

Xiuyan Yang and Yingchun Luo*

Crystal structure of *catena*-poly[(μ_2 -4,4'-bipyridine- $\kappa^2N:N'$)-bis(4-bromobenzoato- κ^1O)zinc(II)], $C_{24}H_{16}Br_2N_2O_4Zn$

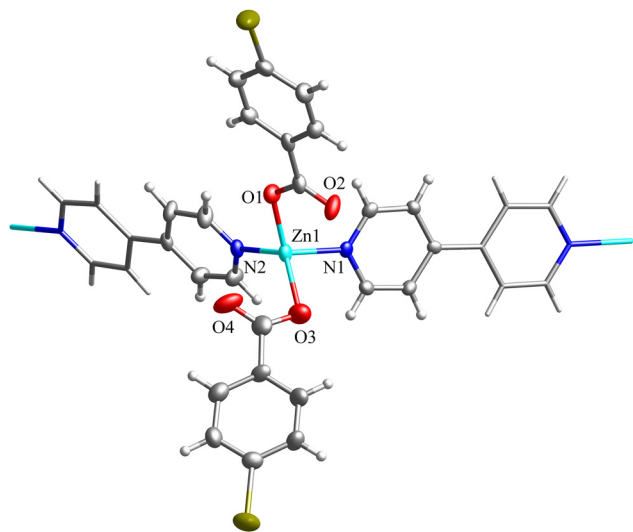


Table 1: Data collection and handling.

Crystal:	Colorless block
Size:	0.18 × 0.16 × 0.15 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	4.50 mm ⁻¹
Diffractometer, scan mode:	Bruker APEX-II, φ and ω -scans
$\theta_{\max}$, completeness:	25°, >99%
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	11890, 4132, 0.040
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 2752
$N(\text{param})_{\text{refined}}$:	298
Programs:	Bruker programs [1], SHELX [2]

Source of material

A mixture of $Zn(NO_3)_2 \cdot 6H_2O$ (1 mmol, 297.5 mg), 4,4'-bipyridine (1 mmol, 156.2 mg), and 4-bromobenzoic acid (2 mmol, 402.0 mg) was dissolved in 1 mL dimethylformamide and 3 mL H_2O and heated in a 15 mL autoclave under autogenous pressure at 413 K for 72 h. After cooling to room temperature, colourless block crystals formed.

Experimental details

Hydrogen atoms were positioned using a riding model, with $C-H = 0.93$ Å and $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$.

Discussion

The design and construction of coordination polymers are of great interest because of their fascinating architectures and interesting properties [3–8]. Using the carboxylic acid and nitrogen containing mixed ligands is the common synthetic strategy for the synthesis of coordination polymers [5, 9–11]. In this study, we selected 4-bromobenzoate and 4,4'-bipyridine as ligands to react with $Zn(NO_3)_2 \cdot 6H_2O$.

In the title compound, there are one Zn(II) cation, two 4-bromobenzoate ligands, and two half of the 4,4'-bipyridine ligand in the asymmetric unit. The atom Zn1 is four coordinated with a distorted tetrahedral geometry by two N atoms from two symmetry 4,4'-bipyridine ligands and two

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Abstract

$C_{24}H_{16}Br_2N_2O_4Zn$, monoclinic, $I2/a$ (no. 15), $a = 7.4191(9)$ Å, $b = 24.458(3)$ Å, $c = 25.810(3)$ Å, $\beta = 92.426(3)^\circ$, $V = 4679.2(10)$ Å³, $Z = 8$, $R_{\text{gt}}(F) = 0.0514$, $wR_{\text{ref}}(F^2) = 0.1710$, $T = 296(2)$ K.

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A part of the title polymeric structure is shown in the figure. Tables 1 and 2 contain details of the measurement method and a list of the atoms including atomic coordinates and displacement parameters.

