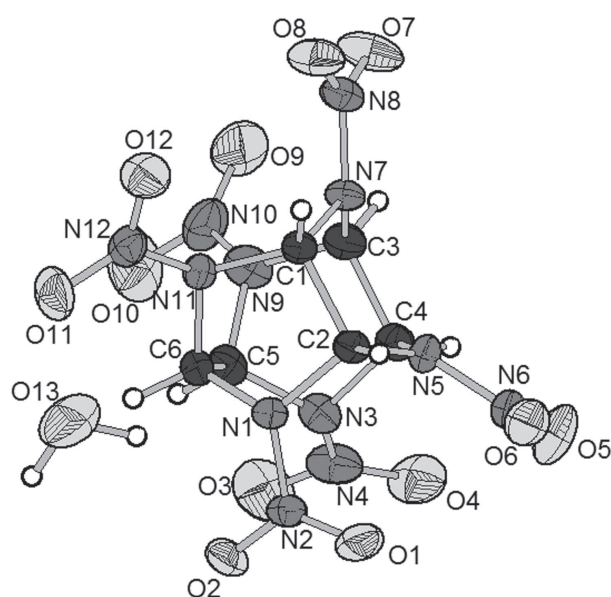


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Crystal structure of 2,4,6,8,10,12-hexanitro-2,4,6,8,10,12-hexaazatetracyclo[5.5.0.0_{5.9}.0_{3.11}]dodecane 1/3 hydrate, C₆H₈N₁₂O₁₃



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Abstract

C₆H₆N₁₂O₁₂ · 0.33 H₂O, orthorhombic, *Pbca* (no. 61), *a* = 9.5152(7) Å, *b* = 13.2551(11) Å, *c* = 23.6736(18) Å, *V* = 2985.8(7) Å³, *Z* = 4, *R*_{gt}(*F*) = 0.0727, *wR*_{ref}(*F*²) = 0.1268, *T* = 293(2) K.

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The crystal structure is shown in the figure. Tables 1–3 contain details of the measurement method and a list of the atoms including atomic coordinates and displacement parameters.

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Table 1: Data collection and handling.

Crystal:	Yellow, block, size 0.20 × 0.21 × 0.22 mm
Wavelength:	Mo K _α radiation (0.71073 Å)
μ:	1.96 cm ^{−1}
Diffractometer, scan mode:	CCD, φ and ω scans
2θ _{max} :	50.1°
<i>N</i> (<i>hkl</i>) _{measured} , <i>N</i> (<i>hkl</i>) _{unique} :	17728, 2637
Criterion for <i>I</i> _{obs} , <i>N</i> (<i>hkl</i>) _{gt} :	<i>I</i> _{obs} > 2 σ(<i>I</i> _{obs}), 2196
<i>N</i> (<i>param</i>) _{refined} :	299
Programs:	SHELX [6], Diamond [7]

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1)	8c	1.2643	0.1508	0.1380	0.027
H(2)	8c	1.1511	0.2272	0.2110	0.027
H(3)	8c	0.9646	0.2484	0.0317	0.034
H(4)	8c	0.8489	0.3200	0.1056	0.034
H(5)	8c	0.7784	0.0349	0.1134	0.035
H(6)	8c	0.9519	−0.0223	0.1736	0.030
H(13A)	8c	1.0640	0.3958	0.0586	0.142
H(13B)	8c	1.0087	0.5057	0.0483	0.142

Source of material

The title compound was obtained as follows: HNIW (hexanitrohexaazaisowurtzitane, CL-20) was dissolved in the mixed solution of water and acetone with heating. The colorless transparent cocrystal compound was formed by slow evaporation of solvent.

Experimental details

Single-crystal X-ray diffraction data were collected using a Bruker-AXS SMART APEX2 CCD diffractometer. The coordinates of the H atoms were idealized and refined using the AFIX 13 or the AFIX 3 option respectively of the SHELX program [6].

Discussion

HNIW is a novel high energetic explosive. ε-CL-20 is the most stable in the known four polymorphs at normal temperatures and pressures, unfortunately, which had

