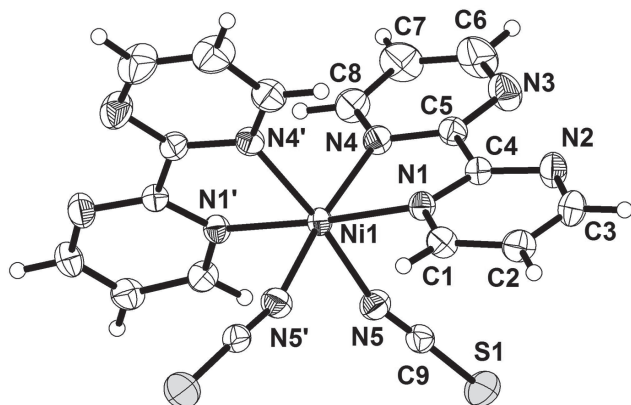


Kwang Ha*

Crystal structure of *cis*-bis(2,2'-bipyrimidine- κ^2 N, N')bis(thiocyanato- κ N)nickel(II), C₁₈H₁₂N₁₀NiS₂

**Table 1:** Data collection and handling.

Crystal:	Yellow block
Wavelength:	Size $0.20 \times 0.12 \times 0.08$ mm Mo $K\alpha$ radiation ($0.71073 \text{ \AA}$)
μ :	11.4 cm^{-1}
Diffractometer, scan mode:	PHOTON 100 CMOS, φ and ω
$2\theta_{\text{max}}$, completeness:	56.8° , $>99\%$
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$, R_{int} :	84487, 2635, 0.050
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 2250
$N(\text{param})_{\text{refined}}$:	141
Programs:	Bruker programs [9], SHELXL [10], ORTEP-3 [11], PLATON [12]

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Abstract

$\text{C}_{18}\text{H}_{12}\text{N}_{10}\text{NiS}_2$, tetragonal, $I4_1/a$ (no. 88), $a = 12.1303(5)$ Å, $c = 28.7471(12)$ Å, $V = 4230.0(4)$ Å³, $Z = 8$, $R_{\text{gt}}(F) = 0.0329$, $wR_{\text{ref}}(F^2) = 0.0854$, $T = 223(2)$ K.

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The crystal structure is shown in the figure. Tables 1 and 2 contain details of the measurement method and a list of the atoms including atomic coordinates and displacement parameters.

Source of material

To a solution of $\text{Ni}(\text{SCN})_2 \cdot 4\text{H}_2\text{O}$ (0.084 g, 0.34 mmol) in MeOH (10 mL) was added 2,2'-bipyrimidine (bpym; 0.108 g, 0.683 mmol) and stirred for 10 min at 50 °C. The formed precipitate was separated by filtration, washed with MeOH and ether and dried at room temperature, to give a yellow powder (0.0747 g). Crystals were obtained by slow evaporation from an *N,N*-dimethylformamide solution at 60 °C.

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$).

Atom	x	y	z	U_{iso}^*/U_{eq}
Ni1	0.5000	0.2500	0.08549(2)	0.02397(10)
S1	0.18497(5)	0.27959(6)	−0.01505(2)	0.05311(18)
N1	0.48201(12)	0.41965(12)	0.08796(5)	0.0261(3)
N2	0.34557(14)	0.55420(14)	0.10439(6)	0.0370(4)
N3	0.23769(15)	0.39851(16)	0.15792(7)	0.0440(4)
N4	0.37278(13)	0.26797(12)	0.13417(5)	0.0281(3)
N5	0.37660(13)	0.24483(13)	0.03621(6)	0.0333(3)
C1	0.53861(16)	0.49300(15)	0.06289(7)	0.0318(4)
H1	0.6062	0.4725	0.0494	0.038*
C2	0.49911(18)	0.59865(16)	0.05651(7)	0.0366(4)
H2	0.5371	0.6503	0.0381	0.044*
C3	0.40206(19)	0.62486(16)	0.07814(7)	0.0396(5)
H3	0.3738	0.6964	0.0743	0.047*
C4	0.38812(14)	0.45409(14)	0.10792(6)	0.0269(3)
C5	0.32795(15)	0.36888(15)	0.13538(6)	0.0288(4)
C6	0.1886(2)	0.3175(2)	0.18221(10)	0.0549(6)
H6	0.1242	0.3342	0.1990	0.066*
C7	0.2281(2)	0.2120(2)	0.18358(9)	0.0523(6)
H7	0.1928	0.1571	0.2012	0.063*
C8	0.32110(18)	0.18927(17)	0.15825(7)	0.0391(5)
H8	0.3490	0.1170	0.1578	0.047*
C9	0.29653(16)	0.25984(15)	0.01548(6)	0.0300(4)

Experimental details

Hydrogen atoms were positioned geometrically and allowed to ride on their parent atoms with $d(\text{C-H}) = 0.94 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The highest peak ($0.84 \text{ e \AA}^{-3}$) and the

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deepest hole ($-0.50 \text{ e } \text{\AA}^{-3}$) in the difference Fourier map are located $0.83 \text{ \AA}$ and $0.67 \text{ \AA}$ from the atom S1, respectively.

Discussion

This contribution is part of my continuing interest in the structural chemistry of 2,2'-bipyrimidine metal complexes [1–8]. The title complex *cis*-[Ni(NCS)₂(bpym)₂] is disposed about a twofold rotation axis running in the [001] direction passing through the Ni atom: the asymmetric unit contains one half of the complex. In the complex, the central Ni(II) is six-coordinated by four pyrimidine N atoms from two chelating bpym ligands and two mutually *cis*-positioned N atoms from two SCN[−] anionic ligands in a distorted octahedral manner. The main contribution to the distortion of the octahedron is made by the tight chelate angle $\angle \text{N1–Ni1–N4} = 78.23(6)^\circ$, which results in non-linear trans axes ($\angle \text{N1–Ni1–N1}^i = 176.06(8)^\circ$ and $\angle \text{N4–Ni1–N5}^i = 172.19(6)^\circ$; symmetry code i: $1 - x, 1/2 - y, z$). The Ni–N bond lengths are almost equal ($2.062(2)$ – $2.095(2) \text{ \AA}$), but the Ni–N bond *trans* to the NCS ligand ($2.095(2) \text{ \AA}$) is slightly longer than the Ni–N bond *trans* to the bpym ligand ($2.071(1) \text{ \AA}$), because of the different trans effects of the ligands. The thiocyanato ligand is almost linear displaying N–C–S bond angle of $178.5(2)^\circ$, and the N atom is bent coordinated to the Ni atom with the Ni–N–C(NCS) bond angle of $164.0(2)^\circ$, characteristic of an N-bonded conformation. The bpym ligand is nearly planar and the dihedral angle between the least-squares planes of the two bpym ligands is $71.96(3)^\circ$. The two pyrimidine rings of the bpym ligand are nearly parallel with the dihedral angle of $4.7(1)^\circ$ between the ring planes. In the crystal structure, several intermolecular π – π interactions between adjacent pyrimidine rings of the ligand are present. The shortest distance between Cg1 (the centroid of ring N1–C4) and Cg1ⁱⁱ (ring N3–C6; symmetry code ii: $5/4 - y, 1/4 + x, 1/4 - z$) is $4.672(1) \text{ \AA}$, and the dihedral angle between the ring planes is $48.89(9)^\circ$.

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