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Crystal structure of *catena*-poly[di- μ_2 -chlorido-1, 10-phenanthroline- $\kappa^2 N, N'$ -cadmium(II)], $C_{12}H_8Cl_2CdN_2$

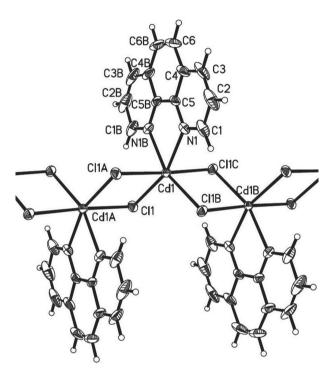


Figure 1: View of the title complex, showing the labeling of the 30% probability ellipsolids. H atoms have been omitted for clarity.

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Abstract

C₁₂H₈Cl₂CdN₂, monoclinic, *C*2/*c* (no. 15), a = 16.969(5) Å, b = 10.515(3) Å, c = 7.238(2) Å, $\beta = 110.710(10)^{\circ}$, V = 1208.0(6) Å³, Z = 4, $R_{\rm gt}(F) = 0.0245$, $wR_{\rm ref}(F^2) = 0.0806$, T = 298(2) K.

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A part of the title structure is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

Table 1: Data collection and handling.

| Crystal: | Colourless block | | |
|---|--|--|--|
| Size: | $0.21\times0.19\times0.18~\text{mm}$ | | |
| Wavelength: | Mo Kα radiation (0.71073 Å) | | |
| μ : | 2.22 mm ⁻¹ | | |
| Diffractometer, scan mode: | Bruker D8 VENTURE PHOTON, $oldsymbol{arphi}$ | | |
| | and ω | | |
| θ_{max} , completeness: | 27.6°, >99% | | |
| $N(hkl)_{measured}$, $N(hkl)_{unique}$, R_{int} : | 16070, 1402, 0.029 | | |
| Criterion for I_{obs} , $N(hkl)_{gt}$: | $I_{\rm obs}>2~\sigma(I_{\rm obs})$, 1303 | | |
| N(param) _{refined} : | 79 | | |
| Programs: | Bruker [1], SHELX [2, 3] | | |

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\mathring{A}^2).

| Atom | х | у | Z | U _{iso} */U _{eq} |
|------|-------------|------------|-------------|------------------------------------|
| Cd1 | 0.5000 | 0.92906(2) | 0.7500 | 0.03231(17) |
| Cl1 | 0.59574(5) | 1.07724(6) | 1.01183(11) | 0.0398(2) |
| N1 | 0.42417(14) | 0.7464(2) | 0.5998(3) | 0.0387(5) |
| C1 | 0.3488(2) | 0.7473(4) | 0.4548(5) | 0.0596(9) |
| H1 | 0.3252 | 0.8245 | 0.3999 | 0.071* |
| C2 | 0.3055(3) | 0.6359(6) | 0.3849(7) | 0.0855(17) |
| H2 | 0.2522 | 0.6397 | 0.2874 | 0.103* |
| С3 | 0.3381(4) | 0.5245(5) | 0.4533(7) | 0.092(2) |
| Н3 | 0.3079 | 0.4505 | 0.4045 | 0.110* |
| C4 | 0.4185(3) | 0.5176(3) | 0.6004(6) | 0.0685(13) |
| C5 | 0.4588(2) | 0.6342(2) | 0.6722(4) | 0.0427(7) |
| C6 | 0.4623(4) | 0.4025(3) | 0.6815(8) | 0.106(3) |
| Н6 | 0.4364 | 0.3250 | 0.6353 | 0.128* |
| | | | | |

Source of material

A N,N-dimethylformamide (DMF) solution (3 mL) of 1,10-phenanthroline (0.0360 g, 0.2 mmol) was placed into the bottom of the tube as a lower layer, and then a pure methanol solvent (7 mL) was added slowly along the tube wall into the tube as a middle layer, finally, a methanol solution (3 mL) of CdCl₂

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(0.0366 g, 0.2 mmol) was also put cautiously along the tube wall into the tube as a upper layer. The resulting three-layer solution was not mixed and was left at room temperature. After three weeks, quality colorless crystals of the title polymer were obtained from the solution and dried in air.

Experimental details

H atoms were generated geometrically and refined as riding atoms with C-H = 0.93 Å and $U_{\rm iso}$ (H) = 1.2 times $U_{\rm eq}$ (C) for aromatic H atoms.

Comment

As is well known, 1,10-phenanthroline, a rigid ligand with a six-element heterocyclic structure containing nitrogen, has a three-ring conjugated planar structure, so that the electron density of nitrogen atoms on the aromatic ring is strengthened [4, 5]. So phenanthroline complexes may be luminous material with good luminous properties. When it is coordinated with metal ions, as a bidentate chelating ligand, novel and stable metal organic complexes with potential and superior optical functions can be formed [6–10].

Single crystal X-ray diffraction analysis reveals that the title structure contain a 1-D polymer. As is shown in the figure, The Cd atom is six-coordinated by two imidazole N atoms of one 1,10-phenanthroline ligand and four bridging chloride ions in a distorted octahedral geometry. The two bond lengths of Cd—N are 2.353(2) Å, the four bond lengths of Cd—Cl are 2.5457(9) and 2.7561(10) Å, respectively. The bond angles around the Cd atom vary from 70.57(12)° (N1—Cd1—N1) to 177.25(3)° (Cl1—Cd1—Cl1). In the crystal, the adjacent molecules are stacked through the aromatic π – π interactions (the average interplanar distance between phenanthroline rings is 3.62 Å) and intermolecular interactions to form a three-dimensional supramolecular network.

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