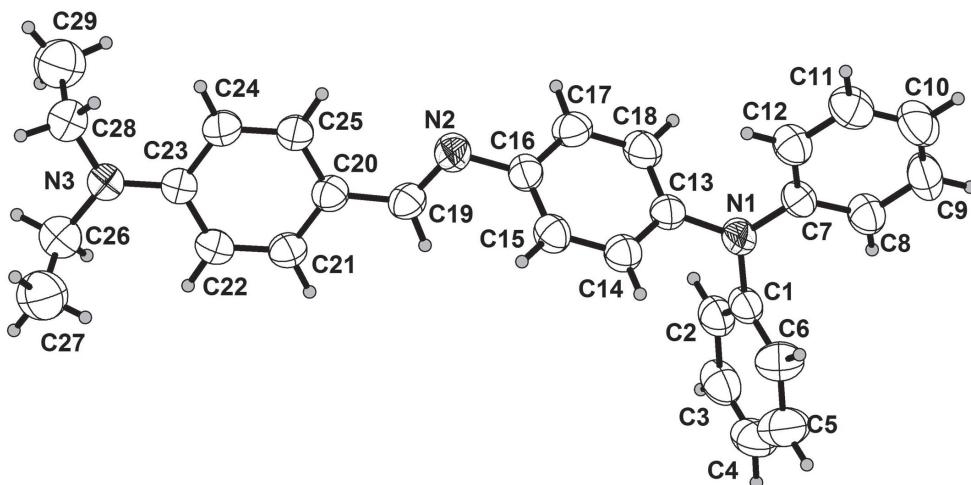


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The crystal structure of (*E*)-4-((4-(diethylamino)benzylidene)amino)-*N,N*-diphenylaniline, C₂₉H₂₉N₃



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Abstract

C₂₉H₂₉N₃, triclinic, $P\bar{1}$ (no. 2), $a = 10.677(4)$ Å, $b = 11.325(4)$ Å, $c = 21.530(8)$ Å, $\alpha = 88.026(5)^\circ$, $\beta = 75.646(5)^\circ$, $\gamma = 71.504(4)^\circ$, $V = 2389.0(15)$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.0663$, $wR_{\text{ref}}(F^2) = 0.2002$, $T = 293(2)$ K.

CCDC no.: 1949288

One of two crystallographically independent molecules of the title structure is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

Source of material

Reagents and solvents used were of commercially available quality. The ethanol solution of *N,N*-diphenylbenzene-1,4-diamine (0.26 g, 1 mmol) was added to a solution containing

Table 1: Data collection and handling.

| | |
|--|--|
| Crystal: | Black needle |
| Size: | 0.30 × 0.10 × 0.10 mm |
| Wavelength: | Mo K α radiation (0.71073 Å) |
| μ : | 0.07 mm ⁻¹ |
| Diffractometer, scan mode: | Bruker SMART, φ and ω |
| θ_{max} , completeness: | 25.0°, 99% |
| $N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} : | 17204, 8351, 0.136 |
| Criterion for I_{obs} , $N(hkl)_{\text{gt}}$: | $I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 4965 |
| $N(\text{param})_{\text{refined}}$: | 581 |
| Programs: | Bruker [1], SHELX [2, 3] |

4-(diethylamino)benzaldehyde (0.18 g, 1 mmol) in 5 mL of absolute ethanol and acetic acid (1 day) under heating and stirring. The mixture was then refluxed for 3 h. Afterwards the mixture was cooled to room temperature. It was pured by ethanol recrystallization in 82% yield as yellow solid. Then the resulting solution had to stand in air. On slow evaporation of the solvent, some black block crystals were obtained after 5 days. ¹H NMR (400 MHz, (CD₃)₂CO) δ [ppm] 8.39 (d, $J = 5.0$ Hz, 1H), 7.75–7.65 (m, 2H), 7.30–7.22 (m, 4H), 7.20–7.11 (m, 2H), 7.07–6.93 (m, 8H), 6.82–6.69 (m, 2H), 3.46 (q, $J = 7.1$ Hz, 4H), 1.16 (t, $J = 7.0$ Hz, 6H). ¹³C NMR (100 MHz, (CD₃)₂CO) δ [ppm] 158.47, 150.13, 148.43, 148.07, 144.95, 130.55, 129.34, 125.31, 124.13, 123.64, 122.59, 121.98, 111.01, 44.18, 12.03. MALDI-TOF: ([M + H]⁺), 516.85. IR (KBr pellet, cm⁻¹): 3447, 1600, 1545, 1495, 1455, 1415, 1375, 1335, 1295, 1255, 1215, 1175, 1135, 1095, 1055, 1015, 975, 935, 895, 855, 815, 775, 735, 695, 655, 615, 575, 535, 495, 455, 415, 375, 335, 295, 255, 215, 175, 135, 105, 65, 55, 45, 35, 25, 15, 5 cm⁻¹.

