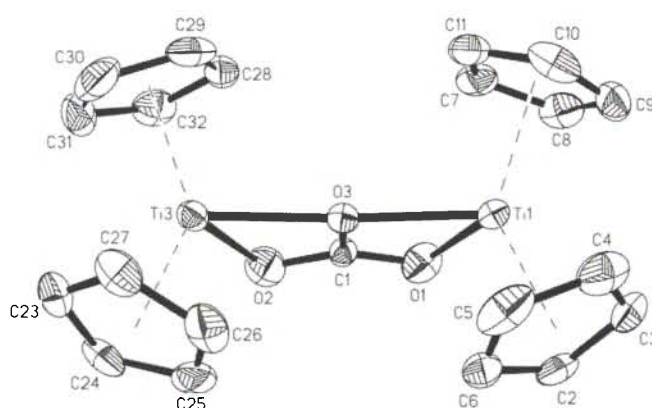


Crystal structure of di(bis(cyclopentadienyl)titanocene) carbonate, $C_{42}H_{40}O_6Ti_4$

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Abstract

$C_{42}H_{40}O_6Ti_4$, monoclinic, $P12_1/n1$ (No. 14), $a = 19.522(4)$ Å, $b = 7.891(2)$ Å, $c = 25.100(5)$ Å, $\beta = 96.52(3)^\circ$, $V = 3841.6$ Å³, $Z = 4$, $R_{obs}(F) = 0.035$, $wR_{all}(F^2) = 0.073$, $T = 293$ K.

Source of material

To a solution of 480 mg (0.58 mmol) $(Cp_2Ti)_2-\mu-(\eta^4-2:3,6:7-PhHC=N-N-CHPh-CHPh-N=N-CHPh)$ in 20 ml of toluene was added 20 µl of H₂O and the solution was then saturated with carbon dioxide. As the brown reaction mixture was stirred for 24 hours at room temperature the color changed to dark blue. The solvent was evaporated in vacuo and the residue was crystallized at 278 K from THF/n-hexane (1:3) to yield 140 mg (58%) of the title compound.

Discussion

On studying the interaction of various titanocene complexes with carbon dioxide, there could be obtained mono-, bi- and tetranuclear species with different substituted Cp-ligands [1–3]. Herein we report on the structure of a postulated binuclear blue carbonate complex of titanocene $(Cp_2Ti)_2(\mu_2, \mu_2-CO_3)$. There are two symmetry independent molecules in the asymmetric unit.

Table 1. Data collection and handling.

