

Smaail Radi*, Said Tighadouini, Taibi Ben Hadda, Mehmet Akkurt, Namık Özdemir, Muhammad Sirajuddin and Yahia N. Mabkhot

Crystal structure of (Z)-1-(1,5-dimethyl-1*H*-pyrazol-3-yl)-3-hydroxybut-2-en-1-one C₉H₁₂N₂O₂

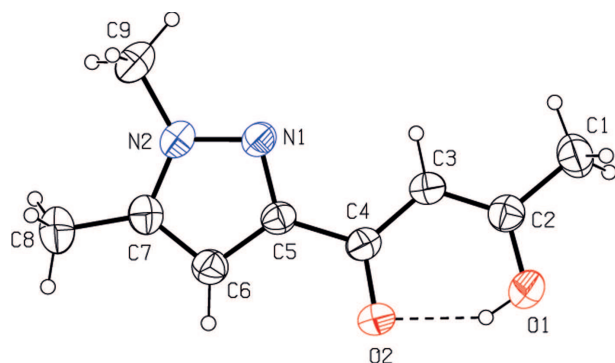


Table 1: Data collection and handling.

Crystal:	Colourless, plate, size 0.040 × 0.317 × 0.540 mm
Wavelength:	Mo K α radiation (0.71073 Å)
μ :	0.91 cm ⁻¹
Diffractometer, scan mode:	STOE IPDS 2, ω scans
$2\theta_{\max}$:	57.92°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	9019, 2509
$N(\text{param})_{\text{refined}}$:	122
Programs:	X-Area [9], SHELX [10]

DOI 10.1515/ncrs-2015-0212

Received February 10, 2016; accepted March 6, 2016; available online March 23, 2016

Abstract

C₉H₁₂N₂O₂, triclinic, $P\bar{1}$ (no. 2), $a = 6.6954(7)$ Å, $b = 7.6325(8)$ Å, $c = 9.8928(11)$ Å, $\alpha = 95.908(9)^\circ$, $\beta = 92.810(9)^\circ$, $\gamma = 108.360(8)^\circ$. $V = 475.49(9)$ Å³, $Z = 2$, $R_{\text{gt}}(F) = 0.0606$, $wR_{\text{ref}}(F^2) = 0.1549$, $T = 296(2)$ K.

CCDC no.: 979896

The figure shows the asymmetric unit of the title structure. Tables 1–3 contain details of the methods used and a list of the atoms including atomic coordinates and displacement parameters.

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1)	2i	0.2341	0.3611	−0.0394	0.118
H(1A)	2i	−0.2924	0.2720	−0.0314	0.124
H(1B)	2i	−0.3091	0.1033	0.0503	0.124
H(1C)	2i	−0.2598	0.0912	−0.1030	0.124
H(3)	2i	−0.0147	0.1783	0.2125	0.072
H(6)	2i	0.7137	0.3781	0.3140	0.073
H(8A)	2i	0.7951	0.3856	0.66740	0.124
H(8B)	2i	0.9125	0.3245	0.5472	0.124
H(8C)	2i	0.7607	0.1740	0.6244	0.124
H(9A)	2i	0.3381	0.2624	0.7143	0.124
H(9B)	2i	0.3913	0.0773	0.6824	0.124
H(9C)	2i	0.1696	0.0870	0.6323	0.124

Source of material

A solution of ethyl 1,5-dimethyl-1*H*-pyrazole-3-carboxylate (12 mmol) in THF (25 mL) was added to a suspension of sodium (22 mmol) in THF (50 mL), and then acetone (1.2 g; 20.6 mmol) was added at 273 K. The resulting mixture was stirred at 273 K for 48 h and the formed residue was filtered, washed with THF, and neutralized with acetic acid to pH 5 after being dissolved in water. The organic layer was extracted with CH₂Cl₂, dried and concentrated in vacuum. The resulting mixture was chromatographed on silica using CH₂Cl₂ as eluant to give the target product. Crystals suitable for X-ray diffraction analysis were obtained by slow evaporation of methanol from the mixture; yield: 37%; M.p. 356–357 K.