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Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

| Atom | x | y | z | <i>U</i> _{iso} */* <i>U</i> _{eq} |
|------|------------|-----------|--------------|--|
| C1 | 0.0858(3) | 1.3253(2) | 0.23642(11) | 0.0568(6) |
| C2 | -0.0257(3) | 1.3344(3) | 0.21327(11) | 0.0659(7) |
| H2 | -0.0551 | 1.2654 | 0.2131 | 0.079* |
| C3 | -0.0944(3) | 1.4449(3) | 0.19034(12) | 0.0773(9) |
| H3 | -0.1696 | 1.4502 | 0.1746 | 0.093* |
| C4 | -0.0520(4) | 1.5462(3) | 0.19080(14) | 0.0889(10) |
| H4 | -0.0980 | 1.6206 | 0.1751 | 0.107* |
| C5 | 0.0581(4) | 1.5388(3) | 0.21428(17) | 0.0960(10) |
| H5 | 0.0861 | 1.6085 | 0.2149 | 0.115* |
| C6 | 0.1275(3) | 1.4289(3) | 0.23684(14) | 0.0787(8) |
| H6 | 0.2027 | 1.4242 | 0.2524 | 0.094* |
| C7 | 0.1808(2) | 1.1991(2) | 0.31988(10) | 0.0515(6) |
| C8 | 0.1000(3) | 1.2881(3) | 0.36818(12) | 0.0647(7) |
| H8 | 0.0302 | 1.3554 | 0.3598 | 0.078* |
| C9 | 0.1231(3) | 1.2769(3) | 0.42888(13) | 0.0812(9) |
| H9 | 0.0685 | 1.3373 | 0.4610 | 0.097* |
| C10 | 0.2247(4) | 1.1787(3) | 0.44250(15) | 0.0855(9) |
| H10 | 0.2396 | 1.1725 | 0.4834 | 0.103* |
| C11 | 0.3042(3) | 1.0897(3) | 0.39519(15) | 0.0780(8) |
| H11 | 0.3732 | 1.0222 | 0.4040 | 0.094* |
| C12 | 0.2822(3) | 1.0996(3) | 0.33455(12) | 0.0646(7) |
| H12 | 0.3366 | 1.0382 | 0.3029 | 0.078* |
| C13 | 0.2159(2) | 1.1067(2) | 0.21203(11) | 0.0539(6) |
| C14 | 0.3095(2) | 1.1167(2) | 0.15639(12) | 0.0584(6) |
| H14 | 0.3389 | 1.1863 | 0.1511 | 0.070* |
| C15 | 0.3590(2) | 1.0232(2) | 0.10872(12) | 0.0584(6) |
| H15 | 0.4207 | 1.0309 | 0.0711 | 0.070* |
| C16 | 0.3183(2) | 0.9183(2) | 0.11608(11) | 0.0533(6) |
| C17 | 0.2316(3) | 0.9060(2) | 0.17374(12) | 0.0611(7) |
| H17 | 0.2082 | 0.8333 | 0.1807 | 0.073* |
| C18 | 0.1796(3) | 0.9999(2) | 0.22080(12) | 0.0635(7) |
| H18 | 0.1196 | 0.9913 | 0.2588 | 0.076* |
| C19 | 0.3485(2) | 0.8595(2) | 0.01129(12) | 0.0561(6) |
| H19 | 0.3091 | 0.9445 | 0.0075 | 0.067* |
| C20 | 0.3887(2) | 0.7798(2) | -0.04595(11) | 0.0544(6) |
| C21 | 0.3597(3) | 0.8327(2) | -0.10230(12) | 0.0600(6) |
| H21 | 0.3161 | 0.9182 | -0.1018 | 0.072* |
| C22 | 0.3930(3) | 0.7635(2) | -0.15826(12) | 0.0649(7) |
| H22 | 0.3701 | 0.8021 | -0.1945 | 0.078* |
| C23 | 0.4610(3) | 0.6356(2) | -0.16165(12) | 0.0649(7) |
| C24 | 0.4921(3) | 0.5812(2) | -0.10483(13) | 0.0725(8) |
| H24 | 0.5375 | 0.4962 | -0.1053 | 0.087* |
| C25 | 0.4559(3) | 0.6525(2) | -0.04931(12) | 0.0660(7) |
| H25 | 0.4771 | 0.6144 | -0.0126 | 0.079* |
| C26 | 0.4371(4) | 0.6177(3) | -0.27436(14) | 0.0891(9) |
| H26A | 0.4263 | 0.5514 | -0.2977 | 0.107* |
| H26B | 0.3481 | 0.6794 | -0.2587 | 0.107* |
| C27 | 0.5279(4) | 0.6747(3) | -0.31668(17) | 0.1066(11) |
| H27A | 0.5381 | 0.7405 | -0.2934 | 0.160* |
| H27B | 0.4902 | 0.7084 | -0.3520 | 0.160* |
| H27C | 0.6153 | 0.6131 | -0.3326 | 0.160* |
| C28 | 0.5802(3) | 0.4315(3) | -0.22361(14) | 0.0821(9) |
| H28A | 0.6298 | 0.4083 | -0.2680 | 0.098* |
| H28B | 0.6460 | 0.4178 | -0.1979 | 0.098* |

