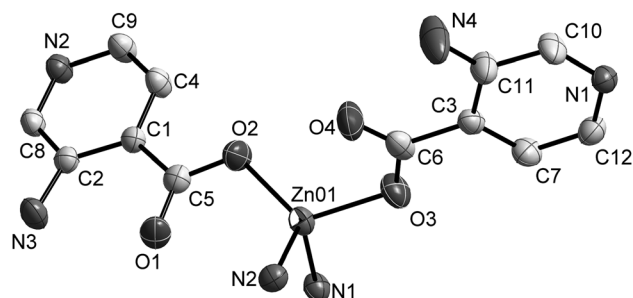


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# The crystal structure of poly [bis( $\mu_2$ -3-aminopyridine-4-carboxylato- $\kappa^2N:O$ ) Zinc(II)], $[\text{Zn}(\text{C}_6\text{H}_5\text{N}_2\text{O}_2)_2]_n$



**Table 1:** Data collection and handling.

|                                                                            |                                                   |
|----------------------------------------------------------------------------|---------------------------------------------------|
| Crystal:                                                                   | Yellow block                                      |
| Size:                                                                      | 0.18 × 0.15 × 0.15 mm                             |
| Wavelength:                                                                | Mo K $\alpha$ radiation (0.71073 Å)               |
| $\mu$ :                                                                    | 1.88 mm <sup>-1</sup>                             |
| Diffractometer, scan mode:                                                 | Bruker APEX-II, $\varphi$ and $\omega$            |
| $\theta_{\text{max}}$ , completeness:                                      | 27.6°, >99 %                                      |
| $N(hkl)_{\text{measured}}$ , $N(hkl)_{\text{unique}}$ , $R_{\text{int}}$ : | 8073, 3023, 0.046                                 |
| Criterion for $I_{\text{obs}}$ , $N(hkl)_{\text{gt}}$ :                    | $I_{\text{obs}} > 2\sigma(I_{\text{obs}})$ , 2424 |
| $N(\text{param})_{\text{refined}}$ :                                       | 191                                               |
| Programs:                                                                  | Bruker [1], Diamond [2], Olex2 [3], SHELX [4, 5]  |

<https://doi.org/10.1515/ncrs-2023-0358>

Received August 2, 2023; accepted September 5, 2023;

published online September 19, 2023

## Abstract

$[\text{Zn}(\text{C}_6\text{H}_5\text{N}_2\text{O}_2)_2]_n$ , orthorhombic,  $P2_12_12_1$  (no. 19),  $a = 7.6139(7)$  Å,  $b = 12.4890(13)$  Å,  $c = 13.9382(13)$  Å,  $V = 1325.4(2)$  Å<sup>3</sup>,  $Z = 4$ ,  $R_{\text{gt}}(F) = 0.0391$ ,  $wR_{\text{ref}}(F^2) = 0.0710$ ,  $T = 296$  K.

CCDC no.: 2285875

Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

## 1 Source of materials

A mixture of  $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$  (0.1 mmol 0.030 g), 3-aminoisonicotinic acid (0.4 mmol 0.055 g),  $\text{H}_2\text{O}$  (8 mL) was sealed in a 25-mL Teflon-lined steel autoclave, and heated at 453 K for 10 days, then slowly cooled to room temperature. Yellow block crystals of title complex could be seen in the product.

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## 2 Experimental details

All hydrogen atoms were placed in calculated positions and refined as riding atoms. The values were set to be 1.2  $U_{\text{eq}}$  of the parent atoms.

## 3 Comment

The complexes of isonicotinic acid and its derivatives have attracted a lot of attentions [6–10]. However, coordination of 3-aminoisonicotinic acid has been less studied. A complex  $[\text{Zn}(\text{C}_6\text{H}_5\text{N}_2\text{O}_2)_2]_n$  of 3-aminoisonicotinic acid has been prepared under hydrothermal reaction conditions. The asymmetric unit of title complex contains one  $\text{Zn}^{2+}$  ion and two  $\text{L}^-$  anions ( $\text{HL} = 3\text{-aminoisonicotinic acid}$ ). Each  $\text{Zn}(\text{II})$  ion has a distorted tetrahedron coordination (see the figure), formed by two carboxylate oxygen atoms and two pyridyl nitrogen atoms of four different bridging  $\text{L}^-$  ligands. The Zn–O distances are nearly equivalent spanning a narrow range of 1.950(3)–1.953(3) Å, and the Zn–N distances vary from 2.027(4) to 2.061(4) Å [11–13]. The N–Zn–O angles are in the range 98.53(15)–119.14(16)°, the N–Zn–N angle is 107.28(15)°, and the O–Zn–O angle is 117.19(15)°. All  $\text{L}^-$  ligands act as  $\mu_2$ -bridges, linking two  $\text{Zn}(\text{II})$  ions.