*Corresponding author: Yingchun Luo, College of Chemical Engineering, Guizhou Minzu University, 550025, Guiyang, Guizhou, P. R. China, E-mail: 1046984620@qq.com. <https://orcid.org/0000-0001-8407-8814>

Xiuyan Yang, College of Chemical Engineering, Guizhou Minzu University, 550025, Guiyang, Guizhou, P. R. China

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

	x	y	z	U _{iso} [*] /U _{eq}
N1	0.3343 (6)	0.39270 (18)	0.91798 (16)	0.0347 (11)
N2	0.4173 (6)	0.30300 (19)	0.83117 (18)	0.0389 (11)
O1	0.0591 (6)	0.36148 (18)	0.81889 (16)	0.0540 (12)
O2	−0.1100 (6)	0.3944 (2)	0.87963 (17)	0.0686 (15)
O3	0.1359 (9)	0.2886 (2)	0.9383 (2)	0.0899 (18)
O4	0.1074 (8)	0.2340 (3)	0.8745 (2)	0.096 (2)
Zn1	0.20946 (9)	0.33098 (3)	0.87544 (3)	0.0397 (2)
Br1	−0.65849 (14)	0.45154 (4)	0.66781 (3)	0.0930 (4)
Br2	−0.06264 (13)	0.05801 (4)	1.07217 (4)	0.0862 (3)
C1	0.3135 (7)	0.4451 (2)	0.9045 (2)	0.0358 (13)
H1	0.252135	0.453118	0.873212	0.043*
C2	0.3793 (7)	0.4882 (2)	0.93497 (19)	0.0314 (12)
H2	0.364100	0.524058	0.923595	0.038*
C3	0.4680 (6)	0.4776 (2)	0.98250 (19)	0.0272 (11)
C4	0.4908 (7)	0.4229 (2)	0.9960 (2)	0.0375 (13)
H4	0.551670	0.413641	1.026999	0.045*
C5	0.4243 (8)	0.3827 (2)	0.9640 (2)	0.0397 (14)
H5	0.441381	0.346501	0.974098	0.048*
C6	0.5737 (9)	0.2843 (3)	0.8502 (2)	0.0542 (17)
H6	0.595445	0.285745	0.885947	0.065*
C7	0.7069 (9)	0.2627 (3)	0.8204 (2)	0.0491 (16)
H7	0.813937	0.249841	0.835930	0.059*
C8	0.6788 (8)	0.2605 (2)	0.7667 (2)	0.0370 (14)
C9	0.5167 (8)	0.2812 (3)	0.7468 (2)	0.0494 (16)
H9	0.492385	0.282006	0.711158	0.059*
C10	0.3914 (8)	0.3008 (3)	0.7803 (3)	0.0529 (17)
H10	0.281561	0.313178	0.766047	0.063*
C11	−0.0794 (8)	0.3867 (2)	0.8340 (2)	0.0438 (15)
C12	−0.2108 (8)	0.4055 (2)	0.7929 (2)	0.0402 (14)
C13	−0.1870 (9)	0.3957 (3)	0.7405 (2)	0.0511 (16)
H13	−0.081027	0.379184	0.730458	0.061*
C14	−0.3169 (10)	0.4099 (3)	0.7030 (3)	0.0596 (19)
H14	−0.298974	0.402650	0.668205	0.072*
C15	−0.4734 (10)	0.4350 (3)	0.7175 (3)	0.0578 (18)
C16	−0.4995 (10)	0.4472 (3)	0.7694 (3)	0.0591 (19)
H16	−0.603679	0.464874	0.779207	0.071*
C17	−0.3674 (8)	0.4324 (3)	0.8060 (2)	0.0486 (16)
H17	−0.383750	0.440747	0.840604	0.058*
C18	0.0985 (10)	0.2440 (4)	0.9194 (3)	0.066 (2)
C19	0.0427 (8)	0.2004 (3)	0.9586 (2)	0.0483 (16)
C20	0.0497 (10)	0.1480 (3)	0.9417 (3)	0.066 (2)
H20	0.075012	0.140893	0.907401	0.079*
C21	0.0191 (10)	0.1046 (3)	0.9755 (3)	0.0623 (19)
H21	0.028166	0.068582	0.964477	0.075*
C22	−0.0248 (9)	0.1165 (3)	1.0255 (3)	0.0556 (18)
C23	−0.0390 (10)	0.1698 (3)	1.0424 (3)	0.061 (2)
H23	−0.072667	0.177433	1.075961	0.073*
C24	−0.0026 (10)	0.2111 (3)	1.0089 (3)	0.064 (2)
H24	−0.008435	0.247175	1.020185	0.076*

O atoms from two different 4-bromobenzoate ligands (see the figure). The Zn—O bond lengths are in the range of 1.948(4)–2.019(5) Å, and the Zn—N bond lengths are 2.063(5), and 2.074(4) Å [9, 12]. Zn(II) cations are linked by 4,4'-bipyridine ligands to form a Zigzag chain structure with the Zn—Zn distances of 11.222 and 11.240 Å.

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Conflict of interest statement: The authors declare no conflicts of interest regarding this article.

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