Table 2 (continued)

| Atom | x | y | z | <i>U</i> _{iso} */* <i>U</i> _{eq} |
|------|-----------|------------|--------------|--|
| C29 | 0.4929(4) | 0.3554(3) | -0.20192(18) | 0.1034(11) |
| H29A | 0.4568 | 0.3677 | -0.1563 | 0.155* |
| H29B | 0.5448 | 0.2692 | -0.2125 | 0.155* |
| H29C | 0.4192 | 0.3788 | -0.2226 | 0.155* |
| C30 | 0.5116(3) | 0.2899(3) | 0.56125(12) | 0.0624(7) |
| C31 | 0.4147(3) | 0.3507(3) | 0.52904(14) | 0.0909(10) |
| H31 | 0.4404 | 0.3583 | 0.4849 | 0.109* |
| C32 | 0.2801(3) | 0.4001(4) | 0.56180(19) | 0.1088(12) |
| H32 | 0.2155 | 0.4415 | 0.5397 | 0.131* |
| C33 | 0.2403(3) | 0.3888(3) | 0.62673(17) | 0.0936(10) |
| H33 | 0.1490 | 0.4224 | 0.6486 | 0.112* |
| C34 | 0.3340(3) | 0.3289(3) | 0.65868(14) | 0.0791(8) |
| H34 | 0.3071 | 0.3212 | 0.7027 | 0.095* |
| C35 | 0.4707(3) | 0.2784(3) | 0.62644(12) | 0.0694(7) |
| H35 | 0.5346 | 0.2368 | 0.6489 | 0.083* |
| C36 | 0.7548(2) | 0.2454(3) | 0.55679(11) | 0.0572(7) |
| C37 | 0.7520(3) | 0.3548(3) | 0.58315(14) | 0.0804(9) |
| H37 | 0.6808 | 0.4276 | 0.5824 | 0.096* |
| C38 | 0.8554(4) | 0.3572(4) | 0.61108(16) | 0.1030(12) |
| H38 | 0.8532 | 0.4316 | 0.6290 | 0.124* |
| C39 | 0.9607(4) | 0.2504(5) | 0.61222(16) | 0.1033(12) |
| H39 | 1.0294 | 0.2518 | 0.6313 | 0.124* |
| C40 | 0.9640(3) | 0.1438(4) | 0.58559(16) | 0.0923(10) |
| H40 | 1.0359 | 0.0714 | 0.5860 | 0.111* |
| C41 | 0.8631(3) | 0.1401(3) | 0.55784(13) | 0.0747(8) |
| H41 | 0.8676 | 0.0653 | 0.5394 | 0.090* |
| C42 | 0.6871(2) | 0.1799(3) | 0.46658(11) | 0.0573(6) |
| C43 | 0.6324(2) | 0.0916(3) | 0.45444(11) | 0.0613(7) |
| H43 | 0.5686 | 0.0712 | 0.4872 | 0.074* |
| C44 | 0.6700(3) | 0.0323(3) | 0.39461(11) | 0.0604(7) |
| H44 | 0.6326 | -0.0283 | 0.3876 | 0.072* |
| C45 | 0.7635(2) | 0.0626(2) | 0.34501(10) | 0.0518(6) |
| C46 | 0.8215(3) | 0.1485(3) | 0.35790(12) | 0.0633(7) |
| H46 | 0.8877 | 0.1669 | 0.3256 | 0.076* |
| C47 | 0.7836(3) | 0.2079(3) | 0.41755(12) | 0.0680(7) |
| H47 | 0.8227 | 0.2670 | 0.4250 | 0.082* |
| C48 | 0.7295(3) | -0.0331(2) | 0.25923(11) | 0.0539(6) |
| H48 | 0.6425 | -0.0240 | 0.2847 | 0.065* |
| C49 | 0.7677(2) | -0.0916(2) | 0.19576(10) | 0.0501(6) |
| C50 | 0.6774(2) | -0.1315(2) | 0.17253(11) | 0.0548(6) |
| H50 | 0.5911 | -0.1215 | 0.1990 | 0.066* |
| C51 | 0.7104(3) | -0.1854(2) | 0.11165(11) | 0.0552(6) |
| H51 | 0.6469 | -0.2118 | 0.0982 | 0.066* |
| C52 | 0.8389(3) | -0.2007(2) | 0.06994(10) | 0.0516(6) |
| C53 | 0.9305(3) | -0.1596(2) | 0.09359(11) | 0.0565(6) |
| H53 | 1.0167 | -0.1683 | 0.0673 | 0.068* |
| C54 | 0.8953(2) | -0.1069(2) | 0.15440(11) | 0.0555(6) |
| H54 | 0.9584 | -0.0808 | 0.1684 | 0.067* |
| C55 | 0.7805(3) | -0.2984(2) | -0.01588(12) | 0.0671(7) |
| H55A | 0.7985 | -0.2903 | -0.0620 | 0.080* |
| H55B | 0.6872 | -0.2475 | 0.0030 | 0.080* |
| C56 | 0.7961(3) | -0.4314(3) | -0.00145(17) | 0.0946(10) |
| H56A | 0.8877 | -0.4824 | -0.0210 | 0.142* |
| H56B | 0.7333 | -0.4576 | -0.0182 | 0.142* |
| H56C | 0.7772 | -0.4396 | 0.0442 | 0.142* |

