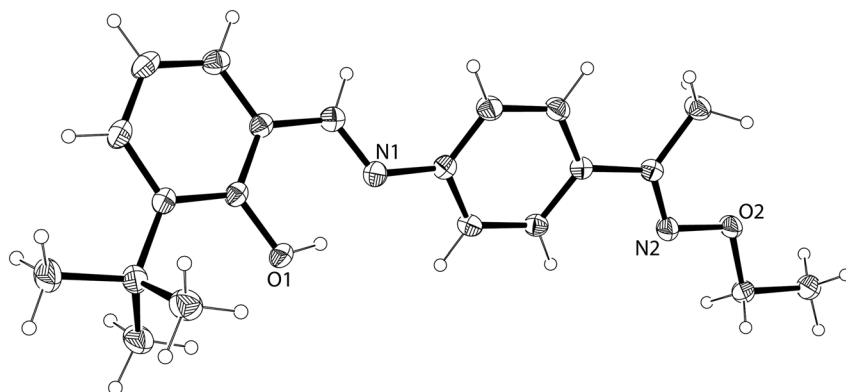


Ji-Xing Zhao*, Da-Qun Chen and Lei-Fang Wu

Synthesis and crystal structure of (*E*)-1-(4-(((*E*)-3-(*tert*-butyl)-2-hydroxybenzylidene)amino)phenyl)ethan-1-one *O*-ethyl oxime, C₂₁H₂₆N₂O₂



<https://doi.org/10.1515/ncrs-2022-0134>

Received March 19, 2022; accepted May 5, 2022;
published online May 31, 2022

Abstract

C₂₁H₂₆N₂O₂, monoclinic, *P*2₁ (no. 4), *a* = 7.1425(7) Å, *b* = 9.2733(8) Å, *c* = 14.3161(13) Å, β = 93.592(4)°, *V* = 946.36(15) Å³, *Z* = 2, *R*_{gt}(*F*) = 0.0461, *wR*_{ref}(*F*²) = 0.1148, *T* = 173(2) K.

CCDC no.: 2170731

Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

Source of material

The title compound was prepared by a method reported earlier [4, 5]. 3-*tert*-Butyl-2-hydroxy-benzaldehyde (178.23 mg, 1 mmol) dissolved in ethanol solution (5 mL) was added to 1-(4-aminophenyl)ethanone *O*-ethyl-oxime (178.21 mg, 1 mmol) dissolved in ethanol (5 mL) at 325 K. The mixture was kept stirring for 6 h. The mixture was cooled to room

Table 1: Data collection and handling.

Crystal:	Plate, yellow
Size:	0.27 × 0.25 × 0.06 mm
Wavelength:	Mo Kα radiation (0.71073 Å)
μ:	0.08 mm ⁻¹
Diffractometer, scan mode:	Bruker D8 venture, φ and ω-scans
θ _{max} , completeness:	27.9°, >99%
<i>N</i> (<i>hkl</i>) _{measured} , <i>N</i> (<i>hkl</i>) _{unique} , <i>R</i> _{int} :	15067, 4486, 0.067
Criterion for <i>I</i> _{obs} , <i>N</i> (<i>hkl</i>) _{gt} :	<i>I</i> _{obs} > 2 σ(<i>I</i> _{obs}), 3423
<i>N</i> (<i>param</i>) _{refined} :	232
Programs:	Bruker programs [1], SHELX [2], OLEX2 [3]

temperature and filtered. The filtrate was recrystallized and several clear yellow crystals were obtained.

Experimental details

Hydrogen atoms were placed in their geometrically idealized positions and constrained to ride on their parent atoms.

Comment

Schiff base compounds are an important class of ligands in coordination chemistry. Schiff base transition metal complexes are attractive because of their show effective applications in various fields of scientific research. For

*Corresponding author: Ji-Xing Zhao, Analysis and Testing Center, Shihezi University, Xinjiang 832003, P. R. China, E-mail: zhaojixing_2016@126.com. <https://orcid.org/0000-0002-1322-1860>

Da-Qun Chen and Lei-Fang Wu, Analysis and Testing Center, Shihezi University, Xinjiang 832003, P. R. China

