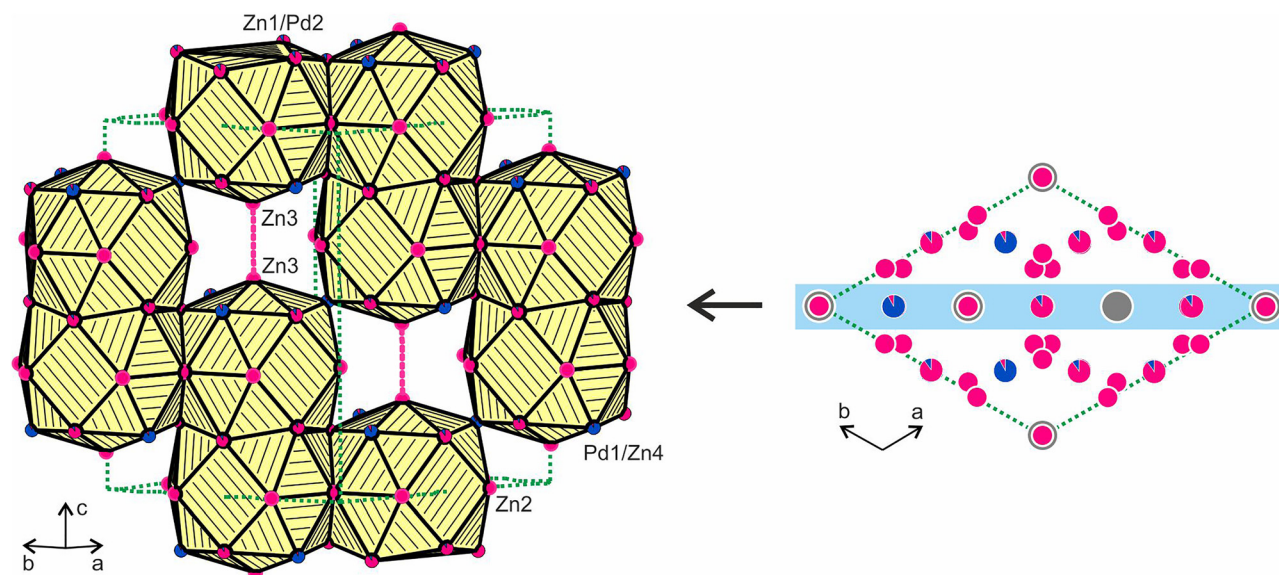


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Crystal structure of $\text{Eu}_2\text{Pd}_{3.37(1)}\text{Zn}_{13.63(1)}$



Projection of the $\text{Eu}_2\text{Pd}_{3.37}\text{Zn}_{13.63}$ structure onto the xy plane (right). Europium, palladium and zinc atoms are drawn as medium grey, blue and magenta circles, respectively. Mixed occupied sites are indicated by segments. A cutout with one row of condensed $\text{Eu}@\text{(Pd/Zn)}_{19}$ polyhedra is shown left.

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Abstract

$\text{Eu}_2\text{Pd}_{3.37}\text{Zn}_{13.63}$, trigonal, $R\bar{3}m$ (no. 166), $a = 9.1693(7)$ Å, $c = 13.2443(13)$ Å, $V = 964.34(14)$ Å³, $Z = 3$, $R_{\text{gt}}(F) = 0.0268$, $wR_{\text{ref}}(F^2) = 0.0287$, $T = 293$ K.

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Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

Source of material

Crystals of the title compound $\text{Eu}_2\text{Pd}_{3.37}\text{Zn}_{13.63}$ were obtained by induction melting of the elements (3:4:12 atomic ratio) in a

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sealed tantalum ampoule in a water-cooled sample chamber. Starting materials were europium ingots (American Elements, 99.99%), palladium sheets (Agosi, 99.99%) and zinc granules (99.9%, Merck). The sample was first heated to 1473 K, kept for 5 min and quenched. The ampoule was then sealed in an evacuated silica tube, heated to 1223 K with a heating rate of 100 K h^{−1}. The sample was kept at 1223 K for 24 h followed by slow cooling (1 K h^{−1}) to room temperature. $\text{Eu}_2\text{Pd}_{3.37}\text{Zn}_{13.63}$ is light grey with metallic luster and stable in air.

Experimental details

The $\text{Eu}_2\text{Pd}_{3.37}\text{Zn}_{13.63}$ crystals were selected from the mechanically fragmented sample of the nominal composition

Table 1: Data collection and handling.

Crystal:	Grey block
Size:	0.030 × 0.040 × 0.075 mm
Wavelength:	Mo Kα radiation (0.71073 Å)
μ:	39.0 mm ^{−1}
Diffractionmeter, scan mode:	IPDS Stoe, ω scans
θ _{max} , completeness:	33.3°, >99%
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	9175, 488, 0.037
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 417
$N(\text{param})_{\text{refined}}$:	26
Programs:	X-Area [1], JANA2006 [2], SUPERFLIP [3, 4]

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$).

Atom	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Eu1	0	0	0.34506(2)	0.01304(7)
Pd1 ^a	0.5	0	0.5	0.01283(13)
Zn1 ^b	0.5	0	0.5	0.01283(13)
Zn2 ^c	0.50154(2)	0.49846(2)	0.15132(4)	0.01548(15)
Pd2 ^d	0.50154(2)	0.49846(2)	0.15132(4)	0.01548(15)
Zn3	0.29377(5)	0	0	0.01484(12)
Zn4	0	0	0.10343(5)	0.01629(16)

^aOccupancy: 0.917(10), ^bOccupancy: 0.083(10), ^cOccupancy: 0.897(8),^dOccupancy: 0.103(8).

