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The crystal structure of aqua-bis{2-bromo-6-((2-(2-phenylacetyl)hydrazineylidene)methyl)phenolato- $\kappa^3 N, O, O'$ }-dimethylformamide- $\kappa^1 O$ -erbium(III) chloride – dimethylformamide – water (1/2/1), $C_{39}H_{49}N_7O_9Br_2ClEr$

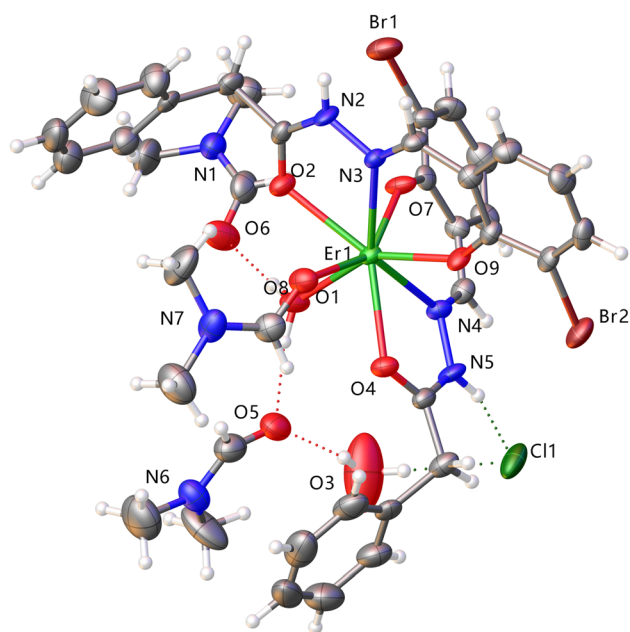


Table 1: Data collection and handling.

Crystal:	Colourless block
Size:	0.14 × 0.11 × 0.09 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	2.46 mm ⁻¹
Diffractometer, scan mode:	Bruker SMART CCD 6000, ω
$\theta_{\max}$, completeness:	25.1°, 99%
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	44,757, 6893, 0.064
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$, 5870
$N(\text{param})_{\text{refined}}$:	536
Programs:	Bruker [1], Olex2 [2], SHELX [3], Diamond [4]

The molecular structure is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

1 Source of materials

The title Er(III) complex was synthesized using the following one-pot method: 0.1502 g phenacetic acid hydrazide (0.1 mmol), 0.2010 g 3-bromo-2-hydroxybenzaldehyde (1.0 mmol), 0.1909 g $ErCl_3 \cdot 6H_2O$ (0.50 mmol), and 0.040 g NaOH (1.0 mmol) were added to the 25 mL ethanol–water–dimethylformamide (v:v:v = 3:1:1) solution with stirring. The mixture was continued to react for 6 h at 80 °C and then cooled to room temperature. After filtering the mixture, the filtrate was allowed to slowly volatilization at room temperature. The yellow block crystals of the title compound were received in 30 days. **Anal. Calcd.** for $C_{39}H_{49}N_7O_9Br_2ClEr$: C, 41.70; H, 4.37; N, 8.73. Found: C, 41.89; H, 4.01; N, 8.52.

2 Experimental details

The hydrogen atoms were positioned geometrically (C–H = 0.93–0.96 Å, N–H = 0.86 Å and O–H = 0.85 Å). Their U_{iso} values were set to $1.2U_{\text{eq}}$ or $1.5U_{\text{eq}}$ of the parent atoms.

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Abstract

$C_{39}H_{49}N_7O_9Br_2ClEr$, triclinic, $P\bar{1}$ (no. 2), $a = 12.3164(9)$ Å, $b = 12.7581(9)$ Å, $c = 14.5023(11)$ Å, $\alpha = 70.873(2)^\circ$, $\beta = 80.281(2)^\circ$, $\gamma = 65.810(2)^\circ$, $V = 1962.4(3)$ Å³, $Z = 1$, $R_g(F) = 0.0383$, $wR_{\text{ref}}(F^2) = 0.0765$, $T = 296.15$ K.

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Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$).

