

Crystal structure of 1-(1-(2,4-dichlorobenzyl)-5-methyl-1*H*-1,2,3-triazol-4-yl)ethanone, C₁₂H₁₁Cl₂N₃O

Marcia Pérez-Ferhmann^I, Victor Kesternich^I, Manuel Cáceres^I, Fernanda Salazar^I, Alejandro Cárdenas^{II}, Fernando Cataldo^{III} and Iván Brito^{*,IV}

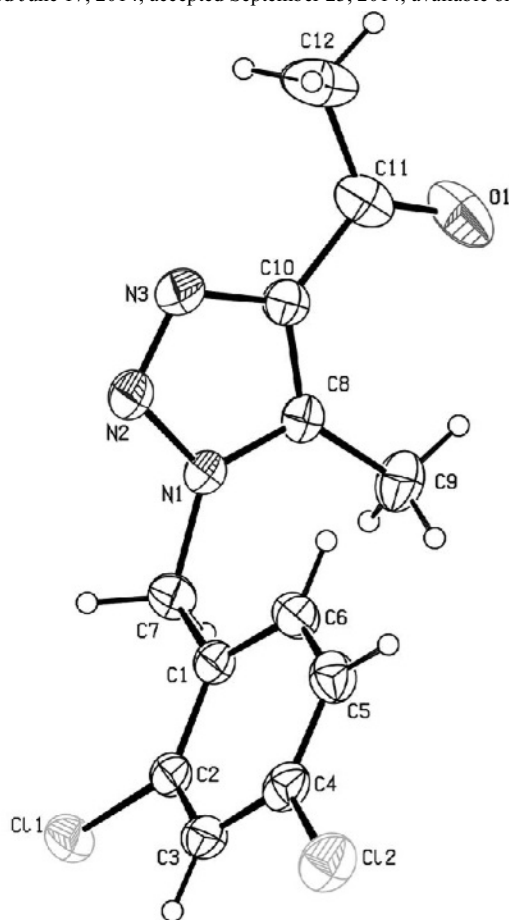
^I Departamento de Química, Universidad Católica del Norte, Casilla 567, Antofagasta, Chile

^{II} Departamento de Física, Facultad de Ciencias Básicas, Universidad de Antofagasta, Casilla 170, Antofagasta, Chile

^{III} Instituto de Química de Recursos Naturales, Universidad de Talca, Casilla 747, Talca, Chile

^{IV} Departamento de Química, Facultad de Ciencias Básicas, Universidad de Antofagasta, Casilla 170, Antofagasta, Chile

Received June 17, 2014, accepted September 23, 2014, available online October 15, 2014, CCDC no. 1267/4179



Abstract

C₁₂H₁₁Cl₂N₃O, triclinic, $P\bar{1}$ (no. 2), $a = 7.0448(14)$ Å, $b = 8.6159(17)$ Å, $c = 11.760(2)$ Å, $\alpha = 98.53(3)^\circ$, $\beta = 99.99(3)^\circ$, $\gamma = 109.55(3)^\circ$, $V = 645.9$ Å³, $Z = 2$, $R_{\text{gt}}(F) = 0.0425$, $wR_{\text{ref}}(F^2) = 0.1247$, $T = 290$ K.

Source of material

To 11.0 g (54.4 mmol) of 2,4-dichlorobenzyl chloride, 4.60 g (70.68 mmol) of sodium azide in 150 ml of acetonitrile were added [1]. The mixture of reaction was refluxed for 15 hours. Thereafter, the mixture is cooled and 50 mL of water was added and extracted with chloroform (3×40 mL). The organic solution was dried with sodium sulfate and distilled at vacuum. The yel-

low oily product obtained was dissolved in 70 mL of DMSO and then, 16.19 mL (156.84 mmol) of acetylacetone and 21.7 g (214.8 mmol) of potassium carbonate were added. The mixture was left to react for 48 hours at room temperature [2]. Finally, 60 mL of saturated solution of NaCl was added and the precipitate formed was washed, dried and crystallized in ethyl acetate, obtaining 12.69 g (yield 84%) of yellow crystals. Analysis: Solid crystalline **m.p.** 385 K **IR** (cm⁻¹): 3076 (C_{Ar}-H); 2974 and 2924 (C_{sp}³-H); 1680 (C=O); 1588 and 1562 (C_{Ar}-C_{Ar}). **¹H-NMR** (CDCl₃, 500 MHz) δ (ppm): 2,50 (3H, s, H15), 2,70 (3H, s, H18), 5,58 (2H, s, H7), 6,79 (1H, d, $J = 8,3$ Hz, H6), 7,21 (1H, d, $J = 8,4$ Hz, H1), 7,46 (1H, s, H3). **¹³C-NMR** (CDCl₃, 125,70 MHz) δ (ppm): 9.17(C15), 27.85(C18), 48.21(C7), 128.12(C1), 129.69(C6), 129.86(C3), 130.59(C5), 133.37(C2), 135.44(C4), 137.28(C14), 144.04(C13), 194.41(C16).

