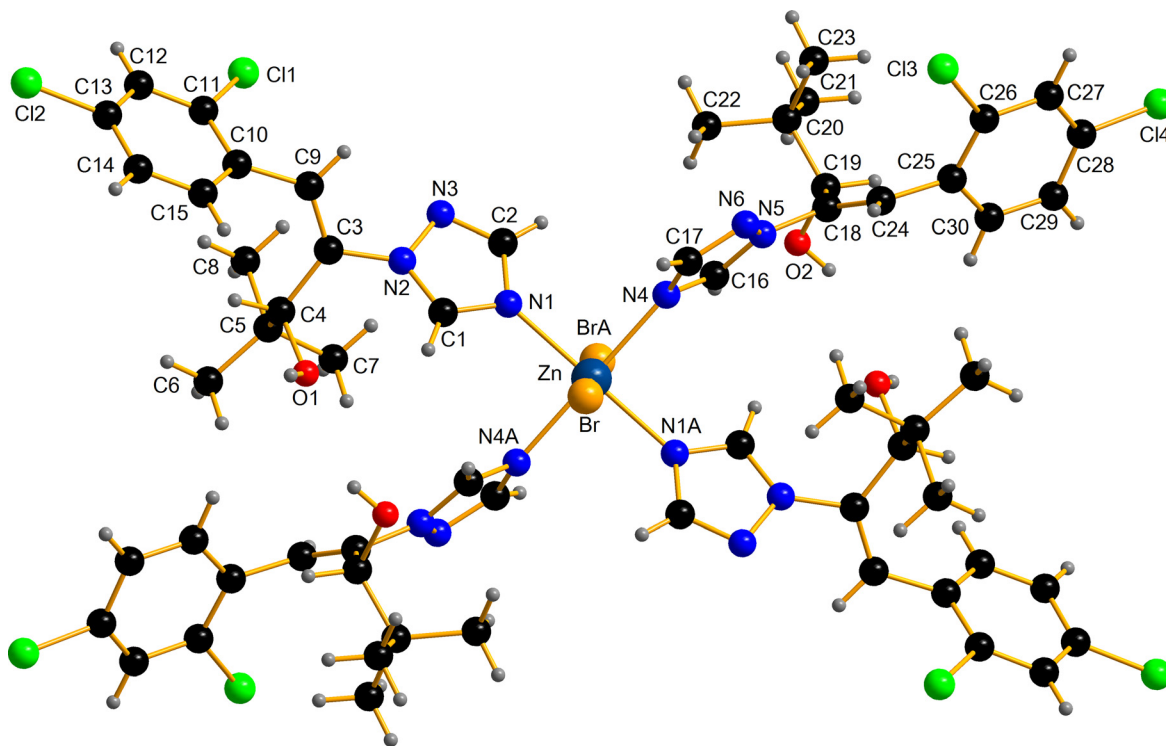


Jie Li*, Biao Yan and Hongya Li

Crystal structure of dibromido-tetra((*E*)-(RS)-1-(2,4-dichlorophenyl)-4,4-dimethyl-2-(1,2,4-triazol-1-yl)pent-1-en-3-ol- κ^1N)zinc(II), $C_{60}H_{68}O_4N_{12}Br_2Cl_8Zn$



<https://doi.org/10.1515/ncrs-2022-0409>

Received August 12, 2022; accepted September 13, 2022;

published online September 22, 2022

Abstract

$C_{60}H_{68}O_4N_{12}Br_2Cl_8Zn$, triclinic, $P\bar{1}$ (no. 2), $a = 9.0149$ (13) Å, $b = 13.671$ (2) Å, $c = 15.168$ (2) Å, $\alpha = 92.280$ (3)°, $\beta = 97.441$ (2)°, $\gamma = 107.862$ (2)°, $V = 1758.0$ (4) Å³, $Z = 1$, $R_{gt}(F) = 0.0417$, $wR_{ref}(F^2) = 0.1005$, $T = 296$ (2) K.