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso} [*] / <i>U</i> _{eq}
O1	−0.0353 (3)	0.5348 (2)	0.22264 (13)	0.0391 (5)
H1	0.025487	0.529506	0.274695	0.059*
N1	0.0338 (3)	0.4763 (2)	0.39732 (15)	0.0350 (5)
C1	−0.2662 (4)	0.4364 (3)	0.31892 (18)	0.0311 (5)
O2	0.9063 (3)	0.4935 (2)	0.74170 (13)	0.0410 (5)
N2	0.7603 (3)	0.5106 (3)	0.67280 (16)	0.0395 (6)
C2	−0.2126 (4)	0.4882 (3)	0.23149 (18)	0.0303 (5)
C3	−0.3403 (4)	0.4909 (3)	0.15240 (19)	0.0316 (6)
C4	−0.5195 (4)	0.4379 (3)	0.1643 (2)	0.0384 (6)
H4	−0.608369	0.437851	0.112034	0.046*
C7	−0.2810 (4)	0.5464 (3)	0.05724 (19)	0.0386 (6)
C6	−0.4486 (4)	0.3846 (3)	0.3263 (2)	0.0378 (6)
H6	−0.485682	0.349011	0.384577	0.045*
C5	−0.5743 (4)	0.3848 (3)	0.2497 (2)	0.0426 (7)
H5	−0.697889	0.349021	0.254794	0.051*
C8	−0.4479 (5)	0.5504 (4)	−0.0160 (2)	0.0541 (9)
H8A	−0.406212	0.588232	−0.075168	0.081*
H8B	−0.546358	0.612942	0.006301	0.081*
H8C	−0.497521	0.452717	−0.025835	0.081*
C9	−0.2055 (5)	0.7008 (4)	0.0657 (2)	0.0531 (8)
H9A	−0.179243	0.736848	0.003538	0.080*
H9B	−0.089795	0.701546	0.106281	0.080*
H9C	−0.299236	0.762744	0.092765	0.080*
C10	−0.1324 (5)	0.4454 (4)	0.0212 (2)	0.0525 (8)
H10A	−0.099111	0.477382	−0.040848	0.079*
H10B	−0.182307	0.347051	0.016990	0.079*
H10C	−0.020449	0.447125	0.064459	0.079*
C11	−0.1353 (4)	0.4311 (3)	0.40019 (18)	0.0340 (6)
H11	−0.175716	0.393032	0.457150	0.041*
C12	0.1696 (4)	0.4600 (3)	0.47326 (17)	0.0320 (6)
C13	0.1791 (4)	0.3419 (3)	0.53270 (19)	0.0390 (7)
H13	0.084394	0.269874	0.527131	0.047*
C14	0.3259 (4)	0.3281 (3)	0.60030 (19)	0.0378 (6)
H14	0.329210	0.247092	0.641016	0.045*
C15	0.4686 (4)	0.4304 (3)	0.60972 (17)	0.0303 (5)
C16	0.4563 (4)	0.5499 (3)	0.54949 (17)	0.0335 (6)
H16	0.550199	0.622488	0.554866	0.040*
C17	0.3101 (4)	0.5634 (3)	0.48277 (18)	0.0349 (6)
H17	0.305259	0.644950	0.442505	0.042*
C18	0.6306 (4)	0.4159 (3)	0.67996 (18)	0.0319 (6)
C19	0.6369 (4)	0.2995 (3)	0.75214 (19)	0.0427 (7)
H19A	0.566277	0.215654	0.727652	0.064*
H19B	0.581134	0.334833	0.808626	0.064*
H19C	0.767611	0.271685	0.767538	0.064*
C20	1.0450 (4)	0.6024 (3)	0.7285 (2)	0.0403 (7)
H20A	0.988715	0.699527	0.733019	0.048*
H20B	1.095478	0.592274	0.666028	0.048*
C21	1.1987 (4)	0.5828 (3)	0.8037 (2)	0.0452 (7)
H21A	1.250590	0.485318	0.799812	0.068*
H21B	1.147964	0.596494	0.865150	0.068*
H21C	1.297892	0.653770	0.795363	0.068*

example, these metal complexes are widely applied in ion recognition [6, 7] luminescence materials [8, 9] and supramolecular constructing [10, 11]. Recently, we have also

investigated the title compound synthesis, structural characterization, and we predict that this compound can form stable complexes with transition metals [12].

The crystal structure of the title compound is shown in the figure. Compared with reported similar structures, all bond lengths and bond angles are within the normal ranges [4, 5, 13]. In the title compound, there is a strong intramolecular O1—H1...N1 hydrogen bond interaction ($d(\text{N1} \cdots \text{H1}) = 1.821 \text{ \AA}$, $d(\text{O1} \cdots \text{H1}) = 0.840 \text{ \AA}$ and $d(\text{O1} \cdots \text{N1}) = 2.576 \text{ \AA}$).

