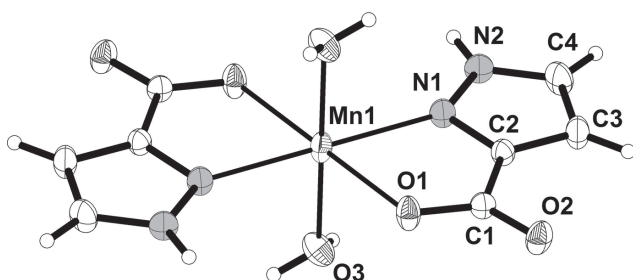


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Crystal structure of *trans*-diaqua-bis(1*H*-pyrazole-3-carboxylato- $\kappa^2 N,O$)manganese(II), $C_8H_{10}N_4O_6Mn$



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Abstract

$C_8H_{10}N_4O_6Mn$, monoclinic, $P2_1/c$ (no. 14), $a = 5.1531(5)$ Å, $b = 11.7052(11)$ Å, $c = 9.4154(7)$ Å, $\beta = 95.1540(10)^\circ$, $V = 565.62(9)$ Å³, $Z = 2$, $R_{gt}(F) = 0.0227$, $wR_{ref}(F^2) = 0.0611$, $T = 298$ K.

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The crystal structure is shown in the figure. The asymmetric unit of the title structure is labelled. Tables 1 and 2 contain details on crystal structure and measurement conditions and a list of the atoms including atomic coordinates and displacement parameters.

Source of material

A mixture of 1*H*-pyrazole-3-carboxylic acid (0.0224 g, 0.2 mmol), $Mn(OAc)_2 \cdot 4H_2O$ (0.0245 g, 0.1 mmol), and distilled water (10 mL) was heated under reflux at 180 °C for 72 h, and then cooled to room temperature. The resulting block crystals of the title compound were obtained and washed with distilled water. The yield is ca. 55% based on Mn.

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Table 1: Data collection and handling.

Crystal:	Colorless block
Size:	$0.38 \times 0.35 \times 0.30$ mm
Wavelength:	Mo $K\alpha$ radiation (0.71073 Å)
μ :	1.20 mm^{-1}
Diffractometer, scan mode:	Bruker SMART, φ and ω -scans
$2\theta_{\max}$, completeness:	25° , >99%
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	2751, 1001, 0.021
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 900
$N(\text{param})_{\text{refined}}$:	89
Programs:	Bruker programs [1], SHELX [2, 3]

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Mn1	0.5000	0.5000	0.5000	0.02409(19)
N1	0.3875(3)	0.68580(12)	0.50559(15)	0.0237(4)
N2	0.2305(3)	0.76013(13)	0.56490(16)	0.0259(4)
H2	0.1274	0.7417	0.6281	0.031*
O1	0.7423(2)	0.58436(11)	0.35255(14)	0.0277(3)
O2	0.8443(2)	0.75116(11)	0.25642(14)	0.0296(3)
O3	0.1991(3)	0.47570(12)	0.32306(15)	0.0322(3)
H3C	0.0513	0.5061	0.3310	0.039*
H3D	0.1757	0.4063	0.2984	0.039*
C1	0.7152(3)	0.69161(15)	0.33378(18)	0.0213(4)
C2	0.5139(3)	0.74888(15)	0.41438(17)	0.0211(4)
C3	0.4345(4)	0.86296(16)	0.4160(2)	0.0297(4)
H3	0.4916	0.9232	0.3624	0.036*
C4	0.2531(4)	0.86614(16)	0.5143(2)	0.0317(5)
H4	0.1626	0.9303	0.5408	0.038*

Experimental details

H atoms attached to C atoms were placed in geometrically idealized positions (C–H 0.93 Å) and treated as riding on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The water H-atoms were located in a difference Fourier map, and were included riding on the O3 atom.

Discussion

Many mononuclear complexes containing the 1*H*-pyrazole-3-carboxylate ligand have been reported, for example:

mononuclear cobalt(II), nickel(II) and zinc complexes with the 1*H*-pyrazole-3-carboxylate ligand [4, 5], but manganese complex based on 3-carboxy-pyrazole have not been reported.

There is one half of a Mn²⁺ cation, one Hpca[−] anion and one water molecule in the asymmetric unit of the title crystal structure. The Mn²⁺ center shows a distorted {MnO₄N₂} octahedral coordination geometry with two nitrogen atoms from two separate Hpca[−] anions, two chelate oxygen atoms from two water molecules and two carboxylic oxygen atoms (O1, O3) from two separate Hpca[−] anions. In the title compound, the equatorial plane is formed by two *N,O*-bidentate 1*H*-pyrazole-3-carboxylate ligands in a *trans* configuration. The axial positions are occupied by two water molecules. The mononuclear complex molecules are arranged in layers parallel to the *ab* plane. The distances of the Mn—O and Mn—N are 2.185(2) Å and 2.253(3) Å, respectively, for the organic ligand, and 2.190(2) Å for the water ligand, which is larger than the homologous Zn complex [5]. The bond angles around the Mn²⁺ ion are in the range of 73.2(1)–167.5(1)°.

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