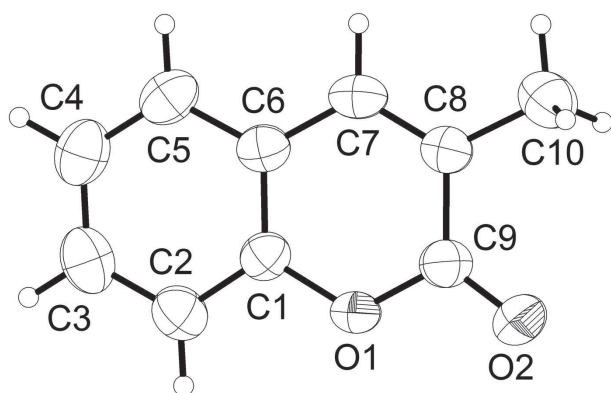


## Open Access

Zeng Fan-Xin, Nie Xu-Liang, Shang-guan Xin-Chen, Yin Zhong-Ping\* and Peng Da-Yong\*

# Crystal structure of 3-methyl-2*H*-chromen-2-one, $C_{10}H_8O_2$

**Table 1:** Data collection and handling.

|                                                       |                                        |
|-------------------------------------------------------|----------------------------------------|
| Crystal:                                              | Colourless, blocks                     |
| Wavelength:                                           | Size $0.22 \times 0.18 \times 0.16$ mm |
| $\mu$ :                                               | Mo $K_{\alpha}$ radiation (0.71073 Å)  |
| Diffractometer, scan mode:                            | 0.9 $cm^{-1}$                          |
| $2\theta_{max}$ , completeness:                       | Bruker APEX-II, $\varphi$ and $\omega$ |
| $N(hkl)_{measured}$ , $N(hkl)_{unique}$ , $R_{int}$ : | 51°, >99%                              |
| Criterion for $I_{obs}$ , $N(hkl)_{gt}$ :             | 5958, 1469, 0.046                      |
| $N(param)_{refined}$ :                                | $I_{obs} > 2 \sigma(I_{obs})$ , 856    |
| Programs:                                             | 110                                    |
|                                                       | SHELX [11]                             |

**Table 2:** Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ ).

| Atom | <i>x</i>     | <i>y</i>    | <i>z</i>    | $U_{iso}^*/U_{eq}$ |
|------|--------------|-------------|-------------|--------------------|
| O1   | −0.03429(17) | 0.22639(7)  | 0.28174(17) | 0.0547(4)          |
| O2   | −0.1471(2)   | 0.10841(8)  | 0.2683(2)   | 0.0758(5)          |
| C10  | 0.2516(3)    | 0.04029(11) | 0.3311(3)   | 0.0669(6)          |
| H10A | 0.3981       | 0.0299      | 0.3469      | 0.100*             |
| H10B | 0.1825       | 0.0193      | 0.2101      | 0.100*             |
| H10C | 0.1963       | 0.0176      | 0.4362      | 0.100*             |
| C8   | 0.2175(3)    | 0.12382(10) | 0.3314(2)   | 0.0465(5)          |
| C9   | 0.0031(3)    | 0.14962(11) | 0.2932(2)   | 0.0512(5)          |
| C7   | 0.3705(3)    | 0.17564(11) | 0.3627(2)   | 0.0500(5)          |
| H7   | 0.5079       | 0.1590      | 0.3881      | 0.060*             |
| C6   | 0.3300(3)    | 0.25549(10) | 0.3584(2)   | 0.0452(4)          |
| C1   | 0.1255(3)    | 0.27903(10) | 0.3142(2)   | 0.0451(5)          |
| C2   | 0.0704(3)    | 0.35431(11) | 0.2989(3)   | 0.0576(5)          |
| H2   | −0.0687      | 0.3685      | 0.2664      | 0.069*             |
| C3   | 0.2248(4)    | 0.40857(12) | 0.3327(3)   | 0.0642(6)          |
| H3   | 0.1903       | 0.4600      | 0.3221      | 0.077*             |
| C4   | 0.4304(4)    | 0.38692(13) | 0.3820(3)   | 0.0668(6)          |
| H4   | 0.5340       | 0.4238      | 0.4071      | 0.080*             |
| C5   | 0.4824(3)    | 0.31146(12) | 0.3943(2)   | 0.0593(5)          |
| H5   | 0.6215       | 0.2974      | 0.4272      | 0.071*             |