CCDC no.: 1060432

*Corresponding author: Jie Li, College of Chemistry and Chemical Engineering, Xinyang Normal University, Xinyang, Henan, 464000, P. R. China, E-mail: lijie@xynu.edu.cn. <https://orcid.org/0000-0002-3104-1222>

Biao Yan and Hongya Li, School of Chemistry and Chemical Engineering, Yulin University, Yulin, Shaanxi, 719000, P. R. China. <https://orcid.org/0000-0002-7159-7911> (B. Yan)

The molecular structure is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

Table 1: Data collection and handling.

Crystal:	Stick, colourless
Size:	0.39 × 0.30 × 0.26 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	1.84 mm ⁻¹
Diffraction, scan mode:	Bruker APEX-II, φ and ω -scans
θ_{max} , completeness:	27.4°, >99%
$N(hkl)_{measured}$, $N(hkl)_{unique}$, R_{int} :	10,124, 7338, 0.023
Criterion for I_{obs} , $N(hkl)_{gt}$:	$I_{obs} > 2 \sigma(I_{obs})$, 5015
$N(param)_{refined}$:	396
Programs:	Bruker programs [1], SHELX [2–4], DIAMOND [5]

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso} */ <i>U</i> _{eq}
Zn	0.000000	0.000000	0.500000	0.03388 (14)
Br	0.24579 (4)	−0.01731 (3)	0.61651 (2)	0.04691 (12)
Cl1	0.11145 (11)	−0.35660 (8)	0.02380 (7)	0.0680 (3)
Cl2	0.69040 (15)	−0.35700 (10)	−0.02233 (9)	0.0991 (4)
Cl3	0.62816 (12)	0.52985 (8)	0.24711 (8)	0.0705 (3)
Cl4	1.16357 (10)	0.46568 (8)	0.18232 (7)	0.0660 (3)
N1	0.0457 (3)	−0.09177 (19)	0.39593 (16)	0.0359 (6)
N2	0.1527 (3)	−0.13447 (19)	0.28551 (16)	0.0348 (6)
N3	−0.0013 (3)	−0.1960 (2)	0.26990 (18)	0.0486 (7)
N4	0.1558 (3)	0.13737 (19)	0.45635 (17)	0.0362 (6)
N5	0.3642 (3)	0.23629 (18)	0.40800 (16)	0.0333 (6)
N6	0.2424 (3)	0.2682 (2)	0.3709 (2)	0.0521 (8)
O1	0.4328 (2)	0.03415 (17)	0.27033 (15)	0.0473 (6)
H1A	0.517860	0.027479	0.290229	0.071*
O2	0.6359 (3)	0.18520 (17)	0.49308 (15)	0.0481 (6)
H2A	0.671088	0.148791	0.463030	0.072*
C1	0.1774 (3)	−0.0746 (2)	0.3600 (2)	0.0381 (7)
H1	0.273867	−0.026973	0.384089	0.046*
C2	−0.0579 (4)	−0.1667 (3)	0.3377 (2)	0.0490 (9)
H2	−0.162590	−0.195567	0.345511	0.059*
C3	0.2625 (3)	−0.1405 (2)	0.22617 (19)	0.0330 (7)
C4	0.3596 (4)	−0.0396 (2)	0.1960 (2)	0.0383 (7)
H4	0.443694	−0.053329	0.167634	0.046*
C5	0.2681 (4)	0.0103 (3)	0.1280 (2)	0.0534 (9)
C6	0.3905 (6)	0.0986 (3)	0.0929 (3)	0.0888 (15)
H6A	0.462362	0.071947	0.065944	0.133*
H6B	0.338073	0.130637	0.049214	0.133*
H6C	0.447664	0.148694	0.141408	0.133*
C7	0.1542 (5)	0.0534 (3)	0.1707 (3)	0.0782 (13)
H7A	0.077203	−0.001679	0.192823	0.117*
H7B	0.211729	0.103529	0.219125	0.117*
H7C	0.102137	0.085471	0.126931	0.117*
C8	0.1792 (6)	−0.0696 (3)	0.0501 (3)	0.0924 (15)
H8A	0.252293	−0.095928	0.023998	0.139*
H8B	0.101974	−0.125237	0.071457	0.139*
H8C	0.127618	−0.037729	0.005910	0.139*
C9	0.2634 (3)	−0.2342 (2)	0.19963 (19)	0.0380 (7)
H9	0.189647	−0.289862	0.218890	0.046*
C10	0.3712 (4)	−0.2581 (2)	0.1422 (2)	0.0365 (7)
C11	0.3129 (4)	−0.3181 (2)	0.0620 (2)	0.0432 (8)
C12	0.4092 (5)	−0.3486 (3)	0.0107 (2)	0.0536 (9)
H12	0.367502	−0.389056	−0.042874	0.064*
C13	0.5682 (5)	−0.3177 (3)	0.0409 (3)	0.0559 (10)
C14	0.6309 (4)	−0.2573 (3)	0.1189 (3)	0.0525 (9)
H14	0.739151	−0.236036	0.137626	0.063*
C15	0.5330 (4)	−0.2281 (2)	0.1697 (2)	0.0436 (8)
H15	0.575832	−0.187614	0.223133	0.052*
C16	0.3106 (3)	0.1594 (2)	0.4588 (2)	0.0356 (7)
H16	0.372532	0.126116	0.491296	0.043*
C17	0.1224 (4)	0.2070 (3)	0.4028 (2)	0.0498 (9)
H17	0.020979	0.211088	0.389451	0.060*
C18	0.5201 (3)	0.2771 (2)	0.3841 (2)	0.0323 (7)
C19	0.6561 (3)	0.2819 (2)	0.4552 (2)	0.0367 (7)
H19	0.750183	0.295626	0.425719	0.044*
C20	0.6924 (4)	0.3677 (3)	0.5315 (2)	0.0445 (8)

