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Preparation of optically active 4-substituted γ-lactones by lipase-catalyzed optical resolution

Abstract: Optically active 4-substituted γ -lactones (**3** and **4**) were synthesized effectively using lipase-catalyzed optical resolution. *N*-methyl-4-hydroxyalkanamides (rac-**1a**-**i**) as substrates were prepared from *N*-methylsuccinimide. The alkylation of *N*-methylsuccinimide using Grignard reagents generated from various alkyl halides followed by reduction resulted in *N*-methyl-4-hydroxyalkanamides. The optical resolution of rac-**1a**-**g** was performed using Novozym 435-catalyzed stereoselective acetylation. The stereoselective preparation of 4-substituted γ -lactones (**3** and **4**) possessing various side chains such as isopentyl, phenyl, and phenethyl groups was achieved with more than 90% enantiopurity.

Keywords: enantioselective acetylation; enzymatic resolution; Grignard reaction; γ -lactone; organolithium reagent.

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Introduction

 γ -Lactones are well-known natural flavor and fragrance compounds [1–3], pheromone components [4–7], and useful building blocks [8–11] for pharmaceutical synthesis. These lactones are present in a wide variety of natural products, such as mango [12, 13], peach [14, 15], strawberry [16, 17], Gouda cheese [18, 19], and other dairy products [20, 21].

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Keita Inoue, Takeshi Osanai, Hayato Okabe and Tetsuo Miyakoshi: Department of Applied Chemistry, School of Science and Technology, Meiji University, 1-1-1 Higashi-mita, Tama-ku, Kawasaki 214-8571, Japan The preparation of optically active compounds has become very important for the development of new biologically active substances containing one or several chiral centers, considering that many chiral drugs [22, 23] and agrochemicals [24, 25] display quite different activity and toxicity profiles with respect to their absolute configuration. Chiral lactones are important components in the synthesis of natural products and biologically active compounds, such as antitumor, antidepressant, and antiviral agents. Ghosh et al. [26] synthesized (+)-cryptophycin 52, a potent antimitotic antitumor agent, by using chiral 4-phenyl- γ -butyrolactone as a building block. This γ -lactone was also used for the preparation of other biologically active compounds [27, 28]. In addition, Kotkar et al. [29] reported the synthesis of (+)-harzialactone starting with chiral 5-phenyl-γ-pentalactone. This lactone exhibits strong antitumor and cytotoxic activities against cultured P388 cells. Here we report the preparation of various chiral γ-lactones by optical resolution using lipase-catalyzed enantioselective acetylation.

Results and discussion

Preparation of N-methyl-4hydroxyalkanamides

We have previously reported the preparation of various *N*-methyl-4-hydroxyalkanamides from *N*-methylsuccinimide by the Grignard reaction and subsequent reductive reaction [30]. In this paper, the introduction of various functional groups was investigated (Scheme 1, Table 1). Alkylations with primary alkyl halides yielded the corresponding *N*-methyl-4-hydroxyalkanamides (*rac-***1a,e,f**) in the yields ranging from 61% to 79% (entries 1, 5, and 6). With aryl bromides, *N*-methyl-4-hydroxyalkanamides (*rac-***1c,d,g**) were synthesized with yields in the range of 87–94% (entries 3, 4, and 7). By contrast, the purification of *N*-methyl-4-hydroxyalkanamides (*rac-***1b,h,i**) obtained from secondary and tertiary alkyl bromides was not satisfactory, and the crude yields were very low (entries 2, 8,

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Scheme 1 Synthesis of N-methyl-4-hydroxyalkanamides with Grignard reagent.

and 9). These decreases in yield result from the side reactions such as the Wurtz reaction and olefin formation, and the low reactivity is caused by steric hindrance [31, 32]. Accordingly, the reactions with organolithium reagents were attempted (Scheme 2). The reactivity of organolithium reagents is higher than that of Grignard reagents and organozinc reagents [33]. Organolithium reagents can generally be used as nucleophiles. Many reports have been published about effective alkylation using organolithium reagents [34-39]. N-methyl-4-hydroxy-5,5-dimethylhexanamide (rac-**1b**) substituted with a tert-butyl group was synthesized with 44% yield using tert-butyllithium.

Lipase-catalyzed enantioselective acetylation of N-methyl-4-hydroxyalkanamides

We have previously reported detailed studies of lipasecatalyzed enantioselective acetylation [30, 40]. The lipase-catalyzed acetylation was performed in diethyl ether using *N*-alkyl-4-hydroxyalkanamides as substrates, vinyl acetate as acyl donor, and Novozym 435 as lipase. These conditions result in high enantioselectivity, and both enantiomers of various γ -lactones have successfully

Scheme 2 Synthesis of N-methyl-4-hydroxy-5,5-dimethylhexanamide rac-1b.

been obtained with more than 99% enantiopurity. In that previous work, lipase screening has been examined using Novozym 435 (immobilized from Candida antarctica), porcine pancreas lipase (PPL) (from porcine pancreas), lipase from pseudomonas fluorescens, immobilized (LPI) (immobilized from Pseudomonas fluorescens), and Lipozym IM (immobilized from *Mucor miehei*) in the acetylation of racemic N-methyl-4-hydroxyundecanamide that is similar in structure to rac-1. With a notable exception of Novozym 435, these lipases exhibit no reactivity toward racemic N-methyl-4-hydroxyundecanamide.

To determine the optimal amount of Novozym 435, the reaction was conducted in diethyl ether at room temperature with different ratios of Novozym 435-1.0 mmol of the substrate. It was concluded that 0.4 g is the optimal amount of Novozym 435-1.0 mmol of the substrate. The acetylation with Novozym 435 proceeds well for all substrates except rac-1b, although the time required to reach approximately 50% conversion varies for different starting materials (Scheme 3, Table 2). We have previously studied Novozym 435-catalyzed acetylation of racemic N-methyl-4-hydroxynonanamide [40]. The 50% conversion was reached in 2 h, and (R)- and (S)- γ -nonalactone were obtained with 98% and more than 99% enantiopurities, respectively. In this work, Novozym 435-catalyzed acetylation of rac-1a required 4 h despite the structural small difference between n-pentyl and isopentyl groups.

Table 1 Preparation of compounds *rac-***1a**–**i** (see Schemes 1 and 2).

Entry	R-X	Product	Yield (%)	Entry	R-X	Product	Yield (%)
1	→	rac- 1a	68	6	Br	rac-1f	61
2	× _{CI}	<i>rac</i> - 1b	10ª	7	Br	rac-1g	87
3	Br	rac- 1c	94	8	Br	<i>rac-</i> 1h	6ª
4	Br	<i>rac-</i> 1d	94	9	Br	rac-1i	10ª
5	CI	rac- 1e	79				

^aCrude product.

Me
$$\stackrel{\text{H}}{\longrightarrow}$$
 $\stackrel{\text{OH}}{\longrightarrow}$ $\stackrel{\text{OH}}{\longrightarrow}$

Scheme 3 Novozym 435-catalyzed stereoselective acetylation of *rac-***1a-g**.

Table 2 Lipase-catalyzed acetylation of rac-1^a and lactonization.

Entry	Substrate	Time (h)	Yield (%)		Yield (%)/enantiomeric excess (% e.e.) ^b / absolute configuration	
			1	2	3	4
1	rac-1a	4	47	47	>99/96/R	>99/>99/5
2 ^c	rac- 1b	24	No reaction		_	_
3	rac- 1c	12	44	51	>99/91/S	>99/94/R
4	rac- 1d	20	38	43	>99/95/ <i>S</i>	>99/93/R
5	rac- 1e	14	48	36	>99/>99/S	>99/95/R
6	rac-1f	7	34	46	>99/>99/R	>99/99/5
7	rac-1g	11	47	50	>99/racemic/-	>99/racemic/-

^aConditions: rac-1, 1.0 mmol; vinyl acetate, 2.0 mmol; Novozym 435, 0.4 g; Et₂O, 20 mL; room temperature.

