

## Short Communication

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Molecular structure of *n*-tributylphosphine telluride

**Abstract:** The molecular of the title compound *n*-Bu<sub>3</sub>P<sub>2</sub>Te was established by X-ray crystallography and compared with the previously known trialkylphosphine tellurides.

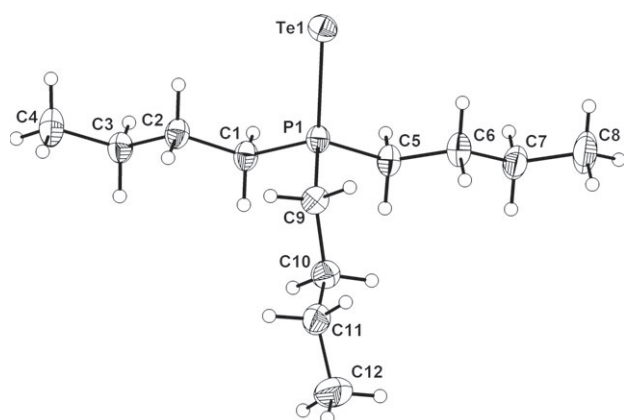
**Keywords:** chalcogenides; trialkylphosphine telluride; X-ray crystallography.

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The title compound *n*-Bu<sub>3</sub>P<sub>2</sub>Te was the first trialkylphosphine telluride ever prepared (Zingaro, 1963). Its characterization was achieved by <sup>125</sup>Te nuclear magnetic resonance and <sup>125</sup>Te Mößbauer spectroscopy (Jones and Sharma, 1987). In organometallic synthesis, *n*-Bu<sub>3</sub>P<sub>2</sub>Te has been utilized as a mild transfer reagent for atomic tellurium (Sasamori et al., 2005; Tajima et al., 2006). A literature search brought forward that only three trialkylphosphine

tellurides have been investigated by X-ray crystallography, namely, *t*-Bu<sub>3</sub>P<sub>2</sub>Te (Kuhn et al., 1987), *i*-Pr<sub>3</sub>P<sub>2</sub>Te (Kuhn et al., 1990), and Me<sub>2</sub>P(Te)CH<sub>2</sub>CH<sub>2</sub>P(Te)Me<sub>2</sub> (Steigerwald et al., 1994). Here, we report the molecular structure of *n*-Bu<sub>3</sub>P<sub>2</sub>Te, which is shown in Figure 1. Selected bond parameters are collected in the caption. Crystal and refinement data are listed in Table 1. The spatial arrangement of the P atom is distorted tetrahedral. The P–Te bond length of *n*-Bu<sub>3</sub>P<sub>2</sub>Te [2.368 (1) Å] compares well with those of *t*-Bu<sub>3</sub>P<sub>2</sub>Te [2.368 (4) Å] (Kuhn et al., 1987), *i*-Pr<sub>3</sub>P<sub>2</sub>Te [2.366 (1) Å] (Kuhn et al., 1990), and Me<sub>2</sub>P(Te)CH<sub>2</sub>CH<sub>2</sub>P(Te)Me<sub>2</sub> [2.357 (2) Å] (Steigerwald et al., 1994) and points to



**Figure 1** Molecular structure of *n*-Bu<sub>3</sub>P<sub>2</sub>Te showing 50% probability ellipsoids and the crystallographic numbering scheme. Selected bond parameters (Å) are as follows: P1–Te1 2.368 (1), P1–C1 1.817 (3), P1–C5 1.826 (3), P1–C9 1.828 (3), C1–P1–Te1 114.78 (9), C5–P1–Te1 113.9 (1), C9–P1–Te1 109.90 (9), C1–P1–C5 102.4 (1), C1–P1–C9 107.1 (1), C5–P1–C9 108.3 (1).

Formula	C <sub>12</sub> H <sub>27</sub> P <sub>2</sub> Te
Formula weight, g/mol	329.91
Crystal system	Triclinic
Crystal size, mm	0.40×0.40×0.40
Space group	<i>P</i> -1
<i>a</i> , Å	7.468 (2)
<i>b</i> , Å	9.503 (3)
<i>c</i> , Å	12.415 (4)
$\alpha$	105.56 (2°)
$\beta$	91.74 (2°)
$\gamma$	110.79 (3°)
<i>V</i> , Å <sup>3</sup>	785.8 (4)
<i>Z</i>	2
$\rho_{\text{calcd}}$ , Mg m <sup>-3</sup>	1.394
<i>T</i> , K	173
$\mu$ (Mo-K $\alpha$ ), mm <sup>-1</sup>	1.966
<i>F</i> (000)	332
$\theta$ range, degrees	2.50–27.49
Index ranges	–9 ≤ <i>h</i> ≤ 8 –11 ≤ <i>k</i> ≤ 11 –16 ≤ <i>l</i> ≤ 16
Number of reflections collected	4454
Completeness to $\theta_{\text{max}}$	98.4%
Number of independent reflections	3561
Number of observed reflections with ( <i>I</i> > 2 $\sigma$ ( <i>I</i> ))	3449
Number of refined parameters	130
GooF ( <i>F</i> <sup>2</sup> )	1.113
<i>R</i> <sub>1</sub> ( <i>F</i> ) ( <i>I</i> > 2 $\sigma$ ( <i>I</i> ))	0.0307
<i>wR</i> <sub>2</sub> ( <i>F</i> <sup>2</sup> ) (all data)	0.0808
( $\Delta/\sigma$ ) <sub>max</sub>	<0.001
Largest difference peak/hole, e Å <sup>-3</sup>	0.898/–1.479

