

Zheng-Kang Xiao and Jian-Mei Lu*

Crystal structure of 3-hydroxy-3-phenyl-1,3-dihydro-2H-indol-2-one, C₁₄H₁₁NO₂

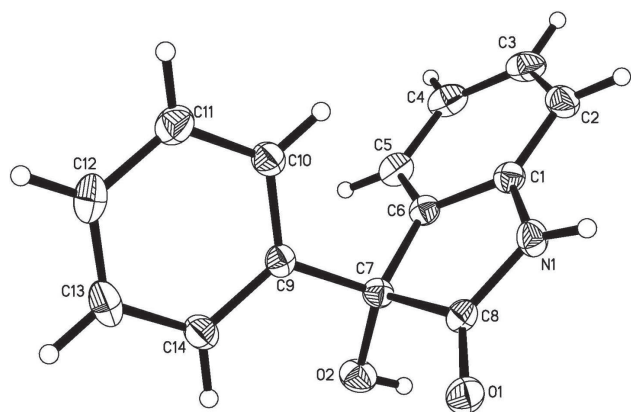


Table 1: Data collection and handling.

Crystal:	Colorless, block, size 0.15 × 0.21 × 0.24 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	0.94 cm ⁻¹
Diffractometer, scan mode:	Bruker APEX II diffractometer, φ and ω scans
$2\theta_{\max}$:	50.98°
$N(hkl)_{\text{measured}}, N(hkl)_{\text{unique}}$:	8413, 1989
$N(\text{param})_{\text{refined}}$:	155
Programs:	SHELX [2], PLATON [3], Bruker programs [4]

DOI 10.1515/ncrs-2015-0244

Received October 22, 2015; accepted February 16, 2016; available online March 17, 2016

Abstract

C₁₄H₁₁NO₂, orthorhombic, *Pbca*, $a = 10.3868(9)$ Å, $b = 7.7695(7)$ Å, $c = 26.576(2)$ Å, $V = 2144.7(3)$ Å³, $Z = 8$, $R_{\text{gt}}(F) = 0.0348$, $wR_{\text{ref}}(F^2) = 0.0911$, $T = 298(2)$ K.

CCDC no.: 1453549

The crystal structure is shown in the figure. Tables 1–3 contain details of the measurement method and a list of the atoms including atomic coordinates and displacement parameters.

Source of material

The title product was obtained by a known method [1]. Under a nitrogen atmosphere, indolin-2-one (0.65 mmol), NHC-Pd(II)-Im complex (1.0 mol%), potassium tertiary butoxide (4.0 equiv), toluene (1.0 mL) and chlorobenzene (0.5 mmol) were successively added. The mixture was stirred vigorously at 100°C for 12 h. Then the mixture was cooled to room temperature and stirred under air atmosphere for another

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1)	8c	0.9486	0.8544	0.5093	0.041
H(2)	8c	1.2527	0.7014	0.6056	0.062
H(2A)	8c	0.8293	0.5414	0.5051	0.047
H(3)	8c	0.8133	0.2808	0.5483	0.054
H(4)	8c	0.9230	0.2354	0.6217	0.054
H(5)	8c	1.0494	0.4517	0.6565	0.046
H(10)	8c	0.8408	0.8086	0.6417	0.041
H(11)	8c	0.7396	0.9731	0.7023	0.051
H(12)	8c	0.8588	1.1512	0.7539	0.055
H(13)	8c	1.0791	1.1649	0.7446	0.056
H(14)	8c	1.1820	1.0003	0.6842	0.044

3 h. The solvent was removed under reduced pressure and the residue was purified by flash column chromatography to give the pure product in 72% yield. Crystal suitable for X-ray diffraction was grown in a mixture of ethyl acetate and petroleum ether.

Experimental details

All hydrogen atoms were introduced using the HFIX command in the SHELXL program [2]. The N–H distance was restrained to 0.86 Å with U_{iso} values to be $1.2U_{\text{eq}}(\text{N})$. The O–H distance was restrained to 0.82 Å with U_{iso} values to be $1.5U_{\text{eq}}(\text{O})$. All other aromatic C–H distances were restrained to 0.93 Å with U_{iso} values to be $1.2U_{\text{eq}}(\text{C})$.

*Corresponding author: Jian-Mei Lu, College of Chemistry and Materials Engineering, Wenzhou University, Chashan University Town, Wenzhou Zhejiang Province, 325035, People's Republic of China, e-mail: ljm@wzu.edu.cn

Zheng-Kang Xiao: College of Chemistry and Materials Engineering, Wenzhou University, Chashan University Town, Wenzhou Zhejiang Province, 325035, People's Republic of China