In addition, (R)- and (S)-7-methyl- γ -octalactones (3a and 4a) derived from optically active 1 and 2 were obtained with 96% and more than 99% enantiomeric excesses. The enantiopurity of (R)-7-methyl- γ -octalactone was slightly low compared with that of (R)- γ -nonalactone. These results show that Novozym 435 exhibits not only low substrate affinity but also low substrate selectivity toward rac-1a, which has a sterically bulky R group. Therefore, a long reaction time can be predicted because the substrate possesses a sterically bulky side chain. An attempted acetylation failed for rac-1b substituted with a bulky tert-butyl group (entry 2). It appears that the substrate affinity of Novozym 435 is low for a substrate with a large steric hindrance around the asymmetric carbon atom. The reaction times for N-methyl-4-hydroxy-4-phenylbutanamide (rac-**1c**, entry 3) and *N*-methyl-4-hydroxy-4-*p*-tolylbutanamide (rac-1d, entry 4) are longer than those for rac-1a (entry 1) and N-methyl-4-hydroxy-6-phenylhexanamide (rac-1f, entry 6). By contrast, rac-1c with a phenyl group and rac-1d with a tolyl group are acetylated by Novozym 435. Phenyl and tolyl groups are more bulky than the isopentyl group; thus, a long reaction time is required to reach 50% conversion. In the case of *rac-***1c** and *rac-***1d**, both enantiomers of 4-phenyl-γ-butyrolactone (3c and 4c) and 4-(p-tolyl)-γbutyrolactone (3d and 4d) were obtained with more than 90% enantiopurities. These optical purities are approximately 5-10% lower than that of rac-1a. These results suggest that the substrate selectivity and the affinity of Novozym 435 toward the high steric hindrance substrate around an asymmetric carbon are low. Naoshima et al. [41] explained the enantioselectivity of lipase by using computer modeling. They measured the C-O distance between the carbonyl carbon atom of the acetyl group in the substrate and the oxygen atom at the active center of lipase. The large difference of the C-O distance among each enantiomer correlated with high enantioselectivity. It appears that for the large steric hindrance around asymmetric carbon atom, the hydroxy group in the substrate and the active center in Novozym 435 are difficult to be approached. It can be suggested that the enantioselectivity of Novozym 435 toward the substrate with large steric hindrance decreases. Although approximately 50% conversion was reached at 11 h for N-methyl-4-hydroxy-4-(p-anisyl)butanamide (rac-1g) with a p-anisyl group, both 4-(p-anisyl)- γ -butyrolactones (3g and 4g) exhibit racemic similarities (entry 7). Phenyl, tolyl, and p-anisyl

^bDetermined by GC using a Chirasil-Dex CB column.

^cAcetylation conducted at 40°C.

groups are structurally similar. However, only the reaction of rac-1g gave racemic γ -lactones (3g and 4g), and this result can be attributed to the presence of the methoxy group. The reaction times required for rac-1c and rac-1d to reach 50% conversion by Novozym 435-catalyzed acetylation were 12 and 20 h, respectively, and that for rac-1g took 11 h. These substrates require comparably long reaction time to reach 50% conversion compared with rac-1a and rac-1f. As mentioned earlier, it was assumed that Novozym 435 shows higher substrate affinity toward the small R group in substrates such as rac-1a and 1f. The reaction time of *rac-***1g** was similar to *rac-***1c**. These results also show that Novozym 435 exhibits low affinity toward the bulky substrates around the asymmetric carbon atom. By contrast, it can be suggested that the high substrate affinity of Novozym 435 is due to hydrogen bonding between the oxygen at the methoxy group and the amino acid residues that constitute the lipase. Both enantiomers of rac-1g can be incorporated into the active site by this hydrogen bonding, which causes the observed lack of enantioselectivity. Novozym 435-catalyzed acetylation of all *rac-***1** except *rac-***1g** progressed with more than 90% enantioselectivity. The stereoselectivity of Novozym 435 varies with structural differences of the R group. The absolute configurations of γ -lactones prepared from hydroxyamide 1 are (R)-form for rac-1a with an isopentyl group and rac-1f with a phenethyl group. By contrast, the corresponding γ -lactones derived from 1 have (S)-configuration in the case of rac-1c with a phenyl group, rac-1d with a tolyl group, and rac-1e with a benzyl group. As shown in Scheme 3, the hydroxyl group is acetylated by Novozym 435 in all substrates. Previously, we have reported a synthetic methodology of the chiral γ -lactones, which combines Novozym 435-catalyzed reaction and Mitsunobu reaction [42, 43]. In this work, Novozym 435-catalyzed stereoselective hydrolysis of N-benzyl-4-acetoxyalkanamides was conducted. The hydrolysis was conducted at 60°C in diisopropyl ether; the reaction progressed with more than 90% enantioselectivity. However, the 50% conversion

was reached after a relatively long period of 24–30 h. In summary, Novozym 435 shows high enantioselectivity for both acetylation and hydrolysis.

Lipase-catalyzed enantioselective hydrolysis of *N*-methyl-4-acetoxyalkanamides

In the acetylation using Novozym 435, rac-1b was inert (Table 2, entry 2). Enantioselectivity was not observed for rac-1g, although approximately 50% conversion was reached in 11 h (Table 2, entry 7). Novozym 435-catalyzed hydrolysis of *rac-2* was investigated to prepare optically active lactones **3b**, **g** and **4b**, **g** (Scheme 4, Table 3). In all cases, the time required for the hydrolysis to reach approximately 50% conversion is much longer than that for acetylation. Novozym 435 is inert toward the hydrolysis of rac-2b as well as the acetylation of rac-1b (entry 2). It appears that the bulky tert-butyl group is not compatible with the substrate specificity of Novozym 435. Although rac-2g is hydrolyzed, the obtained lactone **3g/4g** is racemic (entry 7). Because rac-**2c** and rac-**2d** are hydrolyzed enantioselectively, it can be suggested that the anisyl group of rac-2g is a factor that does not promote enantioselectivity. The reaction mechanism of lipase was reported [44, 45].

Conclusions

N-methyl-4-hydroxyalkanamides with various side chains were prepared in a high yield by the addition reaction of organometallic reagents with N-methylsuccinimide. Novozym 435-catalyzed acetylation of all rac-1 except rac-1b and rac-1g the respective products 2 with more than 90% enantioselectivity. Both enantiomers of γ -lactones with various side chains were successfully synthesized with enantiopurities more than 90%.

Me
$$\stackrel{\text{H}}{\longrightarrow}$$
 $\stackrel{\text{OH}}{\longrightarrow}$ $\stackrel{\text{OH}}{\longrightarrow}$

Scheme 4 Novozym 435-catalyzed stereoselective hydrolysis of rac-2a-g.

Table 3 Lipase-catalyzed hydrolysis of rac-2^a and lactonization.

Entry	Substrate	Time (h)	Yield (%)		Yield (%)/enantiomeric excess (% e.e.) ^b / absolute configuration	
			1	2	4	3
1	rac- 2a	33	41	44	>99/98/5	>99/79/R
2	rac- 2b	48	No reaction		_	_
3	rac- 2c	48	37	56	>99/91/R	>99/72/S
4	rac- 2d	48	43	50	>99/92/R	>99/75/S
5	rac- 2e	48	42	53	>99/95/R	>99/70/S
6	rac- 2f	48	44	46	>99/91/5	>99/96/R
7	rac- 2g	48	49	43	>99/racemic/-	>99/racemic/-

^aConditions: rac-2, 1.0 mmol; MeOH, 3.0 mmol; Novozym 435, 0.4 g; Et₂O, 20 mL; 40°C.

Experimental

Column chromatography was conducted with silica gel FL60D (Fuji Silvsia Chemical Ltd., Aichi, Japan). Thin-layer chromatography was performed with silica gel F-254 on aluminum plates (Merck Ltd., Darmstadt, Germany). 1H NMR (500 MHz) and 13C NMR (126 MHz) spectra were recorded in CDCl, on a JNM-ECA-500 spectrometer (JEOL, Tokyo, Japan), using an internal standard of tetramethylsilane and the central peak of CDCl₃ (77 ppm). Near-infrared spectra were recorded in KBr pellets on an FT-IR 460plus spectrophotometer (JASCO Corp., Tokyo, Japan). Enantiomeric excesses of γ-lactones were determined using a Perkin Elmer Auto System XL gas chromatograph equipped with a chiral capillary column CycloSil B (30 m \times 0.25 mm i.d. \times 0.25 μ m film thickness; Agilent Technologies, Santa Clara, CA, USA). The carrier gas was helium. Optical rotations were measured on a Jasco P-1010 spectropolarimeter (JASCO Corp.), and the reported data refer to the Na-line value using a 25 mL cuvette. High-resolution mass spectral (HRMS) analyses were performed on an AccuTOF GCv 4G instrument (JEOL, Tokyo, Japan). Novozym 435 immobilized lipase from C. antarctica was obtained as a gift from Novozymes A/S (Paraná, Brazil).

