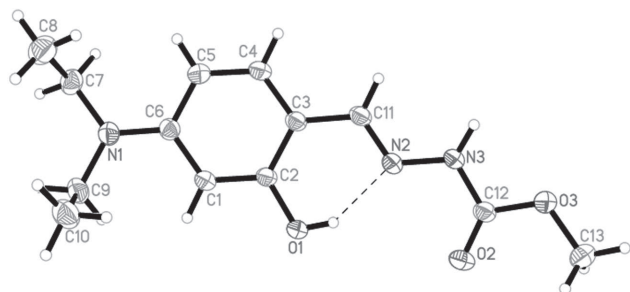


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Crystal structure of methyl (*E*)-2-(4-(diethylamino)-2-hydroxybenzylidene)hydrazine-1-carboxylate, $C_{13}H_{19}N_3O_3$



DOI 10.1515/ncrs-2016-0401

Received December 15, 2016; accepted May 3, 2017; available online May 18, 2017

Abstract

$C_{13}H_{19}N_3O_3$, monoclinic, $P2_1/c$ (no. 14), $a = 11.594(5)$ Å, $b = 11.311(5)$ Å, $c = 11.688(5)$ Å, $\beta = 117.347(19)^\circ$, $V = 1361.4(10)$ Å³, $Z = 4$, $R_{gt}(F) = 0.0479$, $wR_{ref}(F^2) = 0.1630$, $T = 296$ K.

CCDC no.: 1522954

The asymmetric unit of the title crystal structure is shown in the figure. Tables 1 and 2 contain details of the measurement method and a list of the atoms including atomic coordinates and displacement parameters.

Source of materials

Methyl hydrazinecarboxylate (0.9 g, 10 mmol) and 4-diethylamino-2-hydroxybenzaldehyde (1.93 g, 10 mmol) were dissolved in stirred methanol (25 mL). The reaction mixture was refluxed for 7–8 h with TLC monitoring (ethyl acetate:hexane = 1:1). The resulting solid was filtered off and recrystallized from ethanol to give the title compound.

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Table 1: Data collection and handling.

Crystal:	Colourless block
Size:	$0.29 \times 0.24 \times 0.21$ mm
Wavelength:	Mo $K\alpha$ radiation (0.71073 Å)
μ :	0.9 cm^{-1}
Diffractometer, scan mode:	Bruker APEX-II, φ and ω
$2\theta_{\max}$, completeness:	60.2° , 98.9%
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	17068, 3958, 0.025
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$, 2992
$N(\text{param})_{\text{refined}}$:	180
Programs:	Bruker programs [1], SHELX [2], ORTEP [3], DIAMOND [4]

Experimental details

The nitrogen bound hydrogen atoms were located from the difference Fourier map, and refined freely. All other H atoms were positioned geometrically (N–H = 0.86 Å and C–H = 0.93 or 0.96 Å) and refined using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$ and $1.5U_{\text{eq}}(\text{C}_{\text{methyl}})$.

Comment

Benzaldehydehydrazone derivatives and their complexes have become a hot topic of coordination chemistry due to their unique structure and catalytic activity [5]. Meanwhile, it is an important intermediate in the synthetic protocol of 1,3,4-oxadiazoles, which have interesting properties [6]. In our recent report, a benzaldehydehydrazone is involved in a key step of the synthesis of dimethyl-3,6-di(aryl)-1,4-dihydro-1,2,4,5-tetrazine-1,4-dicarboxylates which was proved to have antitumour activity [7]. We have been interested in this type of derivatives and reported several crystallographic studies of the products [8–11]. As a continuation of our work, an unsymmetrical benzaldehydehydrazone derivative, (*E*)-2-(4-(diethylamino)-2-hydroxybenzylidene)hydrazine-1-carboxylate was synthesized from methyl hydrazinecarboxylate and the corresponding aromatic aldehyde.

