

X-RAY STRUCTURE OF TRIS(DIISOBUTYLDITHIOCARBAMATO)PHENYLTIN(IV)

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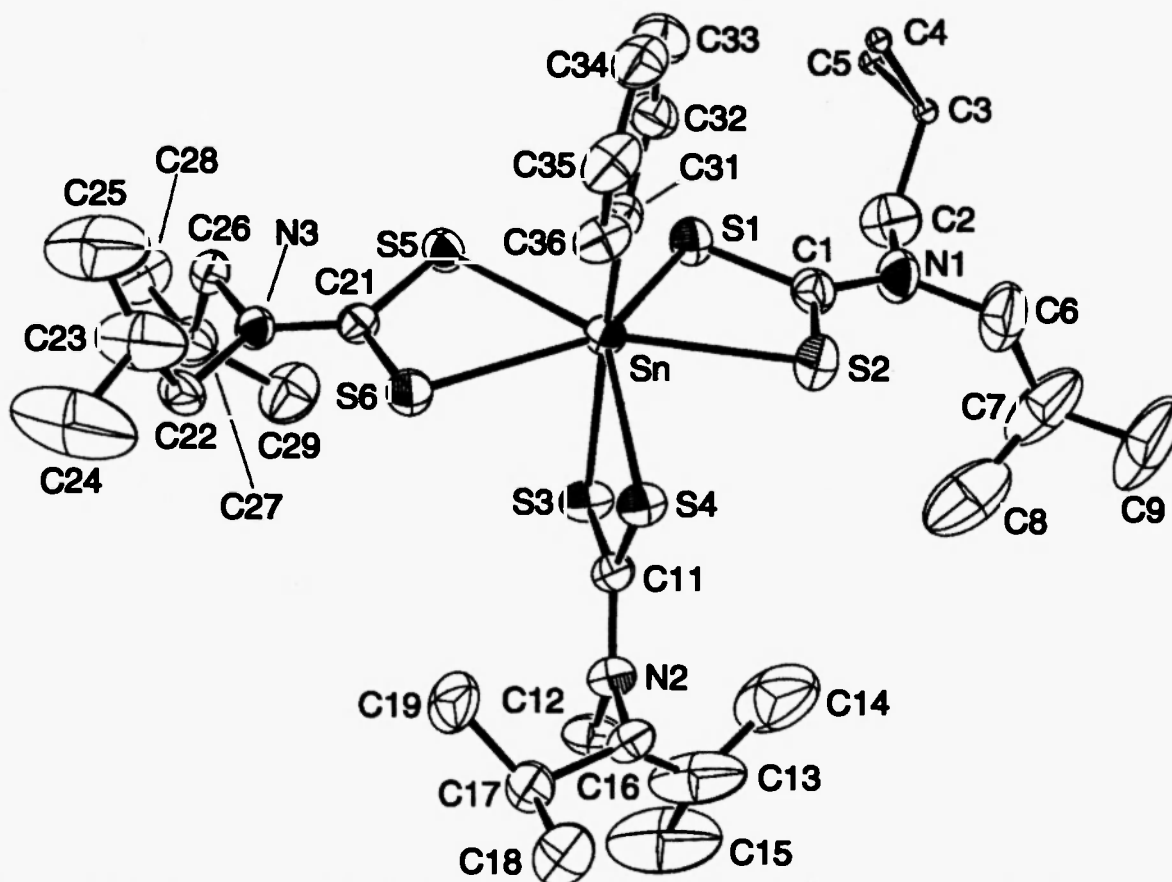


Figure 1. Molecular structure (50% displacement ellipsoids) of $[\text{PhSn}(\text{S}_2\text{CN}^i\text{Bu}_2)_3]$. Selected bond distances and angles: Sn-S1 2.615(2), Sn-S2 2.709(1), Sn-S3 2.491(1), Sn-S4 2.729(1), Sn-S5 2.642(1), Sn-S6 2.774(2), Sn-C31 2.152(4), S1-C1 1.731(4), S2-C1 1.701(5), S3-C11 1.737(4), S4-C11 1.713(4), S5-C21 1.721(4), S6-C21 1.703(4), C1-N1 1.331(6), N2-C11 1.338(5), N3-C21 1.342(5) Å; S1-Sn-S2 66.49(4), S1-Sn-S3 91.13(4), S1-Sn-S4 138.20(4), S1-Sn-S5 76.32(3), S1-Sn-S6 141.07(4), S1-Sn-C31 101.0(1), S2-Sn-S3 87.34(4), S2-Sn-S4 75.92(4), S2-Sn-S5 142.75(4), S2-Sn-S6 150.66(4), S2-Sn-C31 91.5(1), S3-Sn-S4 68.90(4), S3-Sn-S5 91.60(4), S3-Sn-S6 83.09(4), S3-Sn-C31 166.2(1), S4-Sn-S5 137.57(4), S4-Sn-S6 74.76(4), S4-Sn-C31 97.5(1), S5-Sn-S6 65.45(4), S5-Sn-C31 97.6(1), S6-Sn-C31 91.4(1), Sn-S1-C1 89.7(2), Sn-S2-C1 87.2(2), Sn-S3-C11 90.0(1), Sn-S4-C11 82.9(1), Sn-S5-C21 90.1(1), Sn-S6-C21 86.1(2), S1-C1-S2 116.6(3), S3-C11-S4 118.2(2) and S5-C21-S6 117.7(2)°.