Preparation of racemic N-methyl-4-hydroxyalkanamides rac-1a,c-g

Organomagnesium reagents were freshly prepared by the slow addition of the corresponding bromides or chlorides (16.5 mmol) in Tetrahydrofuran (THF) (50 mL) onto magnesium turnings (0.37 g, 15.0 mmol) previously activated with a crystal of iodine. Under vigorous stirring and cooling with an ice bath, N-methylsuccinimide (1.13 g, 10.0 mmol) dissolved in THF (30 mL) was then added to the suspension, and the mixture was stirred for 8 h at room temperature. The residual Grignard reagent was hydrolyzed by gradual addition of ice (10 g) and saturated aqueous NH, Cl (100 mL). The organic phase was separated, and the aqueous phase was extracted with CHCl, (4×50 mL). The combined extracts were washed with water and dried with MgSO₄. The solvent was evaporated under reduced pressure, and the residue was not purified. NaBH, (0.76 g, 20 mmol) in methanol (20 mL) was added to the crude product with stirring, and the mixture was stirred for 1 h at room temperature. The solvent was evaporated under reduced pressure, and water (50 mL) was added. The aqueous phase was extracted with CHCl, (4×50 mL). The combined extracts were washed with water and dried with MgSO. The solvent was evaporated, and the residue was purified by flash chromatography on silica eluting with EtOAc to give the corresponding N-methyl-4-hydroxyalkanamide 1.

N-methyl-4-hydroxy-7-methyloctanamide (1a) Yield 68%; colorless solid; mp 62–63°C (dec); R_r =0.33 (eluent: CHCl₃-MeOH, 10:1, v/v); IR: 3293 (N-H, O-H), 2954 (-CH₂), 2934 (-CH₂-), 2871 (-CH₃), 2846 (-CH₂-), 1645 cm⁻¹ (-NH<u>C=O</u>); ¹H NMR: δ 0.88 (d, 6H, J = 6.9 Hz, -CH(C<u>H</u>,)), 1.19 (m, 1H, -CH,CH(CH,)), 1.32 (m, 1H, -CH,CH(CH,)), 1.46 (m, 2H, 1.66 (m, 1H, -NHC(=0)CH₂CH₂-), 1.87 (m, 1H, -NHC(=0)CH₂CH₂-), 2.36 (m, 2H, -NHC(=0)C \underline{H}_{3} -), 2.81 (d, 3H, J = 4.6 Hz, -NHC \underline{H}_{3}), 2.99 (br, 1H, -OH), 3.60 (m, 1H, -CH(OH)-), 5.79 (br, 1H, -NH-); ¹³C NMR: δ 22.6, 22.6 (-CH(CH₂)₂), 26.4 (-NHCH₂), 28.1 (-CH(CH₂)₂), 32.5 (-CH₂-), 33.2 (-C(=0)CH₂-), 34.8 (-CH₂CH-), 35.6 (-CH₂-), 71.8 (-CHOH), 174.4 (-NH<u>C</u>(=0)-). HRMS (FI). Calcd for $C_{10}H_{22}NO_{2}$ (M+H)+: m/z 188.1651. Found: m/z 188.1645.

N-methyl-4-hydroxy-4-phenylbutanamide (1c) Yield colorless oil; R_i=0.35 (eluent: CHCl₃-MeOH, 10:1, v/v); IR: 3328 (O-H, N-H), 3109 (Ar, C-H), 2937 (CH₂), 2877 (CH₂), 1649 (-NC=O), 1495, 1450 (Ar, C=C), 760, 702 cm⁻¹ (Ar, C-H); ¹H NMR: δ 2.02 (m, 2H, -NHC(=O) $CH_{2}CH_{3}$ -), 2.30 (t, 2H, J = 6.9 Hz, -NHC(=0) CH_{3} -), 2.75 (d, 3H, J = 4.6Hz, -NHCH3), 4.28 (br, 1H, -OH), 4.73 (m, 1H, -CH(OH)-), 6.02 (br, 1H, -NH-), 7.22-7.36 (m, 5H, -Ph); 13 C NMR: δ 26.4 (-NHCH₂), 32.7 $(-C(=0)CH_{,-})$, 34.3 $(-CH_{,-})$, 73.5 (-CHOH), 125.7, 127.3, 128.3, 144.4 (-Ph), 174.3 (-NH \underline{C} (=0)-). HRMS (FD). Calcd for C₁₁H₁₆NO₂ (M)⁺: m/z 193.1103. Found: *m/z* 193.1096.

N-methyl-4-hydroxy-4-(p-tolyl)butanamide (1d) Yield 94%; colorless oil; R_e =0.24 (eluent: CHCl₃-MeOH, 10:1, v/v); IR: 3321 (O-H, N-H), 3039 (Ar, C-H), 2937 (CH₂), 2905 (CH₂), 1645 (-NC=O), 1568, 1515 (Ar, C=C), 816 cm⁻¹ (Ar, C-H); ¹H NMR: δ 1.96 (m, 2H, -NHC(=O)CH, CH₂-), 2.27 (t, 2H, J = 6.9 Hz, -NHC(=0)CH₂-), 2.31 (s, 3H, -PhCH₂), 2.70 (d, 3H, J =4.6 Hz, -NHC \underline{H}_{2}), 4.55 (br, 1H, -O \underline{H}), 4.64 (q, 1H, J = 3.7, 4.1 Hz, -C \underline{H} (OH)-), 6.43 (br, 1H, -NH-), 7.10 (d, 2H, J = 7.8 Hz, -Ph-), 7.19 (d, 2H, J = 8.2 Hz, -Ph-); ¹³C NMR: δ 21.0 (-Ph<u>C</u>H₃), 26.2 (-NH<u>C</u>H₃), 32.7 (-<u>C</u>H₃-), 34.5 (-<u>C</u>H₃-), 73.2 (-CHOH), 125.6, 128.9, 136.7, 141.4 (-Ph-), 174.4 (-NHC(=0)-). HRMS (FD). Calcd for $C_{12}H_{18}NO_{2}$ (M)+: m/z 207.1259. Found: m/z 207.1257.

^bDetermined by GC using a CycloSil B column.

N-methyl-4-hydroxy-5-phenylpentanamide (1e) Yield 79%; colorless solid; mp 100-101°C (dec); R_e=0.30 (eluent: CHCl₂-MeOH, 10:1, v/v); IR: 3412, 3370 (N-H), 3285 (O-H), 3094, 3058, 3033 (Ar, C-H), 2944 (-CH₃), 2926 (-CH₂-), 1651 (-NH<u>C=O</u>), 1495, 1454 cm⁻¹ (Ar, C=C); 1 H NMR: δ 1.69 (m, 1H, -NHC(=0)CH,CH,-), 1.89 (m, 1H, -NHC(=0) $CH_{2}CH_{2}$ -), 2.34 (t, 2H, J = 6.9 Hz, $-NHC(=0)CH_{2}$ -), 2.73–2.85 (m, 5H, -CH,Ph, -NHCH,), 3.21 (br, 1H, -OH), 3.84 (m, 1H, -CH(OH)-), 5.92 (br, 1H, -NH-), 7.17–7.35 (m, 5H, -Ph); 13 C NMR: δ 26.3 (-NHCH₂), 31.8 (-CH₂-), 33.1 (-C(=0)CH₂-), 44.2 (-CH₂Ph), 72.3 (-CHOH), 126.4, 128.5, 129.4, 138.4 (-Ph), 174.2 (-NHC(=0)-). HRMS (FD). Calcd for C, H, NO $(M+H)^+$: m/z 208.1338. Found: m/z 208.1305.

N-methyl-4-hydroxy-6-phenylhexanamide (1f) Yield 61%; colorless solid; mp 40–41°C (dec); R_{ϵ} =0.29 (eluent: CHCl₂-MeOH, 10:1, v/v); IR: 3315 (N-H, O-H), 3085, 3064, 3029 (Ar, C-H), 2950 (-CH₂), 2929 (-CH₂-), 2878 (-CH₂), 2861 (-CH₂-), 1645 (-NHC=O), 1497, 1456 cm⁻¹ (Ar, C=C); ¹H NMR: δ 1.74 (m, 2H, -CH,CH,Ph), 1.86 (m, 2H, -NHC(=0)CH,CH,-), 2.36 $(m, 2H, -NHC(=0)CH_{-}), 2.68-2.88 (m, 2H, -CH_{2}Ph), 2.80 (d, 3H, J = 4.6)$ Hz, -NHCH₂), 3.23 (br, 1H, -OH), 3.66 (m, 1H, -CH(OH)-), 5.70 (br, 1H, -NH-), 7.13–7.36 (m, 5H, -Ph); ¹³C NMR: δ 26.4 (-NHCH₂), 32.1 (-CH₂Ph), 32.5 (- $\underline{C}H_2$ -), 33.1 (- $C(=0)\underline{C}H_2$ -), 39.4 (- $\underline{C}H_2$ -), 70.8 (- $\underline{C}HOH$), 125.8, 128.3, 128.4, 128.6, 129.0, 142.1 (-Ph), 174.3 (-NHC(=0)-). HRMS (FD). Calcd for $C_{12}H_{20}NO_{2}(M)^{+}$: m/z 221.1416. Found: m/z 221.1405.

