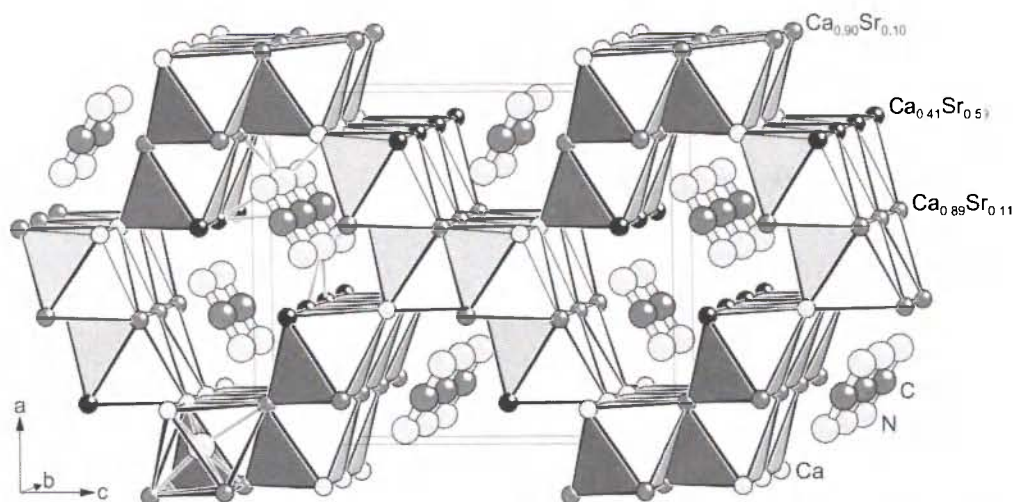


Crystal structure of tetra(calcium/strontium) cyanamide dinitride, $\text{Ca}_{4-x}\text{Sr}_x[\text{CN}_2]\text{N}_2$ ($x = 0.80$)

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Abstract

$\text{CCa}_{3.20}\text{N}_4\text{Sr}_{0.80}$, orthorhombic, $Pnma$ (No. 62), $a = 11.677(1)$ Å, $b = 3.6470(4)$ Å, $c = 13.911(1)$ Å, $V = 592.4$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.029$, $wR_{\text{ref}}(F^2) = 0.057$, $T = 293$ K.

Source of material

Yellow-green single crystals of $\text{Ca}_{4-x}\text{Sr}_x[\text{CN}_2]\text{N}_2$ ($x = 0.80(1)$) were obtained as a minor phase by reaction of Ca_3N_2 (from Ca and N_2 at 1123 K), Sr_2N (from Sr and N_2 at 1123 K), Li and Fe in iron crucibles under N_2 . Carbon was probably introduced either by leaching out of the iron crucible or due to impurities on the surface of Ca- or Sr-pieces. The mixtures were heated to 1123 K (100 K/h), annealed for 36 h, and cooled down (50 K/h) to ambient temperature. Direct synthesis of single phase powder samples was possible by annealing stoichiometric mixtures of Ca_3N_2 , Sr_2N and C at a temperature of 1123 K.

Discussion

In an early investigation on the system Ca-N a high temperature phase $\gamma\text{-Ca}_3\text{N}_2$ was reported [1]. It has been shown recently [2, 3], that this phase has to be regarded as $\text{Ca}_4[\text{CN}_2]\text{N}_2$. Here we report on a strontium substituted phase. The structure is built up by a

framework of corner and edge sharing $(\text{Ca/Sr})_6\text{N}$ octahedra; the cyanamide ions are stacked within channels of the framework. The average bond lengths in the cyanamide ions (1.226 Å) are similar to those reported for $\text{Ca}_4[\text{CN}_2]\text{N}_2$ (1.234 Å) [2] and $\text{Ca}[\text{CN}_2]$ (1.224 Å) [4]. The extent of substitution of Ca by Sr depends on the location of the cations in the framework; average distances $d(\text{Ca/Sr}-\text{N})$ (2.416 Å, 100% Ca – 2.603 Å, 41% Ca) increase with increasing Sr content and are in the range observed in the binary nitrides (Ca_3N_2 : 2.465 Å [5], Sr_2N : 2.612 Å [6]).

Table 1. Data collection and handling.

Crystal:	yellow-green, irregular, size $0.1 \times 0.1 \times 0.1$ mm
Wavelength:	Mo K_α radiation (0.71070 Å)
μ :	99.78 cm ⁻¹
Diffractionmeter, scan mode:	Siemens P4, $\omega/2\theta$
$2\theta_{\text{max}}$:	59.98°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	2643, 987
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 861
$N(\text{param})_{\text{refined}}$:	60
Programs:	SHELXS-97 [7], SHELXL-97 [7], DIAMOND [8]

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Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	Occ.	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
Sr(1)	4c	0.587(3)	0.38560(4)	1/4	0.10800(4)	0.0144(2)	0.0147(2)	0.0172(3)	0	−0.0013(2)	0
Sr(3)	4c	0.110(3)	0.37968(6)	1/4	0.53740(5)	0.0139(4)	0.0132(4)	0.0140(4)	0	−0.0006(2)	0
Sr(4)	4c	0.102(2)	0.14778(6)	1/4	0.23978(5)	0.0122(3)	0.0140(4)	0.0178(4)	0	−0.0009(3)	0
Ca(1)	4c	0.413	0.38560	1/4	0.10800	0.0144	0.0147	0.0172	0	−0.0013	0
Ca(2)	4c		0.10587(7)	1/4	0.82415(6)	0.0152(4)	0.0134(4)	0.0160(4)	0	0.0037(3)	0
Ca(3)	4c	0.890	0.37968	1/4	0.53740	0.0139	0.0132	0.0140	0	−0.0006	0
Ca(4)	4c	0.898	0.14778	1/4	0.23978	0.0122	0.0140	0.0178	0	−0.0009	0
N(1)	4c		0.0028(3)	1/4	0.1184(3)	0.014(2)	0.013(2)	0.022(2)	0	−0.000(1)	0
N(2)	4c		0.2279(3)	1/4	0.6830(3)	0.014(1)	0.016(2)	0.017(2)	0	−0.001(1)	0
N(3)	4c		0.4449(3)	1/4	0.8707(3)	0.019(2)	0.035(2)	0.045(3)	0	0.002(2)	0
N(4)	4c		0.2585(3)	1/4	0.9517(3)	0.024(2)	0.032(2)	0.024(2)	0	−0.001(2)	0
C(1)	4c		0.3508(4)	1/4	0.9094(3)	0.027(2)	0.019(2)	0.013(2)	0	−0.007(2)	0

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