Table 2: (continued)

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso} */ <i>U</i> _{eq}
C21	0.8522 (4)	0.3724 (4)	0.5848 (3)	0.0746 (12)
H21A	0.843463	0.307746	0.609781	0.112*
H21B	0.930840	0.385654	0.545937	0.112*
H21C	0.881986	0.426668	0.632074	0.112*
C22	0.5674 (4)	0.3454 (3)	0.5946 (2)	0.0643 (11)
H22A	0.559419	0.280618	0.619054	0.096*
H22B	0.597466	0.399350	0.642205	0.096*
H22C	0.467230	0.342474	0.562006	0.096*
C23	0.7067 (5)	0.4700 (3)	0.4922 (3)	0.0759 (13)
H23A	0.607172	0.467282	0.458923	0.114*
H23B	0.736534	0.524256	0.539478	0.114*
H23C	0.785389	0.483242	0.453341	0.114*
C24	0.5313 (3)	0.3099 (2)	0.3038 (2)	0.0408 (8)
H24	0.439440	0.309393	0.267666	0.049*
C25	0.6826 (3)	0.3475 (2)	0.2679 (2)	0.0374 (7)
C26	0.7396 (4)	0.4471 (2)	0.2420 (2)	0.0394 (7)
C27	0.8854 (4)	0.4839 (3)	0.2134 (2)	0.0437 (8)
H27	0.921369	0.550897	0.196654	0.052*
C28	0.9750 (4)	0.4196 (3)	0.2104 (2)	0.0431 (8)
C29	0.9195 (4)	0.3183 (3)	0.2302 (2)	0.0492 (9)
H29	0.978866	0.274125	0.224627	0.059*
C30	0.7741 (4)	0.2840 (3)	0.2583 (2)	0.0468 (8)
H30	0.736135	0.215701	0.271336	0.056*

Source of material

The (*E*)-(*RS*)-1-(2,4-dichlorophenyl)-4,4-dimethyl-2-(1,2,4-triazol-1-yl)pent-1-en-3-ol (0.653 g, 0.200 mmol) was dissolved in methanol (10 mL). Then, Zinc bromide (0.225 g, 0.100 mmol) was added. Reaction was allowed to stir at room temperature for 2 h. Colorless stick crystal of the title compound was obtained by slow evaporation from propanone.