N-methyl-4-hydroxy-4-(p-anisyl)butanamide (1g) Yield 87%; colorless oil; R_c=0.24 (eluent: CHCl_c-MeOH, 10:1, v/v); IR: 3313 (O-H, N-H), 3106 (Ar, C-H), 2938 (CH₂), 2836 (CH₂), 1645 (-NC=O), 1563, 1513 (Ar, C=C), 834 cm⁻¹ (Ar, C-H); ¹H NMR: δ 2.05 (m, 2H, -NHC(=O) CH_2CH_3 -), 2.32 (t, 2H, J = 6.9 Hz, -NHC(=0) CH_3 -), 2.81 (d, 3H, J = 4.6 Hz, -NHCH₂), 3.61 (br, 1H, -OH), 3.80 (s, 3H, -PhOCH₂), 4.72 (t, 1H, J = 6.0Hz, $-C\underline{H}(OH)$ -), 5.67 (br, 1H, $-N\underline{H}$ -), 6.87 (d, 2H, J = 8.2 Hz, -Ph-), 7.28 (d, 2H, J = 9.6 Hz, -Ph-); ¹³C NMR: $\delta 26.4$ (-NHCH₃), 32.9 (-C(=0)CH₃-), 34.3 (-CH,-), 55.3 (-PhOCH,), 73.2 (-CHOH), 113.7, 126.9, 136.6, 158.9 (-Ph), 174.1 (-NH \underline{C} (=O)-). HRMS (FD). Calcd for $C_{12}H_{18}NO_3$ (M)+: m/z 223.1208. Found: m/z 223.1207.

Preparation of N-methyl-4-hydroxy-5,5-dimethylhexanamide (1b)

A solution of tert-butyllithium in pentane (6.29 mL, 10.0 mmol) was added dropwise to a solution of N-methylsuccinimide (1.13 g. 10.0 mmol) in THF (20 mL) at -78°C under argon atmosphere, and the mixture was stirred at the same temperature for 2 h. The reaction mixture was poured onto saturated NH₂Cl at 0°C and extracted with EtOAc (5×50 mL). The combined organic layers were washed with water, dried with MgSO, and concentrated in vacuo. The crude product was not purified. NaBH, (0.76 g, 20.0 mmol) was added to a solution of the crude product in MeOH (20 mL) at 0°C , and the mixture was stirred for 1 h. The solvent was evaporated, and water (50 mL) was added to the residue. The aqueous phase was extracted with EtOAc (5×50 mL), and the combined organic layers were washed with water, dried with MgSO, and concentrated in vacuo. The crude product was purified by crystallization from *n*-hexane to give N-methyl-4-hydroxy-5,5-dimethylhexanamide (1b, 0.76 g, 44%) as a colorless solid; mp 117-118°C (dec); R_e=0.30 (eluent: CHCl_e-MeOH, 10:1, v/v); IR: 3292 (N-H, O-H), 2954 (-CH₃), 2932 (-CH₃-), 2868 (-CH₃), 2846 (-CH₂-), 1645 cm⁻¹ (-NH<u>C=O</u>); ¹H NMR: δ 0.91 (s, 9H, -C(C<u>H₂)₃</u>), 1.60 (m, 1H, -NHC(=0)CH₂C \underline{H}_{2} -), 1.89 (m, 1H, -NHC(=0)CH₂C \underline{H}_{2} -),

2.37 (m, 2H, -NHC(=0)C \underline{H}_3 -), 2.63 (br, 1H, -O \underline{H}), 2.82 (d, 3H, J=4.6Hz, -NHCH₂), 3.20 (m, 1H, -CH(OH)-), 5.66 (br, 1H, -NH-); ¹³C NMR: δ 25.6 (-C(CH₂)₃), 26.4 (-NHCH₃), 27.0 (-CH₂-), 34.1 (-C(=0)CH₂-), 35.0 $(-\underline{C}(CH_3)_3)$, 79.7 $(-\underline{C}HOH)$, 174.6 $(-NH\underline{C}(=O)-)$. HRMS (ESI). Calcd for $C_0H_{20}NO_2(M+H)^+$: m/z 174.1494. Found: m/z 174.1513.

General procedure for Novozym 435-catalyzed acetylation and lactonization

Novozym 435 (0.4 g) was added to a racemic mixture of N-methyl-4-hydroxyalkanamides (1, 1.0 mmol) and vinyl acetate (2.0 mmol) in diethyl ether (20 mL). After stirring at room temperature for a period specified in Table 2, the reaction mixture was filtered. Concentration under a reduced pressure followed by flash chromatography of the residue on silica gel (eluent: EtOAc) afforded the corresponding optically active N-methyl-4-hydroxyalkanamide 1a-g and N-methyl-4-acetoxyalkanamide 2a-g. Then NaOH (2.0 g) was added to a solution of 1a-g or 2a-g in methanol (20 mL), and the mixture was heated under reflux for 3 h. After cooling, methanol was removed under reduced pressure and water (50 mL) was added. The aqueous phase was acidified with 10% HCl to pH 3.0, and the mixture was stirred for 8 h then extracted with EtOAc (4×20 mL). The combined organic layers were washed with water, dried with MgSO,, and concentrated under reduced pressure. The residue was purified by flash chromatography eluting with n-hexane-EtOAc, 4:1, to give the corresponding γ-lactone (**3** or **4**) as a colorless oil. The enantiomeric excess was measured by using the chiral GC analysis. The absolute configurations of 3 and 4 were determined by optical activity compared with the literature data.

N-methyl-4-acetoxy-7-methyloctanamide (2a) Colorless $R_{e}=0.33$ (eluent: CHCl₂-MeOH, 10:1, v/v); IR: 3298 (N-H), 2957 (-CH₂), 2933 (-CH₂-), 2871 (-CH₂), 1738 (-OC=O), 1649 (-NHC=O), 1242 cm⁻¹ (-OC=0); H NMR: δ 0.87 (d, 6H, J = 6.3 Hz, $-CH(CH_3)$), 1.18 (m, 2H, -CH,CH,CH(CH,),), 1.53 (m, 3H, -CH,CH,CH,CH,CH,), 1.85 (m, 1H, $-NHC(=O)CH_1CH_2$ -), 1.94 (m, 1H, $-NHC(=O)CH_1CH_2$ -), 2.06 (s, 3H, $-OC(=O)CH_{2}$, 2.18 (m, 2H, -NHC(=O)CH₂-), 2.80 (d, 3H, J = 4.6 Hz, -NHCH₂), 4.84 (quint, 1H, I = 3.4, 9.2 Hz, -CHOC(=0)CH₂), 5.87 (s, 1H, $-N\underline{H}$ -); ¹³C NMR: δ 21.2 (-C(=0) \underline{C} H₃), 22.4, 22.5 (-CH(\underline{C} H₃)₂), 26.3 $(-NHCH_3)$, 27.8 $(-CH_3CH(CH_3)_3)$, 30.2 $(-CH_3-)$, 32.1 $(-C(=0)CH_3-)$, 32.5 $(-\underline{CH}_{3})$, 34.3 $(-\underline{CH}_{3})$, 74.1 $(-\underline{CHOC}(=0)CH_{3})$, 171.3 $(-\underline{OC}(=0)$ CH₃), 172.9 (-NH \underline{C} (=O)-). HRMS (FI). Calcd for C₁₂H₂₄NO₃ (M)⁺: m/z229.1678. Found: *m/z* 229.1662.

N-methyl-4-acetoxy-4-phenylbutanamide (2c) Colorless oil; R_{r} =0.33 (eluent: CHCl₃-MeOH, 10:1, v/v); IR: 3302 (N-H), 3035 (Ar, C-H), 2941 (CH₂, CH₂), 1736 (-O<u>C=O</u>), 1651 (-NH<u>C=O</u>), 1556, 1493 (Ar, C=C), 1240 (C-O-C), 760, 700 cm⁻¹ (Ar, C-H); ¹H NMR: δ 2.05 (s, 3H, $-OC(=O)CH_{2}$, 2.16 (m, 2H, -NHC(=O)CH₂CH₂-), 2.24 (m, 2H, -NHC(=O) $C\underline{H}_{2}$ -), 2.76 (d, 3H, J = 4.6 Hz, -NHC \underline{H}_{3}), 5.75 (m, 1H, -C \underline{H} OC(=0)CH₂), 7.20–7.41 (m, 5H, -Ph); 13 C NMR: δ 21.2 (-C(=0) $\underline{\text{CH}}_3$), 26.3 (-NH $\underline{\text{CH}}_3$), 32.0 $(-\underline{CH}_{2}^{-})$, 32.3 $(-C(=0)\underline{CH}_{2}^{-})$, 75.3 $(-\underline{CHOC}(=0)CH_{3}^{-})$, 126.3, 128.0, 128.5, 139.8 (-Ph), 170.5 (-OC(=O)CH₂), 172.6 (-NHC(=O)-). HRMS (FD). Calcd for $C_{13}H_{18}NO_3$ (M)+: m/z 235.1208. Found: m/z 235.1194.