*Corresponding author: Smaail Radi, Department of Chemistry, Faculty of Sciences, Mohamed Premier University, Oujda 60000, Morocco, e-mail: radi_smaail@yahoo.fr

Said Tighadouini and Taibi Ben Hadda: Department of Chemistry, Faculty of Sciences, Mohamed Premier University, Oujda 60000, Morocco

Mehmet Akkurt: Department of Physics, Faculty of Sciences, Erciyes University, 38039 Kayseri, Turkey

Namık Özdemir: Department of Physics, Faculty of Arts and Sciences, Ondokuz Mayıs University, 55139 Samsun, Turkey

Muhammad Sirajuddin: Department of Chemistry, University of Science & Technology, Bannu, Pakistan

Yahia N. Mabkhot: Department of Chemistry, College of Science, King Saud University, P.O. Box 2455, Riyadh 1451, Saudi Arabia

Table 3: Fractional coordinates and atomic displacement parameters (Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
O(1)	2i	0.1092(2)	0.3321(2)	−0.0671(1)	0.0770(9)	0.107(1)	0.0560(8)	0.0321(9)	0.0070(7)	0.0188(8)
O(2)	2i	0.4218(2)	0.3925(2)	0.1072(1)	0.0608(8)	0.104(1)	0.0670(8)	0.0196(7)	0.0154(6)	0.0352(8)
N(1)	2i	0.2620(2)	0.2152(2)	0.4169(2)	0.0632(9)	0.062(1)	0.0519(8)	0.0202(7)	0.0068(7)	0.0125(7)
N(2)	2i	0.3994(3)	0.2154(2)	0.5212(2)	0.075(1)	0.0603(9)	0.0505(8)	0.0259(8)	0.0061(7)	0.0105(7)
C(1)	2i	−0.2376(3)	0.1722(4)	−0.0188(2)	0.067(1)	0.101(2)	0.078(1)	0.027(1)	−0.008(1)	0.001(1)
C(2)	2i	−0.0075(3)	0.2498(3)	0.0239(2)	0.067(1)	0.065(1)	0.059(1)	0.0266(9)	0.0081(9)	0.0044(9)
C(3)	2i	0.0754(3)	0.2360(3)	0.1501(2)	0.057(1)	0.067(1)	0.058(1)	0.0182(8)	0.0123(8)	0.0139(8)
C(4)	2i	0.2944(3)	0.3072(2)	0.1873(2)	0.061(1)	0.057(1)	0.0525(9)	0.0234(8)	0.0118(8)	0.0117(8)
C(5)	2i	0.3857(3)	0.2848(2)	0.3202(2)	0.0573(9)	0.051(1)	0.0533(9)	0.0199(7)	0.0103(7)	0.0093(7)
C(6)	2i	0.5988(3)	0.3282(3)	0.3632(2)	0.059(1)	0.065(1)	0.062(1)	0.0230(9)	0.0089(8)	0.0101(9)
C(7)	2i	0.6036(3)	0.2820(3)	0.4932(2)	0.065(1)	0.058(1)	0.063(1)	0.0272(9)	−0.0002(8)	0.0027(8)
C(8)	2i	0.7841(4)	0.2925(4)	0.5918(2)	0.084(1)	0.093(2)	0.076(1)	0.038(1)	−0.012(1)	0.005(1)
C(9)	2i	0.3176(4)	0.1554(4)	0.6484(2)	0.103(2)	0.095(2)	0.056(1)	0.035(1)	0.012(1)	0.022(1)

Experimental details

The hydroxyl H atom and H atoms bonded to C atoms were positioned geometrically and were refined using a riding model with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$ for hydroxyl and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for the others.

Discussion

A new generation of highly promising inhibitors bearing β -keto-enol functionality has emerged. Their versatile utility in medicinal chemistry is firmly established including anti-HIV [1, 2], antitumor [3–5], anti-influenza [6], antioxidant [7], and anti-inflammatory [8] activities. In this study we coupled ethyl pyrazole carboxylate with acetone to produce the title compound in acceptable yield.