Author contributions: All the authors have accepted responsibility for the entire content of this submitted manuscript and approved submission.

Research funding: This project was supported by the Open Project Program of Shihezi University (KX000203).

Conflict of interest statement: The authors declare no conflicts of interest regarding this article.

References

1. Bruker. APEX2, SAINT and SADABS; Bruker AXS Inc.: Madison, WI, USA, 2012.
2. Sheldrick G. M. Crystal structure refinement with SHELXL. *Acta Crystallogr.* 2015, *C71*, 3–8.
3. Dolomanov O. V., Bourhis L. J., Gildea R. J., Howard J. A. K., Puschmann H. OLEX2: a complete structure solution, refinement and analysis program. *J. Appl. Crystallogr.* 2009, *42*, 339–341.
4. Ma J. X., Wei P., Li Q. L., Fu T., Zhao J. X., Zhao L. Crystal structure of (*E*)-1-(4-(((*E*)-2-bromo-6-hydroxybenzylidene)amino) phenyl) ethan-1-one *O*-methyl oxime, C₁₆H₁₅BrN₂O₂. *Z. Kristallogr. N. Cryst. Struct.* 2020, *235*, 19–20.
5. Fu T., Ma J. X., Li Q. L., Wei P., Zhao J. X., Zhao L. Crystal structure of (*E*)-1-{4-[[4-fluoro-2-hydroxybenzylidene)amino] phenyl} ethanone *O*-methyl oxime, C₁₆H₁₅FN₂O₂. *Z. Kristallogr. N. Cryst. Struct.* 2020, *235*, 293–295.
6. Wu Y., Ding W. M., Li J., Guo G., Zhang S. Z., Jia H. R., Sun Y. X. A highly selective turn-on fluorescent and naked-eye colourimetric dual-channel probe for cyanide anions detection in water samples. *J. Fluoresc.* 2021, *31*, 437–446.
7. Ding W. M., Wu Y., Zhang S. Z., Li J., Xu L., Sun Y. X. A dual-channel ‘turn-on’ fluorescent chemosensor for high selectivity and sensitivity detection of CN[−] based on a coumarin–Schiff base derivative in an aqueous system. *Luminescence* 2021, *36*, 1–11.
8. Wang F., Gao L., Zhao Q., Zhang Y., Dong W. K., Ding Y. J. A highly selective fluorescent chemosensor for CN[−] based on a novel bis(salamo)-type tetraoxime ligand. *Spectrochim. Acta A.* 2018, *190*, 111–115.
9. Hao J., Li X. Y., Wang L., Zhang Y., Dong W. K. Luminescent and electrochemical properties of four novel butterfly-shaped hetero-pentanuclear [Zn₄Ln] complexes constructed from a bis(salamo)-type ligand. *Spectrochim. Acta A.* 2018, *204*, 388–402.
10. Sun Y. X., Zhao Y. Y., Li C. Y., Yu B., Guo J. Q., Li J. Supramolecular cobalt(II) and copper(II) complexes with Schiff base ligand:

- syntheses, characterizations and crystal structures. *Chin. J. Inorg. Chem.* 2016, 32, 913–920.
11. Chang J., Zhang H. J., Jia H. R., Sun Y. X. Binuclear nickel(II) and zinc(II) complexes based on 2-amino-3-hydroxy-pyridine Schiff base: syntheses, supramolecular structures and spectral properties. *Chin. J. Inorg. Chem.* 2018, 34, 2097–2107.
 12. Fu T., Ma J. X., Wei P., Li Q. L., Zhao J. X., Zhao L. Synthesis and crystal structure of bis{2-bromo-6-(((4-(1-(methoxyimino) ethyl) phenyl)imino)methyl) phenolato- κ^2N,O }cobalt(II)-dichloromethane(1/1), $C_{34}H_{32}Br_2Cl_4CoN_4O_4$. *Z. Kristallogr. N. Cryst. Struct.* 2020, 235, 59–60.
 13. Ma B. Y., Fu T., Jin Y., Wei P., Zhao J. X., Zhao L. Synthesis and crystal structure of and crystal structure of 1-{4-[(2-bromo-6-hydroxy-benzylidene) amino]phenyl}ethanone, $C_{15}H_{12}BrNO_2$. *Z. Kristallogr. N. Cryst. Struct.* 2020, 235, 1093–1094.