O6	0.6358 (4)	0.38627 (16)	0.3514 (2)	0.0248 (9)
O7	0.8481 (4)	0.42707 (17)	0.3392 (2)	0.0288 (10)
O8	0.8946 (4)	0.36458 (17)	0.1701 (2)	0.0280 (10)
O9	0.6796 (4)	0.32430 (17)	0.1688 (2)	0.0305 (10)
O10	0.9188 (4)	0.48624 (16)	0.2102 (2)	0.0278 (10)
O11	1.0712 (4)	0.50565 (15)	0.2961 (2)	0.0239(10)
O18	1.2009 (3)	0.31453 (14)	0.22682 (19)	0.0196 (8)
H18A	1.190447	0.306198	0.183953	0.029*
H18B	1.183987	0.282718	0.252823	0.029*
O13	0.9118 (4)	0.30711 (15)	0.3059 (2)	0.0242 (9)
H13A	0.966630	0.278136	0.303797	0.036*
H13B	0.827120	0.305276	0.287157	0.036*
O14	0.6150 (4)	0.26242 (15)	0.3004 (2)	0.0250 (9)
O15	0.6233 (3)	0.44527 (15)	0.2172 (2)	0.0222 (9)
H15A	0.663738	0.471007	0.240164	0.033*
H15B	0.611437	0.451907	0.171814	0.033*
N1	0.4474 (5)	0.2346 (2)	0.4431 (3)	0.0498 (17)
H1A	0.500895	0.206943	0.459449	0.060*
H1B	0.463952	0.249337	0.401782	0.060*
N2	0.2119 (5)	0.2420 (2)	0.5932 (3)	0.0292 (12)
N3 ^a	0.1029 (9)	0.5117 (4)	0.0600 (4)	0.036 (3)
H3A ^a	0.055540	0.540824	0.042599	0.044*
H3B ^a	0.082424	0.498311	0.101494	0.044*
N4	0.3488 (5)	0.4930 (2)	−0.0866 (3)	0.0318 (13)
N5	0.6720 (5)	0.2251 (2)	0.0826 (3)	0.0416 (15)
H5A	0.653680	0.234067	0.126133	0.050*
H5B	0.633608	0.194446	0.063716	0.050*
N6	0.8905 (6)	0.2677 (3)	−0.0666 (3)	0.0455 (16)
N7	1.1154 (5)	0.61980 (19)	0.3327 (3)	0.0278 (12)
H7A	1.135478	0.584279	0.344883	0.033*
H7B	1.156977	0.648773	0.353677	0.033*
N8	0.8775 (5)	0.70374 (19)	0.2196 (3)	0.0258 (12)
N9	0.5324 (5)	0.3836 (3)	0.4886 (3)	0.0489 (16)
H9A	0.474985	0.374649	0.522530	0.059*
H9B	0.533580	0.362783	0.450327	0.059*
N10	0.6961 (6)	0.5074 (2)	0.5742 (3)	0.0402 (15)
N11	0.3938 (5)	0.1409 (2)	0.1704 (3)	0.0329 (13)
H11A	0.349967	0.114278	0.146312	0.039*
H11B	0.375878	0.177462	0.162787	0.039*
N12	0.6206 (5)	0.0448 (2)	0.2743 (3)	0.0293 (12)
C1	0.2514 (5)	0.3300 (2)	0.3918 (3)	0.0157 (12)
C2	0.2404 (6)	0.2990 (2)	0.4618 (3)	0.0185 (12)
C3	0.3371 (6)	0.2548 (2)	0.4824 (3)	0.0209 (13)
C4	0.3159 (6)	0.2291 (3)	0.5487 (3)	0.0269 (14)
H4	0.380522	0.200365	0.562882	0.032*

Table 2: (continued)

