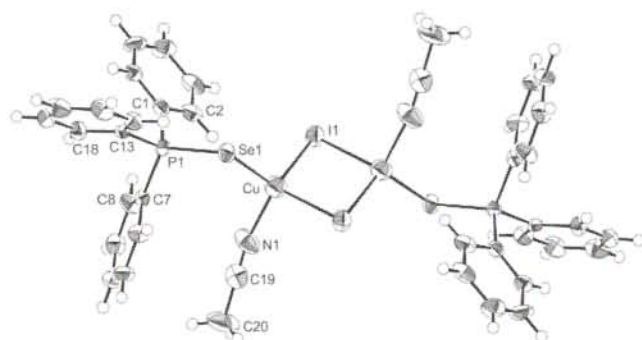


Crystal structure of di- μ -iodo-bis{(methylcyanide-*N*)(triphenylselenophosphorane)copper(I)}, [CuI(Ph₃PSe)(CH₃CN)]₂

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**Table 1.** Data collection and handling.

Crystal:	colorless block, size 0.08 × 0.13 × 0.19 mm
Wavelength:	Mo K α radiation (0.7107 Å)
μ :	45.37 cm ⁻¹
Diffractionmeter, scan mode:	Rigaku AFC6R, $\omega/2\theta$
$2\theta_{\max}$:	55°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	5143, 4921
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 3 \sigma(I_{\text{obs}})$, 2741
$N(\text{param})_{\text{refined}}$:	226
Programs:	DIRDIF [2], TEXSAN [3]

Abstract

C₂₀H₁₈CuINPSe, monoclinic, $P12_1/n1$ (No. 14), $a = 12.288(5)$ Å, $b = 9.517(8)$ Å, $c = 18.125(7)$ Å, $\beta = 109.26(3)^\circ$, $V = 2001.0$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.054$, $wR(F) = 0.059$, $T = 293$ K.

Source of material

CuI (0.100 g, 0.5 mmol) was suspended in a dry acetone/acetonitrile mixture (1:1 v/v), to it was added Ph₃PSe (0.180 g, 0.5 mmol) in ethanol and the resulting solution refluxed for 5 h. The clear solution was filtered and allowed to solidify. The creamish-yellow compound was recrystallised from an acetonitrile-ethanol mixture. The yield of colorless crystals was 70 % (mp 422 K – 423 K).

Discussion

The centrosymmetric, dinuclear structure comprises two μ_2 -iodo atoms that form almost symmetric Cu–I bonds, $d(\text{Cu}–\text{I}) = 2.646(2)$ Å and $d(\text{Cu}–\text{I}') = 2.700(2)$ Å. The remaining donor atoms in the distorted tetrahedral environment of the Cu center are a Se atom, $d(\text{Cu}–\text{Se}) = 2.459(2)$ Å, and an N, derived from an acetonitrile molecule, $d(\text{Cu}–\text{N}) = 2.02(1)$ Å. The range of tetrahedral angles is 102.0(4)° to 118.48(7)°, the largest being for I–Cu–Se. Related structures have been reviewed [1].

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(2)	4e	–0.0015	0.2621	0.1756	0.0466
H(3)	4e	0.1670	0.2720	0.2843	0.0487
H(4)	4e	0.1532	0.2871	0.4113	0.0549
H(5)	4e	–0.0269	0.2882	0.4240	0.0543
H(6)	4e	–0.1940	0.2828	0.3168	0.0381
H(8)	4e	–0.1003	0.5270	0.1726	0.0486
H(9)	4e	–0.1187	0.7370	0.1012	0.0499
H(10)	4e	–0.2705	0.7695	–0.0142	0.0554
H(11)	4e	–0.4119	0.5940	–0.0578	0.0482
H(12)	4e	–0.3907	0.3849	0.0119	0.0436
H(14)	4e	–0.4278	0.0799	0.1307	0.0379
H(15)	4e	–0.5909	0.0841	0.1656	0.0509
H(16)	4e	–0.6298	0.2711	0.2341	0.0556
H(17)	4e	–0.4960	0.4606	0.2683	0.0563
H(18)	4e	–0.3370	0.4606	0.2278	0.0411
H(20a)	4e	–0.0933	0.5632	–0.1327	0.0989
H(20b)	4e	–0.2033	0.4777	–0.1760	0.0989
H(20c)	4e	–0.2040	0.5825	–0.1106	0.0989

Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Cu	4e	–0.0788(1)	0.1232(2)	0.02169(9)	0.051(1)	0.043(1)	0.046(1)	0.0006(8)	0.0136(8)	0.0015(8)
I(1)	4e	0.13153(7)	0.04272(9)	0.10614(5)	0.0343(4)	0.0442(5)	0.0373(4)	0.0032(4)	0.0105(3)	–0.0058(4)
Se(1)	4e	–0.2358(1)	0.0905(1)	0.07617(7)	0.0395(7)	0.0264(6)	0.0321(6)	–0.0006(5)	0.0101(5)	–0.0075(5)

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Table 3. Continued.

