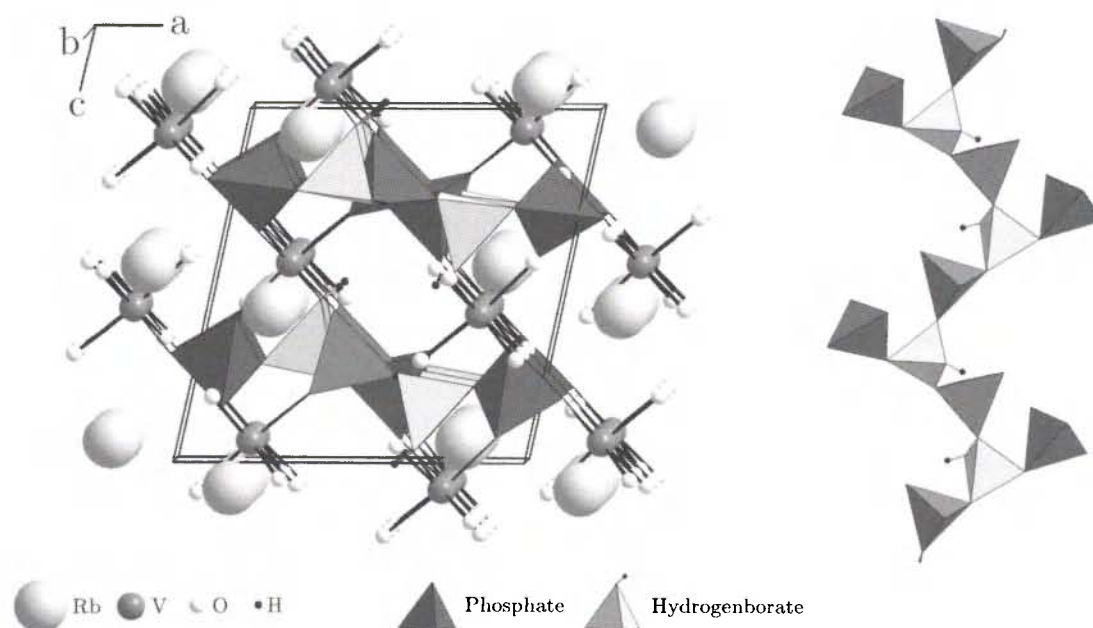


# Crystal structure of rubidium vanadium(III) *catena*-[monohydrogen-monoborate-bis(monophosphate)], $\text{RbV}[\text{BP}_2\text{O}_8(\text{OH})]$

H. Engelhardt, H. Borrmann and R. Kniep\*

Max-Planck-Institut für Chemische Physik fester Stoffe, Pirnaer Landstr. 176, D-01257 Dresden, Germany

Received December 9, 1999, CSD-No. 409468



## Abstract

$\text{BHO}_9\text{P}_2\text{RbV}$ , monoclinic,  $P12_1/c1$  (No. 14),  $a = 9.3789(5)$  Å,  $b = 8.3296(4)$  Å,  $c = 9.6701(3)$  Å,  $\beta = 102.0781(8)^\circ$ ,  $V = 738.7$  Å<sup>3</sup>,  $Z = 4$ ,  $R_{\text{gt}}(F) = 0.100$ ,  $wR_{\text{ref}}(F^2) = 0.111$ ,  $T = 293$  K.

## Source of material

$\text{RbV}[\text{BP}_2\text{O}_8(\text{OH})]$  was prepared under mild hydrothermal conditions from mixtures of a 50 wt.% aqueous solution of  $\text{RbOH}$ ,  $\text{VCl}_3$ ,  $\text{H}_3\text{BO}_3$  and  $\text{H}_3\text{PO}_4$  (85%) in the molar ratio 1:1:1:2 (conc. solution, pH 0–1, 7.5 g – 16 g total weight). The syntheses were carried in teflon autoclaves ( $T = 438$  K, three days) under autogenous pressure. The chemical composition was confirmed by ICP-AES and EDX analyses, the water content was determined by thermogravimetric measurements. FT-IR spectroscopic investigations show the presence of OH groups and prove the absence of hydrogen bonds.