Comment

The tin atom is coordinated by three bidentate dithiocarbamate ligands and the *ipso*-carbon atom of a phenyl group, and thereby exists in a CS_6 donor set that defines a distorted pentagonal bipyramidal geometry. In this description, the S(3) and C(31) atoms occupy axial positions so that the S(3)-Sn-C(31) angle is 166.19(12)°. The deviations of the S(1), S(2), S(4), S(5) and S(6) from the least-squares plane through them are 0.161(1), -0.296(1), 0.212(1), 0.042(1) and -0.181(1) Å, respectively with the Sn atom lying 0.3549(3) out of this plane in the direction of the

C(31) atom. The two dithiocarbamato ligands in the pentagonal plane form more symmetric Sn-S bonds than the dithiocarbamato ligand that spans equatorial and axial sites; the shortest Sn-S distance involves the sulfur atom in the axial position, *i.e.* S(3). The overall coordination geometry matches closely those found in related $\text{RSn}(\text{S}_2\text{CNEt}_2)_3$ systems [1], *i.e.* R = Ph [2, 3]; R = Me [4]; and R = ⁿBu [5].

Experimental

Preparation: The compound was prepared using a literature procedure [2]. Colourless crystals were obtained from the slow evaporation of a $\text{CH}_2\text{Cl}_2/\text{EtOH}$ (1/1) mixture; m.p. 155–156 °C. ^1H NMR: δ 7.79–8.40 m (H-*o*), 7.24–7.44 m (H-*m*, -*p*), 3.58 d (CH_2 , 7.2 Hz), 2.27–2.41 m (CH), 0.92 d (CH_3). ^{13}C NMR: δ 157.3, 130.8, 127.9, 128.2 (C α -C δ , phenyl), 202.4 (Cq), 63.1 (CH_2), 26.9 (CH), 20.2 (CH_3). ^{119}Sn NMR: δ -827.2 ppm. IR (KBr disk) $\nu(\text{C-N})$ 1475, $\nu(\text{C-S})$ 991 cm^{-1} .

Crystallography: The C(3)-C(5) atoms of a butyl group were found to be disordered so that two positions (50% site occupancy) were found for each of C(3) and C(5). The atoms of this group were refined isotropically and hydrogen atoms were not included. The residual electron density peak (1.42 $\text{e} \text{ \AA}^{-3}$) was located in this region. An empirical absorption correction was applied [6].

Table 1. Crystallographic data for $[\text{PhSn}(\text{S}_2\text{CN}^i\text{Bu}_2)_3]$

Formula	$\text{C}_{33}\text{H}_{59}\text{N}_3\text{S}_6\text{Sn}$	Formula weight	808.9
Crystal system	triclinic	Space group	$P\bar{3}21$
<i>a</i> , Å	12.556(6)	<i>b</i> , Å	16.888(5)
<i>c</i> , Å	10.381(2)	α , °	96.92(2)
β , °	106.92(2)	γ , °	78.00(2)
<i>V</i> , Å ³	2056(1)	<i>Z</i>	2
Crystal size, mm	0.08 x 0.15 x 0.37	Diffraction	Rigaku AFC7R
Temperature, K	173	$\mu(\text{Mo-K}\alpha)$, cm^{-1}	9.51
<i>D</i> _{calcd} , g cm^{-3}	1.307	<i>F</i> (000)	848
θ_{max} , °	27.5	No. reflns meas., unique	9849, 9422
No. reflns with $I \geq 2\sigma(I)$	5845	<i>R</i> , <i>wR</i> (F^2 , obs. data)	0.040, 0.098
<i>R</i> , <i>wR</i> (F^2 , all data)	0.101, 0.118	No. parameters	382
Weighting scheme	$w = 1/[\sigma^2(F_o^2) + (0.0489P)^2 + 1.6732P]$ where $P = (F_o^2 + 2F_c^2)/3$		
GoF	1.01	ρ , $\text{e} \text{ \AA}^{-3}$	1.42
Programs used	DIFABS [6], teXsan [7], DIRDIF [8], SHELXL97 [9], ORTEP [10]		
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