Table 1. Data collection and handling.

Crystal:	white blocks, size 0.07×0.11×0.16 mm
Wavelength:	Mo K_{α} radiation (0.71073 Å)
μ :	4.93 cm ⁻¹
Diffractometer, scan mode:	Enraf Nonius CAD4, non-profiled ω
$2\theta_{\text{max}}$:	52.01°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	5082, 2545
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 2092
$N(\text{param})_{\text{refined}}$:	166
Programs:	SHELX, PLATON, XCAD4, OLEX [8–11]

Experimental details

H atoms were located in the difference Fourier map, but refined with fixed individual displacement parameters, using a riding mode with C–H distances of 0.93 Å (for aromatic rings), 0.97 Å (for CH₂), 0.96 Å (for CH₃), with $U_{\text{iso}}(\text{H})$ values of 1.2 $U_{\text{eq}}(\text{C})$ (for CH in aromatic and methylene group), and 1.5 $U_{\text{eq}}(\text{C})$ (for methyl group).

Discussion

1,2,3-Triazoles are an important class of heterocyclic compounds due to their wide range of applications as pharmaceutical agents [3, 4]. The diversity of chemical structures of the 1,2,3-triazole family and their useful biological activities made these compounds attractive targets in synthetic organic chemistry and many studies have been reported on the synthesis of these compounds [5, 7]. The title compound crystallizes in space group $P\bar{1}$. The bond lengths and angles are in the expected ranges. The dihedral angles between the mean plane of the benzene ring and the triazole ring is 86.64(12)° [for the non-H atoms, r.m.s. deviations = 0.006 and 0.002 Å, respectively]. The crystal packing is stabilized by a weak intermolecular C–H...Oⁱ hydrogen bond, [symmetry code

* Correspondence author (e-mail: ivanbritob@yahoo.com)

(i): $-x+1, -y+2, -z+1$] which links the molecules into centrosymmetric ring with graph-set notation $R_2^2(18)$ [10]. The packing also features π – π stacking interactions between triazole rings [centroid–centroidⁱⁱ distance 3.6770(13) Å, symmetry code (ii) $-x+1, -y+1, -z+1$] and between benzene rings [centroid–centroidⁱⁱⁱ distance 3.8279(15) Å, symmetry code (iii): $-x, -y+1, -z$].

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(3)	2i	0.1602	0.4135	−0.1406	0.061

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Cl(1)	2i	0.17456(8)	0.20684(6)	0.01875(5)	0.0546(3)	0.0450(3)	0.0705(4)	0.0137(2)	0.0084(2)	−0.0027(2)
Cl(2)	2i	0.2356(1)	0.74072(9)	−0.17872(5)	0.0732(4)	0.0917(5)	0.0634(4)	0.0379(3)	0.0206(3)	0.0302(3)
O(1)	2i	0.5387(4)	0.9439(3)	0.6552(2)	0.122(2)	0.076(1)	0.082(1)	0.027(1)	0.045(1)	−0.007(1)
N(1)	2i	0.4945(2)	0.5826(2)	0.3595(1)	0.0377(8)	0.0492(9)	0.0493(8)	0.0179(7)	0.0081(6)	0.0067(7)
N(2)	2i	0.7031(3)	0.6251(2)	0.3794(2)	0.0402(9)	0.065(1)	0.072(1)	0.0227(8)	0.0129(8)	0.0020(9)
N(3)	2i	0.7860(3)	0.7430(2)	0.4754(2)	0.0440(9)	0.059(1)	0.069(1)	0.0170(8)	0.0036(8)	0.0044(9)
C(1)	2i	0.3287(3)	0.5293(2)	0.1471(2)	0.0338(9)	0.0460(9)	0.052(1)	0.0146(7)	0.0109(7)	0.0038(8)
C(2)	2i	0.2436(3)	0.4243(2)	0.0342(2)	0.0311(8)	0.0443(9)	0.058(1)	0.0121(7)	0.0111(7)	0.0017(8)
C(3)	2i	0.2140(3)	0.4859(3)	−0.0661(2)	0.0363(9)	0.061(1)	0.050(1)	0.0168(9)	0.0099(8)	0.0005(9)
C(4)	2i	0.2666(3)	0.6586(3)	−0.0532(2)	0.039(1)	0.067(1)	0.054(1)	0.0221(9)	0.0161(8)	0.0165(9)
C(5)	2i	0.3480(3)	0.7670(3)	0.0565(2)	0.053(1)	0.049(1)	0.064(1)	0.0186(9)	0.0189(9)	0.0122(9)
C(6)	2i	0.3796(3)	0.7013(2)	0.1554(2)	0.051(1)	0.048(1)	0.051(1)	0.0156(9)	0.0128(8)	0.0025(8)
C(7)	2i	0.3635(3)	0.4552(2)	0.2538(2)	0.050(1)	0.045(1)	0.057(1)	0.0141(8)	0.0086(9)	0.0051(8)
C(8)	2i	0.4444(3)	0.6754(2)	0.4429(2)	0.045(1)	0.049(1)	0.049(1)	0.0219(8)	0.0158(8)	0.0160(8)
C(9)	2i	0.2271(4)	0.6565(3)	0.4439(2)	0.052(1)	0.079(2)	0.085(2)	0.031(1)	0.028(1)	0.024(1)
C(10)	2i	0.6324(3)	0.7781(2)	0.5169(2)	0.052(1)	0.046(1)	0.048(1)	0.0179(8)	0.0102(8)	0.0122(8)
C(11)	2i	0.6779(4)	0.9095(3)	0.6238(2)	0.088(2)	0.049(1)	0.049(1)	0.017(1)	0.016(1)	0.0131(9)
C(12)	2i	0.9003(5)	0.9967(4)	0.6900(2)	0.115(2)	0.062(2)	0.067(2)	0.007(2)	−0.016(2)	0.006(1)