$3\text{Eu}:4\text{Pd}:12\text{Zn}$. The crystal quality was first tested through Laue photographs (Buerger camera, image plate detection system). Single crystal X-ray diffraction was performed at room temperature on a Stoe IPDS-II diffractometer (graphite monochromatized MoK_α radiation; oscillation mode). A numerical absorption correction was applied. The starting atomic parameters were deduced with the charge-flipping algorithm [3] implemented in Superflip [4] and the structure was refined on F^2 with the Jana2006 software package [2], with anisotropic displacement parameters for all atoms. Separate refinement of the occupancy parameters indicated small degrees of Pd/Zn mixing for the 9d and 18h sites, which were refined as least-squares variables in the final cycles. The $\text{Eu}_2\text{Pd}_{3.37}\text{Zn}_{13.63}$ crystal is a reverse/obverse twin with a domain ratio of 0.971(4):0.029(4).

Comment

$\text{Eu}_2\text{Pd}_{3.37}\text{Zn}_{13.63}$ crystallizes with the $\text{Th}_2\text{Zn}_{17}$ type structure [5], space group $R\bar{3}m$ and completes the series of $\text{RE}_2\text{Pd}_{3-x}\text{Zn}_{14+x}$ compounds [6]. Comparison with the $\text{RE}_2\text{Pd}_{3-x}\text{Zn}_{14+x}$ phases reveals two peculiarities: (i) the solid solution of the europium compound tends towards higher palladium content and (ii) the cell volume is larger than that of the lanthanum representative, indicating divalent europium. The $\text{Th}_2\text{Zn}_{17}$ type structure has four crystallographically independent zinc sites and allows for different coloring variants (Table 3). Typical binary examples are $\text{Ce}_2\text{Fe}_{17}$ [7], $\text{Zr}_2\text{Be}_{17}$ [8] or $\text{Ba}_2\text{Mg}_{17}$ [9]. Ternary ordered variants have been observed for $\text{Ce}_2\text{Co}_{15}\text{Al}_2$ [7] and $\text{Gd}_2\text{Co}_3\text{Zn}_{14}$ [6]. $\text{Eu}_2\text{Pd}_{3.37}\text{Zn}_{13.63}$ is close to the latter ordering variant, however, the 9d and 18h sites still show small degrees of Pd/Zn and Zn/Pd mixing, respectively. Such small homogeneity ranges have been observed for most of the many ternary $\text{Th}_2\text{Zn}_{17}$ type

Table 3: Examples of coloring variants for $\text{Th}_2\text{Zn}_{17}$ type representatives (space group $R\bar{3}m$, Pearson code hR57 and Wyckoff sequence hfd c^2).

Compound	6c	6c	9d	18f	18h	Ref.
$\text{Th}_2\text{Zn}_{17}$	Th	Zn	Zn	Zn	Zn	[5]
$\text{Ce}_2\text{Fe}_{17}$	Ce	Fe	Fe	Fe	Fe	[7]
$\text{Zr}_2\text{Be}_{17}$	Zr	Be	Be	Be	Be	[8]
$\text{Ba}_2\text{Mg}_{17}$	Ba	Mg	Mg	Mg	Mg	[9]
$\text{Ce}_2\text{Co}_{15}\text{Al}_2$	Ce	Al	Co	Co	Co	[7]
$\text{Gd}_2\text{Co}_3\text{Zn}_{14}$	Gd	Zn	Co	Zn	Zn	[6]

representatives [10]. The europium atoms in the $\text{Eu}_2\text{Pd}_{3.37}\text{Zn}_{13.63}$ structure have the coordination number 19 by palladium and zinc. These $\text{Eu}(\text{Pd}/\text{Zn})_{19}$ polyhedra are condensed via common hexagon faces and these double units are stacked in ABC sequence (rhombohedral space group symmetry). The coordination polyhedron around europium is a slightly distorted hexa-capped hexagonal prisms with an additional Zn3 atom. This coordination reminds of the CaCu_5 type [11], one of the basic binary intermetallic structure types. Indeed, the $\text{Th}_2\text{Zn}_{17}$ type derives from the CaCu_5 structure [12] by an ordered replacement of 2/3 of the calcium atoms by europium and of 1/3 by a Zn_3 dumb-bell while the remaining palladium and zinc atoms replace the copper atoms. The Zn_3 dumb-bells are emphasized in the figure between adjacent $\text{Eu}(\text{Pd}/\text{Zn})_{19}$ polyhedra. The Zn3–Zn3 distances of 274 pm are close to the Zn–Zn distances in *hcp* zinc (six times 266 and six times 291 pm) [13]. The Pd–Zn distances within the three-dimensional $[\text{Pd}_{3.37}\text{Zn}_{13.63}]$ network of $\text{Eu}_2\text{Pd}_{3.37}\text{Zn}_{13.63}$ range from 260 to 278 pm, which is slightly longer than the sum of the covalent radii [14] of 253 pm for Pd + Zn. This is similar to the series of dimorphic equiatomic REPdZn phases [15].

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