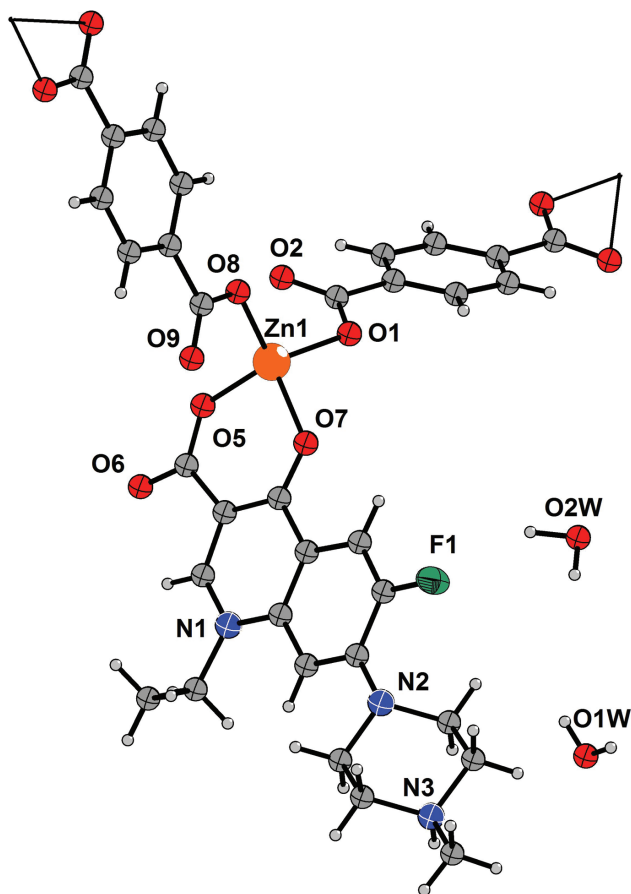


Yong-Jun Hu, Lie Jin, Ling Zhu and Tian Zhou*

Crystal structure of *catena*-poly[(μ_2 -benzene-1,4-dicarboxylato- $\kappa^2 O:O'$)-(1-ethyl-6-fluoro-7-(4-methylpiperazin-1-ium-1-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylato- $\kappa^2 O,O'$)zinc(II)] 1.25 hydrate, $C_{25}H_{26.5}N_3O_{8.25}FZn$



$\beta = 109.785(6)^\circ$, $\gamma = 93.257(4)^\circ$, $V = 1261.05(12) \text{ \AA}^3$, $Z = 2$, $R_{\text{gt}}(F) = 0.0344$, $wR_{\text{ref}}(F^2) = 0.0783$, $T = 293 \text{ K}$.

CCDC no.: 1571276

Table 1: Data collection and handling.

Crystal:	Colorless, prisms
Size:	$0.20 \times 0.16 \times 0.15 \text{ mm}$
Wavelength:	Mo $K\alpha$ radiation ($0.71073 \text{ \AA}$)
μ :	1.04 mm^{-1}
Diffractometer, scan mode:	Bruker FRAMBO, φ and ω -scans
$2\theta_{\text{max}}$, completeness:	25° , $>99\%$
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	8432, 4431, 0.026
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 3772
$N(\text{param})_{\text{refined}}$:	352
Programs:	Bruker programs [1], SHELX [2]

A part of the title crystal structure is shown in the figure. Tables 1 and 2 contain details on crystal structure and measurement conditions and a list of the atoms including atomic coordinates and displacement parameters.

Source of material

A mixture of $Zn(\text{CH}_3\text{COO})_2 \cdot 2\text{H}_2\text{O}$ (0.25 mmol), perfloxacin (0.5 mmol), 1,4-benzenedicarboxylic acid (0.25 mmol), sodium hydroxide (0.5 mmol) and water (15 mL) was stirred for 30 min in air. The mixture was then transferred to a 25 mL Teflon-lined hydrothermal bomb. The bomb was heated to 443 K for 96 h under autogenous pressure. After cooling, colorless single crystals of title compound were obtained.

