

# FACEMASK COMFORT ENHANCEMENT WITH GRAPHENE OXIDE FROM RECOVERED CARBON WASTE TYRES

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## Abstract:

Commercial disposable facemasks have lower breathability and may cause discomfort after hours of wearing them. Graphene oxide (GO) nanoparticles offer a significant opportunity to improve the breathability of facemask materials. Hence, the current study aimed to investigate the feasibility of incorporating GO in facemask materials. The GO was synthesized from recovered carbon of waste carbon tyre. In this study, two concentrations of GO (0.01 and 0.02%) were used to enhance the comfort properties of the polypropylene (PP) facemask fabric. The GO-coated PP facemask fabrics were characterized for scanning electron microscopy, energy dispersive X-ray, and Raman spectroscopy. The comfort properties were determined using air permeability, water vapour permeability, and moisture management test. Raman analysis revealed distinctive peaks corresponding to GO at approximately 1,300 and 1,500  $\text{cm}^{-1}$ . The GO displayed bumping pieces of particles and a textured surface, with a diameter ranging from 30 to 80 nm. The result of mercury porosimetry shows that the PP fabric coated with 0.02% of GO provided a higher pore diameter and porosity at approximately 21.31  $\mu\text{m}$  and 82.79%, respectively. Due to its high pore diameter and porosity, the PP filter facemask fabric coated with 0.02% GO demonstrated enhanced air permeability, water vapour permeability, and moisture management. These results suggested that the sample possesses favourable breathability properties as compared to the sample without GO. By undertaking this study, GO synthesized from the waste carbon tyre was developed, which can enhance the breathability of fabric materials.

## Keywords:

Graphene oxide, melt-blown polypropylene, comfort properties

## 1. Introduction

Coronavirus disease 2019, known as COVID-19, is a pneumonia disease confirmed by the World Health Organization on 12 January 2020 before an outbreak in all countries. COVID-19 is caused by the severe acute respiratory syndrome coronavirus 2 (SARS-CoV-2) that infects the respiratory tract, which can cause human death [1]. It also negatively affected the economy of all countries due to the emergency lockdown in most countries to minimize the disease from spreading quickly to any human being.

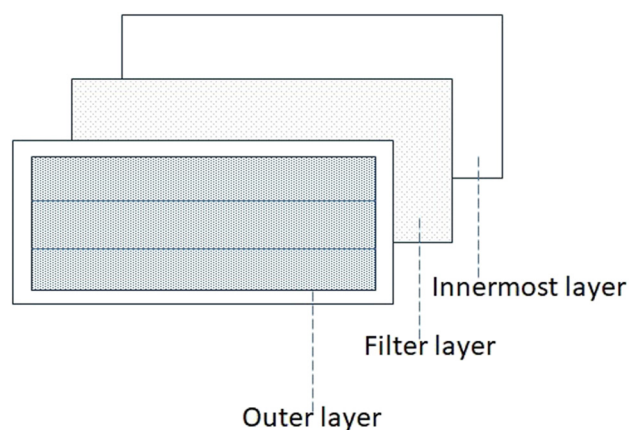
Malaysia also had the highest cases of COVID-19 among Southeast Asians [2]. Since COVID-19 is known as a contagious disease, the frontliners, such as doctors and nurses, use personal protective equipment (PPE) to prevent any infectious diseases from the patients. Facemasks are used to control the transmission of droplets by the wearer [3]. Facemasks can be reusable or disposable based on the type of the mask material. The disposable facemask is mainly made of hydrophobic polymers, such as polypropylene (PP). According to several authors, the three-ply facemask consists of three essential layers, which are the outer layer, filter layer, and innermost layer [4,5]. In Figure 1, the outer layer is waterproof, and it prevents fluids,

such as mucosalivary droplets that are produced from respiratory activities, such as sneezing and coughing. Meanwhile, the filter layer or the centre layer is a melt-blown nonwoven fabric. It prevents particles or germs larger than the filter layer from entering either direction. Finally, the innermost layer comprises absorbent materials that capture mucosalivary droplets from the user. The filter layer exhibits a crucial component of a facemask. Hence, the current study aimed to enhance the filter layer of the facemask.

Most of the facemasks are not comfortable when worn by the wearer due to the discomfort caused by facial heat, especially in hot and humid weather. According to Matuschek et al., one of the main arguments against wearing a mask is that breathing can cause the mask to become damp. When there is a lot of moisture, the masks become airtight. As a result, unfiltered air is inhaled and exhaled around the edges, reducing the protective effect for both the wearer and the environment [5]. One alternative way to improve the mask material's comfort property is by incorporating it with graphene oxide (GO). Carbon black is found to be a good source for GO production.