N-methyl-4-acetoxy-4-(p-tolyl)butanamide (2d) Colorless oil; R_{e} =0.33 (eluent: CHCl₂-MeOH, 10:1, v/v); IR: 3300 (N-H), 3029 (Ar, C-H) 2943 (CH₂, CH₂), 1738 (-OC=O), 1651 (-NHC=O), 1556, 1520 (Ar, C=C),

1240 (C-O-C), 818 cm⁻¹ (Ar, C-H); ¹H NMR: δ 2.02 (s, 3H, -OC(=0)CH₂), 2.13 (m, 2H, -NHC(=0)CH,CH,-), 2.24 (m, 2H, -NHC(=0)CH,-), 2.31 (s, 3H, -PhC \underline{H}_3), 2.72 (d, 3H, J = 4.6 Hz, -NHC \underline{H}_3), 5.69 (m, 1H, -C \underline{H} OC(=0) CH_{3}), 6.18 (br, 1H, -N \underline{H} -), 7.12 (d, 2H, J = 7.8 Hz, -Ph-), 7.19 (d, 2H, J = 7.8 Hz, -Ph-); 13 C NMR: δ 20.9 (-OC(=0)CH₂), 21.0 (-PhCH₂), 26.1 (-NHCH₂), $31.7 (-C(=0)CH_{2}), 32.1 (-CH_{2}), 75.1 (-CHOC(=0)CH_{3}), 126.2, 129.0, 136.7,$ 137.6 (-Ph-), 170.3 (-OC(=O)CH₂), 172.7 (-NHC(=O)-). HRMS (FD). Calcd for $C_{14}H_{20}NO_{2}$ (M)+: m/z 249.1365. Found: m/z 249.1368.

N-methyl-4-acetoxy-5-phenylpentanamide (2e) Colorless oil; R_c =0.33 (eluent: CHCl₃-MeOH, 10:1, v/v). IR: 3308 (N-H), 3086, 3062, 3029 (Ar, C-H), 2938 (-CH₂), 1735 (-O<u>C</u>=<u>O</u>), 1646 (-NH<u>C</u>=<u>O</u>), 1496, 1455 (Ar, C=C), 1242 cm⁻¹ (-O<u>C=O</u>); ¹H NMR: δ 1.88 (m, 1H, -NHC(=O) $CH_{2}C\underline{H}_{3}$ -), 2.13 (m, 1H, -NHC(=0)CH₂CH₃-), 1.98 (s, 3H, -OC(=0)CH₃), 2.21 (m, 2H, -NHC(=0)C \underline{H}_3 -), 2.77 (d, 3H, J = 4.6 Hz, -NHC \underline{H}_3), 2.87 (m, 2H, -CH,Ph), 5.07 (m, 1H, -CHOC(=0)CH,), 5.79 (s, 1H, -NH-), 7.15-7.39 (m, 5H, -Ph); ¹³C NMR: $\delta 21.0 (-OC(=0)CH_2)$, $26.2 (-NHCH_2)$, $29.7 (-CH_2)$, 32.6 (-C(=0)CH₂-), 40.7 (-CH₂Ph), 74.4 (-CHOC(=0)CH₂), 126.5, 128.3, 129.3, 137.0 (-Ph), 170.9 (-OC(=O)CH₂), 172.7 (-NHC(=O)-). HRMS (FD). Calcd for $C_{1/4}H_{20}NO_3$ (M+H)+: m/z 250.1443. Found: m/z 250.1444.

N-methyl-4-acetoxy-6-phenylhexanamide (2f) Colorless oil; R_c =0.33 (eluent: CHCl₂-MeOH, 10:1, v/v); IR: 3297 (N-H), 3087, 3062, 3026 (Ar, C-H), 2943, 2864 (-CH₃), 1736 (-OC=O), 1644 (-NHC=O), 1496, 1454 (Ar, C=C), 1242 cm⁻¹ (-OC=O); ¹H NMR: δ 1.79–2.01 (m, 4H, -CH₂CH(OC(=0)CH₂)CH₂-), 2.04 (s, 3H, -OC(=0)CH₂), 2.19 (m, 2H, $-NHC(=O)CH_{2}$ -), 2.63 (m, 2H, $-CH_{2}$ Ph), 2.78 (d, 3H, J = 4.6 Hz, $-NHCH_{2}$), 4.92 (m, 1H, -CHOC(=0)CH₂), 5.94 (s, 1H, -NH₂), 7.11–7.35 (m, 5H, -Ph); 13 C NMR: δ 21.1(-OC(=0)CH₂), 26.3(-NHCH₂), 30.2(-CH₂Ph), 31.6(-CH₂-), 32.4 (-C(=0)CH₂-), 35.9 (-CH₂CH₂Ph), 73.5 (-CHOC(=0)CH₂), 125.9, 128.2, 128.4, 141.8 (-Ph), 171.3 ($-OC(=O)CH_{2}$), 173.0 (-NHC(=O)-). HRMS (FD). Calcd for $C_{15}H_{22}NO_3$ (M)+: m/z 263.1521. Found: m/z 263.1517.

N-methyl-4-acetoxy-4-(p-anisyl)butanamide (2g) Colorless oil; R_{e} =0.33 (eluent: CHCl₂-MeOH, 10:1, v/v); IR: 3306 (N-H), 3094 (Ar, C-H), 2938 (CH₂), 2838 (CH₂), 1736 (-OC=O), 1648 (-NHC=O), 1613, 1516 (Ar, C=O), 1241 (C-O-C), 832 cm⁻¹ (Ar, C-H); 1 H NMR: δ 2.05 (s, 3H, $-OC(=O)CH_{2}$, 2.14 (m, 2H, -NHC(=O)CH₂CH₂-), 2.24 (m, 2H, -NHC(=O) $C\underline{H}_{3}$ -), 2.79 (d, 3H, J = 4.6 Hz, -NHC \underline{H}_{3}), 3.80 (s, 3H, -PhOC \underline{H}_{3}), 5.54 (s, 1H, -NH-), 5.71 (t, 1H, J = 5.0 Hz, -CHOC(=0)CH₃), 6.87 (d, 2H, J = 8.7Hz, -Ph-), 7.26 (d, 2H, J = 6.9 Hz, -Ph-); ¹³C NMR: $\delta 21.3$ (-OC(=0) $\underline{C}H_3$), 26.3 (-NHCH₂), 31.8 (-C(=0)CH₂-), 32.6 (-CH₂-), 55.3 (-PhOCH₂), 75.1 $(-\underline{C}HOC(=0)CH_3)$, 113.9, 127.9, 131.9, 159.4 (-Ph-), 170.5 $(-O\underline{C}(=0)CH_3)$, 172.4 (-NH<u>C</u>(=0)-). HRMS (FD). Calcd for $C_{14}H_{20}NO_{4}(M)^{+}$: m/z 265.1314. Found: m/z 265.1313.

7-Methyl-y-octalactone (3a, 4a) Colorless oil; R_e=0.25 (eluent: *n*-Hexane-EtOAc, 4:1, v/v); IR: 2951 (CH₂), 2868 (CH₂), 1776 (-O<u>C=O</u>), 1184 cm⁻¹ (C-O-C); ¹H NMR: δ 0.90 (d, 6H, J = 6.3 Hz, -CH(C $\underline{\text{H}}_3$)₂), 1.24 (m, 1H, $-CH_2CH(CH_2)_2$), 1.36 (m, 1H, $-CH_2CH(CH_2)_2$), 1.52–1.66 (m, 2H, -CH,CH,CH,(CH,),), 1.74 (m, 1H, -CH,CH,CH,CH,CH,),, 1.86 (m, 1H, $-OC(=O)CH_{2}CH_{3}$ -), 2.33 (m, 1H, $-OC(=O)CH_{2}CH_{3}$ -), 2.54 (t, 2H, J=8.0Hz, $-OC(=O)CH_2CH_3$ -), 4.47 (quint, 1H, J = 6.3, 7.4 Hz, $-OCH(CH_3-)CH_3$ -); ¹³C NMR: δ 22.4 (-CH(<u>C</u>H₂)₂), 27.8 (-<u>C</u>H(CH₃)₂), 28.0 (-C(=0)CH₂CH₂-), 28.8 (-C(=0)CH₂CH₂-), 33.4 (-CHCH₂CH₂-), 34.1 (-CH₂CH(CH₂)₂-), 81.3 $(-O\underline{C}HCH_3-)$, 177.26 $(-\underline{C}(=O)O-)$. HRMS (FI). Calcd for $C_0H_{17}O_3$ (M+H)+: *m/z* 157.1229. Found: *m/z* 157.1202.