## Discussion

$\text{RbV}[\text{BP}_2\text{O}_8(\text{OH})]$  (hydrated borophosphate, B:P < 1; [1]) is an isotype of  $\text{RbFe}[\text{BP}_2\text{O}_8(\text{OH})]$  [2] and  $\text{CsFe}[\text{BP}_2\text{O}_8(\text{OH})]$  [3]. The anionic partial structure contains open-branched vierer-einfach chains  $[\text{BP}_2\text{O}_8(\text{OH})]^{4-}$ , which are formed by alternating

hydrogenborate and phosphate tetrahedra sharing common corners (figure right). Bond lengths and angles within the anionic chains are consistent with related borophosphates (see [1–3] and refs. therein).  $\text{V}^{3+}$  is in an octahedral coordination by oxygen sites ( $\text{VO}_5(\text{OH})$ ) of three neighbouring tetrahedral chains. By this, an overall three-dimensional linkage of tetrahedra and octahedra is established.  $\text{Rb}^+$  is irregularly coordinated by nine oxygen ligands and one OH group.  $\text{RbV}[\text{BP}_2\text{O}_8(\text{OH})]$  is the second known borophosphate besides  $\text{CsV}_3(\text{H}_2\text{O})_2[\text{B}_2\text{P}_4\text{O}_{16}(\text{OH})_4]$  [4] with vanadium in the oxidation state +3.

**Table 1.** Data collection and handling.

|                                                         |                                                                                                                                                |
|---------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------|
| Crystal:                                                | green, needle, size $0.01 \times 0.03 \times 0.12$ mm                                                                                          |
| Wavelength:                                             | Mo $K_\alpha$ radiation (0.71070 Å)                                                                                                            |
| $\mu$ :                                                 | $83.49 \text{ cm}^{-1}$                                                                                                                        |
| Diffractionmeter, scan mode:                            | Rigaku AFC7-CCD, $0 < \phi < 360^\circ$ , $\Delta\phi = 0.5^\circ$ , $60^\circ$ $\omega$ -scan, $\Delta\omega = 0.5^\circ$ , $\chi = 90^\circ$ |
| $2\theta_{\text{max}}$ :                                | $59.5^\circ$                                                                                                                                   |
| $N(hkl)_{\text{measured}}$ , $N(hkl)_{\text{unique}}$ : | 9165, 1672                                                                                                                                     |
| Criterion for $I_{\text{obs}}$ , $N(hkl)_{\text{gt}}$ : | $I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$ , 1584                                                                                             |
| $N(\text{param})_{\text{refined}}$ :                    | 131                                                                                                                                            |
| Programs:                                               | SHELXS-97 [5], SHELXL-97 [6], DIAMOND [7]                                                                                                      |

\* Correspondence author (e-mail: engelhar@cpgs.mpg.de)

**Table 2.** Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

| Atom | Site | <i>x</i> | <i>y</i> | <i>z</i> | <i>U</i> <sub>iso</sub> |
|------|------|----------|----------|----------|-------------------------|
| H(1) | 4e   | 0.36(1)  | 0.58(2)  | 0.00(1)  | 0.06(5)                 |