The compound crystallizes with one independent molecule in the asymmetric unit. In the crystal structure, the aryl ring C(1)/C(2)/C(3)/C(4)/C(5)/C(6) is almost coplanar

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso} [*] / <i>U</i> _{eq}
N2	0.48723(10)	0.36020(8)	0.24475(10)	0.0442(2)
O1	0.57485(9)	0.17518(7)	0.39031(8)	0.0501(2)
H1	0.5384	0.2116	0.3222	0.075*
C1	0.70334(11)	0.20682(9)	0.61165(11)	0.0410(2)
H1A	0.7099	0.1252	0.6220	0.049*
C3	0.61876(10)	0.37703(9)	0.46903(11)	0.0382(2)
C11	0.54562(10)	0.42820(9)	0.34214(12)	0.0414(3)
H11	0.5411	0.5098	0.3315	0.050*
C4	0.68182(11)	0.44845(9)	0.57802(12)	0.0449(3)
H4	0.6748	0.5301	0.5677	0.054*
C6	0.76706(10)	0.27989(10)	0.72080(11)	0.0404(3)
N3	0.41933(11)	0.40789(9)	0.12330(10)	0.0490(3)
C2	0.63149(10)	0.25331(9)	0.48985(11)	0.0379(2)
O3	0.30823(10)	0.38593(8)	−0.08885(9)	0.0557(3)
N1	0.83844(10)	0.23193(9)	0.84182(10)	0.0489(3)
C12	0.36091(11)	0.32926(10)	0.02564(12)	0.0446(3)
C5	0.75395(12)	0.40329(10)	0.69998(12)	0.0469(3)
H5	0.7945	0.4546	0.7694	0.056*
O2	0.35480(10)	0.22431(8)	0.03875(10)	0.0600(3)
C7	0.90316(13)	0.30488(13)	0.95615(12)	0.0526(3)
H7A	0.9085	0.2607	1.0296	0.063*
H7B	0.8501	0.3742	0.9462	0.063*
C9	0.86748(14)	0.10614(12)	0.85665(13)	0.0539(3)
H9A	0.7881	0.0626	0.8056	0.065*
H9B	0.8971	0.0847	0.9462	0.065*
C13	0.24988(16)	0.30997(14)	−0.19979(14)	0.0643(4)
H13A	0.3026	0.2407	−0.1856	0.096*
H13B	0.2434	0.3514	−0.2742	0.096*
H13C	0.1647	0.2871	−0.2135	0.096*
C10	0.96982(16)	0.06824(15)	0.81657(17)	0.0677(4)
H10A	0.9420	0.0899	0.7284	0.102*
H10B	0.9814	−0.0159	0.8257	0.102*
H10C	1.0506	0.1068	0.8704	0.102*
C8	1.03847(15)	0.34520(16)	0.98522(17)	0.0705(4)
H8A	1.0927	0.2774	0.9975	0.106*
H8B	1.0741	0.3924	1.0621	0.106*
H8C	1.0343	0.3914	0.9145	0.106*
H20	0.4233(16)	0.4836(17)	0.1058(17)	0.065(5)*

with the O atoms of the hydroxy group within 0.013 Å. The dihedral angle between the aryl ring and the plane C(11)/C(12)/C(13)/O(2)/O(3)/N(2)/N(3) is 8.6°. The angles of C(7)—C(8)—N(1) and C(10)—C(9)—N(1) are almost similar with 114.94(12)° and 114.15(12)°, respectively. The intramolecular

N—H···O hydrogen bonding is most predominant and observed mostly in molecules having a CH=N—NH chain in which the hydrogen atom of the secondary amine acts as a proton donor. In the crystal, weak intermolecular hydrogen bonds further stabilize the structure.

Acknowledgements: The authors gratefully acknowledge the support from higher education classroom teaching reform project of Zhejiang Province (No. kg2015694) and Zhejiang Natural Science Foundation (LY13B020001) and Zhejiang Province Education Science Program (2017SCG031).

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