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U_{iso}^*/U_{eq}
C1	0.8902 (4)	0.1835 (4)	0.4217 (3)	0.0281 (11)
C01B	0.7476 (6)	1.1388 (6)	−0.1487 (4)	0.0561 (17)
C2	0.9117 (5)	0.0809 (4)	0.3945 (4)	0.0346 (13)
C3	1.0195 (5)	0.0324 (5)	0.3457 (4)	0.0445 (15)
H3	1.036039	−0.035343	0.326178	0.053*
C4	1.0997 (5)	0.0847 (5)	0.3270 (4)	0.0433 (14)
H4	1.172532	0.052292	0.295603	0.052*
C5	1.0726 (5)	0.1873 (5)	0.3550 (4)	0.0364 (13)
H5	1.128520	0.222604	0.340741	0.044*
C6	0.8223 (6)	0.0295 (5)	0.4192 (5)	0.0517 (16)
H6	0.835014	−0.037817	0.400443	0.062*
C7	0.7223 (6)	0.0763 (5)	0.4680 (5)	0.0535 (17)
H7	0.666556	0.040719	0.483303	0.064*
C8	0.6984 (5)	0.1794 (5)	0.4972 (4)	0.0379 (13)
C9	0.7815 (4)	0.2352 (4)	0.4731 (3)	0.0283 (11)
C10	0.6614 (4)	0.3838 (5)	0.5432 (3)	0.0311 (12)
H10	0.647295	0.453679	0.558474	0.037*
C11	0.5743 (5)	0.3346 (5)	0.5706 (4)	0.0403 (14)
H11	0.504683	0.370385	0.604166	0.048*
C12	0.5932 (5)	0.2337 (6)	0.5473 (4)	0.0487 (16)
H12	0.535691	0.200215	0.564789	0.058*
C13	0.7568 (5)	0.5764 (5)	0.3036 (4)	0.0303 (12)
C14	0.6755 (4)	0.6567 (5)	0.2192 (4)	0.0366 (13)
H14A	0.605586	0.637212	0.227269	0.044*
H14B	0.649683	0.740038	0.217678	0.044*
C15	0.6849 (5)	0.7079 (5)	0.0451 (4)	0.0350 (13)
C16	0.5700 (5)	0.7894 (5)	0.0354 (4)	0.0495 (16)
H16	0.520653	0.801581	0.090173	0.059*
C17	0.5287 (5)	0.8535 (5)	−0.0579 (4)	0.0540 (16)
H17	0.450330	0.908324	−0.064871	0.065*
C18	0.5997 (5)	0.8387 (5)	−0.1405 (4)	0.0409 (14)
C19	0.7166 (5)	0.7561 (5)	−0.1281 (4)	0.0378 (13)
H19	0.766553	0.745098	−0.182741	0.045*
C20	0.7591 (5)	0.6912 (5)	−0.0376 (4)	0.0370 (13)
H20	0.837352	0.635982	−0.030637	0.044*
C21	0.5579 (6)	0.9029 (6)	−0.2414 (5)	0.0605 (18)
C22	0.4297 (8)	0.9878 (9)	−0.2550 (6)	0.131 (4)
H22A	0.415346	1.023191	−0.323458	0.196*
H22B	0.413340	1.049826	−0.224921	0.196*
H22C	0.378685	0.944609	−0.225183	0.196*
C23	0.9526 (4)	0.7049 (4)	0.3744 (3)	0.0263 (11)
C24	0.9001 (5)	0.7596 (4)	0.2755 (4)	0.0391 (14)
H24A	0.938076	0.704223	0.236469	0.047*
H24B	0.815694	0.774631	0.282447	0.047*
C25	0.8752 (5)	0.9278 (5)	0.1356 (4)	0.0413 (14)
C26	0.8770 (6)	1.0419 (5)	0.0976 (4)	0.0589 (18)
H26	0.905826	1.072260	0.133922	0.071*
C27	0.8361 (6)	1.1098 (5)	0.0060 (4)	0.0595 (18)
H27	0.836077	1.186939	−0.018401	0.071*
C28	0.7950 (5)	1.0669 (5)	−0.0510 (4)	0.0410 (14)
C29	0.7964 (5)	0.9515 (5)	−0.0123 (4)	0.0458 (15)
H29	0.769054	0.920363	−0.049083	0.055*
C30	0.8373 (5)	0.8824 (5)	0.0791 (4)	0.0473 (15)
H30	0.839362	0.804463	0.102887	0.057*
C31	0.7543 (8)	1.2601 (6)	−0.1950 (5)	0.086 (2)

Table 2: (continued)