Experimental details

The carbon-bound H atoms were positioned geometrically ($\text{C}-\text{H} = 0.93\text{--}0.97 \text{ \AA}$) and were included in the refinement in the riding model approximation, with $U(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The H on nitrogen atom were located in a difference Fourier map, and were refined with a distance restraint of $\text{N}-\text{H} = 0.86(1)$ or $0.85(1) \text{ \AA}$ and with $U(\text{H}) = 1.5U_{\text{eq}}(\text{N}, \text{O})$. During the refinement it became clear that O2w is not fully occupied (0.25).

Comment

Perfloxacin (Hpef), $\text{C}_{17}\text{H}_{20}\text{FN}_3\text{O}_3$ (systematic name 1-ethyl-6-fluoro-7-(4-methylpiperazin-4-ium-1-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylic acid) is a member of the fluoro-

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Abstract

$\text{C}_{25}\text{H}_{26.5}\text{N}_3\text{O}_{8.25}\text{FZn}$, Triclinic, $P\bar{1}$ (no. 2) $a = 10.2889(6) \text{ \AA}$, $b = 10.6034(5) \text{ \AA}$, $c = 12.9576(8) \text{ \AA}$, $\alpha = 105.984(5)^\circ$,

*Corresponding author: Tian Zhou, School of Environmental and Biological Engineering, Guangdong University of Petrochemical Technology, Maoming, 525000, Guangdong Province, P. R. China, e-mail: zhoutiangsh@163.com

Yong-Jun Hu: School of Life Sciences, Changchun Normal University, Changchun 130032, Jilin Province, P. R. China

Lie Jin and Ling Zhu: School of Chemical Engineering, Guangdong University of Petrochemical Technology, Maoming, 525000, Guangdong Province, P. R. China

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	x	y	z	<i>U</i> _{iso} */ <i>U</i> _{eq}
Zn1	0.28138(3)	0.36474(3)	0.29389(3)	0.02868(10)
F1	0.32095(16)	0.92156(15)	0.72440(14)	0.0527(5)
O1	0.3882(2)	0.27672(18)	0.39987(16)	0.0447(5)
O2	0.3335(2)	0.1043(2)	0.24347(19)	0.0559(6)
O5	0.08543(18)	0.29088(16)	0.19723(15)	0.0364(4)
O6	−0.1434(2)	0.28099(18)	0.13667(16)	0.0449(5)
O7	0.23676(17)	0.50723(16)	0.40580(15)	0.0322(4)
O8	0.3850(2)	0.3984(2)	0.20392(18)	0.0514(6)
O9	0.2847(2)	0.5723(2)	0.17822(19)	0.0548(6)
N1	−0.1506(2)	0.60011(18)	0.38281(17)	0.0254(4)
N2	0.0696(2)	1.00136(19)	0.71101(17)	0.0295(5)
N3	0.0165(2)	1.21639(19)	0.87265(18)	0.0313(5)
H3	−0.0489	1.1718	0.8884	0.038*
C1	0.4492(3)	0.0753(2)	0.4270(2)	0.0307(6)
C2	0.5274(3)	0.1368(3)	0.5427(2)	0.0348(6)
H2	0.5466	0.2292	0.5718	0.042*
C3	0.4226(3)	−0.0626(2)	0.3842(2)	0.0352(6)
H3A	0.3711	−0.1052	0.3062	0.042*
C7	0.3860(3)	0.1556(3)	0.3489(3)	0.0342(6)
C9	−0.0248(3)	0.3340(2)	0.2057(2)	0.0282(6)
C10	−0.0117(2)	0.4557(2)	0.3043(2)	0.0245(5)
C11	0.1139(2)	0.5313(2)	0.3933(2)	0.0238(5)
C12	0.0997(2)	0.6473(2)	0.4773(2)	0.0235(5)
C13	−0.0325(2)	0.6818(2)	0.4702(2)	0.0243(5)
C14	−0.1368(3)	0.4939(2)	0.3051(2)	0.0259(5)
H14	−0.2184	0.4418	0.2472	0.031*
C15	−0.2933(2)	0.6332(2)	0.3656(2)	0.0301(6)
H15A	−0.3015	0.6691	0.4402	0.036*
H15B	−0.3622	0.5529	0.3225	0.036*
C16	−0.3226(3)	0.7333(3)	0.3012(3)	0.0434(7)
H16A	−0.4150	0.7533	0.2914	0.065*
H16B	−0.3162	0.6972	0.2268	0.065*
H16C	−0.2551	0.8132	0.3443	0.065*
C17	−0.0426(3)	0.7970(2)	0.5501(2)	0.0271(5)
H17	−0.1307	0.8180	0.5448	0.033*
C18	0.0734(3)	0.8792(2)	0.6356(2)	0.0271(5)
C19	0.2055(3)	0.8405(2)	0.6421(2)	0.0329(6)
C20	0.2183(3)	0.7299(2)	0.5670(2)	0.0300(6)
H20	0.3067	0.7081	0.5748	0.036*
C21	0.1208(3)	1.0138(2)	0.8346(2)	0.0354(6)
H21A	0.2070	0.9776	0.8551	0.042*
H21B	0.0520	0.9633	0.8502	0.042*
C22	0.1467(3)	1.1579(2)	0.9067(2)	0.0363(6)
H22A	0.1765	1.1645	0.9877	0.044*
H22B	0.2208	1.2071	0.8958	0.044*
C23	−0.0392(3)	1.1968(2)	0.7470(2)	0.0329(6)
H23A	0.0264	1.2467	0.7280	0.039*
H23B	−0.1269	1.2306	0.7263	0.039*
C24	−0.0630(3)	1.0518(2)	0.6786(2)	0.0334(6)
H24A	−0.1320	1.0019	0.6944	0.040*
H24B	−0.0981	1.0412	0.5966	0.040*
C25	0.0372(3)	1.3587(2)	0.9397(2)	0.0435(7)
H25A	−0.0493	1.3920	0.9157	0.065*
H25B	0.1075	1.4083	0.9264	0.065*
H25C	0.0667	1.3682	1.0205	0.065*

