

THE CRYSTAL STRUCTURE OF DIMETHYLVINYLSILOXYGERMATRANE

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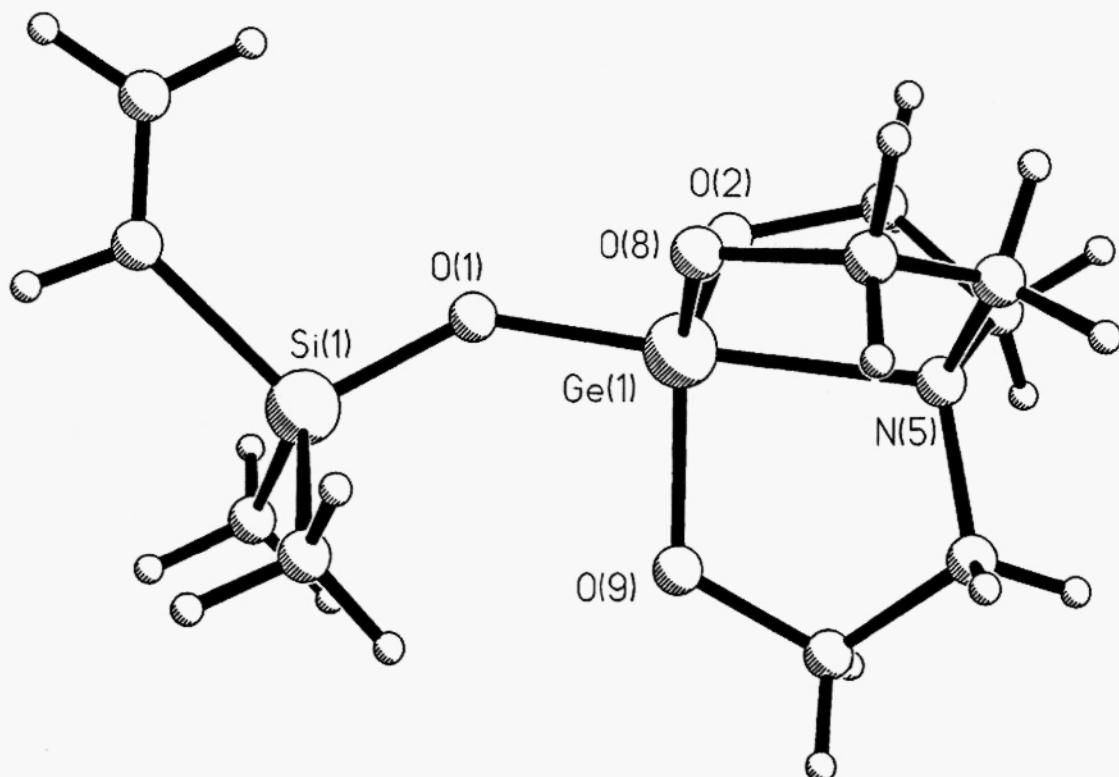


Figure 1. Molecular structure of dimethylvinylsiloxygermatrane

Comment

Over past two decades, germatranes have attracted considerable attention due to their five-coordinated structure. The intramolecular N \rightarrow Ge bond length in germatranes (2.01 – 2.29 Å) depends on the electronic effects of the substituents bound to the germanium atom [1 – 6]. Previously for the structural analogues of the title compound it was demonstrated that the intramolecular N \rightarrow Ge bond length of various thienylsiloxygermatranes changes from 2.105(4) Å to 2.156(6) Å [7].

Vinyldimethylethoxysilane reacts with anhydrous hydroxygermatrane in the presence of catalytic amount of sodium hydroxide to give the title compound in 82% yield. Its X-ray crystal structure (Figure 1) reveals that germanium atom is five-coordinated and adopts trigonal bipiramidal geometry as in other known germatranes. Three oxygen atoms are positioned in the equatorial fashion [O(8)-Ge(1)-O(9) 118.7(3) $^\circ$, O(8)-Ge(1)-O(2) 119.2(3) $^\circ$, O(9)-Ge(1)-O(2) 119.2(3) $^\circ$]. The siloxy group and the nitrogen atom occupy apical positions [N(5)-Ge(1)-O(1) 176.9(3) $^\circ$] with N \rightarrow Ge transannular bond of 2.132(7) Å. The Si(1)-O(1)-Ge(1) angle equals 136.5(4) $^\circ$. This value is slightly increased as compared with trimethylsiloxygermatrane [133.0 (4) $^\circ$] and lies between 134.8(3) $^\circ$ for (2-thienyl)dimethylsiloxygermatrane and 139.6(3) $^\circ$ for di(2-thienyl)methylsiloxygermatrane, being considerably different from 180. $^\circ$ [tri(2-thienyl)siloxygermatrane] [7].