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
P(1)	4e	−0.2383(2)	0.2695(3)	0.1474(2)	0.026(1)	0.021(2)	0.022(1)	−0.002(1)	0.007(1)	−0.004(1)
N(1)	4e	−0.0930(9)	0.317(1)	−0.0257(7)	0.045(7)	0.042(8)	0.09(1)	0.004(6)	0.028(7)	0.020(7)
C(1)	4e	−0.1153(9)	0.275(1)	0.2360(6)	0.030(6)	0.022(6)	0.032(6)	0.003(5)	0.012(5)	0.003(5)
C(2)	4e	−0.007(1)	0.269(1)	0.2266(6)	0.051(8)	0.022(7)	0.030(7)	−0.004(6)	−0.006(6)	−0.002(5)
C(3)	4e	0.0937(9)	0.274(1)	0.2912(7)	0.018(6)	0.039(7)	0.052(8)	0.003(5)	−0.004(6)	0.008(6)
C(4)	4e	0.086(1)	0.283(1)	0.3664(7)	0.054(8)	0.042(8)	0.029(7)	−0.007(7)	−0.003(6)	−0.002(6)
C(5)	4e	−0.021(1)	0.284(1)	0.3732(6)	0.060(9)	0.050(9)	0.020(6)	0.000(7)	0.006(6)	0.006(6)
C(6)	4e	−0.1211(9)	0.281(1)	0.3092(6)	0.028(6)	0.033(7)	0.030(6)	0.005(5)	0.004(5)	0.004(5)
C(7)	4e	−0.2439(8)	0.436(1)	0.0983(6)	0.015(5)	0.024(6)	0.029(6)	−0.002(4)	0.011(4)	0.000(5)
C(8)	4e	−0.163(1)	0.540(1)	0.1255(7)	0.039(7)	0.030(7)	0.047(7)	−0.002(6)	0.005(6)	0.001(6)
C(9)	4e	−0.174(1)	0.665(1)	0.0827(7)	0.042(7)	0.036(8)	0.047(8)	−0.005(6)	0.016(6)	−0.005(6)
C(10)	4e	−0.265(1)	0.685(1)	0.0147(7)	0.071(9)	0.029(7)	0.043(8)	0.005(7)	0.025(7)	0.010(6)
C(11)	4e	−0.348(1)	0.581(1)	−0.0117(6)	0.043(7)	0.046(8)	0.029(6)	0.007(6)	0.007(6)	0.012(6)
C(12)	4e	−0.3350(9)	0.457(1)	0.0303(6)	0.035(6)	0.036(7)	0.036(7)	−0.007(6)	0.007(5)	0.003(6)
C(13)	4e	−0.3667(9)	0.268(1)	0.1758(6)	0.031(6)	0.023(6)	0.018(5)	0.003(5)	0.007(5)	0.003(5)
C(14)	4e	−0.4425(9)	0.159(1)	0.1582(6)	0.029(6)	0.027(7)	0.038(7)	−0.005(5)	0.011(5)	0.002(5)
C(15)	4e	−0.539(1)	0.161(1)	0.1793(7)	0.044(8)	0.027(7)	0.054(8)	−0.009(6)	0.013(6)	0.003(6)
C(16)	4e	−0.562(1)	0.270(1)	0.2195(8)	0.034(7)	0.051(9)	0.056(9)	0.004(7)	0.017(6)	0.020(7)
C(17)	4e	−0.483(1)	0.384(1)	0.2390(7)	0.052(8)	0.053(9)	0.038(7)	0.025(7)	0.018(6)	0.010(7)
C(18)	4e	−0.3883(9)	0.383(1)	0.2162(6)	0.034(6)	0.033(7)	0.039(7)	0.006(5)	0.016(6)	0.002(6)
C(19)	4e	−0.121(1)	0.405(1)	−0.0692(8)	0.033(7)	0.045(9)	0.064(9)	−0.009(7)	0.019(7)	−0.001(8)
C(20)	4e	−0.159(1)	0.517(2)	−0.1272(9)	0.11(1)	0.05(1)	0.08(1)	0.00(1)	0.02(1)	0.030(9)

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