Approximately 57,391 tonnes of waste tyres were produced annually in Malaysia [6]. Aside from carbon waste tyres, various





**Figure 1.** Illustration of three layers of face masks.

studies have found that agricultural waste may be efficiently employed to generate a value-added product with improved performance attributes [7]. This waste has great potential to substitute carbon-based resources such as GO. Hence, the current study synthesized the GO from waste carbon tyres using a modified Hummers' method.

The GO demonstrates excellent filler for thermally conductive materials that can be easily coupled with fabrics to produce textiles and clothing with improved thermal comfort. In addition, Alen et al. reported that GO could increase the permeability of membrane materials. This is due to the gas particles' capability to pass through GO surfaces [8]. This statement shows the feasibility of incorporating GO in the facemask, which can enhance the air and water vapour permeabilities of the facemask. In addition, several studies have explored the thermal comfort properties of textile-based materials [9,10].

In the current study, GO was incorporated in the filter layer of a facemask material. The PP filter facemask fabric was sonicated in a dispersed GO suspension. The GO-coated PP filter facemask fabrics were tested for comfort properties, such as air permeability, water vapour permeability, and moisture management test (MMT). By undertaking the study, the comfort properties of the GO-coated facemask materials could be determined. Hence, enhanced properties of PP filter facemask fabric were developed, which will benefit those who wear facemasks.

## 2. Experimental

### 2.1. Materials

The materials used in this research were a single layer of PP melt-blown filter facemask fabric (Fibertex Personal Care Sdn. Bhd.) and GO. The average fabric weight obtained from ten measurements taken at various locations on the fabric was approximately 142.86 g/m<sup>2</sup>. The GO was synthesized by the modified Hummers' method, as described earlier in previous work [11]. Prior to the coating process, the GO was dispersed in distilled water to form a GO suspension. This GO dispersion contributes to achieving a uniform coating on the fabric.

### 2.2. Preparation of GO suspension

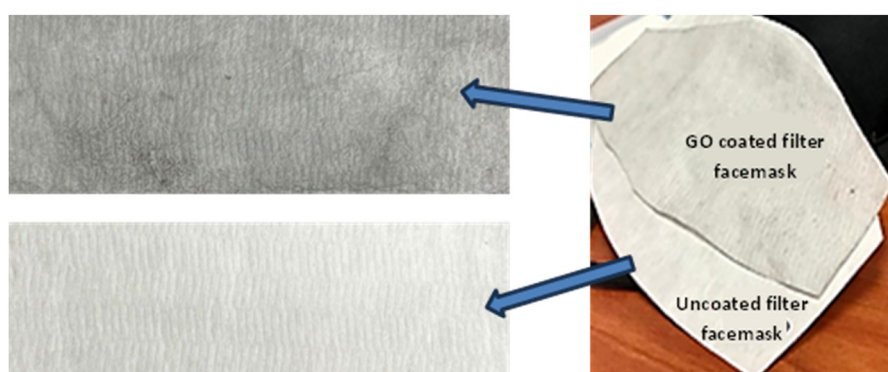
The GO suspension was prepared by combining 0.1 and 0.2 g GO nanoparticles with 1,000 ml distilled water to produce two concentrations, which were 0.01 and 0.02%, respectively. The suspension was sonicated for 60 min at 40°C using a sonicate ultrasonic bath (Elma Ultrasonic S10H, Germany) to generate a homogeneous dispersion of the GO nanoparticles in distilled water.

### 2.3. Formation of the GO-coated PP filter facemask fabric

The PP melt-blown filter facemask was immersed into the dispersed 0.01 and 0.02% GO suspension for 30 min under 30 amplitude sonication and later was dried at 80°C for 2 h. The GO-coated PP filter facemask was then rinsed with deionized water and ethanol to remove any unfixed GO from the filter facemask. Finally, the GO incorporated with the PP melt-blown filter facemask was oven-dried overnight in a vacuum oven at 35°C. The GO-coated PP filter facemask and uncoated PP filter facemask are illustrated in Figure 2.

### 2.4. Characterization of the GO-coated PP filter facemask fabric using scanning electron microscopy (SEM), energy-dispersive X-ray (EDX), and Raman spectroscopy

The presence of GO from recovered carbon was tested using Raman spectroscopy (Horiba Jobin Yvon, HR800, Japan).