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U_{iso}^*/U_{eq}
H31A	0.722513	1.293400	−0.258935	0.128*
H31B	0.835911	1.252224	−0.200264	0.128*
H31C	0.708895	1.312266	−0.155437	0.128*
C32	0.8198 (4)	0.5562 (4)	0.6351 (3)	0.0233 (11)
C33	0.7224 (4)	0.5834 (5)	0.7126 (4)	0.0341 (12)
H33A	0.748977	0.526125	0.775489	0.041*
H33B	0.703777	0.663110	0.717281	0.041*
C34	0.5227 (4)	0.5928 (4)	0.7509 (3)	0.0298 (12)
C35	0.5207 (4)	0.5993 (5)	0.8441 (4)	0.0383 (13)
H35	0.588194	0.595306	0.867948	0.046*
C36	0.4175 (5)	0.6118 (5)	0.9018 (4)	0.0389 (13)
H36	0.416530	0.615863	0.964784	0.047*
C37	0.3150 (4)	0.6186 (4)	0.8682 (4)	0.0318 (12)
C38	0.3184 (4)	0.6158 (5)	0.7737 (4)	0.0359 (13)
H38	0.249938	0.623752	0.748488	0.043*
C39	0.4207 (4)	0.6014 (5)	0.7161 (4)	0.0372 (13)
H39	0.421547	0.597513	0.653031	0.045*
C40	0.2085 (5)	0.6253 (5)	0.9346 (5)	0.0399 (14)
C41	0.0923 (5)	0.6615 (6)	0.8918 (5)	0.0608 (18)
H41A	0.094526	0.599346	0.867162	0.091*
H41B	0.077769	0.734455	0.839387	0.091*
H41C	0.029683	0.674089	0.941207	0.091*
N1	0.7624 (3)	0.3364 (3)	0.4968 (3)	0.0241 (9)
N2	0.9713 (3)	0.2373 (3)	0.4007 (3)	0.0276 (9)
O1	0.8125 (3)	0.5002 (3)	0.5821 (2)	0.0260 (7)
O2	0.9015 (3)	0.5922 (3)	0.6327 (2)	0.0342 (8)
O3	0.6192 (3)	0.5764 (3)	0.6877 (2)	0.0382 (9)
O4	0.2159 (3)	0.6014 (4)	1.0220 (3)	0.0573 (11)
O5	0.9376 (3)	0.6092 (3)	0.4262 (2)	0.0247 (7)
O6	1.0093 (3)	0.7475 (3)	0.4040 (2)	0.0328 (8)
O7	0.9162 (4)	0.8685 (3)	0.2278 (3)	0.0499 (11)
O8	0.7010 (5)	1.1016 (4)	−0.1916 (3)	0.0879 (17)
O9	0.6224 (4)	0.8850 (5)	−0.3117 (3)	0.0786 (15)
O10	0.7390 (3)	0.6389 (3)	0.1311 (3)	0.0456 (10)
O11	0.8579 (3)	0.5004 (3)	0.2889 (2)	0.0345 (8)
O12	0.7201 (3)	0.5891 (3)	0.3862 (2)	0.0363 (9)
O14	1.0169 (8)	0.5585 (11)	0.1300 (8)	0.220 (5)
H14C	1.087155	0.514904	0.115454	0.331*
H14D	0.997885	0.514963	0.183153	0.331*
Yb1	0.91664 (2)	0.42751 (2)	0.45313 (2)	0.02433 (8)
O13	0.6677 (8)	0.8076 (8)	0.4780 (6)	0.139 (2)*
H13A	0.736095	0.809539	0.476669	0.208*
H13B	0.622840	0.839166	0.521038	0.208*

3 Comment

Erbium complexes may show novel structures [5] and excellent properties in catalysis [6], dual visible/near-infrared luminescence [7], luminescent electrochromic device [8], single-molecule magnet [9]. As an important class of poly-podand, aroylhydrazone ligands have been used to construct diverse metal complexes [10]. We have been working on the synthesis, structure and properties of aroylhydrazone metal

complexes [11–16]. In this work, we report a new aroylhydrazone Er(III) complex based on phenacetic acid hydrazide, 3-bromo-2-hydroxybenzaldehyde, $ErCl_3 \cdot 6H_2O$, and NaOH as starting materials.

The molecular structure of Er(III) complex is shown in the figure. The asymmetric unit of the title compound consists of one Er(III) ion, two 2 monoanionic hydrazone ligands, one coordinated dimethylformamide molecule, one coordinated water molecule, two uncoordinated dimethylformamide molecules, one uncoordinated water molecule, and one chloride ion. The Er(III) atom is eight-coordinated by four oxygen atoms (O2, O4, O7, O9) and two nitrogen atoms (N3, N4) from two different tridentate 2-bromo-6-(phenylacetyl-hydrazonomethyl)-phenylate ligands, one oxygen atom (O8) from one coordinated dimethylformamide molecule, one oxygen atom (O1) from one coordinated water molecule. The bond lengths of Er–O and Er–N are in the range of 2.184(4)–2.388(4) Å and 2.491(4)–2.506(5) Å, respectively. The dihedral angles of plane 1 (C2–C3–C4–C5–C6–C14) and plane 2 (C1–C13–C12–C11–C10–C15), plane 3 (C16–C17–C18–C19–C20–C21) and plane 4 (C25–C26–C27–C28–C29–C30) are 61.41° and 56.67°, respectively.