Table 2 (continued)

| Atom | <i>x</i> | <i>y</i> | <i>z</i> | <i>U</i> _{iso} */* <i>U</i> _{eq} |
|------|-----------|-------------|--------------|--|
| C57 | 1.0035(3) | -0.2605(2) | -0.03587(12) | 0.0679(7) |
| H57A | 1.0279 | -0.3291 | -0.0670 | 0.081* |
| H57B | 1.0736 | -0.2782 | -0.0124 | 0.081* |
| C58 | 0.9995(3) | -0.1429(3) | -0.07073(13) | 0.0781(8) |
| H58A | 0.9297 | -0.1245 | -0.0937 | 0.117* |
| H58B | 1.0863 | -0.1537 | -0.1004 | 0.117* |
| H58C | 0.9801 | -0.0754 | -0.0404 | 0.117* |
| N1 | 0.1570(2) | 1.20921(19) | 0.25841(9) | 0.0642(6) |
| N2 | 0.3626(2) | 0.82293(19) | 0.06678(10) | 0.0622(6) |
| N3 | 0.4965(3) | 0.5650(2) | -0.21777(11) | 0.0933(9) |
| N4 | 0.6504(2) | 0.2397(2) | 0.52836(9) | 0.0714(7) |
| N5 | 0.8085(2) | 0.0066(2) | 0.28226(9) | 0.0588(5) |
| N6 | 0.8730(2) | -0.2525(2) | 0.00897(9) | 0.0628(6) |

3413s, 1596s, 1586m, 1525m, 1404w, 1317m, 1270m, 1317m, 1174s, 835m, 820m, 745m, 699s, 592m.

Experimental details

All H-atoms were placed in calculated positions and treated as riding: C—H = 0.93–0.97 Å, with *U*_{iso}(H) = 1.2 or 1.5 *U*_{eq} (parent C-atom).

Comment

Schiff bases bearing a bridged C=N structure have a wide range of applications in detection of metal ions, because they always have weak fluorescence signals for the C=N isomerization, but when the C=N group is coordinated with metal ions to form the complexes, the C=N isomerization is inhibited that results in much stronger fluorescence signals [4–6]. Among these compounds, the excellent electronic and optical properties of triphenylamine-based probes are studied widely for their butterfly-like structure, low ionization potential, high electron affinity and good UV light harvesting features [6, 7]. As part of an ongoing study concerning the optical properties of Schiff bases [8], the title compound is reported here.

The asymmetric unit of the title structure consists of two independent molecules of title molecule. In both independent molecules, the C=N bond adopts *E* or *trans* configurations. These core fragments of the title molecule are almost planar with torsion angles C20—N19—N2—C16 and

C49—N48—N5—C45 of 178.7(2)° and -179.9(2)°, respectively, the dihedral angles between the mean planes of the phenyl rings are 55.3(2)° and 151.1(1)°, respectively. Geometric parameters are in general as expected [9].

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