**Figure 2.** Photographs of the GO-coated PP filter facemask and uncoated PP filter facemask.

Meanwhile, the morphological structures of the GO particles and GO-coated PP filter facemask were characterized using SEM (Hitachi TM3000 Tabletop SEM, USA). The GO-coated PP filter facemasks were cut into 1 cm × 1 cm size dimensions followed by gold coating for 60 s before observation to increase the conductivity of the samples. The magnifications used were 500× and 10,000× with a constant acceleration voltage of 5 kV. The elemental analysis of the samples was tested using the EDX analysis.

## 2.5. Pore diameter and porosity analysis

The pore diameter and porosity of the GO-coated PP filter facemask fabric and uncoated PP filter facemask fabric were tested using a mercury porosimeter (AutoPore V-Micromeritics). The mercury porosimeter employs a pressurized chamber (up to 60,000 psi) to force mercury to intrude into the voids of the fabrics. As the pressure is applied, mercury fills the larger pores, followed by smaller pores. The pore diameter of the GO-coated PP filter facemask fabric and uncoated filter facemask fabric was determined automatically by the porosimeter using the Washburn equation [1]:

$$D = \frac{(-4\gamma\cos\theta)}{P}, \quad (1)$$

where  $D$  is the pore diameter ( $\mu\text{m}$ ),  $\gamma$  is the surface tension of mercury (485 mN/m or 485 dynes/cm),  $\theta$  is the contact angle of mercury (approximately 130°), and  $P$  is the applied pressure (psi).

## 2.6. Air permeability, water vapour permeability, and MMT

The air permeability test was conducted using Mesdan Air Tronic 3240B (Italy) based on ASTM D737-18 in metres per second (m/s) and the air volume at about 10 l. The test area was 10 cm<sup>2</sup>, with a pressure value of about 100 Pa. For water vapour permeability (WVP), the test was conducted using the WVP tester (SDL Atlas, M261, USA) based on the ISO 8096 standard method, and the WVP of the samples was calculated using the following equation [2]:

$$\text{WVP} = \frac{24M}{At}, \quad (2)$$

where  $M$  is the water vapour loss (g) in the expose time period  $t$  (h), and  $A$  is the area of the uncovered specimen (m<sup>2</sup>). The MMT was conducted based on the American Textile and Colorists Association standard test methods 195-2009 using an MMT (SDL Atlas, USA). The MMT aims to measure the transfer of liquid solutions and the distribution in a sample. The size of each sample was 80 mm × 80 mm. The top and bottom sensors were placed on the sample. The saline solution was pumped onto the upper surface of the fabric to simulate a drop of liquid sweat. The MMT software determined the moisture management properties of the sample.

## 3. Results and discussion

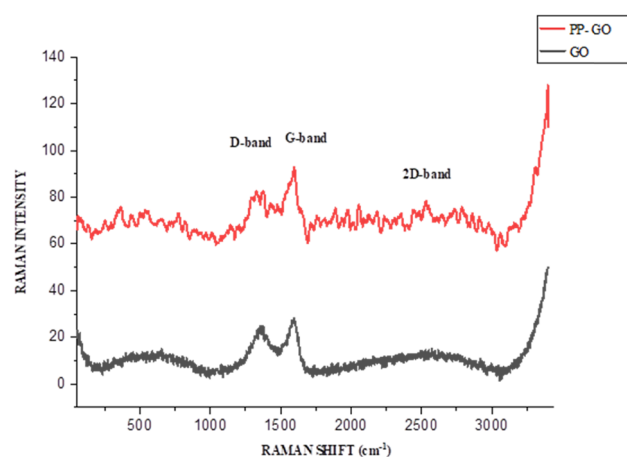
### 3.1. Characterization of GO from recovered carbon of waste carbon tyre using Raman spectroscopy

Raman spectroscopy explains the conversion of recovered carbon to GO. Raman analyses are essentially identical to the characteristic peak of GO by two peaks at around 1,300 and 1,500 cm<sup>-1</sup> (D band and G band) [12].

Based on Figure 3, the Raman spectra of the obtained GO display two peaks at 1,367 and 1,593 cm<sup>-1</sup>, corresponding to the D-band and G-band. The results show that GO was successfully synthesized from the recovered carbon of waste tyre.  $I_D/I_G$  ratios can be determined by the heights or areas of the peak. The widening of peak D is related to the distribution of agglomerates with different orders and dimensions; thus, the information about the aromatic rings is less distorted in the maximum intensity than in width, which depends on the disorder [13].

Figure 3 shows the same pattern of Raman peak for GO onto the PP filter facemask fabric. The D band and G band peaks of Raman spectroscopy from the fabric sample successfully proved the attachment of GO. It shows that oxygen on the GO structure helps to improve the compatibility of the GO and PP fabric.

