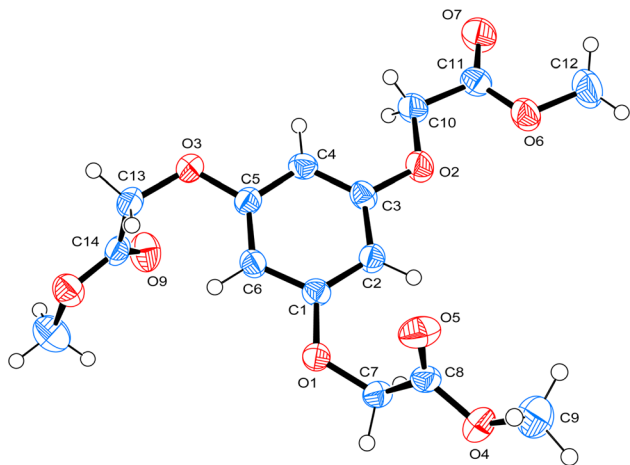


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The crystal structure of trimethyl 2,2',2''-(benzene-1,3,5-triyltris(oxy))triacetate, C₁₅H₁₈O₉



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Abstract

C₁₅H₁₈O₉, triclinic, $P\bar{1}$ (no. 2), $a = 8.4962(7)$ Å, $b = 8.6835(7)$ Å, $c = 11.9767(10)$ Å, $\alpha = 90.9650(10)^\circ$, $\beta = 94.9460(10)^\circ$, $\gamma = 115.9310(10)^\circ$, $V = 790.21(11)$ Å³, $Z = 2$, $R_{gt}(F) = 0.0318$, $wR_{ref}(F^2) = 0.0855$, $T = 296(2)$ K.

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The molecular structure is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

Source of material

All chemicals were purchased from commercial sources and used without further purification. The 2,2',2''-(benzene-

Table 1: Data collection and handling.

Crystal:	Colorless block
Size:	0.25 × 0.23 × 0.21 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	0.12 mm ⁻¹
Diffraction, scan mode:	Bruker APEX-II, φ and ω
$\theta_{\max}$, completeness:	26.0°, >99%
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	8728, 3099, 0.015
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$, 2828
$N(\text{param})_{\text{refined}}$:	221
Programs:	Bruker [1], SHELX [2, 3], WinGX/ORTEP [4]

1,3,5-triyltris(oxy))triacetic acid (5 mmol, 1.501 g) was dissolved in anhydrous methanol (50 mL), then add 5 drops of anhydrous tin tetrachloride. The mixture was refluxed for 8 h. When the reaction vessel had cooled to room temperature, the reaction mixture was filtered. The solvent was removed by evaporation *in vacuo*, the title product was obtained by recrystallization from methanol.

Experimental details

All H-atoms bonded to C atoms were placed geometrically and refined using a riding model with common isotropic displacement factors $U_{\text{iso}}(\text{H}) = 1.2$ or $1.5 U_{\text{eq}}(\text{parent C-atom})$.

Comment

The phenoxyacetic acid and its esters are plant growth regulators and herbicides [5, 6]. They are mainly used as compounds, which can be modified to find new active compounds. Unlike benzoic acid and fatty acids, phenoxyacetic acid derivatives have flexible and varied coordination modes and spatial conformations, and other structural characteristics [7, 8]. Thus, the synthesis and crystal structure study of the title compound are of great significance.

Single-crystal structure analysis revealed that the title compound crystallized in the triclinic space group $P\bar{1}$. The ORTEP diagram is presented in Figure. Bond lengths and angles are all in the expected ranges. The bond length of O1–C7, O2–C10, O3–C13 are 1.4149(14), 1.4115(15),

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Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	x	y	z	U _{iso} [*] /U _{eq}
O1	0.52503 (11)	0.70930 (10)	0.50102 (7)	0.0397 (2)
O2	0.63951 (13)	1.14824 (11)	0.24293 (8)	0.0482 (3)
O3	1.02660 (11)	1.24728 (11)	0.57614 (7)	0.0421 (2)
O4	0.22899 (12)	0.39083 (12)	0.29436 (8)	0.0477 (2)
O5	0.52030 (12)	0.54099 (12)	0.29972 (9)	0.0508 (3)
O6	0.54858 (14)	1.21785 (12)	0.04808 (8)	0.0508 (3)
O7	0.74230 (16)	1.49418 (13)	0.06792 (9)	0.0649 (3)
O8	0.96858 (13)	1.12425 (12)	0.85781 (8)	0.0462 (2)
O9	0.86020 (14)	1.27747 (14)	0.75727 (8)	0.0555 (3)
C1	0.62078 (15)	0.87017 (14)	0.46557 (9)	0.0315 (2)
C2	0.57818 (15)	0.93173 (14)	0.36800 (9)	0.0339 (3)
H2	0.477872	0.864181	0.320249	0.041*
C3	0.68998 (16)	1.09846 (15)	0.34291 (9)	0.0342 (3)
C4	0.83886 (16)	1.20146 (14)	0.41271 (10)	0.0352 (3)
H4	0.912146	1.312084	0.394558	0.042*
C5	0.87659 (15)	1.13509 (15)	0.51142 (9)	0.0328 (3)
C6	0.77001 (15)	0.97087 (14)	0.53843 (9)	0.0336 (3)
H6	0.796996	0.927772	0.604089	0.040*
C7	0.36280 (15)	0.59877 (15)	0.43992 (10)	0.0353 (3)
H7A	0.299643	0.664766	0.416927	0.042*
H7B	0.292133	0.512744	0.488577	0.042*
C8	0.38512 (15)	0.51064 (14)	0.33782 (10)	0.0341 (3)
C9	0.2291 (3)	0.2954 (2)	0.19511 (14)	0.0704 (5)
H9A	0.262084	0.370214	0.134353	0.106*
H9B	0.113549	0.203994	0.175545	0.106*
H9C	0.311793	0.248377	0.209148	0.106*
C10	0.73863 (18)	1.32012 (15)	0.21726 (11)	0.0418 (3)
H10A	0.861718	1.344456	0.219399	0.050*
H10B	0.728460	1.396129	0.273453	0.050*
C11	0.67627 (17)	1.35388 (16)	0.10326 (11)	0.0410 (3)
C12	0.4900 (2)	1.2438 (2)	−0.06459 (13)	0.0650 (4)
H12A	0.431381	1.316162	−0.060333	0.097*
H12B	0.410085	1.135098	−0.101503	0.097*
H12C	0.589792	1.297453	−0.106345	0.097*
C13	1.06662 (15)	1.19376 (16)	0.68125 (10)	0.0385 (3)
H13A	1.189042	1.265329	0.707941	0.046*
H13B	1.050442	1.076366	0.672214	0.046*
C14	0.95169 (15)	1.20471 (15)	0.76702 (10)	0.0373 (3)
C15	0.8712 (2)	1.1341 (2)	0.94821 (13)	0.0658 (4)
H15A	0.910775	1.251990	0.973129	0.099*
H15B	0.889932	1.071978	1.009410	0.099*
H15C	0.748106	1.084864	0.922318	0.099*