Experimental details

Hydrogen atoms were placed in their geometrically idealized positions and constrained to ride on their parent atoms, with C–H = 0.96 Å (methyl), *U*_{iso}(H) = 1.5 *U*_{eq}(C), C–H = 0.98 Å (methine), *U*_{iso}(H) = 1.2 *U*_{eq}(C), C–H = 0.93 Å (aromatic and alkenyl), *U*_{iso}(H) = 1.2 *U*_{eq}(C), and O–H = 0.82 Å (hydroxyl), *U*_{iso}(H) = 1.5 *U*_{eq}(O).

Comment

As a broad spectrum systemic 1,2,4-triazole fungicide, diniconazole (*E*)-(*RS*)-1-(2,4-dichlorophenyl)-4,4-dimethyl-2-(1*H*-1,2,4-triazol-1-yl) pent-1-en-3-ol, has been widely used to control various fungi in many kinds of crops by inhibiting

steroid demethylation [6, 7]. Complexation of pesticides with transition metals is a valuable way to improve bioactivity of original pesticides [8–16]. Zinc is a transition metal essential to the life of crops. Many triazole fungicides zinc complexes have been reported [13–19], the complexes synthesized shows better antifungal activities than the ligand [13–16]. In this paper, a new diniconazole Zn(II) complex is described.

The asymmetric unit of the title structure contains a half Zn(II) cation, one bromine anion and two diniconazole molecules to construct a mononuclear complex. The Zn(II) cation is six-coordinated by two bromine anions and nitrogen atoms of triazole cycles of four organic ligands. The Zn–N length (Zn–N1 = 2.136 (2) Å, Zn–N4 = 2.159 (2) Å) and Zn–Br length (Zn–Br = 2.7257 (4) Å), the N1–Zn–N4, N1–Zn–N4ⁱ, N1–Zn–Br, N4–Zn–Br, N1–Zn–Brⁱ, and N4–Zn–Brⁱ bond angles of 90.27 (9)°, 89.73 (9)°, 92.61 (7)°, 89.49 (7)°, 87.39 (7)°, and 90.51 (7)°, respectively (symmetry code: (i) $-x, -y, -z + 1$). Zn(II) cation is located in a special position at center of symmetry, so the bond angle of N1–Zn–N1ⁱⁱ, N4–Zn–N4ⁱⁱ, and Br–Zn–Brⁱⁱ is 180°. This is in agreement with closely related Zn(II) complexes reported [20]. The complexes are bound by O1–H1A...Brⁱⁱ ($d_{O1...Br} = 3.248$ (2) Å, 166.4°), and O2–H2A...Brⁱⁱ ($d_{O2...Br} = 3.290$ (2) Å, 171.5°), (symmetry code: (ii) $-x + 1, -y, -z + 1$) hydrogen bonds forming a one-dimensional (1D) chain along the *a*-axis.

Author contributions: All the authors have accepted responsibility for the entire content of this submitted manuscript and approved submission.

Research funding: The work was supported by National Natural Science Foundation of China (22003055, 22163012), Key Laboratory Project Foundation of Shaanxi Provincial Education Department in China (20JS158), Joint Fund of the Yulin University and the Dalian National Laboratory for Clean Energy (YLU-DNL Fund 2021007), Nanhu Scholars Program for Young Scholars of the Xinyang Normal University.

Conflict of interest statement: The authors declare no conflicts of interest regarding this article.