In this work, the values of the  $I_D/I_G$  ratios were calculated using intensity values as listed in Table 1. The  $I_D/I_G$  ratio measures the degree of disorder and is inversely proportional to the mean size of the sp<sup>2</sup> clusters present in the material. Thus, the higher the  $I_D/I_G$  ratio in carbon materials, the greater the structural disorder. The GO reduction process can be expressed in the Raman spectra by changing the relative intensity of the D and G bands. As shown in Table 1, the  $I_D/I_G$  ratio of GO from recovered carbon waste is higher than that of the GO-coated PP filter facemask fabric, suggesting that new (or more) graphite domains were formed, and the number of sp<sup>2</sup> clusters increased during the reduction. This indicates an efficient reduction, collaborating with



**Figure 3.** The Raman spectra of GO from carbon waste tyre and GO on the PP filter facemask fabric.



**Table 1.** Raman spectroscopy of GO from recovered carbon waste and GO-coated PP filter facemask fabric

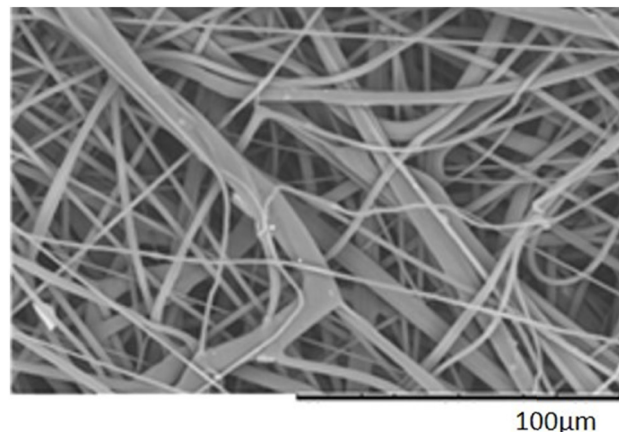
| Samples                             | D Peak ( $\text{cm}^{-1}$ ) | G Peak ( $\text{cm}^{-1}$ ) | $I_D/I_G$ Ratio |
|-------------------------------------|-----------------------------|-----------------------------|-----------------|
| GO from recovered carbon waste      | 1367.37                     | 1598.78                     | 0.92            |
| GO-coated PP filter facemask fabric | 1326.87                     | 1594.93                     | 0.71            |

the results obtained by the surface energy characterizations. The value of  $I_D/I_G$  indicates the GO with a higher degree of graphitization and less defect is produced [14,15]. From this analysis, the GO-coated PP filter facemask fabric was further investigated for comfort properties at different GO concentrations.

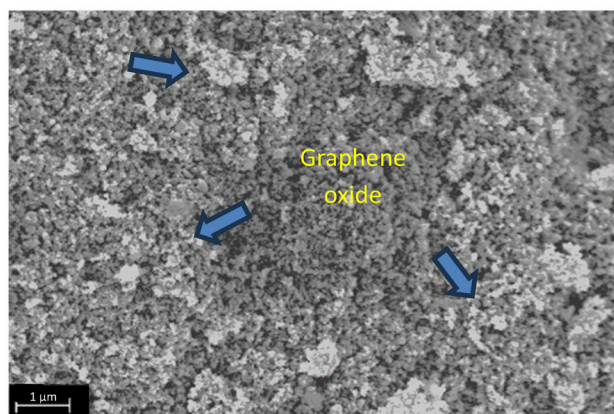
### 3.2. Morphological structure and elementary contents of GO and GO-coated PP filter facemask

Figure 4a and b depicts our synthesized GO powder made from the recovered carbon of waste carbon tyre. As shown in Figure 4b, the image of GO reveals a circular structure trailed together on its surface. Additionally, it exhibits bumping pieces of particles and a rough surface with a diameter of between 30 and 80 nm.

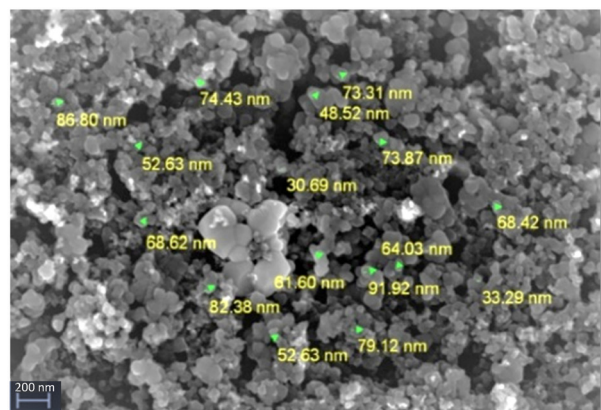
Figure 5 illustrates the SEM pictures of the uncoated, 0.01% GO-coated PP filter facemask fabric, and 0.02% GO-coated PP