Each  $\text{Zn}(\text{II})$  ions is connected to four adjacent  $\text{Zn}(\text{II})$  ions via four  $\mu_2$   $\text{L}^-$  ligands, forming the 3-D polymeric structure of the title complex. Identical adamantanoid cages can be found in the structure. In the cages, the Zn–Zn edges are

**Table 2:** Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ ).

| Atom | x           | y           | z           | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|-------------|-------------|-------------|----------------------------------|
| Zn01 | 0.40340 (6) | 0.55320 (4) | 0.72064 (4) | 0.02899 (15)                     |
| O1   | 0.2312 (5)  | 0.6572 (3)  | 0.8818 (2)  | 0.0409 (9)                       |
| O2   | 0.2883 (4)  | 0.6927 (2)  | 0.7283 (3)  | 0.0404 (8)                       |
| O3   | 0.5196 (5)  | 0.5171 (3)  | 0.6003 (3)  | 0.0447 (10)                      |
| N1   | 0.9465 (5)  | 0.4970 (3)  | 0.3320 (3)  | 0.0294 (10)                      |
| O4   | 0.6962 (5)  | 0.6497 (3)  | 0.6392 (3)  | 0.0545 (11)                      |
| N2   | −0.1933 (4) | 0.9498 (3)  | 0.7895 (3)  | 0.0314 (8)                       |
| N3   | −0.0059 (6) | 0.7805 (3)  | 0.9737 (3)  | 0.0405 (11)                      |
| H3A  | −0.073203   | 0.799180    | 1.020349    | 0.049*                           |
| H3B  | 0.075379    | 0.733814    | 0.982894    | 0.049*                           |
| C1   | 0.0739 (6)  | 0.7973 (3)  | 0.8055 (3)  | 0.0249 (10)                      |
| C2   | −0.0277 (6) | 0.8248 (4)  | 0.8858 (3)  | 0.0268 (11)                      |
| C3   | 0.7595 (6)  | 0.5489 (5)  | 0.4982 (3)  | 0.0297 (10)                      |
| C4   | 0.0406 (5)  | 0.8518 (4)  | 0.7211 (4)  | 0.0327 (10)                      |
| H4   | 0.109485    | 0.837361    | 0.667545    | 0.039*                           |
| C5   | 0.2073 (6)  | 0.7102 (4)  | 0.8071 (3)  | 0.0296 (11)                      |
| C6   | 0.6543 (7)  | 0.5755 (5)  | 0.5859 (4)  | 0.0359 (13)                      |
| C7   | 0.7158 (7)  | 0.4618 (4)  | 0.4418 (3)  | 0.0362 (12)                      |
| H7   | 0.621118    | 0.419031    | 0.459420    | 0.043*                           |
| C8   | −0.1603 (6) | 0.9004 (4)  | 0.8724 (3)  | 0.0307 (12)                      |
| H8   | −0.230379   | 0.917560    | 0.924867    | 0.037*                           |
| C9   | −0.0902 (6) | 0.9264 (3)  | 0.7138 (3)  | 0.0355 (11)                      |
| H9   | −0.108348   | 0.961438    | 0.655730    | 0.043*                           |
| C10  | 0.9884 (7)  | 0.5818 (4)  | 0.3831 (4)  | 0.0424 (14)                      |
| H10  | 1.080351    | 0.624565    | 0.361673    | 0.051*                           |
| C11  | 0.9015 (8)  | 0.6114 (4)  | 0.4684 (3)  | 0.0444 (13)                      |
| C12  | 0.8088 (6)  | 0.4369 (4)  | 0.3607 (3)  | 0.0366 (12)                      |
| H12  | 0.776369    | 0.377456    | 0.324578    | 0.044*                           |
| N4   | 0.9611 (8)  | 0.6992 (4)  | 0.5176 (4)  | 0.106 (3)                        |
| H4A  | 0.983279    | 0.749859    | 0.477434    | 0.127*                           |
| H4B  | 0.879417    | 0.719749    | 0.556284    | 0.127*                           |

8.7976(9) or 8.8427(9) Å long, the Zn–Zn–Zn angles vary between 90.437(6) and 126.638(7)°, and the maximal dimensions of the cage is 22.8417(22) Å. Large cavities can be found in the structure of title complex, inducing 3-fold interpenetration of the networks. The separated distances from one net to other interpenetrating one is 7.6139(10) Å.

