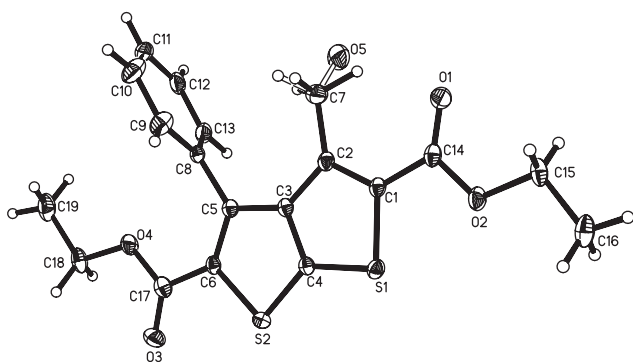


Yahia N. Mabkhot*, Assem Barakat, Salim S. Al-Showiman, Wolfgang Frey, Shams Uzzaman and Hazem A. Ghabbour

Crystal structure of diethyl-3-methyl-4-phenylthieno[2,3-*b*]thiophene-2,5-dicarboxylate, $C_{19}H_{18}O_4S_2$



DOI 10.1515/ncrs-2015-0174

Received December 15, 2015; accepted February 12, 2016; available online March 23, 2016

Abstract

$C_{19}H_{17.75}O_{4.13}S_2$, monoclinic, $P2_1/n$ (no. 14), $a = 7.1921(6)$ Å, $b = 18.6304(14)$ Å, $c = 13.3739(11)$ Å, $\beta = 92.681(4)^\circ$, $V = 1790.0(2)$ Å³, $Z = 4$, $R_{gt}(F) = 0.0478$, $wR_{ref}(F^2) = 0.0913$, $T = 100$ K.

CCDC no.: 1453161

***Corresponding author: Yahia N. Mabkhot**, Department of Chemistry, College of Science, King Saud University, P. O. Box 2455, Riyadh 11451, Saudi Arabia, e-mail: yahia@ksu.edu.sa

Assem Barakat: Department of Chemistry, College of Science, King Saud University, P. O. Box 2455, Riyadh 11451, Saudi Arabia; and Department of Chemistry, Faculty of Science, Alexandria University, P. O. Box 426, Ibrahimia, Alexandria 21321, Egypt

Salim S. Al-Showiman: Department of Chemistry, College of Science, King Saud University, P. O. Box 2455, Riyadh 11451, Saudi Arabia

Wolfgang Frey: Institut für Organische Chemie, Universität Stuttgart, Pfaffenwaldring 55, Stuttgart 70569, Germany

Shams Uzzaman: Department of Chemistry, Aligarh Muslim University, Aligarh 202 002, India

Hazem A. Ghabbour: Department of Pharmaceutical Chemistry, College of Pharmacy, King Saud University, P. O. Box 2457, Riyadh 11451, Kingdom of Saudi Arabia; and Department of Medicinal Chemistry, Faculty of Pharmacy, Mansoura University, Mansoura 35516, Egypt

Table 1: Data collection and handling.

Crystal:	Colourless, prism, size $0.11 \times 0.15 \times 0.49$ mm
Wavelength:	Mo K_α radiation (0.71073 Å)
μ :	3.19 cm ⁻¹
Diffractometer, scan mode:	Bruker Kappa APEXII Duo, $\omega + \varphi$ Scans
$2\theta_{max}$:	56.66°
$N(hkl)_{measured}$, $N(hkl)_{unique}$:	16128, 4455
Criterion for I_{obs} , $N(hkl)_{gt}$:	$I_{obs} > 2 \sigma(I_{obs})$, 2809
$N(param)_{refined}$:	311
Programs:	SAINT program [14], SHELX [15]