References

1. Bruker. APEX2, SAINT and SADABS; Bruker AXS Inc.: Madison, Wisconsin, USA, 2009.
2. Sheldrick G. M. *SHELXTL* – Integrated space-group and crystal-structure determination. *Acta Crystallogr.* 2015, *A71*, 3–8.
3. Sheldrick G. M. Crystal structure refinement with *SHELXL*. *Acta Crystallogr.* 2015, *C71*, 3–8.
4. Sheldrick G. M. A short history of *SHELX*. *Acta Crystallogr.* 2008, *A64*, 112–122.
5. Brandenburg K. DIAMOND; Crystal Impact GbR: Bonn, Germany, 2006.
6. Funaki Y., Ishiguri Y., Kato T., Tanaka S. Structure-activity relationships of vinyl triazole fungicides. *J. Pestic. Sci.* 1984, *9*, 229–236.
7. Takano H., Oguri Y., Kato T. Mode of action of (E)-1-(2,4-dichlorophenyl)-4,4-dimethyl-2-(1,2,4-triazol-1-yl)-1-penten-3-ol (S-3308) is *ustilago maydis*. *J. Pestic. Sci.* 1983, *8*, 575–582.
8. Li J., Ren G.-Y., Zhang Y., Yang M.-Y., Ma H.-X. Two Cu(II) complexes of 1,2,4-triazole fungicides with enhanced antifungal activities. *Polyhedron* 2019, *157*, 163–169.
9. Li J., Liu H., Guo Z., Yang M., Song J., Ma H. Two Cu(II)-triadimenol complexes as potential fungicides: synergistic actions and DFT calculations. *RSC Adv.* 2018, *8*, 2933–2940.
10. Li J., Zhang Y., Yang M., Ma H. Two novel Co(II) and Ni(II) complexes of tebuconazole with enhanced antifungal activities. *RSC Adv.* 2017, *7*, 33364–33372.
11. Li J., Teng X., Yan B., Yang M., Song J., Ma H. Two Cu(II) complexes of triadimenol: crystal structure, antifungal activities and structure–activity relationship. *New J. Chem.* 2015, *39*, 6997–7003.
12. Li J., Teng X., Yan B., Guan Y., Yang M., Song J., Ma H. Synergistic actions between tebuconazole ligand and Cu(II) cation: reasons for the enhanced antifungal activity of four Cu(II) complexes based on the fungicide tebuconazole. *New J. Chem.* 2015, *39*, 9550–9556.
13. Li J., Xi T., Yan B., Yang M.-Y., Ma H.-X., Song J.-R. Crystal structure and antifungal activities of a new Zn(II) complex based on triadimenol. *Chin. J. Struct. Chem.* 2015, *34*, 1825–1829.
14. Ren G.-Y., Li J., Zhou J.-H., Yan B., Ren Y.-H., Sun X.-H., Ma H.-X. Enhanced antifungal activities of four Zn(II) complexes based on uniconazole. *Appl. Organomet. Chem.* 2018, *32*, e4169.
15. Ren G.-Y., Li J., Wei X.-Z., Zhou J.-H., Yan B., Guo Z.-Q., Ren Y.-H., Sun X.-H., Ma H.-X. The enhanced biological activities of three Zn(II) complexes based on different 1,2,4-triazole fungicides. *Appl. Organomet. Chem.* 2018, *32*, e4474.
16. Chen X., Yang C. Synthesis, controlled release properties, and increased antifungal activities of novel *cis*- and *trans*-racemate complexes of propiconazole. *J. Agric. Food Chem.* 2009, *57*, 2441–2446.
17. Gao J.-S., Ma D.-S., Ma Z.-G., Chen G.-R., Hou Y.-J., Ye L. The syntheses and characterizations for zinc diniconazole complexes. *J. Mol. Sci.(Chin.)* 2001, *17*, 17–22.
18. Evans P. D., Schmalzl K. J., Forsyth C. M., Fallon G. D., Schmid S., Bendixen B., Heimdal S. Formation and structure of metal complexes with the fungicides tebuconazole and propiconazole. *J. Wood Chem. Technol.* 2007, *47*, 120–127.
19. Nie X.-L., Song X.-Y., Huang C.-G., Gong L., Shang-Guan X.-C. Synthesis and crystal structures of complexes of zinc, silver, copper assembled by uniconazole. *J. Chem. Crystallogr.* 2017, *27*, 243–256.
20. Bourne S. A., Kilkenny M., Nassimbeni L. R. One- and two-dimensional coordination polymers of zinc(II) with pyrazine. Solid state reactions and decomposition kinetics of the interconversion reactions. *J. Chem. Soc., Dalton Trans.* 2001, *2001*, 1176–1179.