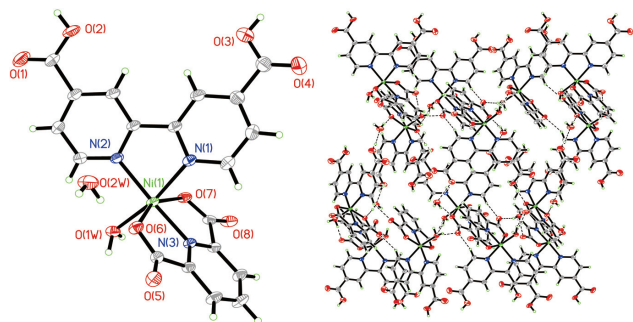


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Crystal structure of biaqua(2,2'-bipyridine-4,4'-dicarboxylato- $\kappa^2 N, N'$)(pyridine-2,6-dicarboxylato- $\kappa^3 O, N, O'$)nickel(II) hydrate, $C_{19}H_{15}N_3NiO_{10}$



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Abstract

$\text{C}_{19}\text{H}_{15}\text{N}_3\text{NiO}_{10}$, monoclinic, $P2_1/c$ (no. 14), $a = 12.7175(10)$ Å, $b = 10.6780(10)$ Å, $c = 15.0802(16)$ Å, $\beta = 96.786(9)^\circ$, $V = 2033.5(3)$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.0358$, $wR_{\text{ref}}(F^2) = 0.0960$, $T = 293(2)$ K.

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The molecular structure is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

Source of material

The $\text{Ni}(\text{CH}_3\text{COO})_2$ (0.08840 g, 0.5 mmol) and pyridine-2,6-dicarboxylic acid (0.04187 g, 0.25 mmol) in 10 mL H_2O /methanol (1:4, *v/v*) reacted for one and a half hours. And 2,2'-bipyridine-4,4'-dicarboxylic acid (0.2442 g, 1 mmol) was put into the solution. The filtrate was kept at 293 K to obtain green block crystals.

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Table 1: Data collection and handling.

Crystal:	Green block
Size:	0.24 × 0.22 × 0.20 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	1.02 mm ⁻¹
Diffractometer, scan mode:	XtaLAB Pro, ω
$\theta_{\max}$, completeness:	29.9°, 99%
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	22810, 5266, 0.029
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 4244
$N(\text{param})_{\text{refined}}$:	302
Programs:	CrysAlis ^{PRO} [1], SHELX [2, 3], WinGX/ORTEP [4]

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$).

Atom	x	y	z	U_{iso}^*/U_{eq}
Ni1	0.84135(2)	0.36710(2)	0.69566(2)	0.02922(8)
C1	0.70667(17)	0.4616(2)	0.83749(16)	0.0438(5)
H1	0.769195	0.461094	0.876300	0.053*
C2	0.61472(18)	0.5023(3)	0.86990(17)	0.0467(5)
H2	0.615442	0.528650	0.928750	0.056*
C3	0.52209(17)	0.5024(2)	0.81175(16)	0.0422(5)
C4	0.52411(16)	0.4613(2)	0.72554(16)	0.0419(5)
H4	0.462215	0.459850	0.685974	0.050*
C5	0.61879(15)	0.4222(2)	0.69800(15)	0.0353(4)
C6	0.62660(15)	0.3810(2)	0.60602(15)	0.0353(4)
C7	0.53850(17)	0.3548(2)	0.54547(16)	0.0424(5)
H5	0.470943	0.359442	0.563060	0.051*
C8	0.55168(19)	0.3218(2)	0.45917(16)	0.0441(5)
C9	0.6533(2)	0.3186(2)	0.43498(16)	0.0466(5)
H7	0.664433	0.301315	0.376353	0.056*
C10	0.73795(18)	0.3413(2)	0.49905(15)	0.0413(5)
H6	0.806100	0.335617	0.482823	0.050*
C11	0.41925(18)	0.5453(2)	0.84244(17)	0.0465(5)
C12	0.4585(2)	0.2899(2)	0.39212(18)	0.0520(6)
C13	1.03089(14)	0.39946(18)	0.61165(13)	0.0305(4)
C14	1.11847(17)	0.3659(2)	0.57033(17)	0.0422(5)
H9	1.166808	0.425989	0.556378	0.051*
C15	1.13241(19)	0.2408(2)	0.55026(18)	0.0493(6)
H10	1.189710	0.216489	0.521283	0.059*
C16	1.06112(17)	0.1522(2)	0.57333(17)	0.0409(5)
H11	1.069421	0.068036	0.559849	0.049*
C17	0.97737(14)	0.19152(18)	0.61682(14)	0.0309(4)
C18	1.00001(15)	0.53119(18)	0.63515(13)	0.0310(4)
C19	0.89373(14)	0.10798(18)	0.64980(15)	0.0324(4)

Table 2 (continued)

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso} [*] / <i>U</i> _{eq}
N1	0.72618(13)	0.37106(16)	0.58367(12)	0.0330(3)
N2	0.71014(12)	0.42301(17)	0.75361(12)	0.0347(4)
N3	0.96324(12)	0.31244(15)	0.63390(11)	0.0284(3)
O1	0.41016(17)	0.5689(3)	0.91797(14)	0.0781(7)
O2	0.34376(14)	0.5535(2)	0.77592(13)	0.0650(6)
H3	0.291102	0.584812	0.793466	0.097*
O3	0.36887(16)	0.2950(3)	0.42567(15)	0.0709(6)
H8	0.320167	0.280056	0.386456	0.106*
O4	0.46925(18)	0.2639(2)	0.31612(14)	0.0711(6)
O5	1.05591(13)	0.61827(14)	0.61420(11)	0.0425(4)
O6	0.91722(11)	0.54046(13)	0.67501(10)	0.0362(3)
O7	0.82414(10)	0.16576(13)	0.68847(10)	0.0357(3)
O8	0.89770(12)	−0.00642(14)	0.63829(13)	0.0482(4)
O1W	0.92419(11)	0.35060(13)	0.82186(10)	0.0365(3)
H1A	0.940876	0.272703	0.833419	0.055*
H1B	0.985521	0.387594	0.824409	0.055*
O2W	0.79751(14)	0.75986(18)	0.68385(14)	0.0595(5)
H2A	0.842230	0.700438	0.686287	0.089*
H2B	0.828800	0.828988	0.677110	0.089*

Experimental details

Data of crystal were collected via CrysAlis^{PRO} at 293 K [1], the structure was solved by SHELXT [2] program and refined by SHELXL [3] within WinGX [4].

Comment

As is shown in the left part of picture, the Ni1 is six-coordinated by one oxygen atom from water, two nitrogen

atoms from one 2,2'-bipyridine-4,4'-dicarboxylic acid ligand, one nitrogen atom and two oxygen atoms from the pyridine-2,6-dicarboxylato ligand (*cf.* the figure). Thus, there is one free water molecule. Most importantly, the weak interactions such as CH $\cdots$ O, OH $\cdots$ O could be found and make the crystal structure become 3D supermolecule structure. All geometric parameters are in the expected ranges and agree with an analogous Cu complex [5].

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