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(7A)	4e	0.058(4)	0.823(2)	1.032(2)	0.038(8)
H(7B)	4e	0.275(4)	0.818(2)	1.093(2)	0.040(1)
H(7C)	4e	0.112(4)	0.842(2)	1.154(2)	0.037(9)
H(7D)	4e	0.230(3)	0.804(8)	1.060(2)	0.0370
H(9)	4e	0.435(3)	0.805(1)	0.864(2)	0.023(6)
H(10)	4e	0.377(4)	0.681(1)	0.848(2)	0.050(8)
H(11)	4e	0.076(3)	0.637(1)	0.815(2)	0.026(6)
H(12)	4e	-0.173(4)	0.717(1)	0.793(2)	0.041(8)
H(13)	4e	-0.123(3)	0.842(1)	0.806(2)	0.027(6)
H(15A)	4e	0.175(3)	1.025(1)	1.406(2)	0.028(7)
H(15B)	4e	0.397(3)	1.016(1)	1.401(1)	0.010(5)
H(16A)	4e	0.222(4)	1.151(1)	1.381(2)	0.042(8)
H(16B)	4e	0.336(4)	1.127(1)	1.473(2)	0.046(8)
H(16C)	4e	0.440(3)	1.140(1)	1.370(2)	0.037(7)
H(18A)	4e	0.060(3)	0.899(1)	0.501(2)	0.016(6)
H(18B)	4e	0.292(3)	0.899(1)	0.505(2)	0.024(6)
H(19A)	4e	0.065(4)	0.774(1)	0.554(2)	0.036(7)
H(19B)	4e	0.175(3)	0.780(1)	0.455(2)	0.047(8)
H(19C)	4e	0.282(4)	0.771(2)	0.562(2)	0.059(9)

The crystal structure is shown in the figure. Tables 1–3 contain details of the measurement method and a list of the atoms including atomic coordinates and displacement parameters.

Table 3: Fractional atomic coordinate and displacement parameters (Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
S(1)	4e	0.28487(7)	1.05569(3)	1.06346(4)	0.0166(3)	0.0158(3)	0.0176(3)	−0.0015(2)	0.0007(2)	−0.0039(2)
O(1)	4e	0.2209(2)	0.91872(8)	1.2820(1)	0.037(1)	0.0214(8)	0.0175(9)	0.0065(7)	0.0000(7)	0.0011(7)
C(1)	4e	0.2400(3)	0.9737(1)	1.1213(2)	0.013(1)	0.016(1)	0.015(1)	0.0039(9)	0.0005(9)	−0.0011(9)
S(2)	4e	0.26054(8)	1.04872(3)	0.82988(4)	0.0195(3)	0.0155(3)	0.0174(3)	−0.0016(2)	0.0042(2)	0.0013(2)
O(2)	4e	0.2822(2)	1.03710(7)	1.2694(1)	0.0250(9)	0.0226(9)	0.0139(9)	0.0018(7)	−0.0011(7)	−0.0045(7)
C(2)	4e	0.2049(3)	0.9184(1)	1.0546(2)	0.017(1)	0.017(1)	0.015(1)	0.0052(9)	0.0004(9)	−0.0006(9)
O(3)	4e	0.2296(2)	1.00775(8)	0.6177(1)	0.035(1)	0.0253(9)	0.019(1)	−0.0029(8)	0.0039(7)	0.0050(7)
C(3)	4e	0.2152(3)	0.9432(1)	0.9536(2)	0.010(1)	0.016(1)	0.014(1)	0.0028(9)	0.0005(8)	−0.0003(9)
O(4)	4e	0.1773(2)	0.88993(8)	0.6422(1)	0.032(1)	0.0241(8)	0.0122(9)	−0.0044(7)	0.0027(7)	−0.0044(7)
C(4)	4e	0.2538(3)	1.0160(1)	0.9487(2)	0.011(1)	0.016(1)	0.013(1)	0.0002(8)	0.0007(8)	−0.0009(9)
C(5)	4e	0.1946(3)	0.9120(1)	0.8559(2)	0.013(1)	0.016(1)	0.015(1)	0.0016(9)	0.0003(9)	−0.0004(9)
C(6)	4e	0.2144(3)	0.9630(1)	0.7827(2)	0.014(1)	0.018(1)	0.012(1)	−0.0014(9)	0.0009(8)	−0.0033(9)
C(7)	4e	0.1545(4)	0.8428(1)	1.0831(2)	0.040(2)	0.020(1)	0.020(2)	−0.002(1)	0.001(1)	0.001(1)
O(5)	4e	0.029(2)	0.8132(7)	1.119(1)	0.026(6)	0.031(6)	0.036(7)	0.001(6)	0.011(6)	−0.003(6)
C(8)	4e	0.1615(3)	0.8337(1)	0.8387(2)	0.027(1)	0.015(1)	0.009(1)	0.001(1)	0.0020(9)	−0.0020(9)
C(9)	4e	0.3074(4)	0.7856(1)	0.8502(2)	0.028(2)	0.023(1)	0.026(2)	0.003(1)	−0.007(1)	−0.004(1)
C(10)	4e	0.2754(4)	0.7119(1)	0.8404(2)	0.043(2)	0.023(1)	0.028(2)	0.012(1)	−0.009(1)	−0.005(1)
C(11)	4e	0.0967(4)	0.6871(1)	0.8207(2)	0.054(2)	0.016(1)	0.018(1)	−0.002(1)	0.008(1)	−0.005(1)
C(12)	4e	−0.0480(4)	0.7348(1)	0.8067(2)	0.035(2)	0.026(1)	0.020(1)	−0.008(1)	0.012(1)	−0.009(1)
C(13)	4e	−0.0163(3)	0.8081(1)	0.8155(2)	0.024(1)	0.020(1)	0.021(1)	−0.002(1)	0.006(1)	−0.006(1)
C(14)	4e	0.2456(3)	0.9713(1)	1.2307(2)	0.014(1)	0.021(1)	0.016(1)	0.0049(9)	−0.0009(9)	−0.004(1)
C(15)	4e	0.2938(4)	1.0422(1)	1.3786(2)	0.022(1)	0.032(1)	0.013(1)	0.004(1)	−0.001(1)	−0.006(1)
C(16)	4e	0.3236(4)	1.1201(1)	1.4034(2)	0.028(2)	0.036(2)	0.023(2)	0.000(1)	−0.002(1)	−0.014(1)
C(17)	4e	0.2085(3)	0.9572(1)	0.6730(2)	0.012(1)	0.025(1)	0.020(1)	0.001(1)	0.0010(9)	−0.000(1)
C(18)	4e	0.1747(4)	0.8761(1)	0.5345(2)	0.027(2)	0.036(1)	0.012(1)	0.001(1)	0.002(1)	−0.006(1)
C(19)	4e	0.1728(4)	0.7959(2)	0.5246(2)	0.032(2)	0.040(2)	0.026(2)	−0.005(1)	0.008(1)	−0.016(1)