**Table 1** Crystal data and structure refinement of *n*-Bu<sub>3</sub>P<sub>2</sub>Te.

a formal bond order of about 1.5 (Kuhn et al., 1987). The distortion is reflected by the mean Te–P–C and C–P–C angles [112.9(1)° and 105.9(1)°] that are in good agreement with those of Me<sub>2</sub>P(Te)CH<sub>2</sub>CH<sub>2</sub>P(Te)Me<sub>2</sub> [113.5(3)° and 105.2(4)°] (Steigerwald et al., 1994). In contrast, the mean Te–P–C and C–P–C angles of *t*-Bu<sub>3</sub>PTe [108.7(4)° and 110.2(5)°] (Kuhn et al., 1987) and *i*-Pr<sub>3</sub>PTe [110.9(1)° and 110.7(1)°] (Kuhn et al., 1990) are much closer to the ideal tetrahedral angle of 109.3°.

## Experimental

The title compound *n*-Bu<sub>3</sub>PTe was prepared by the reaction of *n*-Bu<sub>3</sub>P with tellurium powder (Zingaro, 1963). Single crystals were obtained by slow evaporation of a toluene solution under the inert conditions of a glove box.

## X-ray crystallography

Intensity data of *n*-Bu<sub>3</sub>PTe were collected on a Siemens P4 diffractometer with graphite-monochromated Mo-Kα (0.7107 Å) radiation. Data were reduced and corrected for absorption (Walker and Stuart, 1983). The structure was solved by direct methods and difference Fourier synthesis with subsequent full-matrix least-squares refinements on *F*<sup>2</sup>, using all data (Sheldrick, 2008). All nonhydrogen atoms were refined using anisotropic displacement parameters. Hydrogen atoms attached to carbon atoms were included in geometrically calculated positions using a riding model. Crystal and refinement data are collected in Table 1. Figures were created using DIAMOND (Brandenburg and Putz, 2006). Crystallographic data (excluding structure factors) for the structural analysis have been deposited with the Cambridge Crystallographic Data Centre, CCDC number 898395. Copies of this information may be obtained free of charge from The Director, CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (Fax: +44-1223-336033; e-mail: deposit@ccdc.cam.ac.uk or <http://www.ccdc.cam.ac.uk>).

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## References

- Brandenburg, K.; Putz, H. DIAMOND V3.1d. Crystal Impact GbR: Bonn, Germany, 2006.
- Jones, C. H. W.; Sharma, R. D. <sup>125</sup>Te NMR and Mößbauer spectroscopy of tellurium-phosphine complexes and tellurocyanates. *Organometallics*. **1987**, *6*, 1419–1423.
- Kuhn, N.; Schumann, H.; Wolmershäuser, G. (*tert*-C<sub>4</sub>H<sub>9</sub>)<sub>3</sub>PTe-ein typisches tellurophosporan? *Z. Naturforsch.* **1987**, *42b*, 674–678.
- Kuhn, N.; Henkel, G.; Schumann, H.; Fröhlich, R. Die Bindungsverhältnisse in Phosphantelluriden. Eine empirische NMR – Studie und die Kristallstruktur von (*iso*-C<sub>3</sub>H<sub>7</sub>)<sub>3</sub>PTe. *Z. Naturforsch.* **1990**, *45b*, 1010–1018.
- Sasamori, T.; Mieda, E.; Takeda, N.; Tokitoh, N. Telluradistibirane and telluradibismirane: three-membered heterocycles of heavier main group elements. *Angew. Chem. Int. Ed.* **2005**, *44*, 3717–3720.
- Sheldrick, G. M. A short history of SHELX. *Acta Crystallogr.* **2008**, *A64*, 112–122.
- Steigerwald, M. L.; Siegrist, T.; Gyorgy, E. M.; Hessen, B.; Kwon, Y. -U.; Tanzler, S. M. Effect of diverse ligands on the course of a molecules-to-solids process and properties on its intermediates. *Inorg. Chem.* **1994**, *33*, 3389–3395.
- Tajima, T.; Takeda, N.; Sasamori, T.; Tokitoh, N. A kinetically stabilized stannanetellone, a tin-tellurium double-bonded compound. *Organometallics*. **2006**, *25*, 3552–3553.
- Walker, N.; Stuart, D. An empirical method for correcting diffractometer data for absorption effects. *Acta Crystallogr.* **1983**, *A39*, 158–166.
- Zingaro, R. A. Tributylphosphine telluride. *J. Organomet. Chem.* **1963**, *1*, 200.