Table 2. continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(5)	2i	0.3813	0.8825	0.0640	0.065
H(6)	2i	0.4367	0.7745	0.2295	0.062
H(7A)	2i	0.2304	0.3958	0.2693	0.063
H(7B)	2i	0.4274	0.3737	0.2363	0.063
H(9A)	2i	0.1661	0.6854	0.3746	0.102
H(9B)	2i	0.2263	0.7301	0.5134	0.102
H(9C)	2i	0.1486	0.5416	0.4443	0.102
H(12A)	2i	0.9090	1.0698	0.7624	0.141
H(12B)	2i	0.9770	1.0622	0.6422	0.141
H(12C)	2i	0.9575	0.9140	0.7079	0.141

References

- Chen, X.; Shi, D.: Synthesis and Biological Activity of Novel Phosphonate Derivatives Containing of Pyridyl and 1,2,3-Triazole Rings Phosphorus, Sulfur, and Silicon. **183** (2008) 1134–1144.
- Melo, J.; Donnic, C.; Augusti, R.; Ferreira, V. F.; Souza, C.; Ferreira, M.; Cunha, A.: 1,2,3-triazolic heterocycles: history, preparations, applications and pharmacological activities. *Quim. Nova.* **29** (2006) 569–579.
- Holla, B. S.; Mahalinga, M.; Karthikeyan, M. S.; Poojary, B.; Akberali, P. M.; Kumari, N. S.: Synthesis, characterization and antimicrobial activity of some substituted 1,2,3-triazoles. *Eur. J. Med. Chem.* **40** (2005) 1173–1178.
- Dabak, K.; Sezer, O.; Akar, A.; Anac, O.: Synthesis and investigation of tuberculosis inhibition activities of some 1,2,3-triazole derivatives. *Eur. J. Med. Chem.* **38** (2003) 215–218.
- Lewis, W. G.; Green, L. G.; Grynszpan, F.; Radic, Z.; Carlier, P. R.; Taylor, P.; Finn, M. G.; Sharpless, K. B.: Click Chemistry In Situ: Acetylcholinesterase as a Reaction Vessel for the Selective Assembly of a Femtomolar Inhibitor from an Array of Building Blocks. *Angew. Chem., Int. Ed.* **41** (2002) 1053–1057.
- Wang, Q.; Chan, T. R.; Hilgraf, R.; Fokin, V. V.; Sharpless, K. B.; Finn, M. G.: Bioconjugation by Copper(I)-Catalyzed Azide-Alkyne [3+2] Cycloaddition. *J. Am. Chem. Soc.* **125** (2003) 3192–3193.
- Link, A. J.; Tirrelli, D. A.: Cell Surface Labeling of *Escherichia coli* via Copper(I)-Catalyzed [3+2] Cycloaddition. *J. Am. Chem. Soc.* **125** (2003) 11164–11165.
- Sheldrick, G. M.: A short history of SHELX. *Acta Crystallogr.* **A64** (2008) 112–122.
- Spek, A. L.: Single-crystal structure validation with the program PLATON. *J. Appl. Crystallogr.* **36** (2003) 7–13.
- Harms, K.; Wocadlo, S.: XCAD4. University of Marburg, Germany (1995).
- Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H.: OLEX2: A complete structure solution, refinement and analysis program. *J. Appl. Cryst.* **42** (2009) 339–341.
- Bernstein, J.; Davis, R. E.; Shimoni, L.; Chang, N.-L.: Patterns in Hydrogen Bonding: Functionality and Graph Set Analysis in Crystals. *Angew. Chem. Int. Ed. Engl.* **34** (1995) 1555–1573.
- Blessing, R. H.: An empirical correction for absorption anisotropy. *Acta Crystallogr.* **A51** (1995) 33–37.