DOI 10.1515/ncrs-2016-0028

Received January 25, 2016; accepted April 26, 2016; available online May 19, 2016

**Abstract**

$C_{10}H_8O_2$ , monoclinic,  $P2_1/c$  (no. 14),  $a = 6.524(3)$  Å,  $b = 17.561(9)$  Å,  $c = 6.989(4)$  Å,  $\beta = 100.058(7)^\circ$ ,  $V = 788.4(7)$  Å<sup>3</sup>,  $Z = 4$ ,  $R_{gt}(F) = 0.0418$ ,  $wR_{ref}(F^2) = 0.1069$ ,  $T = 296(2)$  K.

**CCDC no.:** 1476569

The crystal structure is shown in the figure. Tables 1 and 2 contain details of the measurement method and a list of the atoms including atomic coordinates and displacement parameters.

**\*Corresponding authors: Yin Zhong-Ping**, Key Laboratory of Natural Product Research and Development/College of Food, Sciences and Engineering, Jiangxi Agricultural University, Nanchang 330045, People's Republic of China, e-mail: yin\_zhongping@163.com; and **Peng Da-Yong**, Collaborative Innovation Center of Jiangxi Typical Trees Cultivation and Utilization/Key Laboratory of Natural Product Research and Development/College of Sciences, Jiangxi Agricultural University, Nanchang 330045, People's Republic of China; and Key Laboratory of Protection and Utilization of Biological Resources in Tarim Basin of Xinjiang Production & Construction Group, Alar 330045, People's Republic of China, e-mail: dayongpeng@163.com

**Zeng Fan-Xin and Shang-guan Xin-Chen:** Key Laboratory of Natural Product Research and Development/College of Food Science and Engineering, Jiangxi Agricultural University, Nanchang 330045, People's Republic of China

**Nie Xu-Liang:** College of Sciences, Jiangxi Agricultural University, Nanchang 330045, People's Republic of China

**Source of material**

The synthesis of 3-methylcoumarin is carried out using salicylaldehyde and propionic anhydride for starting material, according to a modified Perkin reaction [9]. All chemicals used were commercially available of AR grade, and were used as received without further purification. The mixture of salicylaldehyde (1 eq), propionic anhydride (3 eq), and  $K_2CO_3$

(0.25 eq) was refluxed in sealed tube until the color of the reaction mixture became brown. The mixture was poured into crushed ice, and the pH was adjusted to 7 using NaHCO<sub>3</sub>. The precipitate was filtered off, washed with water, dried, and recrystallized from a mixture of hexane/ethyl acetate (4:1) to give 420.5 mg of 3-methyl-2H-chromen-2-one as colourless needles: mp 89–91°C. The colourless crystals of the title compound were obtained by slowly evaporating of an ethanol solution at room temperature. The crystalline product was filtered, washed with ethanol and dried. Analysis calculated for C<sub>10</sub>H<sub>8</sub>O<sub>2</sub>: C, 74.99; H, 5.03; O, 19.98 %; found: C, 75.16; H, 5.12; O, 19.75 %.

### Experimental details

All H atoms were included in calculated positions and refined as riding atoms, with C–H = 0.96–0.97 Å with  $U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{C})$  for methyl H atoms and 1.2  $U_{\text{eq}}(\text{C})$  for all other H atoms.

