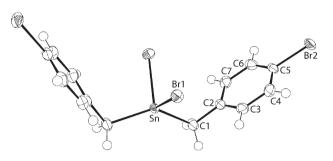
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Crystal structure of dibromidobis(4-bromobenzyl) tin(IV), $C_{14}H_{12}Br_4Sn$



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Abstract

 $C_{14}H_{12}Br_4Sn$, monoclinic, I2/a (no. 15), a = 12.6379(2) Å, b = 4.9674(1) Å, c = 26.1845(4) Å, $\beta = 94.507(1)^\circ$, V = 1638.71(5) Å³, Z = 4, $R_{gt}(F) = 0.0142$, $wR_{ref}(F^2) = 0.0358$, T = 100 K.

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The molecular structure of the title complex is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

Source of material

The melting point (uncorrected) was measured on an Mel-Temp II melting point apparatus. 1H and $^{13}\{^1H\}C$ NMR spectra were recorded in DMSO-d₆ solution on a Bruker Ascend 400 MHz NMR (Billerica, MA, USA) spectrometer.

Di(4-bromobenzyl)tin dibromide was prepared from the direct reaction of 4-bromobenzyl bromide (Merck, 0.50 g, 2.0 mmol) and metallic tin powder (Sigma-Aldrich, 0.12 g, 1.0 mmol) in toluene according to a literature procedure [5]. After refluxing for 5 h, fine white powders were observed in

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Table 1: Data collection and handling.

Colourless prism
$0.07 \times 0.03 \times 0.02~\text{mm}$
Cu Κα radiation (1.54184 Å)
$23.7 \ \text{mm}^{-1}$
XtaLAB Synergy, ω
67.0°, >99%
10027, 1450, 0.031
$I_{\rm obs} > 2 \ \sigma(I_{\rm obs})$, 1418
87
CrysAlis ^{PRO} [1], SHELX [2, 3], WinGX/ORTEP [4]

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\mathring{A}^2).

Atom	х	у	Z	U _{iso} */U _{eq}
Sn	0.250000	-0.00473(4)	0.500000	0.01254(7)
Br1	0.39552(2)	0.32357(5)	0.52646(2)	0.01689(7)
Br2	0.41779(2)	0.70829(5)	0.27602(2)	0.02498(8)
C1	0.29003(19)	-0.1656(5)	0.42709(7)	0.0202(5)
H1A	0.350663	-0.291059	0.432852	0.024*
H1B	0.228858	-0.268688	0.411326	0.024*
C2	0.31845(18)	0.0518(5)	0.39113(8)	0.0163(4)
С3	0.42172(17)	0.1525(5)	0.39318(7)	0.0188(5)
Н3	0.473217	0.086032	0.418448	0.023*
C4	0.45068(18)	0.3473(5)	0.35910(8)	0.0193(5)
H4	0.521338	0.413915	0.360804	0.023*
C5	0.37503(19)	0.4434(5)	0.32252(8)	0.0179(4)
C6	0.27165(18)	0.3493(5)	0.31938(8)	0.0195(5)
Н6	0.220605	0.416788	0.294006	0.023*
C7	0.24382(18)	0.1548(5)	0.35391(7)	0.0189(5)
H7	0.172856	0.090381	0.352247	0.023*

the mixture and the contents of the reaction vessel became grey. The obtained slurry was filtered after cooling and the residue was extracted with acetone/toluene. Colourless crystals suitable for crystallographic studies were obtained from the slow evaporation of the filtrate. Yield: 0.32 g (52%). M. pt: 453–455 K. 1 H NMR (DMSO-d₆, p.p.m.): δ 7.19–7.21 (d, 4H, PhH, 3 J = 8.32 Hz), 7.38–7.40 (d, 4H, PhH, 3 J = 8.32 Hz), 3.10 (s, 4H, Ph—CH₂). 13 C{ 1 H} NMR (DMSO-d₆, p.p.m.): δ 50.8 (–CH₂), 118.1, 139.9, 132.1, 139.4 (Ph—C).

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Experimental details

The C-bound H placed atoms were geometrically (C-H = 0.95 - 0.99 Å)refined and as riding with $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}({\rm C}).$

Comment

Interest in diorganotin dihalido molecules similar to the title compound [6, 7] relate to the ability of the diorganotin centre to increase its coordination number via intra- and intermolecular Sn··· X secondary [8] interactions. Besides structural consequences, the formation of intramolecular $Sn \cdots X$ interactions has implications for the stability and function of organotin species investigated as potential anti-tumour agents [9].

The molecule is illustrated in the figure (70% displacement ellipsoids; unlabelled atoms are related by the symmetry operation (i) 1/2 - x, y, 1 - z). The title complex has crystallographic two-fold symmetry with the tin atom lying on the two-fold axis. The tin atom is tetrahedrally coordinated by two bromido atoms [Sn-Br1 = 2.5143(2) Å]and two methylene-carbon ligands [2.166(2) Å] derived from the 4-bromobenzyl substituents. The resulting Br₂C₂ donor set exhibits significant deviations from the regular tetrahedral geometry with the angles correlating with steric effects. Thus, the narrowest angle of 99.125(12)° is subtended by the bromido atoms and the widest angle of 136.70(12)° by the methylene-carbon atoms. From symmetry, the phenyl rings are orientated towards the same side of the molecule but, are splayed as seen in the dihedral angle between them of 88.7(5)°.

In the molecular packing, a three-dimensional architecture is sustained by a combination of side-on $C-Br \cdots \pi$, secondary Sn··· Br interactions and Br··· Br halogen bonding. Globally, molecules assemble into columns parallel to the b-axis, with connections between them being of the type $C-Br \cdots \pi$ $[C5-Br2 \cdots Cg(C2-C7)^{ii}: Br2 \cdots Cg(C2-C7)^{ii}]$ $(C7)^{ii} = 3.5754(10) \text{ Å}$ with angle at $Br2 = 92.65(7)^{\circ}$ for symmetry operation (ii) x, 1+y, z] and $Sn \cdots Br$ $[Sn \cdots Br^{ii} = 3.8464(3) \text{ Å}]$. Interactions between columns to form layers in the ab-plane involve the coordinated bromido ligands $[Br1 \cdots Br1^{iii} = 3.5414(4) \text{ Å for (iii): } 1 - x, 1 - y, 1 - z].$

The layers are connected along the c-axis by bromo··· bromo interactions [Br2···Br2^{iv} = 3.5761(4) Å for (iv): 1 - x, -1/2 + y,

The crystal structure of the isostructural all-chloro derivative of the title compound is known [10]. The Cl-Sn-Cl and C-Sn-C angles in the literature known compound are 98.68(3) and 136.15(12)°, respectively, both marginally smaller than in the present structure.

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