Table 3: Atomic displacement parameters (Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
O(1)	8c	0.9171(2)	0.2056(1)	0.27540(7)	0.055(1)	0.049(1)	0.0358(8)	−0.0096(8)	0.0168(8)	−0.0159(8)
N(1)	8c	1.0027(2)	0.1165(1)	0.20449(6)	0.0227(8)	0.0268(8)	0.0202(8)	−0.0006(7)	0.0024(6)	0.0007(7)
C(1)	8c	1.1627(2)	0.1509(1)	0.13189(7)	0.0222(9)	0.026(1)	0.0194(9)	−0.0004(8)	−0.0004(7)	0.0010(8)
O(2)	8c	0.8063(2)	0.0715(1)	0.24738(7)	0.0311(8)	0.0354(8)	0.0455(9)	−0.0047(7)	0.0116(7)	0.0070(7)
N(2)	8c	0.9015(2)	0.1326(1)	0.24494(7)	0.0282(9)	0.0339(9)	0.0234(8)	0.0020(8)	0.0048(7)	0.0030(8)
C(2)	8c	1.0844(2)	0.2027(1)	0.18262(8)	0.0226(9)	0.026(1)	0.0197(9)	−0.0028(8)	−0.0011(7)	−0.0003(8)
O(3)	8c	0.5820(2)	0.1288(2)	0.1218(1)	0.0260(9)	0.080(2)	0.096(2)	−0.013(1)	−0.010(1)	0.008(1)
N(3)	8c	0.8023(2)	0.1813(1)	0.13822(7)	0.0190(8)	0.0345(9)	0.0342(9)	0.0008(7)	−0.0036(7)	0.0067(8)
C(3)	8c	0.9798(2)	0.2101(2)	0.06674(8)	0.028(1)	0.035(1)	0.0229(9)	−0.0003(9)	−0.0022(8)	0.0038(9)
O(4)	8c	0.6221(2)	0.2884(2)	0.12955(8)	0.038(1)	0.066(1)	0.063(1)	0.0227(9)	0.0062(9)	0.010(1)
N(4)	8c	0.6560(2)	0.2013(2)	0.12776(8)	0.025(1)	0.062(1)	0.037(1)	0.006(1)	0.0003(8)	0.010(1)
C(4)	8c	0.8999(2)	0.2596(2)	0.11779(8)	0.026(1)	0.029(1)	0.030(1)	0.0036(8)	−0.0001(8)	0.0064(8)
O(5)	8c	0.8764(2)	0.4209(1)	0.1830(1)	0.077(1)	0.038(1)	0.088(2)	0.026(1)	0.014(1)	−0.011(1)
N(5)	8c	1.0002(2)	0.2840(1)	0.16099(7)	0.0307(9)	0.0212(8)	0.0293(9)	0.0003(7)	0.0016(7)	−0.0047(7)
C(5)	8c	0.8542(2)	0.0844(2)	0.11757(9)	0.027(1)	0.031(1)	0.031(1)	−0.0073(9)	−0.0031(8)	−0.0003(9)
O(6)	8c	1.0570(2)	0.3816(1)	0.23404(7)	0.072(1)	0.045(1)	0.043(1)	−0.0206(9)	0.0123(9)	−0.0211(8)
N(6)	8c	0.9744(2)	0.3680(1)	0.19602(9)	0.053(1)	0.0251(9)	0.044(1)	−0.0064(9)	0.018(1)	−0.0077(9)
C(6)	8c	0.9732(2)	0.0452(1)	0.15888(8)	0.028(1)	0.0221(9)	0.025(1)	−0.0011(8)	0.0012(8)	0.0008(8)
O(7)	8c	1.1848(2)	0.2392(2)	−0.00816(6)	0.044(1)	0.086(1)	0.0242(8)	−0.0065(9)	0.0011(7)	0.0202(8)
N(7)	8c	1.1268(2)	0.2053(1)	0.08092(7)	0.0242(8)	0.0318(9)	0.0187(8)	0.0001(7)	0.0024(7)	0.0034(7)
O(8)	8c	1.3401(2)	0.1734(1)	0.04730(7)	0.0293(9)	0.065(1)	0.0398(9)	0.0043(8)	0.0088(7)	0.0090(8)
N(8)	8c	1.2240(2)	0.2044(1)	0.03673(7)	0.033(1)	0.039(1)	0.0237(9)	−0.0065(8)	0.0040(7)	0.0029(8)
N(9)	8c	0.9169(2)	0.1100(2)	0.06323(7)	0.037(1)	0.049(1)	0.0213(8)	−0.0121(9)	−0.0089(7)	−0.0011(8)
O(9)	8c	0.9079(3)	0.1106(2)	−0.02962(8)	0.109(2)	0.066(1)	0.0290(9)	−0.002(1)	0.002(1)	−0.0107(9)
N(10)	8c	0.8675(2)	0.0714(2)	0.01389(9)	0.063(1)	0.042(1)	0.038(1)	0.003(1)	−0.006(1)	−0.013(1)
O(10)	8c	0.7895(3)	−0.0008(2)	0.01781(9)	0.125(2)	0.064(1)	0.063(1)	−0.039(2)	−0.015(1)	−0.018(1)
O(11)	8c	1.1563(2)	−0.1088(1)	0.16057(8)	0.051(1)	0.0287(8)	0.064(1)	0.0057(7)	0.0011(9)	0.0129(8)
N(11)	8c	1.1072(2)	0.0475(1)	0.12995(7)	0.0269(9)	0.0220(8)	0.0273(8)	0.0033(7)	0.0027(7)	0.0017(7)
O(12)	8c	1.3239(2)	−0.0144(1)	0.12873(8)	0.0325(9)	0.048(1)	0.061(1)	0.0137(8)	0.0092(8)	0.0115(8)
N(12)	8c	1.2037(2)	−0.0313(1)	0.14178(7)	0.038(1)	0.029(1)	0.0288(9)	0.0086(8)	−0.0011(8)	0.0023(8)
O(13)	8c	0.9918(9)	0.4354(6)	0.0411(3)	0.111(6)	0.071(4)	0.102(6)	−0.038(4)	−0.051(5)	0.004(4)

high sensitivity and high production cost. Based on the cocrystallization [1–5], we therefore present a cocrystal of CL-20/H₂O to synthesize a new cocrystal explosive. We provide a method for the security of energetic materials. The crystal structure reveals that the cocrystal is made up of 0.33 H₂O and two 2,4,6,8,10,12-hexanitro-2,4,6,8,10,12-hexaazatetracyclo[5.5.0.0_{5,9}.0_{3,11}]dodecane molecules. In crystal structure CL-20 builds up intermolecular hydrogen bonds to water molecules. Hydrogen bonds between all molecules form a two-dimensional cocrystal.

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