4-Phenyl-y-butyrolactone (3c, 4c) Colorless oil; R,=0.15 (eluent: n-hexane-EtOAc, 4:1, v/v); IR: 3033 (Ar, C-H), 2950 (CH₂), 1776 (-OC=O), 1606, 1496 (Ar, C=C), 1176 (C-O-C), 760, 700 cm⁻¹ (Ar, C-H); ¹H NMR: δ 2.10 (m, 1H, -OC(=0)CH₂CH₂-), 2.63-2.72 (m, 3H, -OC(=0) $C\underline{H}_{5}C\underline{H}_{5}$ -), 5.52 (t, 1H, J = 6.9 Hz, $-OC\underline{H}(CH_{5})Ph$), 7.31–7.46 (m, 5H, -Ph); 13 C NMR: δ 28.9 (-C(=0)CH, CH,-), 31.0 (-C(=0)CH, CH,-), 81.2 (-OCHPh), 125.2, 128.4, 128.7, 139.3 (-Ph), 176.9 (-C(=0)O-). HRMS (ESI). Calcd for $C_{10}H_{11}O_{2}(M)^{+}$: m/z 162.0681. Found: $(M)^{+}$, 162.0653.

4-(p-Tolyl)-γ-butyrolactone (3d, 4d) Colorless oil; R_.=0.18 (eluent: *n*-hexane-EtOAc, 4:1, v/v); IR: 3025 (Ar, C-H), 2985 (CH₂), 2949 (CH₂), 1774 (-OC=O), 1616, 1518 (Ar, C=C), 1176 (C-O-C), 806 cm⁻¹ (Ar, C-H); ¹H NMR: δ 2.18 (m, 1H, -OC(=0)CH₂CH₂-, 2.35 (s, 3H, -PhCH₂), 2.56–2.69 (m, 3H, $-OC(=O)CH_3CH_3$ -), 5.47 (t, 1H, J = 7.3 Hz, $-OCH(CH_3)Ph$ -), 7.11– 7.25 (m, 4H, -Ph-); 13 C NMR: δ 21.1 (-PhCH₃), 29.0 (-C(=0)CH₂CH₃-), 30.9 (-C(=O)CH,CH,-), 81.3 (-OCHPh-), 125.3, 129.3, 132.2, 138.3 (-Ph-), 177.0 (-C(=0)O). HRMS (FI). Calcd for $C_{11}H_{13}O_{2}$ (M)+: m/z 176.0837. Found: m/z 176.0834.

5-Phenyl-γ-pentalactone (3e, 4e) Colorless oil; R_.=0.15 (eluent: n-hexane-EtOAc, 4:1, v/v). IR: 3030 (Ar, C-H), 2943 (CH₂), 1774 (-OC=O), 1603, 1496 (Ar, C=C), 1178 (C-O-C), 750, 702 cm⁻¹ (Ar, C-H); ¹H NMR: δ 1.96 (m, 1H, -OC(=0)CH₂CH₂-), 2.25 (m, 1H, -OC(=0)CH₂CH₂-), 2.42 (m, 2H, -OC(=O)CH,CH,-), 2.93 (q, 1H, J = 6.0, 6.4 Hz, -CHCH,Ph),3.08 (q, 1H, J = 6.0, 6.4 Hz, -CHCH, Ph), 4.74 (quint, 1H, J = 6.9, 6.4 Hz, $-OCH(CH_2-)CH_2Ph)$, 7.18–7.38 (m, 5H, -Ph); ¹³C NMR: δ 27.1 (-C(=O) $CH_{,C}H_{,-}$), 28.6 (-C(=0) $\underline{C}H_{,C}CH_{,-}$), 41.3 (-CH $\underline{C}H_{,P}$ h), 80.8 (-O $\underline{C}HCH_{,P}$ h), 127.0, 128.6, 129.4, 135.8 (-Ph), 177.0 (-C(=0)0-). HRMS (FI). Calcd for $C_{11}H_{12}O_{2}(M)^{+}$: m/z 176.0837. Found: m/z 176.0812.

6-Phenyl-γ-hexalactone (3f, 4f) Colorless oil; R₌=0.18 (eluent: *n*-hexane-EtOAc, 4:1, v/v); IR: 3028 (Ar, C-H), 2943 (CH₂), 1770 (-OC=O), 1603, 1495 (Ar, C=C), 1180 (C-O-C), 750, 702 cm⁻¹ (Ar, C-H); ¹H NMR: δ 1.82-1.96 (m, 2H, -C(=0)CH₂CH₂CH(O-)CH₂-), 2.05 (m, 1H, -CHCH₂ CH₂Ph), 2.31 (m, 1H, -OC(=0)CH₂CH₂-), 2.53 (m, 2H, -OC(=0)CH₂CH₂-), 2.73 (m, 1H, $-CH_{2}$ Ph), 2.83 (m, 1H, $-CH_{2}$ Ph), 4.47 (quint, 1H, J = 6.9, 6.9Hz, $-C(=O)CH_3CH_3CH_4(O-)CH_3-$), 7.16–7.36 (m, 5H, -Ph); ¹³C NMR: δ 27.9 (-C(=0)CH,CH,-), 28.8 (-C(=0)CH,CH,-), 31.6 (-CHPh), 37.3 (-OCHCH,-), 79.8 (-OCHCH,-), 126.1, 128.4, 128.5, 140.7 (-Ph), 177.1 (-C(=0)O-). HRMS (FI). Calcd for $C_{12}H_{15}O_{2}$ (M)+: m/z 190.0994. Found: m/z 190.0977.

4-(p-Anisyl)-γ-butyrolactone (3g, 4g) Colorless oil; R_.=0.10 (eluent: *n*-hexane-EtOAc, 4:1, *v/v*); IR: 3037 (Ar, C-H), 2956 (CH₂), 2937 (CH₂), 1773 (-OC=O), 1613, 1517 (Ar, C=C), 1176 (C-O-C), 837 cm⁻¹ (Ar, C-H); ¹H NMR: δ 2.20 (m, 1H, -OC(=0)CH,CH,-), 2.57-2.71 (m, 3H, -OC(=0)CH,CH,-), 3.28 (s, 3H, -PhOC \underline{H} ₂), 5.47 (t, 1H, J = 7.4 Hz, -C(=0)CH₂CH₂CH₂C \underline{H} (O-)Ph-), 6.92 (d, 2H, J = 8.6 Hz, -Ph-), 7.27 (d, 2H, J = 7.4 Hz, -Ph-); ¹³C NMR: δ 29.2 (-C(=O) CH,CH,-), 30.9 (-C(=0)CH,CH,-), 55.3 (-PhOCH,), 81.3 (-OCHPh-), 114.1, 126.9, 131.3, 165.3 (-Ph-), 177.1 (-C(=0)O-). HRMS (FI). Calcd for C₁₁H₁₃O₃ (M)+: m/z 192.0786. Found: m/z 192.0758.

General procedure for Novozym 435-catalyzed hydrolysis

Racemic N-methyl-4-acetoxyalkanamides rac-2a-g were prepared almost quantitatively from N-methyl-4-hydroxyalkanamides rac-1a-g by acetylation using acetic anhydride [30]. Briefly, a mixture of racemic N-methy-5-acetoxyalkanamide (rac-2a-g, 1.0 mmol), methanol (3.0 mmol, 0.10 g), Novozym 435 (0.4 g), and diethyl ether (20 mL) was stirred at 40°C for a period specified in Table 3, then filtered to remove Novozym 435, and concentrated. The purification of

the crude product by silica gel column chromatography eluenting with EtOAc gave optically active N-methyl-4-hydroxyalkanamide 1a-g and N-methyl-4-acetoxyalkanamide 2a-g. Lactonization is described earlier

Determination of enantiomeric excess

Enantiomeric excesses of optically active γ -lactone 3 and 4 were measured by chiral GC. General GC conditions: chiral column, CycloSil B; injector temperature, 250°C; detector temperature, 250°C; He gas, 2.0 mL/min.

7-Methyl-γ-octalactone 3a and 4a: Oven temperature, 140°C (isothermal); retention time, 8.5 min for (R)-3a, 8.8 min for (S)-4a.