Source of material

The chemical reagents and solvents used in this study are commercially available and were used as purchased. The synthesis of diethyl 3-methyl-4-phenylthieno[2,3-*b*]thiophene-2,5-dicarboxylate was carried out according to procedure reported by Comel & Kirsch [1]. This target compound was obtained as colorless crystals (ethanol), Yield: 86%; m.p. 130 °C; ¹H-NMR (400 MHz, CDCl₃): δ = 1.17 (t, 3H, CH₃, J = 7.1 Hz), 1.39 (t, 3H, CH₃, J = 7.1 Hz), 2.22 (s, 3H, CH₃), 4.07 (q, 2H, CH₂, J = 7.1 Hz), 4.26 (q, 2H, CH₂, J = 7.1 Hz), 7.30 (m, 2H, Ph), 7.43 ppm (m, 3H, Ph); ¹³C-NMR (100 MHz, CDCl₃): δ = 14.4 (CH₃), 15.9 (CH₃), 16.5 (CH₃), 63.6 (2CH₂), 125.2 (2 CAr), 127.1 (CAr), 128.5 (2 CAr), 131.8 (CAr), 133.2 (CAr), 135.2 (CAr), 139.8 (CAr), 142.9 (CAr), 145.7 (CAr), 149.0 (CAr), 161.8 (CO₂), 162.8 ppm (CO₂).

Experimental details

Data were collected on a Bruker APEX-II CCD area diffractometer. Cell refinement and data reduction were carried out by Bruker SAINT [14]. There is a disorder situation of the moiety attached on C2 (see figure). The main part of the disorder is formed by the methyl group and the minor part represents an aldehyde function. The ratio of the population factors methyl/aldehyde is 7:1.

Discussion

Thieno[2,3-*b*]thiophene scaffold represents the key pharmacophore of several pharmaceutically agents. Thus, several thieno [2,3-*b*] thiophene were reported to exhibit marked chemotherapeutic activities such as antibacterial, antiviral, antitumor, anti-glaucoma drugs, β-glucuronidase, α-glucosidase inhibition, and anti-oxidant potential [2–13]. In continues of our research program we report her crystal structure of title compound elucidated by X-ray diffraction analysis.

In the title compound, C₁₉H₁₈O₄S₂, the thieno [2,3-*b*] thiophene nucleus is tetra-substituted at C1, C2, C5 and C6. All bond lengths and angles are in the expected ranges.

Acknowledgements: The authors extend their sincere appreciation to the Deanship of Scientific Research at king Saud University for its funding this Prolific Research group (PRG-1437-29).

