

Guang-Hui Zhang*

Crystal structure of 2-amino-4-(3,4-difluorophenyl)-5-oxo-5,6,7,8-tetrahydro-4H-chromene-3-carbonitrile, $C_{16}H_{12}N_2O_2F_2$

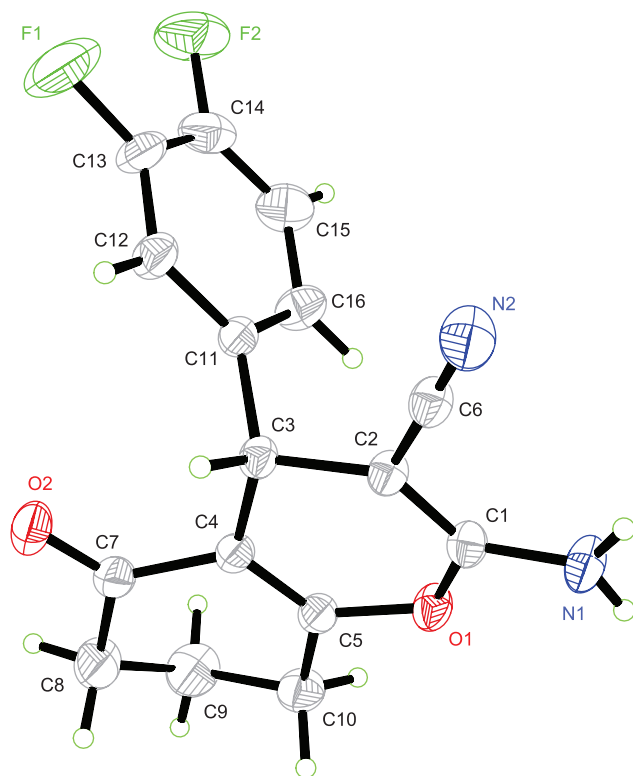


Table 1: Data collection and handling.

Crystal:	Colorless, prism, size $0.16 \times 0.17 \times 0.19$ mm
Wavelength:	Mo K_{α} radiation ($0.71073 \text{ \AA}$)
μ :	1.11 cm^{-1}
Diffractometer, scan mode:	Area detector, φ and ω scans
$2\theta_{\max}$:	50°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	5321, 2528
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 1841
$N(\text{param})_{\text{refined}}$:	200
Programs:	SHELX [6]

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(3A)	8f	0.1391	1.0562	0.4594	0.045
H(8A)	8f	−0.1702	0.8688	0.3585	0.091
H(8B)	8f	−0.1160	0.7951	0.4219	0.091
H(9A)	8f	−0.0690	0.7784	0.2929	0.092
H(9B)	8f	−0.1178	0.6744	0.3298	0.092
H(10A)	8f	0.0335	0.6394	0.4029	0.066
H(10B)	8f	0.0638	0.6493	0.3320	0.066
H(12A)	8f	0.1136	1.2509	0.4100	0.061
H(15A)	8f	0.1779	1.1156	0.2066	0.080
H(16A)	8f	0.1844	0.9811	0.2956	0.067
H(1A)	8f	0.4293	0.8221	0.4668	0.064
H(1B)	8f	0.3806	0.7073	0.4364	0.064

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Abstract

$C_{16}H_{12}F_2N_2O_2$, monoclinic, $C2/c$ (no. 15), $a = 13.2109(11) \text{ \AA}$, $b = 10.957(1) \text{ \AA}$, $c = 20.2646(18) \text{ \AA}$, $\beta = 101.411(9)^{\circ}$, $V = 2875.4(4) \text{ \AA}^3$, $Z = 8$, $R_{\text{gt}}(F) = 0.0529$, $wR_{\text{ref}}(F^2) = 0.1649$, $T = 293 \text{ K}$.

The crystal structure is shown in the figure, Tables 1–3 contain details of the measurement method and a list of

the atoms including atomic coordinates and displacement parameters.

Source of material

The title compound was synthesized according to a reported procedure [5]. A mixture of 3,5-cyclohexanedione (10 mmol), 3,4-difluorobenzaldehyde (10 mmol), malononitrile (10 mmol) and 4-(dimethylamino)pyridine (DMAP) (1 mmol) in ethanol (100 mL) was refluxed for 2–3 h and then cooled to room temperature. After filtering the precipitates, they were sequentially washed with ice-cooled water and ethanol and then dried under vacuum.