4-Phenyl-γ-butyrolactone **3c** and **4c**: Oven temperature, 150°C (isothermal); retention time, 17.2 min for (S)-3c, 19.0 min for (R)-4c.

4-p-Tolyl-γ-butyrolactone 3d and 4d: Oven temperature, 150°C (isothermal); retention time, 27.0 min for (*S*)-**3d**, 30.4 min for (*R*)-**4d**.

5-Phenyl-γ-pentalactone **3e** and **4e**: Oven temperature, 140°C

(isothermal); retention time, 27.3 min for (S)-3e, 28.2 min for (R)-4e. 6-Phenyl-γ-hexalactone **3f** and **4f**: Oven temperature, 140°C

(isothermal); retention time, 76.0 min for (R)-3f, 77.8 min for (S)-4f.

4-p-Anisyl-γ-butyrolactone **3g** and **4g**: Oven temperature, 160°C (isothermal); retention time, 39.0 min for 3f, 42.6 min for 4f.

Specific rotation of optically active amides 1 and 2

The absolute configuration and the enantiomeric excesses were determined by using the values of the corresponding lactones 3 and 4.

(R)-N-methyl-4-hydroxy-7-methyloctanamide [(R)-1a]: $[\alpha]^{25}_{p}$ =-4.8 (c 1.0, MeOH, 96% e.e.).

(S)-N-methyl-4-hydroxy-4-phenylbutanamide [(S)-1c]: $[\alpha]^{25}$ =-40.7 (c 1.0, CHCl., 91% e.e.).

(S)-N-methyl-4-hydroxy-4-(p-tolyl)butanamide [(S)-1d]: $[\alpha]^{25}$ -48.6 (c 1.0, CHCl₂, 95% e.e.).

(S)-N-methyl-4-hydroxy-5-phenylpentanamide [(S)-1e]: $[\alpha]^{25}$ = +4.7 (c 1.0, CHCl₂, >99% e.e.).

(R)-N-methyl-4-hydroxy-6-phenylhexanamide [(R)-1f]: $[\alpha]^{25}$ =+8.4 (c 1.0, MeOH, >99% e.e.).

(S)-N-methyl-4-acetoxy-7-methyloctanamide [(S)-2a]: $[\alpha]^{25}_D = +8.0$ (c 1.0, MeOH, >99% e.e.).

(*R*)-*N*-methyl-4-acetoxy-4-phenylbutanamide [(*R*)-**2c**]: $[\alpha]^{25}_{p}$ = +58.8 (c 1.0, CHCl₂, 94% e.e.).

(*R*)-*N*-methyl-4-acetoxy-4-(*p*-tolyl)butanamide [(R)-2d]: $[\alpha]^{25}$ +65.7 (c 1.0, CHCl₃, 93% e.e.).

(*R*)-*N*-methyl-4-acetoxy-5-phenylpentanamide [(R)-2e]: $[\alpha]^{25}$ +13.1 (c 1.0, CHCl₃, 95% e.e.).

(S)-N-methyl-4-acetoxy-6-phenylhexanamide [(S)-2f]: $[\alpha]^{25}_{D}$ =+8.1 (c 1.0, MeOH, 99% e.e.).

Assignment of absolute configuration for lactones 3 and 4

The absolute configuration of γ -lactones 3 and 4 was determined by comparison of the sign of the measured specific rotation with that in the literature.

(S)-7-Methyl- γ -octalactone **4a**: $[\alpha]^{25}_{p}$ =-46.7 (c 1.0, MeOH, >99% e.e); Lit. [α]²⁰_D=-38.6° (c 0.26, MeOH) [46].

(R)-4-Phenyl- γ -butyrolactone **4c**: $[\alpha]_{p}^{25}$ =+32.8 (*c* 1.0, CHCl₂, 94% e.e.); Lit. $[\alpha]_{D}^{25} = +20.3^{\circ}$ (62% e.e.) [47].

(R)-4-p-Tolyl- γ -butyrolactone **4d**: $[\alpha]^{25}_{p}$ =+25.7 (c 1.0, CHCl₃, 93% e.e.); Lit. $[\alpha]^{25}_{p}$ =+10.8° (91% e.e.) [48].

(S)-5-Phenyl- γ -pentalactone **3e**: $[\alpha]^{25}_{p}$ =+19.0 (c 1.0, CHCl₃, >99% e.e.); Lit. $[\alpha]_{p}^{25} = +24.7^{\circ}$ (c 1, CHCl₃, 97% e.e.) [29].

(*R*)-6-Phenyl- γ -hexalactone **3f**: $[\alpha]_{D}^{25}$ =+65.8 (*c* 1.0, MeOH, >99% e.e.); Lit. $[\alpha]^{25}_{D}$ = +39.2° (c 0.2–1.0, MeOH, >99% e.e.) [46].

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References

- [1] Cooke, R. C.; Leeuwen, K. A. V.; Capone, D. L.; Gawel, R.; Elsey, G. M.; Sefton, M. A. Odor detection thresholds and enantiomeric distributions of several 4-alkyl substituted γ-lactones in Australian red wine. J. Agric. Food Chem. 2009, 57, 2462-2467.
- [2] Tamogami, S.; Awano, K.; Kitahara, T. Analysis of the enantiomeric ratios of chiral compounds in absolute jasmine. Flavour Fragr. J. 2001, 16, 161-163.
- [3] Sarrazin, E.; Frerot, E.; Bagnoud, A.; Aeberhardt, K.; Rubin, M. Discovery of new lactones in sweet cream butter oil. J. Agric. Food Chem. 2011, 59, 6657-6666.
- [4] Leal, W. S.; Wojtasek, H.; Kuwahara, J. F. P. S.; Saito, H.; Hasegawa, M. Perireceptor events in pheromone perception in scarab beetles. J. Asia Pac. Entomol. 1998, 1, 1-8.
- Sabitha, G.; Yadagiri, K.; Yadav, J. S. Synthesis of the Junus integer pheromone (4R,9Z)-9-octadecen-4-olide. Tetrahedron Lett. 2007, 48, 1651-1652.
- [6] Sabitha, G.; Fatima, N.; Reddy, E. V.; Yadav, J. S. Prins and RCM protocols for the synthesis of the pheromones of the giant white butterfly Idea leuconoe. Tetrahedron Lett. 2008, 49, 6087-6089.
- [7] Brown, H. C.; Kulkarni, S. V.; Racherla, U. S. Chiral synthesis via organoboranes. 39. A facile synthesis of γ -substituted- γ butyrolactones in exceptionally high enantiomeric purity. J. Org. Chem. 1994, 59, 365-369.
- [8] Gogoi, S.; Argade, N. P. An efficient PS-catalyzed chemo-, regio- and enantioselective hydrolysis of (±)-2,3-di-O-acetyl-2-C-methyl-D-erythrono-1,4-lactone: a facile preparation of bioactive natural products (-)-saccharinic acid lactone and potassium (2R,3R)-2,3,4-trihydroxy-2-methylbutanoate. Tetrahedron Asymmetry 2006, 17, 927-932.
- [9] Zoute, L.; Lemonnier, G.; Nguyen, T. M.; Quirion, J. C.; Jubault, P. Convenient preparation of α -bromo- α -fluoro- β hydroxyesters building blocks from aldehydes, ketones and lactones. Tetrahedron Lett. 2011, 52, 2473-2475.
- [10] Smith, N. D.; Goodman, M. Enantioselective synthesis of α -methyl-D-cysteine and lanthionine building blocks via α -methyl-D-serine- β -lactone. *Org. Lett.* **2003**, *5*, 1035–1037.
- [11] Stangeland, E. L.; Sammakia, T. Use of thiazoles in the halogen dance reaction: application to the total synthesis of WS75624 B. J. Org. Chem. 2004, 69, 2381-2385.