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C5	0.1210 (6)	0.2839 (3)	0.5719 (3)	0.0296 (15)
H5	0.046426	0.293797	0.602089	0.036*
C6	0.1314 (6)	0.3127 (3)	0.5087 (3)	0.0253 (14)
H6	0.065431	0.341665	0.496927	0.030*
C7	0.2865 (6)	0.4183 (2)	0.1197 (3)	0.0179 (12)
C8	0.3049 (6)	0.4428 (2)	0.0468 (3)	0.0204 (13)
C9	0.2134 (6)	0.4862 (3)	0.0205 (3)	0.0272 (14)
H9 ^b	0.135727	0.499106	0.046892	0.033*
C10	0.2420 (7)	0.5094 (3)	−0.0459 (4)	0.0351 (16)
H10	0.181564	0.538634	−0.063087	0.042*
C11	0.4343 (6)	0.4511 (3)	−0.0616 (3)	0.0348 (16)
H11	0.509144	0.438386	−0.090008	0.042*
C12	0.4173 (6)	0.4258 (3)	0.0041 (3)	0.0243 (14)
H12 ^a	0.481224	0.397333	0.019876	0.029*
C13	0.7369 (6)	0.4174 (2)	0.3733 (3)	0.0190 (12)
C14	0.7974 (6)	0.3347 (2)	0.1421 (3)	0.0174 (12)
C15	0.8287 (5)	0.3098 (2)	0.0700 (3)	0.0200 (12)
C16	0.7620 (6)	0.2595 (3)	0.0435 (3)	0.0243 (13)
C17	0.7982 (7)	0.2409 (3)	−0.0258 (4)	0.0373 (17)
H17	0.753572	0.207506	−0.043628	0.045*
C18	0.9557 (7)	0.3152 (3)	−0.0399 (4)	0.0413 (17)
H18	1.022357	0.334542	−0.067942	0.050*
C19	0.9289 (6)	0.3366 (3)	0.0267 (3)	0.0323 (15)
H19	0.978297	0.369447	0.043111	0.039*
C20	0.9793 (6)	0.5221 (2)	0.2500 (3)	0.0197 (13)
C21	0.9451 (5)	0.5865 (2)	0.2433 (3)	0.0192 (13)
C22	1.0182 (6)	0.6301 (2)	0.2805 (3)	0.0206 (13)
C23	0.9805 (6)	0.6886 (2)	0.2635 (3)	0.0234 (13)
H23	1.032371	0.718655	0.284970	0.028*
C24	0.8056 (6)	0.6609 (3)	0.1858 (3)	0.0281 (14)
H24	0.731770	0.670806	0.154862	0.034*
C25	0.8390 (5)	0.6025 (2)	0.1960 (3)	0.0221 (13)
H25	0.789617	0.573695	0.170927	0.027*
C26	0.7216 (5)	0.4468 (2)	0.4443 (3)	0.0221 (13)
C27	0.6210 (5)	0.4304 (2)	0.4954 (3)	0.0197 (13)
C28	0.6141 (7)	0.4621 (3)	0.5581 (3)	0.0361 (16)
H28	0.546609	0.450637	0.591448	0.043*
C29	0.7917 (8)	0.5230 (3)	0.5259 (4)	0.0449 (19)
H29	0.851187	0.554587	0.536135	0.054*
C30	0.8077 (6)	0.4950 (3)	0.4614 (3)	0.0343 (16)
H30	0.875733	0.508203	0.429238	0.041*
C31	0.5475 (6)	0.2296 (2)	0.2586 (3)	0.0200 (13)
O16	0.4587 (4)	0.25024 (15)	0.2148 (2)	0.0241 (10)
C33	0.5704 (5)	0.1648 (2)	0.2622 (3)	0.0190 (12)
C34	0.4915 (6)	0.1252 (2)	0.2208 (3)	0.0199 (13)
C35	0.5183 (6)	0.0655 (3)	0.2320 (3)	0.0278 (14)
H35	0.460816	0.038411	0.208341	0.033*
C36	0.6973 (6)	0.0833 (3)	0.3112 (4)	0.0305 (15)
H36	0.769784	0.069407	0.340718	0.037*
C37	0.6739 (6)	0.1422 (2)	0.3075 (3)	0.0221 (13)
H37	0.727830	0.167549	0.335649	0.027*
N3A ^b	0.5295 (12)	0.3863 (5)	0.0206 (6)	0.030 (4)
H3AA ^b	0.597268	0.380750	−0.009397	0.036*
H3AB ^b	0.529864	0.367833	0.060574	0.036*

^aOccupancy: 0.596(9), ^bOccupancy: 0.404(9).

four oxygen atoms of four bridging L^- ligands, two chelating oxygen atoms of one L^- ligand and three oxygen atoms of three coordinated aqua molecules. The environment of the Nd^{2+} ion is similar to Nd^{3+} . Each of $\text{Nd}^{2+}(\text{III})$ ions is surrounded by two chelating oxygen atoms of a L^- ligand, four oxygen atoms of four μ_2 -bridging L^- ligands and three water oxygen atoms, forming a nine-coordination geometry, too. The Nd–O distances vary from 2.359(4) to 2.918(3) Å [12–14]. The O–Nd–O angles are in the range 51.35(12)–145.45(13)°. Among the six L^- ligands of the asymmetric unit, four act as μ_2 -bridges, linking $\text{Nd}^{2+}(\text{III})$ and $\text{Nd}^{2+}(\text{III})$ ions, and two chelate to $\text{Nd}^{2+}(\text{III})$ ions. While two of the four coordinated aqua molecules also play the roles of bridges, connecting $\text{Nd}^{2+}(\text{III})$ and $\text{Nd}^{2+}(\text{III})$ ions, and the other two act as monodentate ligands.