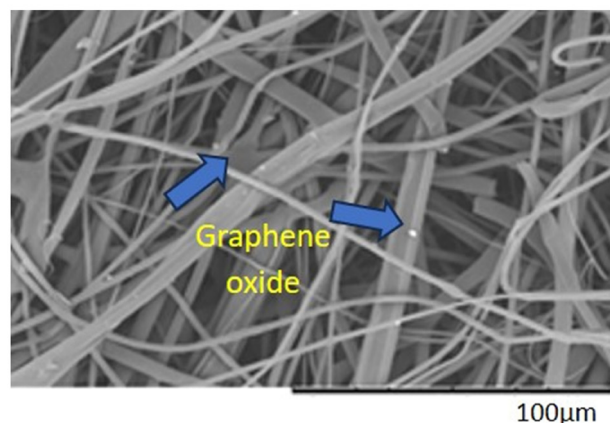
(a)



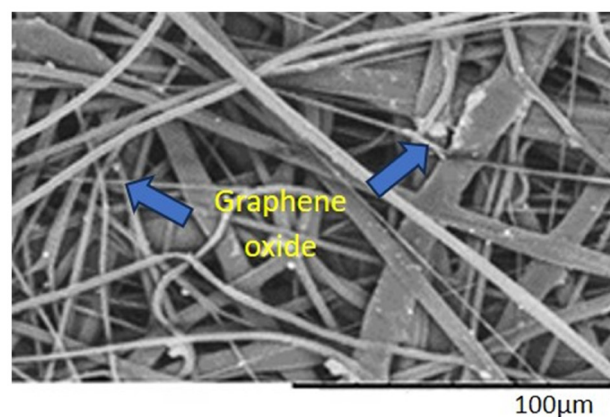
(a)



(b)



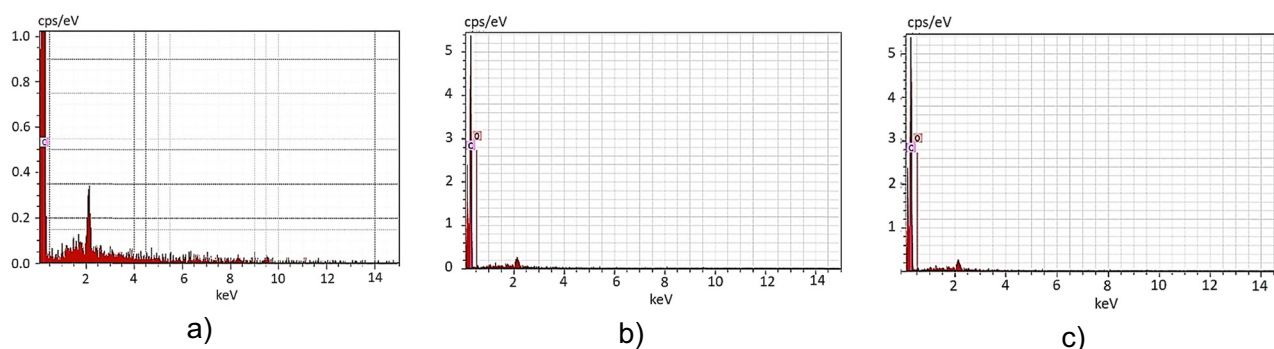
(b)



(c)

**Figure 4.** (a) The SEM micrograph of GO obtained at 10,000 $\times$  magnification and (b) size of GO under 30,000 $\times$  magnification.

**Figure 5.** SEM images of (a) uncoated samples, (b) 0.01% GO-coated PP filter facemask fabric, and (c) 0.02% GO-coated PP filter facemask fabric.



The percentage (%) of carbon and oxygen:

|    | Carbon (C) | Oxygen (O) |
|----|------------|------------|
| a) | 100        | 0          |
| b) | 94.21      | 5.79       |
| c) | 93.75      | 6.25       |

**Figure 6.** EDX analysis of (a) uncoated samples, (b) 0.01% GO-coated PP filter facemask fabric, and (c) 0.02% GO-coated PP filter facemask fabric.

filter facemask fabric, revealing that the GO had effectively adhered to the filter fabric. The increasing percentage of GO indicates that the GO is increasingly attached to the fabric. Consequently, GO exhibited superior compatibility with non-woven materials, which led to good air permeability, water vapour permeability, and moisture management.