### Discussion

Coumarins, considered as cinnamic acid derivatives, are a kind of aromatic secondary metabolites, which are widely existed in plants [1]. These compounds have been reported to exhibit a variety of pharmacological activities, such as anti-virus, anti-HIV, anti-neoplasm, anti-cancer, anti-coagulant agents [2–4]. The synthesis of coumarin and its derivatives have attracted increasing attention due to their extensive biological applications [5–8]. In this paper, we report the synthesis and crystal structure of 3-methyl-2H-chromen-2-one. In the molecule of the title compound (Figure), bond lengths and angles within coumarin are very similar to those given in the literature for structurally related derivatives [10]. The molecular conformation is characterized by the O2–C9–C8–C7 and O2–C9–O1–C1 torsion angles of 177.9° and 177.8°, respectively. The dihedral angle of benzene ring and pyranose ring is 1.7°.

**Acknowledgements:** We are grateful for the financial support from the Research Foundation of Key Laboratory of

Protection and Utilization of Biological Resources in Tarim Basin of Xinjiang Production & Construction Group [BRYB1502] and the National Natural Science Foundation of China (Grant No. 21562022, 31460436).

### References

1. Burger, A.: Burger's Medicinal Chemistry. Wiley-Interscience, New York. 1995.
2. Sardari, S.; Mori, Y.; Horita, K.; Micetich, R. G.; Nishibe, S.; Daneshtalab, M.: Synthesis and antifungal activity of coumarins and angular furanocoumarins. *Bioorg. Med. Chem.* **7** (1999) 1933–1940.
3. Liu, B.; Xie, L. G.; Xu, X. H.; Li, Y. H.: Synthesis, crystal structure and herbicidal activity of 3-benzoyl-4-hydroxycoumarin derivatives. *Chin. J. Org. Chem.* **31** (2011) 2067–2073.
4. Sun, Y. F.; Song, H. C.; Sun, X. Z.; XU, Z. L.: Synthesis and crystal structure of new 3-substituted-6-arylazocomarins. *Chin. J. Org.* **23** (2003) 162–166.
5. Jiang, H.; Zou, H.; Xia, P.; Zhang, Q.: Synthesis and molecular structure of 3-(p-Methoxyphenyl)-4-hydroxymethyl-6-bromo-7-hydroxycoumarin co-crystallized with methanol. *Chin. J. Struct. Chem.* **27** (2008) 1423–1426.
6. Lu, L.; Zhang, W. J.: Crystal structure of (E)-N'-((7-(diethylamino)-2-oxo-2H-chromen-3-yl)methylene)-2-hydroxybenzohydrazide, C<sub>21</sub>H<sub>21</sub>N<sub>3</sub>O<sub>4</sub>. *Z. Kristallogr. NCS.* **224** (2009) 413–414.
7. Li, X. C.; Wang, B. C.; Son, Y. A.; Wang, J. G.; Wang, S.: Crystal structure of 8-methoxy-2-oxo-2H-chromene-3-carboxylic acid, C<sub>11</sub>H<sub>8</sub>O<sub>5</sub>. *Z. Kristallogr. NCS.* **225** (2010) 65–66.
8. Song, L. M.; Gao, J. H.: Synthesis, crystal structure and photoluminescence of ethyl coumarin-3-carboxylate derivatives. *Chin. J. Struct. Chem.* **33** (2014) 57–64.
9. Li, X. C.; Yang, L.; Zhang, X.; Li, L. F.; Zhou, Y.; Son, Y. A.: Crystal structure of 7-(diethylamino)-2-oxo-2H-chromene-3-carbaldehyde, C<sub>14</sub>H<sub>15</sub>NO<sub>3</sub>. *Z. Kristallogr. NCS.* **229** (2014) 241–242.
10. Florekova, L.; Flasik, R.; Stankovicova, H.; Gaplovsky, A.: Efficient synthesis of 3-methyl-2H-chromen-2-one. Classic versus microwave conditions. *Synth. Commun.* **41** (2011) 1514–1519.
11. Sheldrick, G. M.: A short history of *SHELX*. *Acta Crystallogr.* **A64** (2008) 112–122.