Crystal:	blue prism, size 0.2 × 0.3 × 0.5 mm
Wavelength:	Mo K_α radiation (0.71073 Å)
μ :	8.48 cm ⁻¹
Diffraction, scan mode:	Stoe IPDS, 100 exposures, $\Delta\phi = 2^\circ$
$2\theta_{max}$:	48.6°
$N(hkl)_{measured}$, $N(hkl)_{unique}$:	10981, 5960
Criterion for I_{obs} , $N(hkl)_{gt}$:	$I_{obs} > 2 \sigma(I_{obs})$, 3473
$N(param)_{refined}$:	469
Programs:	SHELXS-86 [4], SHELXL-93 [5]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U_{iso}
H(2)	4e	0.5481(2)	0.1172(5)	0.4034(2)	0.063
H(3)	4e	0.5052(2)	0.1450(5)	0.3079(2)	0.071
H(4)	4e	0.5804(2)	-0.0225(6)	0.2554(2)	0.086
H(5)	4e	0.6738(2)	-0.1471(5)	0.3186(2)	0.087
H(6)	4e	0.6527(2)	-0.0630(5)	0.4105(2)	0.075
H(7)	4e	0.7272(2)	0.5379(5)	0.3252(2)	0.070
H(8)	4e	0.5989(2)	0.5554(5)	0.3088(2)	0.074
H(9)	4e	0.5608(2)	0.3364(6)	0.2399(2)	0.083
H(10)	4e	0.6655(3)	0.1888(6)	0.2157(2)	0.086
H(11)	4e	0.7662(2)	0.3143(6)	0.2680(2)	0.078
H(13)	4e	0.9301(2)	0.2656(6)	0.7218(2)	0.088
H(14)	4e	0.9893(2)	0.4277(6)	0.6561(2)	0.096
H(15)	4e	0.9263(3)	0.3768(7)	0.5675(2)	0.109
H(16)	4e	0.8275(3)	0.1935(7)	0.5779(2)	0.099
H(17)	4e	0.8280(2)	0.1283(5)	0.6737(2)	0.089
H(18)	4e	0.8923(2)	0.8285(6)	0.6692(2)	0.095
H(19)	4e	0.7640(2)	0.8028(6)	0.6636(2)	0.090
H(20)	4e	0.7383(3)	0.5778(8)	0.7255(2)	0.099
H(21)	4e	0.8475(4)	0.4627(7)	0.7693(2)	0.115
H(22)	4e	0.9423(3)	0.6206(7)	0.7359(2)	0.109
H(23)	4e	0.9718(2)	-0.1031(5)	0.4600(1)	0.065
H(24)	4e	0.8881(2)	-0.0248(5)	0.5229(1)	0.071
H(25)	4e	0.7699(2)	-0.0795(5)	0.4769(2)	0.078
H(26)	4e	0.7819(2)	-0.1937(5)	0.3867(2)	0.078
H(27)	4e	0.9057(2)	-0.2040(5)	0.3752(2)	0.071
H(28)	4e	0.8479(2)	0.4392(5)	0.3528(2)	0.066
H(29)	4e	0.8898(2)	0.1909(6)	0.3060(2)	0.077
H(30)	4e	0.9862(2)	0.0650(6)	0.3647(2)	0.101
H(31)	4e	1.0036(2)	0.2352(7)	0.4480(2)	0.100
H(32)	4e	0.9189(2)	0.4693(6)	0.4398(2)	0.079
H(33)	4e	0.5119(2)	0.1446(5)	0.5515(2)	0.080
H(34)	4e	0.5780(2)	0.0694(5)	0.6379(2)	0.066
H(35)	4e	0.7021(2)	0.0722(5)	0.6252(2)	0.074
H(36)	4e	0.7115(3)	0.1552(5)	0.5304(2)	0.091
H(37)	4e	0.5930(3)	0.2006(6)	0.4866(2)	0.100
H(38)	4e	0.6426(2)	0.7015(7)	0.6481(2)	0.097
H(39)	4e	0.6023(3)	0.4558(8)	0.6988(2)	0.101
H(40)	4e	0.5034(3)	0.3299(8)	0.6434(3)	0.123
H(41)	4e	0.4813(3)	0.4909(9)	0.5603(3)	0.126
H(42)	4e	0.5669(3)	0.7228(7)	0.5627(3)	0.116