Table 3: Fractional coordinates and atomic displacement parameters (Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
N(1)	8c	0.9816(1)	0.8123(1)	0.53632(4)	0.0410(6)	0.0339(6)	0.0269(5)	−0.0005(5)	−0.0048(4)	0.0040(4)
O(1)	8c	1.12093(9)	1.0324(1)	0.55566(3)	0.0464(6)	0.0338(5)	0.0365(5)	−0.0073(4)	0.0026(4)	0.0054(4)
O(2)	8c	1.21913(8)	0.7669(1)	0.62590(4)	0.0277(5)	0.0461(6)	0.0493(6)	0.0050(4)	−0.0047(4)	−0.0034(4)
C(1)	8c	0.9519(1)	0.6478(2)	0.55575(5)	0.0295(6)	0.0297(6)	0.0328(6)	0.0023(5)	0.0032(5)	−0.0035(5)
C(2)	8c	0.8738(1)	0.5231(2)	0.53504(5)	0.0330(7)	0.0437(8)	0.0420(7)	−0.0007(6)	−0.0006(6)	−0.0126(6)
C(3)	8c	0.8648(1)	0.3682(2)	0.56115(6)	0.0379(8)	0.0344(7)	0.064(1)	−0.0080(6)	0.0114(7)	−0.0153(7)
C(4)	8c	0.9301(2)	0.3410(2)	0.60551(6)	0.0456(8)	0.0271(7)	0.0620(9)	−0.0006(6)	0.0142(7)	0.0018(6)
C(5)	8c	1.0064(1)	0.4691(2)	0.62627(5)	0.0400(7)	0.0316(7)	0.0432(8)	0.0049(6)	0.0045(6)	0.0050(5)
C(6)	8c	1.0166(1)	0.6232(2)	0.60078(5)	0.0287(6)	0.0270(6)	0.0324(6)	0.0035(5)	0.0034(5)	−0.0018(5)
C(7)	8c	1.0876(1)	0.7867(2)	0.61411(5)	0.0268(6)	0.0288(6)	0.0298(6)	0.0006(5)	−0.0019(5)	0.0019(5)
C(8)	8c	1.0678(1)	0.8948(2)	0.56539(4)	0.0305(6)	0.0291(6)	0.0281(6)	0.0020(5)	0.0049(5)	−0.0003(5)
C(9)	8c	1.0221(1)	0.8866(1)	0.65653(4)	0.0333(6)	0.0255(6)	0.0244(6)	0.0004(5)	−0.0021(5)	0.0049(5)
C(10)	8c	0.8894(1)	0.8800(2)	0.66238(5)	0.0338(7)	0.0345(7)	0.0353(7)	−0.0014(5)	−0.0015(5)	−0.0038(5)
C(11)	8c	0.8286(1)	0.9787(2)	0.69867(5)	0.0401(7)	0.0428(8)	0.0437(7)	0.0039(6)	0.0070(6)	−0.0030(6)
C(12)	8c	0.8996(2)	1.0850(2)	0.72946(5)	0.062(1)	0.0419(8)	0.0335(7)	0.0080(7)	0.0051(7)	−0.0062(6)
C(13)	8c	1.0311(2)	1.0928(2)	0.72386(5)	0.062(1)	0.0417(8)	0.0360(7)	−0.0043(7)	−0.0092(7)	−0.0091(6)
C(14)	8c	1.0930(1)	0.9942(2)	0.68762(5)	0.0377(7)	0.0387(7)	0.0346(7)	−0.0054(6)	−0.0057(5)	−0.0004(5)

Discussion

3-Substituted 3-hydroxyoxindoles are attractive compounds in organic synthesis because they are frequently found in many alkaloids, some natural substances and in pharmaceutically and biologically active compounds [5–7]. The bond lengths and angles derived from the title structure are within normal ranges. Adjacent molecules are connected by O—H···O hydrogen bonds [3.0925(13) Å], and N—H···O hydrogen bonds [2.9267(14) Å], which may stabilize the crystal structure.

Acknowledgements: I acknowledge the financial support from the National Science Foundation of China (No. 21471115) for the publication fee.

References

1. Xiao, Z.-K.; Yin, H.-Y.; Shao, L.-X.: N-Heterocyclic carbene-palladium(II)-1-methylimidazole complex catalyzed α -arylation of oxindoles with aryl chlorides and aerobic oxidation of the products in a one-pot procedure. *Org. Lett.* **15** (2013) 1254–1257.
2. Sheldrick, G. M.: A short history of SHELX. *Acta Cryst.* **A64** (2008) 112–122.
3. Spek, A. L.: Structure validation in chemical crystallography. *Acta Crystallogr.* **D65** (2009) 148–155.
4. Bruker: Bruker AXS SADABS, Version 2.03, SAINT, Version 6.02, Bruker AXS Inc. Madison, Winsconsin, USA, 2002.
5. Trost, B. M.; Brennan, M. K.: Asymmetric syntheses of oxindole and indole spirocyclic alkaloid natural products. *Synthesis* **18** (2009) 3003–3025.
6. Peddibhotla, S.: 3-Substituted-3-hydroxy-2-oxindole, an emerging new scaffold for drug discovery with potential anti-cancer and other biological activities. *Curr. Bioact. Compd.* **5** (2009) 20–38.
7. Badillo, J. J.; Hanhan, N. V.; Franz, A. K.: Enantioselective synthesis of substituted oxindoles and spirooxindoles with applications in drug discovery. *Curr. Opin. Drug Discovery Dev.* **13** (2010) 758–776.