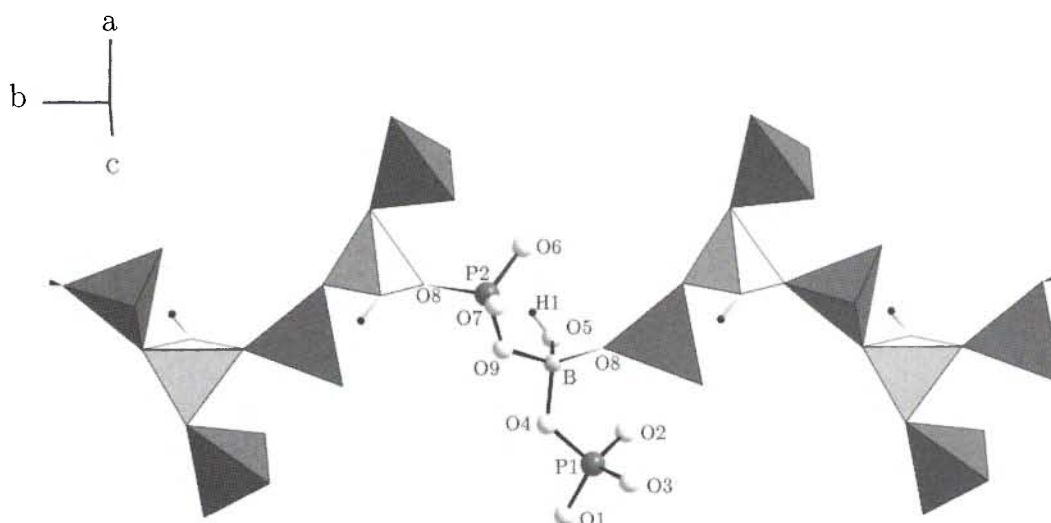


Crystal structure of caesium iron(III) *catena*-[monohydrogenmonoborate-bis(monophosphate)], CsFe[BP₂O₈(OH)]

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Abstract

BCsFeHO₉P₂, monoclinic, *P*12₁/*c*1 (No. 14), *a* = 9.351(2) Å, *b* = 8.631(4) Å, *c* = 9.763(2) Å, β = 102.58(2)°, *V* = 769.0 Å³, *Z* = 4, *R*_g(*F*) = 0.024, *wR*(*F*²) = 0.083, *T* = 293 K.

Source of material

CsFe[BP₂O₈(OH)] was prepared by hydrothermal reaction of 2.78 g of a 50 wt.% aqueous solution of CsOH, 2.51 g FeCl₃·6H₂O, 0.57 g H₃BO₃ and 2.14 g H₃PO₄ (85%) (molar ratio 1 : 1 : 1 : 2, conc. solution, pH 1) in teflon autoclaves at *T* = 433 K–443 K for six days.

Discussion

The crystal structure of CsFe[BP₂O₈(OH)] (hydrated borophosphate, B:P < 1; [1]) contains open-branched vierer-einfach chains [BP₂O₈(OH)]^{4–}, which are formed by alternating hydrogenborate and phosphate tetrahedra sharing common corners. The caesium-iron(III) compound is an isotype of RbFe[BP₂O₈(OH)][2]. Fe³⁺ is in an octahedral coordination (FeO₅(OH)) by oxygen sites of three neighbouring tetrahedral chains, while Cs⁺ is irregularly coordinated by ten oxygen sites. Bond lengths and angles within the anionic partial structure are consistent with related borophosphates (see [1, 2] and refs. therein).

Table 1. Data collection and handling.

Crystal:	amethyst-like (pale-violet), monoclinic prism, size 0.04 × 0.03 × 0.25 mm
Wavelength:	Mo K _α radiation (0.71073 Å)
μ:	70.68 cm ^{–1}
Diffractometer, scan mode:	Siemens P4, ω
2θ _{max} :	60°
<i>N</i> (<i>hkl</i>) _{measured} , <i>N</i> (<i>hkl</i>) _{unique} :	5058, 2237
Criterion for <i>I</i> _{obs} , <i>N</i> (<i>hkl</i>) _{gt} :	<i>I</i> _{obs} > 2 σ(<i>I</i> _{obs}), 2056
<i>N</i> (<i>param</i>) _{refined} :	131
Programs:	SHELXS-97 [3], SHELXL-97 [4]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1)	4e	0.382(9)	0.592(7)	0.025(8)	0.05(2)

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Table 3. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
Cs	4e	0.19523(2)	0.61015(3)	0.55238(2)	0.0247(2)	0.0199(2)	0.0218(2)	−0.00106(7)	0.0020(1)	−0.00393(7)
Fe	4e	0.20561(5)	0.15542(5)	0.42773(4)	0.0096(2)	0.0107(2)	0.0095(2)	−0.0003(1)	0.0014(2)	0.0004(1)
B	4e	0.3413(4)	0.5433(4)	0.1948(4)	0.010(1)	0.011(1)	0.014(1)	0.000(1)	−0.002(1)	−0.000(1)
P(1)	4e	0.07373(8)	0.42853(9)	0.20068(8)	0.0087(3)	0.0102(3)	0.0094(3)	−0.0005(2)	0.0013(2)	0.0007(2)
P(2)	4e	0.41972(8)	0.23297(9)	0.21319(8)	0.0083(3)	0.0098(3)	0.0111(3)	0.0007(2)	0.0013(2)	0.0005(2)
O(1)	4e	0.0748(2)	0.0050(3)	0.3137(2)	0.0097(9)	0.018(1)	0.018(1)	−0.0038(8)	0.0020(8)	−0.0053(8)
O(2)	4e	0.0863(3)	0.3325(3)	0.0735(2)	0.017(1)	0.018(1)	0.0102(9)	−0.0022(8)	0.0073(8)	−0.0029(8)
O(3)	4e	0.1011(3)	0.3333(3)	0.3358(2)	0.015(1)	0.014(1)	0.014(1)	0.0041(8)	0.0017(8)	0.0038(8)
O(4)	4e	0.1940(2)	0.5588(3)	0.2225(2)	0.0104(9)	0.013(1)	0.017(1)	−0.0010(7)	0.0040(8)	−0.0023(8)
O(5)	4e	0.3236(3)	0.5373(3)	0.0406(2)	0.018(1)	0.015(1)	0.0098(9)	−0.0040(8)	0.0018(8)	−0.0004(8)
O(6)	4e	0.3411(3)	0.1310(3)	0.2992(3)	0.016(1)	0.0123(9)	0.021(1)	−0.0012(8)	0.0085(9)	−0.0002(8)
O(7)	4e	0.3601(3)	0.2182(3)	0.0566(2)	0.019(1)	0.016(1)	0.0084(9)	0.0041(9)	−0.0005(8)	−0.0008(8)
O(8)	4e	0.4206(3)	0.4056(2)	0.2620(2)	0.016(1)	0.0098(9)	0.014(1)	0.0023(8)	−0.0008(8)	0.0004(7)
O(9)	4e	0.5845(2)	0.1857(3)	0.2455(3)	0.0092(9)	0.0114(9)	0.020(1)	0.0010(7)	0.0012(8)	0.0034(8)

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