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Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
C(1)	4e	0.7408(2)	0.2708(4)	0.4170(1)	0.031(2)	0.033(2)	0.037(2)	−0.002(2)	0.003(2)	0.001(2)
C(2)	4e	0.5676(2)	0.0682(5)	0.3750(2)	0.050(2)	0.058(3)	0.054(2)	−0.023(2)	0.022(2)	−0.005(2)
C(3)	4e	0.5436(2)	0.0830(5)	0.3218(2)	0.039(2)	0.065(3)	0.073(3)	−0.017(2)	0.002(2)	0.004(2)
C(4)	4e	0.5858(2)	−0.0096(6)	0.2925(2)	0.086(3)	0.078(4)	0.054(3)	−0.044(3)	0.014(3)	−0.023(2)
C(5)	4e	0.6377(2)	−0.0801(5)	0.3277(2)	0.065(3)	0.031(3)	0.127(4)	−0.008(2)	0.039(3)	−0.019(3)
C(6)	4e	0.6260(2)	−0.0325(5)	0.3789(2)	0.054(3)	0.050(3)	0.081(3)	−0.021(2)	−0.006(2)	0.024(2)
C(7)	4e	0.6989(2)	0.4710(5)	0.3014(2)	0.066(3)	0.048(3)	0.064(3)	−0.015(2)	0.012(2)	0.014(2)
C(8)	4e	0.6271(2)	0.4814(5)	0.2924(2)	0.069(3)	0.035(3)	0.085(3)	0.020(2)	0.028(2)	0.020(2)
C(9)	4e	0.6060(2)	0.3590(6)	0.2540(2)	0.058(3)	0.074(3)	0.070(3)	−0.014(2)	−0.016(2)	0.034(3)
C(10)	4e	0.6647(3)	0.2764(6)	0.2404(2)	0.111(4)	0.064(3)	0.040(2)	0.011(3)	0.012(3)	0.007(2)
C(11)	4e	0.7206(2)	0.3462(6)	0.2695(2)	0.056(3)	0.083(4)	0.059(3)	0.011(2)	0.021(2)	0.030(2)
C(12)	4e	0.7415(2)	0.5284(4)	0.5729(1)	0.036(2)	0.032(2)	0.046(2)	−0.002(2)	0.007(2)	0.002(2)
C(13)	4e	0.9166(2)	0.2719(6)	0.6852(2)	0.070(3)	0.075(4)	0.070(3)	0.033(3)	−0.013(3)	0.005(3)
C(14)	4e	0.9500(2)	0.3612(6)	0.6487(2)	0.039(2)	0.065(4)	0.136(5)	0.009(2)	0.014(3)	−0.005(3)
C(15)	4e	0.9148(3)	0.3333(7)	0.5997(2)	0.089(4)	0.106(5)	0.085(4)	0.047(3)	0.044(3)	0.019(3)
C(16)	4e	0.8598(3)	0.2313(7)	0.6055(2)	0.070(4)	0.077(4)	0.097(4)	0.036(3)	−0.012(3)	−0.043(3)
C(17)	4e	0.8601(2)	0.1942(5)	0.6585(2)	0.058(3)	0.034(3)	0.131(5)	0.010(2)	0.013(3)	0.009(3)
C(18)	4e	0.8679(2)	0.7536(6)	0.6885(2)	0.067(3)	0.041(3)	0.129(4)	−0.012(2)	0.010(3)	−0.022(3)
C(19)	4e	0.7961(2)	0.7389(6)	0.6853(2)	0.079(4)	0.050(3)	0.093(4)	0.014(2)	0.001(3)	−0.018(3)
C(20)	4e	0.7821(3)	0.6141(8)	0.7196(2)	0.072(3)	0.097(5)	0.079(3)	−0.007(3)	0.015(3)	−0.048(3)
C(21)	4e	0.8429(4)	0.5500(7)	0.7442(2)	0.150(6)	0.079(4)	0.055(3)	−0.004(4)	−0.003(3)	−0.021(3)
C(22)	4e	0.8956(3)	0.6373(7)	0.7254(2)	0.084(4)	0.067(4)	0.111(4)	0.017(3)	−0.035(3)	−0.039(3)
C(23)	4e	0.9241(2)	−0.1106(5)	0.4532(1)	0.053(2)	0.047(3)	0.059(2)	0.013(2)	−0.005(2)	0.005(2)
C(24)	4e	0.8774(2)	−0.0671(5)	0.4884(1)	0.086(3)	0.052(3)	0.038(2)	0.018(2)	0.003(2)	0.015(2)
C(25)	4e	0.8111(2)	−0.0981(5)	0.4625(2)	0.063(3)	0.042(3)	0.094(3)	0.007(2)	0.028(3)	0.032(2)
C(26)	4e	0.8179(2)	−0.1612(5)	0.4121(2)	0.073(3)	0.029(3)	0.087(3)	−0.005(2)	−0.017(3)	0.006(2)
C(27)	4e	0.8870(2)	−0.1678(5)	0.4057(2)	0.082(3)	0.038(3)	0.056(3)	0.010(2)	0.002(2)	−0.002(2)
C(28)	4e	0.8842(2)	0.3687(5)	0.3659(2)	0.046(2)	0.052(3)	0.066(3)	−0.011(2)	0.002(2)	0.026(2)
C(29)	4e	0.9074(2)	0.2299(6)	0.3398(2)	0.078(3)	0.073(4)	0.045(2)	−0.019(3)	0.024(2)	0.007(2)
C(30)	4e	0.9609(2)	0.1600(6)	0.3724(2)	0.053(3)	0.072(4)	0.136(5)	0.011(2)	0.045(3)	0.028(3)
C(31)	4e	0.9707(2)	0.2553(7)	0.4188(2)	0.047(3)	0.091(4)	0.106(4)	−0.025(3)	−0.024(3)	0.048(3)
C(32)	4e	0.9234(2)	0.3853(6)	0.4144(2)	0.074(3)	0.056(3)	0.068(3)	−0.025(3)	0.009(2)	0.001(2)
C(33)	4e	0.5597(2)	0.1377(5)	0.5582(2)	0.057(3)	0.065(3)	0.077(3)	−0.022(2)	−0.001(2)	0.002(2)
C(34)	4e	0.5964(2)	0.0951(5)	0.6063(2)	0.063(3)	0.049(3)	0.055(2)	−0.016(2)	0.020(2)	0.005(2)
C(35)	4e	0.6656(2)	0.0972(5)	0.5992(2)	0.052(3)	0.032(3)	0.098(3)	−0.003(2)	−0.003(2)	0.005(2)
C(36)	4e	0.6710(3)	0.1434(5)	0.5463(2)	0.083(3)	0.046(3)	0.111(4)	−0.021(2)	0.068(3)	−0.024(3)
C(37)	4e	0.6048(3)	0.1682(6)	0.5221(2)	0.132(5)	0.072(4)	0.045(2)	−0.047(3)	0.010(3)	−0.007(2)
C(38)	4e	0.6058(2)	0.6305(7)	0.6368(2)	0.054(3)	0.063(4)	0.126(4)	0.017(3)	0.014(3)	−0.036(3)
C(39)	4e	0.5834(3)	0.4931(8)	0.6652(2)	0.099(4)	0.086(4)	0.074(3)	0.031(3)	0.042(3)	−0.008(3)
C(40)	4e	0.5284(3)	0.4237(8)	0.6342(3)	0.061(4)	0.088(5)	0.169(6)	0.007(3)	0.062(4)	−0.021(5)
C(41)	4e	0.5160(3)	0.5129(9)	0.5880(3)	0.057(3)	0.097(5)	0.157(6)	0.035(4)	−0.010(4)	−0.034(4)
C(42)	4e	0.5638(3)	0.6421(7)	0.5894(3)	0.091(4)	0.059(4)	0.144(5)	0.040(3)	0.034(4)	0.015(3)
O(1)	4e	0.6777(1)	0.3353(3)	0.4089(1)	0.060(2)	0.073(2)	0.073(2)	−0.004(1)	0.014(1)	−0.006(2)
O(2)	4e	0.7918(1)	0.2812(3)	0.4565(1)	0.073(2)	0.067(2)	0.063(2)	−0.000(2)	0.006(2)	0.000(1)
O(3)	4e	0.7559(1)	0.1731(3)	0.37660(7)	0.037(1)	0.032(2)	0.030(1)	0.0013(9)	0.0071(9)	−0.0039(9)
O(4)	4e	0.6861(2)	0.5278(4)	0.5366(1)	0.081(2)	0.070(2)	0.074(2)	−0.011(2)	0.005(2)	0.004(2)
O(5)	4e	0.7330(1)	0.4368(3)	0.61613(8)	0.039(1)	0.035(2)	0.038(1)	−0.003(1)	0.009(1)	0.009(1)
O(6)	4e	0.8032(1)	0.5964(4)	0.5745(1)	0.069(2)	0.081(2)	0.093(2)	−0.004(2)	0.011(2)	0.023(2)
Ti(1)	4e	0.65469(3)	0.21577(7)	0.33289(2)	0.0329(3)	0.0305(4)	0.0370(3)	−0.0018(3)	0.0084(3)	−0.0018(3)
Ti(2)	4e	0.83821(3)	0.48597(8)	0.65077(2)	0.0401(4)	0.0314(4)	0.0480(4)	0.0011(3)	0.0052(3)	0.0022(3)
Ti(3)	4e	0.85901(3)	0.12645(8)	0.41541(2)	0.0388(4)	0.0332(4)	0.0357(3)	0.0007(3)	0.0028(3)	0.0044(3)
Ti(4)	4e	0.62640(3)	0.38128(8)	0.58799(2)	0.0377(4)	0.0394(4)	0.0464(4)	0.0012(3)	0.0093(3)	0.0004(3)