The crystal structure is quite interesting and displays the formation of an intramolecular OH–O hydrogen bond (see the figure). The structure leads to two independent N1, O2 and O1, O2-bidentate units having two different geometries with a very attractive evidence for even semi-combination probably existing between them, leading to a bis-bidentate N1, O2, O1-unit. However, both are held together by means of intra-electrostatic forces and van der Waals interactions. The oxygen atoms O1, O2 are separated by a distance of 2.538 Å. This distance is always found in the case of antibacterial pharmacophore site. Hence, this compound constitutes a promising anti-HIV agent. The whole molecule is almost planar, with an rms deviation of 0.002 Å for all non-hydrogen atoms. In the crystal, π - π stacking interactions [centroid-centroid distance = 3.5901(12) Å] between the pyrazole rings stabilize the molecular packing.

Acknowledgements: The authors would like to extend their sincere appreciation to the Deanship of Scientific Research at King Saud University for its funding this Prolific Research group (PRG-1437-29).

References

- Pommier, Y.; Johnson, A. A.; Marchand, C.: Integrase inhibitors to treat HIV/AIDS. *Nat. Rev. Drug Discov.* **4** (2005) 236–248.
- Egbertson, S. S.; Anthony, N. J.; Summa, V.: From diketo acids to heterocyclic templates: history of HIV integrase medicinal chemistry at Merck West Point and Merck Rome (IRBM) leading to discovery of raltegravir. In: *Pharmaceutical & Medicinal Chemistry* (Ed. Neamati, N.), pp 197–230, Wiley, 2011.
- Goldgur, Y.; Craigir, R.; Cohen, G. H.; Fujiwara, T.; Yoshinaga, T.; Fujishita, T.; Sugimoto, H.; Endo, T.; Murai, H.; Davies, D. R.: Structure of the HIV-1 integrase catalytic domain complexed with an inhibitor: A platform for antiviral drug design. *Proc. Natl. Acad. Sci. USA*, **96** (1999) 13040–13043.
- Tan, K. L.; Ali, A.; Du, Y.; Fu, H.; Jin, H. X.; Chin, T. M.; Khan, M.; Go, M. L.: Synthesis and evaluation of bisbenzylidenedioxotetrahydro-thiopranones as activators of endoplasmic reticulum (ER) stress signaling pathways and apoptotic cell death in acute promyelocytic leukemic cells. *J. Med. Chem.* **57** (2014) 5904–5918.
- Liang, G.; Shao, L.; Wang, Y.; Zhao, C.; Chu, Y.; Xiao, J.; Zhao, Y.; Li, X.; Yang, S.: Exploration and synthesis of curcumin analogues with improved structural stability both in vitro and in vivo as cytotoxic agents. *Bioorg. Med. Chem.* **17** (2009) 2623–2631.
- Ishikawa, Y.; Fujii, S.: Binding mode prediction and inhibitor design of anti-influenza virus diketo acids targeting metalloenzyme RNA polymerase by molecular docking. *Bioinformation* **6** (2011) 221–225.
- Anand, P.; Thomas, S. G.; Kunnumakkara, A. B.; Sundaram, C.; Harikumar, K. B.; Sung, B.; Tharakan, S. T.; Misra, K.; Priyadarsini, I. K.; Rajasekharan, K. N.; Aggarwal, B. B.: Biological activities of curcumin and its analogues (congeners) made by man and mother nature. *Biochem. Pharmacol.* **76** (2008) 1590–1611.
- Liang, G.; Li, X.; Chen, L.; Yang, S.; Wu, X.; Studer, E.; Gurley, E.; Hylemon, P. B.; Ye, F.; Li, Y.; Zhou, H.: Synthesis and anti-inflammatory activities of monocarbonyl analogues of curcumin. *Bioorg. Med. Chem. Lett.* **18** (2008) 1525–1529.
- Stoe & Cie: X-Area & X-RED32 Software; Stoe & Cie GmbH: Darmstadt, Germany, 2002.
- Sheldrick, G. M.: A short history of SHELX. *Acta Crystallogr. A* **64** (2008) 112–122.