Table 2 (continued)

Atom	x	y	z	<i>U</i> _{iso} */ <i>U</i> _{eq}
C26	0.3597(3)	0.4914(3)	0.1588(2)	0.0358(6)
C27	0.4312(3)	0.4940(3)	0.0748(2)	0.0314(6)
C28	0.4132(3)	0.5907(3)	0.0203(2)	0.0357(6)
H28	0.3547	0.6518	0.0335	0.043*
C29	0.5181(3)	0.4033(3)	0.0538(2)	0.0344(6)
H29	0.5303	0.3378	0.0898	0.041*
O1W	0.1688(3)	0.8904(2)	1.0674(2)	0.0926(10)
H1WA	0.1841	0.8401	1.0106	0.139*
H1WB	0.2406	0.9496	1.1095	0.139*
O2w ^a	0.5753(14)	0.9214(14)	0.9416(11)	0.124(5)
H2WA ^a	0.5346	0.9939	0.9619	0.185*
H2WB ^a	0.5167	0.8963	0.8577	0.185*

^aOccupancy: 0.2500.

quinolone (flq) family of antibiotics, widely used in veterinary clinical practice because of its wide antibiotic spectrum and its excellent bactericidal activity [4]. Our interest in perfloxacin and related fluoroquinolones is focused on their potential as multidentate/bridging ligands in coordination chemistry. The cobalt, manganese and zinc complexes of perfloxacin anion had been reported [5–7].

In the title compound, each Zn^{II} center atom is tetra-coordinated (ZnO₄), displaying a slightly distorted tetrahedron geometry defined by one bidentate O,O-bonded 1-ethyl-6-fluoro-7-(4-methylpiperazin-1-yl)-4-oxo-quinoline-3-carboxylate (pef) anions (Zn-O5, Zn-O7), two unidentate O-bonded 1,4-benzenedicarboxylate anions (Zn-O1, Zn-O8) (*cf.* the figure). The 1,4-benzenedicarboxylate ligands act as bridging ligands, together with two zinc(II) cations to form a coordination polymer.

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