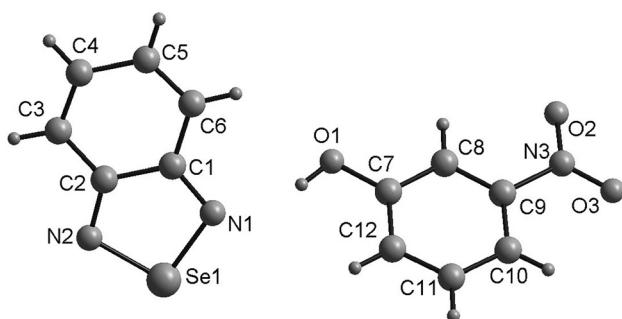


Jian-Ge Wang*

Crystal structure of 3-nitrophenol-2,1,3-benzoselenadiazole (1/1), $C_{12}H_9N_3O_3Se$

**Table 1:** Data collection and handling.

Crystal:	Colorless block
Size:	0.25 × 0.24 × 0.21 mm
Wavelength:	Mo $K\alpha$ radiation (0.71073 Å)
μ :	3.04 mm ⁻¹
Diffractometer, scan mode:	SuperNova, ω
θ_{\max} , completeness:	28.4°, >99%
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	9024, 2709, 0.033
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$, 2124
$N(\text{param})_{\text{refined}}$:	173
Programs:	CrysAlis ^{Pro} [1], SHELX [2, 3], Olex2 [4]

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Abstract

$C_{12}H_9N_3O_3Se$, monoclinic, $P2_1/c$ (no. 14), $a = 3.9106(3)$ Å, $b = 23.5453(14)$ Å, $c = 13.4611(8)$ Å, $\beta = 92.697(6)$ °, $V = 1238.07(14)$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.0432$, $\omega R_{\text{ref}}(F^2) = 0.0786$, $T = 290$ K.

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The asymmetric unit of the title crystal structure is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

1 Source of materials

All reagents and solvents were used as obtained without further purification. The methanol solution (10 mL) of 3-nitrophenol (0.10 mmol, 0.0139 g) was slowly added to another methanol solution (10 mL) of 2,1,3-benzoselenadiazole (0.1 mmol, 0.0183 g). The mixture was stirred for 2 h at room temperature. The solution was filtered, and the filtrate was kept at the room temperature. After 1 week colorless crystals of the title compound were obtained.

2 Experimental details

The hydrogen atoms were assigned with common isotropic displacement factors $U_{\text{iso}}(\text{H}) = 1.2$ times U_{eq} (C, aromatic ring), $U_{\text{iso}}(\text{H}) = 1.5$ times $U_{\text{eq}}(\text{OH})$. The final refinement by using geometrical restraints, with C–H = 0.93 Å (aromatic ring), O–H = 0.82 Å (OH).

3 Comment

With their unique structures and interesting physico-chemical properties, supramolecular compounds attract much attention [5–7]. In general, the different components are linked together by one or more intermolecular interactions such as hydrogen bonds, forming high ordered supramolecular cocrystal structures [8–10]. As known, the nitrogen atoms of polyaza heteroaromatic compounds can serve as weak base to accept protons from carboxylic acid to form acid-base conjugate pair incorporating charge transfer interactions, which enable the potential characteristic of the construction of cocrystals. In this paper, we report the reaction product of the nitrogen heterocyclic compound 2,1,3-benzoselenadiazole together with 3-nitrophenol.

The asymmetric unit of the title structure contains one 2,1,3-benzoselenadiazole and one 3-nitrophenol. The dimers of two 2,1,3-benzoselenadiazole molecules are formed by a pair of N–Se···N chalcogen bonds. And the adjacent dimers are connected by O–H···N hydrogen bonds and N–Se···O chalcogen bonds to form a two dimensional

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Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	x	y	z	<i>U</i> _{iso} */* <i>U</i> _{eq}
Se1	0.44256 (9)	0.55392 (2)	0.89226 (2)	0.04105 (13)
N1	0.2846 (7)	0.55800 (11)	0.7653 (2)	0.0421 (7)
N2	0.3121 (7)	0.48169(11)	0.90576 (19)	0.0395 (7)
C1	0.1481 (8)	0.50795 (13)	0.7420 (2)	0.0345 (7)
C2	0.1642 (8)	0.46575 (13)	0.8197 (2)	0.0326 (7)
C3	0.0239 (9)	0.41080 (14)	0.7998 (3)	0.0444 (8)
H3	0.0307	0.3831	0.8491	0.053*
C4	-0.1188 (9)	0.39948 (15)	0.7089 (3)	0.0477 (9)
H4	-0.2091	0.3636	0.6959	0.057*
C5	-0.1344 (9)	0.44102 (15)	0.6326 (3)	0.0479 (9)
H5	-0.2365	0.4318	0.5709	0.057*
C6	-0.0055 (9)	0.49346 (15)	0.6471 (3)	0.0453 (9)
H6	-0.0162	0.5200	0.5958	0.054*
O1	0.5223 (7)	0.61149 (10)	0.59512 (18)	0.0574 (7)
H1	0.4605	0.6031	0.6506	0.086*
O2	0.8942 (8)	0.73963 (12)	0.3359 (2)	0.0691 (8)
O3	0.7645 (8)	0.82264 (11)	0.3821 (2)	0.0724 (9)
N3	0.7669 (8)	0.77224 (14)	0.3933 (2)	0.0489 (8)
C7	0.4997 (8)	0.66826 (13)	0.5824 (2)	0.0353 (7)
C8	0.6323(8)	0.69083 (13)	0.4970 (2)	0.0362 (8)
H8	0.7305	0.6674	0.4505	0.043*
C9	0.6148 (8)	0.74837 (13)	0.4830 (2)	0.0338 (7)
C10	0.4733 (9)	0.78504 (15)	0.5490 (3)	0.0455 (9)
H10	0.4659	0.8240	0.5376	0.055*
C11	0.3425 (9)	0.76131 (16)	0.6331 (3)	0.0490 (9)
H11	0.2439	0.7848	0.6793	0.059*
C12	0.3546 (8)	0.70396 (15)	0.6501 (2)	0.0416 (8)
H12	0.2651	0.6890	0.7073	0.050*

network. Bond lengths and angles are all in the expected ranges [11].

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