The EDX analysis was performed to analyse the elementary composition of GO on the fabric. Figure 6a shows the presence of carbon with a percentage of 100% in the uncoated PP filter fabric. Figure 6b reveals that the GO-coated PP filter facemask fabrics have carbon and oxygen percentages of 94.21 and 5.79%, respectively. Meanwhile, the carbon and oxygen percentages in Figure 6c are 93.7 and 6.25%, respectively, suggesting the presence of GO in the fabric. The existence of oxygen and carbon in GO was also reported by Han et al. [16].

As stated by Bonnia et al., the elemental composition of carbon (C) is higher than the oxygen (O) element, which shows an excellent composition of GO. A similar finding was observed in the current study where the elemental composition of carbon (C) was also higher than oxygen (O). As shown in Figure 6b and c, the percentages of GO are as follows: 0.01% of GO exhibits 94.21% of carbon (C) and 5.79% of oxygen (O). Meanwhile, 0.02% GO provides 93.75% carbon (C) and 6.25% oxygen (O), respectively.

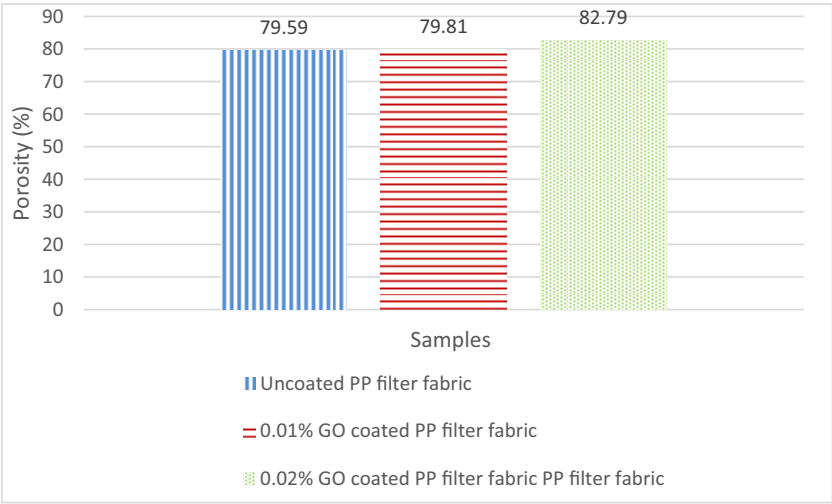
The EDX analysis shows that oxygen element appears for the coated sample, proving that GO was successfully attached to the fabric. Increasing the oxygen content along with an increasing percentage of GO applied to the fabric may enhance the moisture management of the PP filter facemask material. Goswami et al. also stated that GO gives good air permeability to the filter facemask, leading to good breathability of the material [17]. Hence, an extensive study on air permeability, water vapour, and moisture management was conducted in the current research work.

### 3.3. Pore diameter and porosity

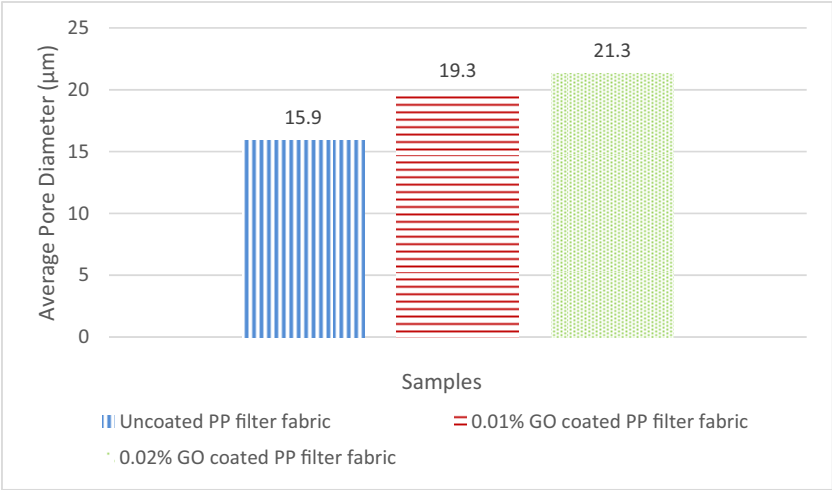
According to Moa and Russell, the pore structure can be defined by the total pore volume (porosity) and pore connectivity [18]. In the current study, GO was found to enhance the porosity and pore diameter of the PP filter facemask fabric. Based on Figure 7a, incorporating GO in PP filter facemask fabric increases the fabric porosity by 79.59–82.79%. Meanwhile, the fabric pore diameter increased from 15.9 to 21.3  $\mu\text{m}$  (Figure 7b). The presence of GO particles may result in partial embedding within the PP filter facemask fabric, which can modify the fabric's structure, ultimately resulting in larger pores. In other words, the GO can enhance the breathability of the PP filter fabric, which is in line with Du et al.. According to Du et al., a facemask with a larger porosity enables air in and out, increasing breathability. They added that the surgical mask has an intermediate porosity of 77%, while the N95 mask has the lowest porosity of 65% [19]. The authors also reported that the reusable mask has the most oversized pore diameter with approximately 46  $\mu\text{m}$ , whereas the N95 mask has the finest pore diameter, 28  $\mu\text{m}$  [19]. In the current study, GO-coated PP filter pore diameter is in a range of 16–21  $\mu\text{m}$ . From this analysis, the results show that the GO-coated PP filter facemask fabrics have the potential to be used in one of the layers of N95 facemask materials.