References

1. Comel, A.; Kirsch, G. J.: Efficient one pot preparation of variously substituted thieno [2,3-*b*] thiophene. *Heterocycl. Chem.* **38** (2001) 1167–1171.

2. Mabkhot, Y. N.; Aldawsari, F. D.; Al-Showiman, S. S.; Barakat, A.; Hadda, T. B.; Mubarak, M. S.; Naz, S.; Ul-Haq, Z.; Rauf, A.: Synthesis, bioactivity, molecular docking and POM analyses of novel substituted thieno[2,3-*b*] thiophenes and related congeners. *Molecules* **20** (2015) 1824–1841.
3. Mabkhot, Y. N.; Barakat, A.; Yousuf, S.; Choudhary, M. I.; Frey, W.; Hadda, T. B.; Mubarak, M. S.: Substituted thieno [2,3-*b*] thiophenes and related congeners: Synthesis, β -glucuronidase inhibition activity, crystal structure, and POM analyses. *Bioorg. Med. Chem.* **22** (2014) 6715–6725.
4. Mabkhot, Y. N.; Aldawsari, F. D.; Al-Showiman, S. S.; Barakat, A.; Choudhary, M. I.; Yousuf, S.: Synthesis, characterization, X-ray crystal structure, and antimicrobial activity of 1,1'-(3,4-diphenylthieno[2,3-*b*]thiophene-2,5-diyl)diethanone. *J. Chem.* **2014** (2014) 201–206.
5. Mabkhot, Y. N.; Al-Showiman, S. S.; Barakat, A.; Choudhary, M. I.; Yousuf, S.: 3,4-Dimethylthieno[2,3-*b*]thiophene-2,5-dicarbonitrile. *Acta Cryst.* **E69** (2013) o1272.
6. Mabkhot, Y. N.; Barakat, A.; Alshahrani, S.: Expeditious and highly efficient protocol for the synthesis of novel diversely substituted thieno[2,3-*b*] thiophene. *J. Mol. Struct.* **1027** (2012) 15–19.
7. Mabkhot, Y. N.; Barakat, A.; Al-Majid, A. M.; Alamar, A. S.; Al-Nahary, T. T.: A novel and expedient approach to new heterocycles containing thiazole, thiazolo[3,2-*a*]pyridine. *Int. J. Mol. Sci.* **13** (2012) 5035–5047.
8. Mabkhot, Y. N.; Barakat, A.; Al-Majid, A. M.; Alshahrani, S. A.: Comprehensive and facile synthesis of some functionalized bis-heterocyclic compounds containing a thieno[2,3-*b*]thiophene motif. *Int. J. Mol. Sci.* **13** (2012) 2263–2275.
9. Mabkhot, Y. N.; Al-Majid, A. M.; Barakat, A.; Alshahrani, S.; Siddiqui, Y.: 1,1'-(3-Methyl-4-phenylthieno[2,3-*b*]thiophene-2,5-diyl)diethanone as a building block in heterocyclic synthesis. Novel synthesis of some pyrazole and pyrimidine derivatives. *Molecules*. **16** (2011) 6502–6511.
10. Mabkhot, Y. N.; Aldawsari, F.; Al-Showiman, S. S.; Barakat, A.; Soliman, S. M.; Choudhary, M. I.; Yousuf, S.; Mubarak, M.; Hadda, T.: Novel enaminone derived from thieno[2,3-*b*]thiophene: Synthesis, X-ray crystal structure, HOMO, LUMO, NBO analyses and biological activity. *Chem. C. J.* **9** (2015) 24–35.
11. Mabkhot, Y. N.; Barakat, A.; Al-Majid, A. M.; Alshahrani, S.; Yousuf, S.; Choudhary, M. I.: Synthesis, reactions and biological activity of some new bis-heterocyclic ring compounds containing sulphur atom. *Chem. Cent. J.* **7** (2013) 112–120.
12. Hanahan, D.; Weinberg, R. A.: The hallmarks of cancer. *Cell* **100** (2000) 57–70.
13. Mac Coss, M., Baillie, T. A.: Organic chemistry in drug discovery. *Science* **303** (2004) 1810–1813.
14. Bruker: APEX2, SAINT and SADABS. Bruker AXS Inc., Madison, WI, USA, 2009.
15. Sheldrick, G. M.: A short history of SHELX. *Acta Crystallogr.* **A64** (2008) 112–122.