*Corresponding author: Guang-Hui Zhang, School of Pharmacy, Shaanxi University of Chinese Medicine, Xi'an, China, e-mail: zhang_guanghui66@126.com

Table 3: Atomic displacement parameters (Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
C(1)	8 <i>f</i>	0.2847(2)	0.8399(2)	0.4308(1)	0.036(1)	0.042(2)	0.041(1)	0.003(1)	0.004(1)	−0.003(1)
C(2)	8 <i>f</i>	0.2629(2)	0.9571(2)	0.4427(1)	0.035(1)	0.035(1)	0.041(1)	0.002(1)	0.002(1)	−0.004(1)
C(3)	8 <i>f</i>	0.1563(2)	1.0121(2)	0.4210(1)	0.037(1)	0.033(1)	0.042(1)	0.003(1)	0.008(1)	−0.004(1)
C(4)	8 <i>f</i>	0.0801(2)	0.9099(2)	0.4027(1)	0.036(1)	0.035(1)	0.045(1)	0.001(1)	0.006(1)	0.004(1)
C(5)	8 <i>f</i>	0.1083(2)	0.7964(2)	0.3909(1)	0.035(1)	0.037(1)	0.050(1)	−0.001(1)	0.005(1)	0.002(1)
C(6)	8 <i>f</i>	0.3438(2)	1.0333(2)	0.4746(1)	0.041(1)	0.039(2)	0.046(1)	0.005(1)	0.005(1)	−0.006(1)
C(7)	8 <i>f</i>	−0.0292(2)	0.9371(3)	0.4004(1)	0.039(1)	0.046(2)	0.063(2)	0.003(1)	0.010(1)	0.012(1)
C(8)	8 <i>f</i>	−0.1047(2)	0.8351(3)	0.3813(2)	0.037(2)	0.057(2)	0.132(3)	−0.005(1)	0.012(2)	0.010(2)
C(9)	8 <i>f</i>	−0.0694(2)	0.7420(3)	0.3365(2)	0.050(2)	0.059(2)	0.110(3)	−0.018(2)	−0.007(2)	−0.005(2)
C(10)	8 <i>f</i>	0.0373(2)	0.6946(3)	0.3660(2)	0.054(2)	0.040(2)	0.071(2)	−0.009(1)	0.009(1)	−0.003(1)
C(11)	8 <i>f</i>	0.1503(2)	1.1016(2)	0.3626(1)	0.030(1)	0.032(1)	0.049(1)	−0.0027(9)	0.001(1)	−0.001(1)
C(12)	8 <i>f</i>	0.1270(2)	1.2225(2)	0.3694(1)	0.055(2)	0.036(2)	0.061(2)	0.004(1)	0.011(1)	−0.003(1)
C(13)	8 <i>f</i>	0.1233(2)	1.3012(3)	0.3166(2)	0.062(2)	0.031(2)	0.089(2)	0.003(1)	0.009(2)	0.012(2)
C(14)	8 <i>f</i>	0.1415(2)	1.2616(3)	0.2569(2)	0.060(2)	0.066(2)	0.064(2)	−0.014(2)	0.003(1)	0.022(2)
C(15)	8 <i>f</i>	0.1649(2)	1.1426(3)	0.2477(2)	0.077(2)	0.069(2)	0.054(2)	−0.010(2)	0.009(2)	0.003(2)
C(16)	8 <i>f</i>	0.1689(2)	1.0628(3)	0.3011(1)	0.070(2)	0.041(2)	0.055(2)	−0.000(1)	0.010(1)	−0.003(1)
O(1)	8 <i>f</i>	0.2095(1)	0.7601(2)	0.40019(9)	0.0394(9)	0.033(1)	0.064(1)	0.0017(7)	0.0058(8)	−0.0095(8)
O(2)	8 <i>f</i>	−0.0563(1)	1.0376(2)	0.4157(1)	0.049(1)	0.050(1)	0.105(2)	0.0141(9)	0.022(1)	0.005(1)
N(1)	8 <i>f</i>	0.3757(2)	0.7830(2)	0.4465(1)	0.043(1)	0.044(1)	0.067(1)	0.011(1)	−0.002(1)	−0.009(1)
N(2)	8 <i>f</i>	0.4092(2)	1.0964(2)	0.5002(1)	0.050(1)	0.054(2)	0.079(2)	−0.004(1)	0.000(1)	−0.018(1)
F(1)	8 <i>f</i>	0.1039(2)	1.4199(2)	0.3230(2)	0.183(3)	0.048(1)	0.154(2)	0.024(1)	0.051(2)	0.024(1)
F(2)	8 <i>f</i>	0.1385(2)	1.3436(2)	0.2062(1)	0.123(2)	0.095(2)	0.091(2)	−0.015(1)	0.012(1)	0.049(1)

Discussion

Dihydropyran and its derivatives have received considerable attention in the past few years for their versatile biological and medical activities. Different substituents at the aromatic ring lead to multiple chemical modifications which generates antioxidant, anti-inflammatory, antibacterial and anticancer activities for these compounds [1, 2]. Recognizing the considerable importance of the compounds, research focused on the synthesis of dihydropyran derivatives [3, 4].

In the crystal structure of the title compound, the pyran ring and the adjacent cyclohexenone ring are both basically planar, and the two planes are also essentially parallel each other. However, the phenyl ring makes a torsion angle to the pyran ring.

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