### 3.4. Air permeability

The uncoated and GO-coated PP filter facemask fabrics were tested for air permeability, and the results are presented in Figure 8. From the analysis, the air permeability of GO-coated PP filter facemask fabrics (0.01 and 0.02%) is 7% higher than that of the uncoated one. In Figure 8, the air permeability of the uncoated, 0.01% GO, and 0.02% GO are 195.28, 210.12, and 209.6 mm/s, respectively. Due to the high porosity of the GO-coated PP filter facemask fabrics as stated in Figure 7a, air was able to penetrate through the fabric samples at a higher rate than that of the uncoated fabric samples.

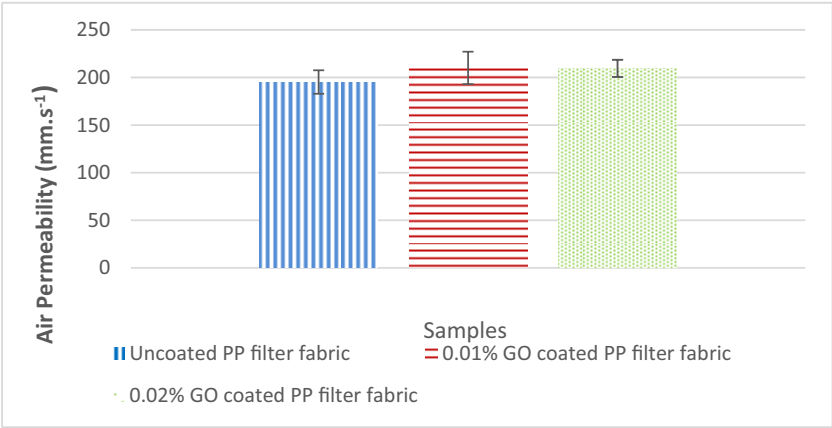


(a)

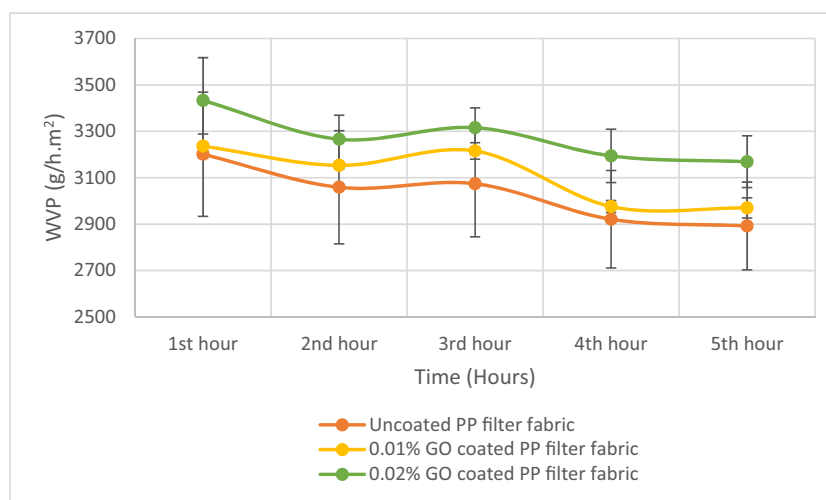


(b)

**Figure 7.** (a) Porosity and (b) average pore diameter size of uncoated PP filter facemask fabric, 0.01% GO-coated PP filter facemask fabrics, and 0.02% GO-coated PP filter facemask fabric, respectively.



**Figure 8.** The air permeability of uncoated PP filter facemask fabric, 0.01% GO-coated PP filter facemask fabric, and 0.02% GO-coated PP filter facemask fabric.



**Figure 9.** The WVP of uncoated PP filter facemask fabric, 0.01% GO-coated PP filter facemask fabric, and 0.02% GO-coated PP filter facemask fabric.