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References

1. Bruker. *SAINT* and *SADABS*; Bruker AXS Inc.: Madison, Wisconsin, USA, 2000.
2. Dolomanov O. V., Bourhis L. J., Gildea R. J., Howard J. A. K., Puschmann H. OLEX2: a complete structure solution, refinement and analysis program. *J. Appl. Cryst.* 2009, 42, 339–341.
3. Sheldrick G. M. Crystal structure refinement with SHELXL. *Acta Crystallogr.* 2015, C71, 3–8.
4. Brandenburg K. *DIAMOND. Visual Crystal Structure Information System. Ver. 3.2*; Crystal Impact: Bonn, Germany, 2012.
5. Klementyeva S. V., Afonin M. Y., Bogomyakov A. S., Gamer M. T., Roesky P. W., Konchenko S. N. Mono- and dinuclear rare-earth chlorides ligated by a mesityl-substituted β -diketiminato. *Eur. J. Inorg. Chem.* 2016, 2016, 3666–3672.
6. Sarkar A., Jana S., Nayek H. P. A pentanuclear Er(III) coordination cluster as a catalyst for selective synthesis of 1,2-disubstituted benzimidazoles. *Appl. Organomet. Chem.* 2021, 35, 1–9.
7. Golezorkhi B., Guénée L., Nozary H., Fürstenberg A., Suffren Y., Eliseeva S. V., Petoud S., Hauser A., Piguet C. Thermodynamic programming of erbium(III) coordination complexes for dual visible/near-infrared luminescence. *Chem. Eur. J.* 2018, 24, 13158–13169.
8. Fernandes M., Freitas V., Pereira S., Leones R., Silva M. M., Carlos L., Fortunato E., Ferreira R., Rego R., de Zea Bermudez V. Luminescent electrochromic devices for smart windows of energy-efficient buildings. *Energies* 2018, 11, 3513.
9. Diamantoula M., Panagiota S. P., Evangelos P., Sotirios C., Mathieu R., Pierre D., Rodolphe C., Spyros P. P. Asymmetric dinuclear lanthanide(III) complexes from the use of a ligand derived from 2-acetylpyridine and picolinoylhydrazide: synthetic, structural and magnetic studies. *Molecules* 2020, 25, E3153.
10. Kuriakose D., Prathapachandra Kurup M. R. Supramolecular frameworks formed via hydrogen bonding and non-covalent interactions and interaction energy calculations of solvent coordinated cis-dioxomolybdenum(VI) complexes derived from ONO donor aroylhydrazones: cytotoxicity studies. *Inorg. Chim. Acta* 2020, 505, 119472.
11. Tai X. S., Wang L. H., Xia Y. P. The crystal structure of poly[(μ_2 -aqua-aqua-(μ_3 -(E)-2-(4-((2-carbamothioylhydrazineylidene) methyl)phenoxy)acetato- κ^3 O:S) sodium(I))], $C_{10}H_{14}N_3O_5SNa$. *Z. Kristallogr. N. Cryst. Struct.* 2022, 237, 277–279.
12. Wang L. H., Wang Z. J., Zhao M. L., Tai X. S., Ouyang J., Li Y. F., Zhang W., Jia W. L. Synthesis, crystal structure of tetra-nuclear macrocyclic Zn(II) complex and its application as catalyst for oxidation of benzyl alcohol. *Bull. Chem. React. Eng. Catal.* 2021, 16, 839–846.
13. Tai X. S., Li P. F., Liu L. L. Synthesis, crystal structure and catalytic activity of a calcium(II) complex with 4-formylbenzene-1,3-disulfonate-isonicotinic acid hydrazone. *Bull. Chem. React. Eng. Catal.* 2018, 13, 429–435.
14. Tai X. S., Wang X. Synthesis, structural characterization and antitumor activity of a Ca(II) coordination polymer based on 4-formyl-1,3-benzenedisulfonate-2-furoic acid hydrazide ligands. *Crystallogr. Rep.* 2017, 62, 242–245.
15. Du L. C., Tai X. S. The crystal structure of *catena*-poly[diaqua-bis(μ_2 -2-((2-phenylacetyl)hydrazineylidene)methyl)benzoato- κ^2 O:O']zinc(II)], $C_{32}H_{30}N_4O_8Zn$. *Z. Kristallogr. N. Cryst. Struct.* 2021, 236, 843–844.
16. Tai X. S., Zhou X. J., Liu L. L., Cao S. H., Wang L. H. The crystal structure of *catena*-poly[(μ_2 -2-((3-bromo-2-oxidobenzylidene)amino)acetato- κ^4 O,N,O':O'')-(dimethylformamide- κ' O)]zinc(II)], $C_{12}H_{13}N_2O_4BrZn$. *Z. Kristallogr. N. Cryst. Struct.* 2020, 235, 901–902.