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References

- Fachinetti, G.; Floriani, C.; Chiesi-Villa, A.; Guastini, C.: Carbon Dioxide Activation. Deoxygenation and Disproportionation of Carbon Dioxide Promoted by Bis(cyclopentadienyl)titanium and -zirconium Derivatives. A Novel Bonding Mode of the Carbonato and a Trimer of the Zirconyl Unit. *J. Am. Chem. Soc.* **101** (1979) 1767–1775.
- Burlakov, V. V.; Rosenthal, U.; Dolgushin, F. M.; Yanovsky, A. I.; Struchkov, Yu. T.; Ellert, O. G.; Shur, B. V.; Vol'pin, M. E.: The formation of dinuclear titanium(III) carbonate complex in the reaction of carbon dioxide with bis(trimethylsilyl)acetylene complex of permethyltitanocene. *Organomet. Chem. USSR* **5** (1992) 593–594.
- Burlakov, V. V.; Dolgushin, F. M.; Yanovsky, A. I.; Struchkov, Yu. T.; Shur, V. B.; Rosenthal, U.; Thewalt, U.: Interaction of carbon dioxide with the bis(trimethylsilyl)acetylene complex of permethyltitanocene: synthesis and structure of the binuclear carbonate complex of permethyltitanocene (Cp₂*Ti)₂(μ₁,μ₂-CO₃). *J. Organomet. Chem.* **522** (1996) 241–247.
- Sheldrick, G. M.: Phase Annealing in SHELX-90: Direct Methods for Larger Structures. *Acta Crystallogr. A* **46** (1999) 467–473.
- Sheldrick, G. M.: SHELXL-93, a program for refining crystal structures. University of Göttingen, Germany 1993.