**Table 3.** Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

| Atom | Site | <i>x</i>  | <i>y</i>  | <i>z</i>   | <i>U</i> <sub>11</sub> | <i>U</i> <sub>22</sub> | <i>U</i> <sub>33</sub> | <i>U</i> <sub>12</sub> | <i>U</i> <sub>13</sub> | <i>U</i> <sub>23</sub> |
|------|------|-----------|-----------|------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|
| Rb   | 4e   | 0.1882(1) | 0.6016(1) | 0.56009(9) | 0.0300(5)              | 0.0231(5)              | 0.0239(5)              | −0.0053(4)             | 0.0041(4)              | −0.0063(4)             |
| V    | 4e   | 0.2107(1) | 0.1502(2) | 0.4338(1)  | 0.0056(6)              | 0.0077(7)              | 0.0067(7)              | −0.0007(5)             | 0.0010(5)              | 0.0004(5)              |
| B    | 4e   | 0.334(1)  | 0.559(1)  | 0.1957(9)  | 0.012(5)               | 0.008(5)               | 0.006(4)               | 0.001(4)               | 0.000(3)               | −0.001(3)              |
| P(1) | 4e   | 0.0740(2) | 0.4353(2) | 0.2075(2)  | 0.005(1)               | 0.007(1)               | 0.006(1)               | −0.0002(8)             | −0.0002(7)             | 0.0014(7)              |
| P(2) | 4e   | 0.4220(2) | 0.2406(2) | 0.2171(2)  | 0.0050(9)              | 0.008(1)               | 0.008(1)               | 0.0006(8)              | 0.0001(7)              | 0.0004(7)              |
| O(1) | 4e   | 0.0761(6) | 0.0085(6) | 0.3073(5)  | 0.007(3)               | 0.011(3)               | 0.012(3)               | −0.002(2)              | 0.002(2)               | −0.003(2)              |
| O(2) | 4e   | 0.0891(6) | 0.3343(6) | 0.0805(5)  | 0.016(3)               | 0.016(3)               | 0.008(3)               | 0.000(2)               | 0.006(2)               | −0.003(2)              |
| O(3) | 4e   | 0.1059(6) | 0.3426(6) | 0.3463(5)  | 0.012(3)               | 0.008(3)               | 0.009(3)               | 0.000(2)               | −0.001(2)              | 0.001(2)               |
| O(4) | 4e   | 0.1874(5) | 0.5768(6) | 0.2232(5)  | 0.008(3)               | 0.009(3)               | 0.011(3)               | −0.001(2)              | 0.000(2)               | −0.001(2)              |
| O(5) | 4e   | 0.3168(7) | 0.5496(7) | 0.0384(6)  | 0.011(3)               | 0.010(3)               | 0.014(3)               | −0.004(2)              | 0.001(2)               | −0.002(2)              |
| O(6) | 4e   | 0.3480(6) | 0.1288(6) | 0.3027(5)  | 0.011(3)               | 0.008(3)               | 0.020(3)               | 0.001(2)               | 0.009(2)               | 0.003(2)               |
| O(7) | 4e   | 0.3658(6) | 0.2223(6) | 0.0597(5)  | 0.015(3)               | 0.013(3)               | 0.009(3)               | 0.003(2)               | −0.001(2)              | −0.001(2)              |
| O(8) | 4e   | 0.4111(6) | 0.4171(6) | 0.2643(5)  | 0.014(3)               | 0.009(3)               | 0.009(3)               | 0.001(2)               | 0.001(2)               | 0.000(2)               |
| O(9) | 4e   | 0.5887(5) | 0.2059(6) | 0.2510(5)  | 0.004(3)               | 0.010(3)               | 0.014(3)               | 0.003(2)               | 0.001(2)               | 0.003(2)               |

**Acknowledgments.** We thank the Pinguin-Stiftung (Düsseldorf), the Fonds der Chemischen Industrie and the BMBF for supporting this work. We are indebted to Ms. Dipl.-Chem. U. Schmidt for the ICP-AES analyses and Ms. Dipl.-Ing. M. Eckert for the work at the scanning electron microscope.

## References

- Knip, R.; Engelhardt, H.; Hauf, C.: A First Approach to Borophosphate Structural Chemistry. *Chem. Mater.* **10** (1998) 2930-2934.
- Knip, R.; Boy, I.; Engelhardt, H.: RbFe[BP<sub>2</sub>O<sub>8</sub>(OH)]: Ein neues Borophosphat mit offen-verzweigten Vierer-Einfach-Tetraederketten. *Z. Anorg. Allg. Chem.* **625** (1999) 1512-1516.
- Engelhardt, H.; Knip, R.: Crystal structure of caesium iron(III) catena-[monohydrogenborate-bis(monophosphate)], CsFe[BP<sub>2</sub>O<sub>8</sub>(OH)] *Z. Kristallogr. NCS* **214** (1999) 443-444.
- Engelhardt, H.; Borrmann, H.; Schnelle, W.; Knip, R.: Das erste Vanadium(III)-Borophosphat: Darstellung und Kristallstruktur von CsV<sub>3</sub>(H<sub>2</sub>O)<sub>2</sub>[B<sub>2</sub>P<sub>4</sub>O<sub>16</sub>(OH)<sub>4</sub>] *Z. Anorg. Allg. Chem.*, publication in preparation.
- Sheldrick, G. M.: SHELXS-97 - Program for the Solution of Crystal Structures. University of Göttingen, Germany 1997.
- Sheldrick, G. M.: SHELXL-97 - Program for Crystal Structure Refinement. University of Göttingen, Germany 1997.
- Brandenburg, K.: Diamond Version 2.1c, 1996-1999 Crystal Impact GbR, Bonn, Germany.