Experimental

The mixture of vinyldimethylethoxysilane (0.02 mol) and hydroxygermatrane (0.02 mol) in dry xylene (40 ml) was refluxed for 5 hours. After cooling to room temperature the title product was filtered off as white solid (Yield 82%). Single crystals were grown from ethanol by slow evaporation of the solvent, mp 144 °C.

¹H NMR (CDCl₃) : 0.17 (6H, s, Me₂Si); 2.89 (6H, t, J=6.1 Hz, CH₂N); 3.86 (6H, t, J=6.1 Hz, CH₂O); 5.64-6.36 (3H, m, CH=CH₂).

²⁹Si NMR (CDCl₃) : 16.8 (Me₂Si).

Table 1. Crystal data and structure refinement for dimethylvinylsiloxygermatrane

Empirical formula	C ₁₀ H ₂₁ GeNO ₄ Si	Index ranges	0 ≤ h ≤ 7, 0 ≤ k ≤ 16, -15 ≤ l ≤ 15
Formula weight	319.96		
Temperature	293(2) K	Independent reflections	1905
Diffractometer	Syntex P2 ₁	Observed reflections, I>2_(I)	1491
Wavelength	0.71069 Å	Refinement method	Full-matrix least-squares on F ²
Crystal system	monoclinic	Restraints / parameters	0 / 155
Space group	P2 ₁ /c	Goodness-of-fit on F ²	1.063
Unit cell dimensions	a = 6.726(2) Å b = 15.276(5) Å c = 14.144(2) Å β = 95.83(2)°	Final R factors [I>_I] Weighting R on F ² Weighting scheme	R = 0.073 [I>_I], R = 0.087 (for all data) wR2 = 0.209
Volume	1445.7(6) Å ³		1/w = σ ² (F ²) + (0.1555P) ² + 0.49P, where P = (F _o ² + 2F _c ²)/3
Z	4	Extinction coefficient	0.024(5)
Density (calculated)	1.470 g/cm ³	Largest diff. peak and hole	1.09 and -0.91 e Å ⁻³
Absorption coefficient	2.204 mm ⁻¹	Programs used	AREN [8], SHELXL-93 [9] PLUTO [10]
F(000)	664		
Crystal size	0.50 x 0.50 x 0.75 mm		
2θ range for data	0 to 45°	Deposition number	CCDC 155997

References

1. E. Lukevics, S. Belyakov, P. Arsenyan, J. Popelis, *J. Organomet. Chem.*, **549** (1997) 163.
2. R. Eujen, E. Petrauskas, A. Roth, D.J. Brauer, *J. Organomet. Chem.*, **613** (2000) 86.
3. E. Lukevics, L. Ignatovich, S. Belyakov, *J. Organomet. Chem.*, **588** (1999) 222.
4. G. S. Zaitseva, M. Nasim, L. I. Livantsova, V. A. Tafeenko, L. A. Aslanov, V. S. Petrosyan, *Heteroatom. Chem.* **1** (1990) 439.
5. S. P. Narula, S. Soni, R. Shankar, R. K. Chadha, *J. Chem. Soc. Dalton Trans.*, (1992) 3055.
6. E. Lukevics, L. Ignatovich, S. Belyakov, L. Hohlova, *Khim. Heterotsikl. Soed.*, **2** (1997) 275.
7. E. Lukevics, S. Belyakov, L. Ignatovich, N. Shilina, *Bull. Soc. Chim. Fr.*, **132** (1995) 545.
8. V. I. Andrianov, *Kristallografiya*, **32** (1987) 228.
9. G. M. Sheldrick, *Crystallogr. Computing*, **6** (1993) 110.
10. S. Motherwell, W. Clegg, *PLUTO. Program for Plotting Molecular and Crystal Structures*. University of Cambridge, UK (1978).

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