Each $\text{Nd}^{2+}(\text{III})$ ion is connected to one $\text{Nd}^{2+}(\text{III})$ ion via two μ_2 - L^- and two water molecules, and to another $\text{Nd}^{2+}(\text{III})$ ion by two μ_2 - L^- - L^- ligands, forming the 1-D zigzag-chain structure of the title complex with nonbonding Nd···Nd separations of 4.32–5.28 Å and Nd–Nd–Nd angles of 155.408(11)°. The zigzag chains stack parallel along the *b* axis, constructing the 3-D architecture of the title complex.

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Author contributions: All the authors have accepted responsibility for the entire content of this submitted manuscript and approved submission.

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References

1. Bruker. SAINT, APEX3 and SADABS; Bruker AXS Inc.: Madison, WI, USA, 2013.
2. Brandenburg K. *DIAMOND. Visual Crystal Structure Information System. Ver. 4.6.8*; Crystal Impact GbR: Bonn, Germany, 2022.
3. Dolomanov O. V., Bourhis L. J., Gildea R. J., Howard J. A. K., Puschmann H. OLEX2: a complete structure solution, refinement and analysis program. *J. Appl. Crystallogr.* 2009, 42, 339–341.
4. Sheldrick G. M. Crystal structure refinement with SHELXL. *Acta Crystallogr.* 2015, C71, 3–8.
5. Sheldrick G. M. SHELXTL - integrated space-group and crystal-structure determination. *Acta Crystallogr.* 2015, A71, 3–8.
6. Xue X., Shao Z., Cui Y., Wen T., Chen J., Geng K., Mi L., Hou H. Water-stable amino-functionalized coordination polymer for efficient Hg^{2+} capture. *Cryst. Growth Des.* 2022, 22, 1412–1420.
7. Ghosh D., Fukushima T., Kobayashi K., Sen S., Kitagawa S., Katod T., Tanaka K. Base assisted C–C coupling between carbonyl and polypyridyl ligands in a Ru–NADH-type carbonyl complex. *Dalton Trans.* 2017, 46, 4373–4381.
8. Ruan M., Li A., Wen Y., Zhou L., Zhang J., Xuan X. Adenine-based Bio-MOFs with high water and acid? Base stability for ammonia capture. *CrystEngComm* 2022, 24, 7420–7426.
9. Gantzer N., Kim M.-B., Robinson A., Terban M. W., Ghose S., Dinnebier R. E., York A. H., Tiana D., Simon C. M., Thallapally P. K. Computation-informed optimization of Ni(PyC)2 functionalization for noble gas separations. *Cell. Rep. Phys. Sci.* 2022, 3, 101025.
10. Alduhaish O., Li B., Arman H., Lin R.-B., Zhao J. C.-G., Chen B. A Two-dimensional microporous metal-organic framework for highly selective adsorption of carbon dioxide and acetylene. *Chinese Chem. Lett.* 2017, 28, 1653–1658.
11. Liu F.-F., Ping Z.-Y., Zhou W.-W., Zhao W. The crystal structure of poly $[(\mu_2\text{-3-Aminopyridine-4-Carboxylate-}\kappa^2\text{Nu: O})_2\text{Zinc(II)}]_n$, $[\text{Zn}(\text{C}_6\text{H}_5\text{N}_2\text{O}_2)_2]_n$. *Z. Kristallogr. N. Cryst. Struct.* 2023, 238, 1131–1132.
12. Tiihonen A., Lahtinen M. In-depth structural analysis of lanthanoid coordination networks based on a flexible tripodal zwitterionic isonicotinate ligand. *CrystEngComm* 2019, 21, 2286–2302.
13. Mondal K. C., Sengupta O., Dutta P., Seehra M., Nayak S. K., Mukherjee P. S. Three-dimensional 3d-4f heterometallic polymers containing both azide and carboxylate as co-LIGANDS. *Inorg. Chim. Acta* 2009, 362, 1913–1917.
14. Feng R., Chen L., Chen Q.-H., Shan X.-C., Gai Y.-L., Jiang F.-L., Hong M.-C. Novel luminescent three-dimensional heterometallic complexes with 2-fold interpenetrating (3,6)-connected nets. *Cryst. Growth Des.* 2011, 11, 1705–1712.