1.4145(14) Å, respectively, which are similar to those reported in the literature [9, 10].

Molecules of the title compound form a set of weak C–H···O intermolecular hydrogen bonds in the crystal phase (Table 3). The geometric characteristics of these hydrogen bonds are close that does not allow to separate out a main structural motif of the crystal packing.

Table 3: Symmetry operations and geometric characteristics of inter-molecular hydrogen bonds.

Hydrogen bond	Symmetry operation	Geometrical characteristics		
		H···A, Å	D···A, Å	D–A···, Å
C2–H···O9	1 – x, 2 – y, 1 – z	2.65	3.529 (1)	157
C7–H7a···O9	1 – x, 2 – y, 1 – z	2.57	3.379 (2)	141
C9–H9b···O8	1 – x, 1 – y, 1 – z	2.64	3.300 (2)	127
C10–H10b···O5	x, 1 + y, z	2.62	3.385 (2)	137
C12–H12a···O7	1 – x, 3 – y, –z	2.65	3.604 (3)	175
C13–H13a···O5	2 – x, 2 – y, 1 – z	2.31	3.233 (1)	158

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Conflict of interest statement: The authors declare no conflicts of interest regarding this article.

References

- [1] Bruker. *SAINT*, *APEX2* and *SADABS*; Bruker AXS Inc.: Madison, WI, USA, 2012.
- [2] Sheldrick G. M. *SHELXTL* – integrated space-group and crystal-structure determination. *Acta Crystallogr.* 2015, *A71*, 3–8.
- [3] Sheldrick G. M. Crystal structure refinement with SHELXL. *Acta Crystallogr.* 2015, *C71*, 3–8.
- [4] Farrugia L. J. WinGX and ORTEP for Windows: an update. *J. Appl. Cryst.* 2012, *45*, 849–854.
- [5] Drzewiecka-Antonik A., Ferenc W., Mirosław B., Osypiuk D., Sarzyński J. Structure, thermal stability and magnetic behavior of Mn(II) complexes with phenoxyacetic acid herbicides. *Polyhedron* 2021, *207*, 115370.
- [6] de Assis Alves T., Fontes Pinheiro P., Miranda Praça-Fontes M., Fonseca Andrade-Vieira L., Barelo Corrêa K., de Assis Alves T., da Cruz F. A., Lacerda Júnior V., Ferreira A., Bastos Soares T. C. Toxicity of thymol, carvacrol and their respective phenoxyacetic acids in *Lactuca sativa* and *Sorghum bicolor*. *Ind. Crops Prod.* 2018, *114*, 59–67.
- [7] Xie Y.-Q., Zhang Y.-M., Li Z.-H., Yao H., Wei T.-B., Shi B.-B., Qu W.-J., Lin Q. Synthesis, crystal structure of a novel metal–organic framework and its catalyzing properties on the selective oxidation of cyclohexene to cyclohexenone. *Inorg. Chim. Acta* 2021, *525*, 120494.
- [8] Chen Z., Gao D.-L., Diao C.-H., Liu Y., Ren J., Chen J., Zhao B., Shi W., Cheng P. Syntheses, structures tuned by 4,4'-bipyridine and magnetic properties of a series of transition metal compounds containing o-carboxylphenoxyacetate acid. *Cryst. Growth Des.* 2012, *12*, 1201–1211.
- [9] Li Y., Qiu D., Gu R., Wang J., Shi J., Li Y. Aryne 1,2,3-trifunctionalization with aryl allyl sulfoxides. *J. Am. Chem. Soc.* 2016, *138*, 10814–10817.
- [10] Gopalakrishnan T., Rao L. M. (*p*-Methylphenoxy)acetic acid. *Acta Crystallogr.* 1981, *B37*, 2085–2087.