- [12] Chidley, H. G.; Kulkarni, R. S.; Pujari, K. H.; Giri, A. P.; Gupta, V. S. Spatial and temporal changes in the volatile profile of Alphonso mango upon exogenous ethylene treatment. Food Chem. 2013, 136, 585-594.
- [13] Macleod, A. J.; Macleod, G.; Snyder, C. H. Volatile aroma constituents of mango (cv Kensington). Phytochemistry 1988, 27, 2189-2193.
- [14] Wang, Y.; Yang, C.; Li, S.; Yang, L.; Wang, Y.; Zhao, J.; Jiang, Q. Volatile characteristics of 50 peaches and nectarines evaluated by HP-SPME with GC-MS. Food Chem. 2009, 116, 356-364.
- [15] Visai, C.; Vanoli, M. Volatile compound production during growth and ripening of peaches and nectarines. Sci. Hortic. **1997**, 70, 15-24.
- [16] Boishebert, V.; Giraudel, J. L.; Montury, M. Characterization of strawberry varieties by SPME-GC-MS and Kohonen self-organizing map. Chemom. Intell. Lab. Syst. 2006, 80, 13-23.
- [17] Lambert, Y.; Demazeau, G.; Largeteau, A.; Bouvier, J. M. Changes in aromatics volatile composition of strawberry after high pressure treatment. Food Chem. 1999, 67, 7-16.
- [18] Alewijn, M.; Smit, B. A.; Sliwinski, E. L.; Wouters, J. T. M. The formation mechanism of lactones in Gouda cheese. Int. Dairy J. 2007, 17, 59-66.
- [19] Leuven, I. V.; Caelenberg, T. V.; Dirinck, P. Aroma characterization of Gouda-type cheeses. Int. Dairy J. 2008, 18, 790-800.
- [20] Rincon-Delgadillo, M. I.; Lopez-Hernandez, A.; Wijaya, I.; Rankin, S. A. Diacetyl levels and volatile profiles of commercial starter distillates and selected dairy foods. J. Dairy Sci. 2012, 95, 1128-1139.
- [21] Cagno, R. D.; Banks, J.; Sheehan, L.; Fox, P. F.; Brechany, E. Y.; Corsetti, A.; Gobbetti, M. Comparison of the microbiological, compositional, biochemical, volatile profile and sensory characteristics of three Italian PDO ewes' milk cheeses. Int. Dairy J. 2003, 13, 961-972.
- [22] Bredikhin, A. A.; Bredikhina, Z. A.; Zakharychev, D. V.; Pashagin, A. V. Chiral drugs related to guaifenesin: synthesis and phase properties of methocarbamol and mephenoxalone. Tetrahedron Asymmetry 2007, 18, 1239-1244.
- [23] Patel, R. N. Microbial/enzymatic synthesis of chiral intermediated for pharmaceuticals. Enzyme Microb. Technol. 2002, 31,
- [24] Tajik, H.; Dadras, A.; Aghabeygi, S. A facile synthesis of novel optically active R,R-2-{4-[2-(4-(5-chloro-3-halopyridin-2-yloxy) phenoxy)propionyloxy]phenoxy}propionic acid esters using cyanuric chlorides as potential herbicide. Chin. Chem. Lett. 2011, 22, 535-538.
- [25] Itoh, H.; Furukawa, Y.; Tsuda, M.; Takeshiba, H. Synthesis and fungicidal activity of enantiomerically pure (R)- and (S)-silicon-containing azole fungicides. Bioorg. Med. Chem. 2004, 12, 3561-3567.
- [26] Ghosh, A. K.; Swanson, L. Enantioselective synthesis of (+)-cryptophycin 52 (LY355703), a potent antimitotic antitumor agent. J. Org. Chem. 2003, 68, 9823-9826.
- [27] Ghosh, A. K.; Shurrush, K.; Kulkarni, S. Asymmetric synthesis of anti-aldol segments via a nonaldol route: synthetic applications to statines and (-)-tetrahydrolipstatin. J. Org. Chem. 2009, 74, 4508-4518.
- [28] Hilborn, J. W.; Lu, Z. H.; Jurgens, A. R.; Fang, Q. K.; Byers, P.; Wald, S. A.; Senanayake, C. H. A practical asymmetric synthe-

- sis of (R)-fluoxetine and its major metabolite (R)-norfluoxetine. Tetrahedron Lett. 2001, 42, 8919-8921.
- [29] Kotkar, S. P.; Suryavanshi, G. S.; Sudalai, A. A short synthesis of (+)-harzialactone A and (R)-(+)-4-hexanolide via prolinecatalyzed sequential α-aminooxylation and Horner-Wadsworth-Emmons olefination of aldehydes. Tetrahedron Asymmetry **2007**, 18, 1795-1798.
- [30] Shimotori, Y.; Nakahachi, Y.; Inoue, K.; Miyakoshi, T. Synthesis of optically active γ -lactones via lipase-catalyzed resolution. Flavour Fragr. J. 2007, 22, 421-429.
- [31] Garst, J. F.; Lawrence, K. E.; Batlaw, R.; Boone, J. R.; Ungváry, F. Magnesium bromide in Grignard reagent formation. Inorg. Chim. Acta 1994, 222, 365-375.
- [32] Tuulmets, A.; Panov, D. Solvation effects in partially solvated Grignard reagents. J. Organometal. Chem. 1999, *575*, 182–186.
- [33] Chalk, A. J. Ring metalation of toluene by butyllithium in the presence of N,N,N',N'-tetramethylethylenediamine. J. Organometal. Chem. 1968, 11, 615-618.
- [34] Lenoir, D.; Dauner, H.; Frank, R. M. Dehydratisierung von aliphatischen Di-tert-butylalkylcarbinolen. Kraftfeld-Rechnungen von sterisch gehinderten Ethylenen. Chem. Ber. 1980, 113, 2636-2647.
- [35] Hellmann, S.; Beckhaus, H. D.; Rüchardt, C. Synthesen, Spektren, Struktur und Spannung hochverzweigter Pentane. Chem. Ber. 1983, 116, 2219-2237.
- [36] Uemura, M.; Yagi, K.; Iwasaki, M.; Nomura, K.; Yorimitsu, H.; Oshima, K. Pentamethylcyclopentadienide in organic synthesis: nucleophilic addition of lithium pentamethylcyclopentadienide to carbonyl compounds and carbon-carbon bond cleavage of the adducts yielding the parent carbonyl compounds. Tetrahedron 2006, 62, 3523-3535.
- [37] Pace, V.; Castoldi, L.; Hoyos, P.; Sinisterra, J. V.; Pregnolato, M.; MªSánchez-Montero, J. Highly regioselective control of 1,2-addition of organolithiums to α , β -unsaturated compounds promoted by lithium bromide in 2-methyltetrahydrofuran: a facile and eco-friendly access to allylic alcohols and amines. Tetrahedron 2011, 67, 2670-2675.
- [38] Basak, D.; Versek, C.; Toscano, D. T.; Christensen, S.; Tuominen, M. T.; Venkataraman, D. Anhydrous proton conductivities of squaric acid derivatives. Chem. Commun. 2012, 48, 5922-5924.
- [39] Harrowven, D. C.; Mohamed, M.; Gonçalves, T. P.; Whitby, R. J.; Bolien, D.; Sneddon, H. F. An efficient flow-photochemical synthesis of 5H-furanones leads to an understanding of torquoselectivity in cyclobutenone rearrangements. Angew. Chem. Int. Ed. 2012, 51, 4405-4408.
- [40] Shimotori, Y.; Miyakoshi, T.; Synthesis of optically active γ-lactones with lipase catalyst. J. Oleo Sci. 2006, 55,
- [41] Naoshima, Y.; Mori, Y.; Yagi, Y. Proceeding of the 15th International Conference on Genome Informatics Poster and Software Demonstrations, 2004; P129-1-P129-2.
- [42] Shimotori, Y.; Miyakoshi, T. Synthesis of (S)- γ -lactones with a combination of lipase-catalyzed resolution and Mitsunobu reaction. Synth. Commun. 2009, 39, 1570-1582.
- [43] Shimotori, Y.; Miyakoshi, T. Combination of Novozym 435-catalyzed hydrolysis and Mitsunobu reaction for

- production of (R)-γ-lactones. Synth. Commun. 2010, 40, 1607-1613.
- [44] Ema, T.; Kobayashi, J.; Maeno, S.; Sakai, T.; Utaka, M. Origin of the enantioselectivity of lipases explained by a stereo-sensing mechanism operative at the transition state. Bull. Chem. Soc. Jpn. 1998, 71, 443-453.
- [45] Raza, S.; Fransson, L.; Hult, K. Enantioselectivity in Candida antarctica lipase B: a molecular dynamics study. Protein Sci. **2001**, *10*, 329-338.
- [46] Ishihara, M.; Tsuneya, T.; Shiota, H.; Shiga, M.; Yokoyama, Y. The absolute configurations of marmelo lactones. Agric. Biol. Chem. 1983, 47, 2121-2122.
- [47] Patil, S. T.; Karnik, A. V. Aluminum chloride-mediated kinetic resolution of racemic γ -substituted- γ -lactones. *Chirality* **2004**, *16*, 336–338.
- [48] Fukuzawa, S.; Seki, K.; Tatsuzawa, M.; Mutoh, K. A facile synthesis of chiral γ -butyrolactones in extremely high enantioselectivity mediated by samarium (II) iodide. J. Am. Chem. Soc. 1997, 119, 1482-1483.