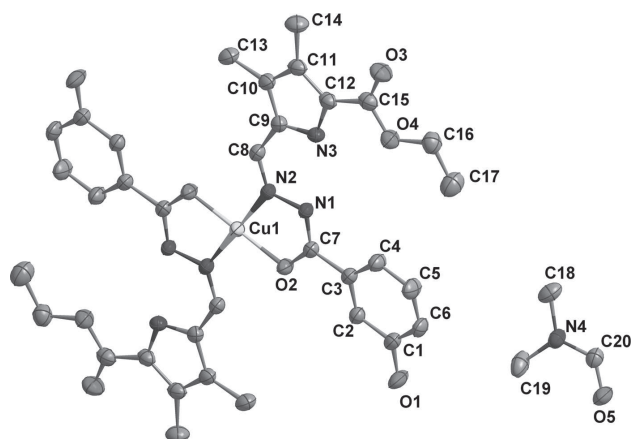


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# Crystal structure of bis-(*N'*-(5-ethoxycarbonyl-3,4-dimethyl-pyrrol-2-yl-methylidene)-3-hydroxybenzohydrazide- $\kappa^2 O, N$ )copper(II) – dimethylformamide (1/2), $C_{40}H_{50}N_8O_{10}Cu$



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## Abstract

$C_{40}H_{50}N_8O_{10}Cu$ , triclinic,  $P\bar{1}$  (no. 2),  $a = 6.8237(9)$  Å,  $b = 12.9229(17)$  Å,  $c = 13.3275(18)$  Å,  $\alpha = 104.949(2)^\circ$ ,  $\beta = 104.049(2)^\circ$ ,  $\gamma = 99.538(2)^\circ$ ,  $V = 1068.3(2)$  Å<sup>3</sup>,  $Z = 1$ ,  $R_{gt}(F) = 0.0497$ ,  $wR_{ref}(F^2) = 0.1343$ ,  $T = 296(2)$  K.

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The crystal structure is shown in the figure. The asymmetric unit is labelled. Tables 1 and 2 contain details on crystal structure and measurement conditions and a list of the atoms including atomic coordinates and displacement parameters.

## Source of material

The hydrazone ligand and the title complex were prepared according to the literature method [3].

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Table 1: Data collection and handling.

Crystal:	Brown block
Size:	0.23 × 0.20 × 0.19 mm
Wavelength:	Mo K $\alpha$ radiation (0.71073 Å)
$\mu$ :	0.58 mm <sup>-1</sup>
Diffractometer, scan mode:	APEX Detector, $\varphi$ and $\omega$
$\theta_{max}$ , completeness:	25.0°, >99%
$N(hkl)_{measured}$ , $N(hkl)_{unique}$ , $R_{int}$ :	5702, 3747, 0.021
Criterion for $I_{obs}$ , $N(hkl)_{gt}$ :	$I_{obs} > 2 \sigma(I_{obs})$ , 2778
$N(param)_{refined}$ :	268
Programs:	SHELX [5], Bruker [6]

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å<sup>2</sup>).