The current analysis agrees with a study conducted by Li et al. The authors observed that the presence of GO resulted in enhanced air permeability and water vapour permeability in their samples [20]. Considering this, the GO-coated PP filter facemask fabric has the potential to be used as a filter layer in facemask materials. As emphasized by Patanaik and Anandjiwala, air permeability plays a crucial role in the effectiveness of nonwovens in various applications, including filters, breathable liners, protective garments, and hygiene clothes [21].

### 3.5. Water vapour permeability

Breathing comfort is critical while using a facemask. It is also affected by the rate of moisture transport through the filter. To understand the breathing comfort properties, the water vapour and MMTs were conducted on the uncoated PP and GO-coated PP filter facemask fabrics. In Figure 9, the uncoated PP and GO-coated PP filter facemask fabrics show a reduction from the first hour to the fifth hour. The WVP of the uncoated PP filter fabric was reduced from 3201.33 to 2892.29 g/h m<sup>2</sup>. Meanwhile, the 0.01% GO-coated PP filter facemask fabric was reduced from 3236.15 to 2970.01 g/h m<sup>2</sup> and the 0.02% GO-coated PP filter fabric was reduced from 3432.88 to 3168.95 g/h m<sup>2</sup>.

The results presented in Figure 9 also show that the 0.02% GO-coated PP filter facemask fabric exhibits the highest WVP, followed by 0.01% GO and uncoated PP fabric samples. The high amount of GO was expected to enhance the WVP of the samples. This improvement was attributed to the transfer of water molecules either across GO or through openings created by surface imperfections in GO. Facemasks incorporated with thermally conductive materials, such as GO, are believed to offer excellent WVP, contributing to enhanced comfort during wear. This is because these conductive materials facilitate the transmission of moisture and heat into the environment [6,22].

According to Huang, water vapour is vital in determining thermal comfort [23]. From the experimental study, it can be

concluded that GO can help to improve the breathability and comfort properties of the facemask materials.

### 3.6. MMT

Moisture control is one of the most crucial performance criteria in the textile industry today. Hence, it has significantly affected the thermophysiological comfort of the human body. The microclimate between the skin and clothes should be thermally stable through moisture management [24]. Furthermore, the overall moisture management capacity (OMMC) is an index that indicates the ability to transport liquid moisture through a sample. As shown in Table 2, the OMMC characteristics of the PP filter facemask fabric had been enhanced with the assistance of GO. The uncoated PP fabric demonstrates the lowest scale of OMMC, which is 3.5 (good to very good rating). The 0.01% GO-coated PP fabric exhibits the second-highest scale, which is 4 (very good rating), and the 0.02% GO-coated fabric shows the highest scale, which is 5 (excellent rating) of OMMC. Due to the hydrophobicity character of the PP, it enables moisture wicking at a low absorption rate and spreading speed [24]. The presence of GO in the PP filter facemask fabric increases the pore volume (voids), resulting in better fabric moisture control with promising moisture management properties.

**Table 2.** OMMC of uncoated PP filter facemask, 0.01% GO coated PP filter facemask fabrics, and 0.02% GO coated PP filter facemasks fabrics

| Samples                                    | Grade | Description       |
|--|-------|-------------------|
| Uncoated PP filter facemask fabric         | 3.5   | Good to very good |
| 0.01% GO-coated PP filter facemask fabric  | 4     | Very good         |
| 0.02% GO-coated PP filter facemask fabric. | 5     | Excellent         |



## 4. Conclusions

The current study successfully incorporated GO on PP filter facemask fabric. The SEM analysis shows the bumping pieces, coarse, and chunky of sharp GO for 0.01 and 0.02% GO-coated PP filter facemask fabrics. Meanwhile, the EDX result analysis also proves the presence of GO in the fabric. Air permeability, WVP, and moisture management determine the comfort properties of the fabric samples. It can be seen that the comfort properties of the PP filter facemask have significantly improved with an additional GO in the fabric. The presence of GO in the PP filter facemask fabric transports moisture into the environment, reducing the accumulation of moisture on the fabric surfaces. Based on the results of the tests, the 0.02% GO-coated PP filter facemask fabric gives an optimal comfort property and breathability due to its high air permeability, WVP, and excellent overall moisture management rate. The current study demonstrates the feasibility of the GO-coated PP filter facemask fabric to be employed as a filter layer in facemask materials. In addition, it was expected that the GO would gradually leach out over time. Consequently, the study proposes conducting further experiments to investigate the adherence of GO to the PP filter facemask.

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**Conflict of interest:** The authors state no funding.

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