**Research ethics:** Not applicable.

**Author contributions:** The authors have accepted responsibility for the entire content of this manuscript and approved its submission.

**Competing interests:** The authors state no conflict of interest.

**Research funding:** NSF (Natural Science Foundation) of Anhui Province (2108085MB53), the NSF for Distinguished Young Scholars of Anhui University (2022AH020087), University NSF of Anhui Province (KJ2020A0647), Research Program of Huainan Normal University (2020XJYB003), Project of Huainan City and Program for Innovation Research Team in Huainan Normal University (XJTD202006).  
**Data availability:** The raw data can be obtained on request from the corresponding author.

## References

1. Bruker. SAINT, APEX3 and SADABS; Bruker AXS Inc.: Madison, WI, USA, 2013.
2. Brandenburg K. *DIAMOND*. Visual Crystal Structure Information System. Ver. 4.6.8; Crystal Impact GbR: Bonn, Germany, 2022.
3. Dolomanov O. V., Bourhis L. J., Gildea R. J., Howard J. A. K., Puschmann H. OLEX2: a complete structure solution, refinement and analysis program. *J. Appl. Cryst.* 2009, 42, 339–341.
4. Sheldrick G. M. Crystal structure refinement with SHELXL. *Acta Crystallogr.* 2015, C71, 3–8.
5. Sheldrick G. M. *SHELXTL* – integrated space-group and crystal-structure determination. *Acta Crystallogr.* 2015, A71, 3–8.
6. Briones D., Fernández B., Calahorra A. J., Fairen-Jimenez D., Sanz R., Martínez F., Orcajo G., Sebastián E. S., Seco J. M., González C. S., Llopis J., Rodríguez-Diéguez A. Highly active anti-diabetic metal-organic framework. *Cryst. Growth Des.* 2016, 16, 537–540.
7. Yang E., Li H.-Y., Wang F., Yang H., Zhang J. Enhancing CO<sub>2</sub> adsorption enthalpy and selectivity via amino functionalization of a tetrahedral framework material. *CrystEngComm* 2013, 15, 658–661.
8. Wang F., Tan Y.-X., Yang H., Kang Y., Zhang J. Open diamondoid amino-functionalized MOFs for CO<sub>2</sub> capture. *Chem. Commun.* 2012, 48, 4842–4844.
9. Conesa-Egea J., Nogal N., Martínez J. I., Fernández-Moreira V., Rodríguez-Mendoza U. R., González-Platas J., Gómez-García C. J., Delgado S., Zamora F., Amo-Ochoa P. Smart composite films of nanometric thickness based on copper-iodine coordination polymers. Toward sensors. *Chem. Sci.* 2018, 9, 8000–8010.
10. Jeon Y., Lee M., Kim H., Park K. Crystal structure of tetraaquabis(4-carboxypyridine-2,6-dicarboxylato- $\kappa^3\text{N},\text{O},\text{O}'$ ) ( $\mu_2$ -4,4'-bipyridine- $\kappa^2\text{N}:\text{N}'$ ) dizinc(II) dihydrate,  $[\text{Zn}_2(\text{C}_8\text{H}_3\text{NO}_6)_2(\text{C}_{10}\text{H}_8\text{N}_2)(\text{H}_2\text{O})_4]\cdot 2\text{H}_2\text{O}$ ,  $\text{C}_{26}\text{H}_{26}\text{N}_4\text{O}_{18}\text{Zn}_2$ . *Z. Kristallogr. N. Cryst. Struct.* 2015, 230, 213–214.
11. Shao Z., Yu C., Xie Q., Wu Q., Zhao Y., Hou H. Porous functionalized MOF self-evolution promoting molecule encapsulation and Hg<sup>2+</sup> removal. *Chem. Commun.* 2019, 55, 13382–13385.
12. Xue X., Shao Z., Cui Y., Wen T., Chen J., Geng K., Mi L., Hou H. Water-stable amino-functionalized coordination polymer for efficient Hg<sup>2+</sup> capture. *Cryst. Growth Des.* 2022, 22, 1412–1420.
13. Tarushi A., Kastanias F., Psychars V., Raptoulou C. P., Psomas G., Kessissoglou D. P. A [24–MC-6] zinc metallacoronate with a nonsteroidal antiinflammatory drug as the constructing ligand. *Inorg. Chem.* 2012, 51, 7460–7462.