Atom	x	y	z	$U_{iso}^*/U_{eq}$
Cu1	1.000000	0.500000	0.000000	0.0476(2)
O1	0.8905(5)	0.1187(2)	0.2604(2)	0.0776(8)
H1B	0.849800	0.084257	0.298261	0.116*
O2	0.9427(4)	0.40078(18)	0.07822(18)	0.0530(6)
O3	0.1448(4)	0.7726(2)	0.2918(2)	0.0793(8)
O4	0.2821(4)	0.6268(2)	0.2928(2)	0.0646(7)
O5	0.2396(4)	-0.0059(2)	0.6170(2)	0.0749(8)
N1	0.7235(4)	0.5099(2)	0.1258(2)	0.0473(7)
N2	0.8059(4)	0.5662(2)	0.0629(2)	0.0473(7)
N3	0.5015(4)	0.6659(2)	0.1603(2)	0.0487(7)
H3A	0.512050	0.607331	0.178387	0.058*
N4	0.2398(4)	0.1341(2)	0.5467(2)	0.0556(7)
C1	0.7859(6)	0.1982(3)	0.2540(3)	0.0552(9)
C2	0.8399(5)	0.2685(3)	0.1970(3)	0.0501(8)
H2A	0.944850	0.259578	0.164595	0.060*
C3	0.7400(5)	0.3517(3)	0.1877(3)	0.0488(8)
C4	0.5812(7)	0.3639(3)	0.2349(3)	0.0698(11)
H4A	0.512540	0.419565	0.229250	0.084*
C5	0.5268(8)	0.2932(3)	0.2898(4)	0.0842(14)
H5B	0.418140	0.300306	0.319809	0.101*
C6	0.6295(7)	0.2118(3)	0.3017(3)	0.0721(11)
H6A	0.593700	0.166229	0.341575	0.086*
C7	0.8059(5)	0.4250(3)	0.1266(2)	0.0450(7)
C8	0.7487(5)	0.6545(3)	0.0534(2)	0.0485(8)
H8A	0.808270	0.689189	0.011151	0.058*
C9	0.6072(5)	0.7064(3)	0.0987(3)	0.0491(8)
C10	0.5470(6)	0.8014(3)	0.0872(3)	0.0521(8)
C11	0.4008(6)	0.8178(3)	0.1430(3)	0.0547(9)
C12	0.3758(5)	0.7331(3)	0.1887(3)	0.0508(8)
C13	0.6296(6)	0.8747(3)	0.0283(3)	0.0669(10)
H13A	0.562447	0.934621	0.032360	0.100*

Table 2 (continued)

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> <sub>iso</sub> <sup>*</sup> / <i>U</i> <sub>eq</sub>
H13B	0.777124	0.903502	0.061368	0.100*
H13C	0.602206	0.832719	−0.046617	0.100*
C14	0.2889(7)	0.9089(3)	0.1510(4)	0.0770(12)
H14A	0.334958	0.956395	0.112160	0.116*
H14B	0.141621	0.877769	0.119799	0.116*
H14C	0.318591	0.950859	0.226060	0.116*
C15	0.2543(6)	0.7157(3)	0.2616(3)	0.0585(9)
C16	0.1776(6)	0.6043(4)	0.3703(3)	0.0716(11)
H16A	0.027731	0.585877	0.337727	0.086*
H16B	0.215738	0.668773	0.434099	0.086*
C17	0.2424(7)	0.5108(4)	0.4011(4)	0.0950(15)
H17A	0.175266	0.494100	0.452480	0.142*
H17B	0.203503	0.447365	0.337419	0.142*
H17C	0.390742	0.529922	0.433398	0.142*
C18	0.1934(7)	0.2401(3)	0.5521(4)	0.0776(12)
H18A	0.226998	0.263997	0.494456	0.116*
H18B	0.047935	0.233755	0.544329	0.116*
H18C	0.274473	0.293087	0.620992	0.116*
C19	0.3166(7)	0.0817(4)	0.4598(4)	0.0892(15)
H19A	0.331837	0.129083	0.415859	0.134*
H19B	0.449238	0.068471	0.490112	0.134*
H19C	0.219781	0.012839	0.415660	0.134*
C20	0.2073(5)	0.0837(3)	0.6171(3)	0.0583(9)
H20	0.155428	0.119374	0.671227	0.070*

### Experimental details

The structure was solved by Direct Methods and refined with the SHELX crystallographic software package [5]. The hydrogen atoms were placed at calculated positions and refined as riding atoms with fixed isotropic displacement parameters.

### Comment

Acylhydrazones are an important class of ligands in coordination chemistry and have been found extensive application in different fields [1]. Our previous work show that acylhydrazone ligands bearing pyrrole units and their

complexes exhibit considerable antibacterial and antitumor activity [2–4]. As part of this study the title complex was synthesized and characterized by X-ray diffraction.

The asymmetric unit consists of one half of a copper(II) complex and one DMF molecule (*cf.* the figure). In the title crystal structure, the central copper ion is coordinated with two enolized acylhydrazone ligands by NO bidentate donor sets, thus giving a distorted planar square coordination geometry. Two neighbouring DMF molecules link with the complex *via* pairs of O—H···O hydrogen bonds. N—H···O hydrogen bonds are also presented in the crystal. Cu—O and Cu—N bond lengths are all in the expected ranges [7].

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