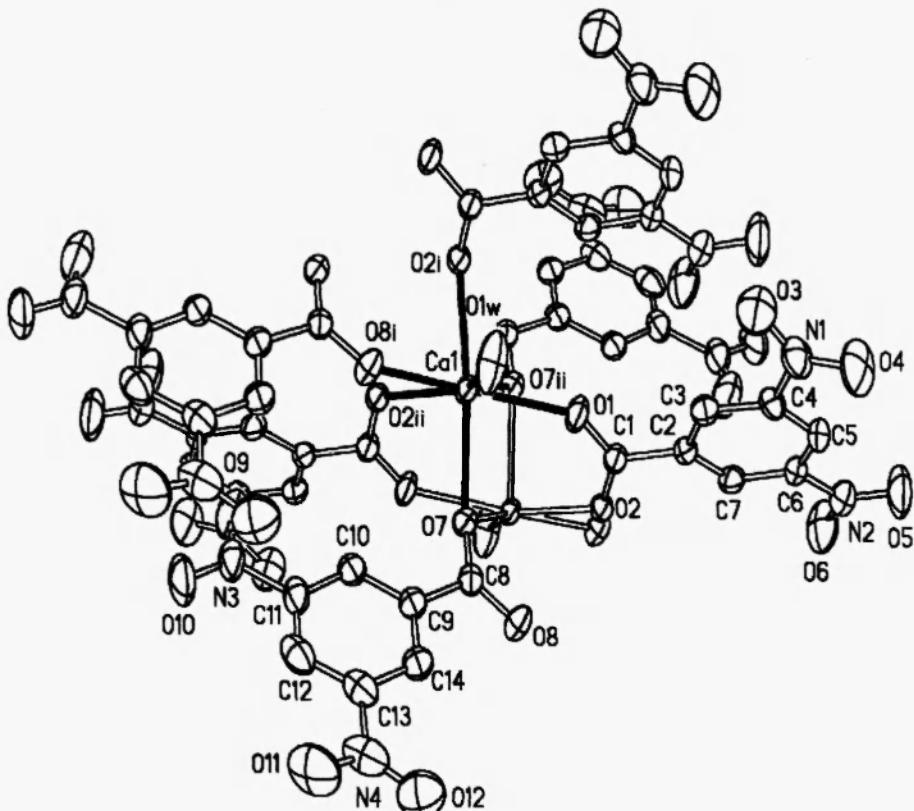


## CRYSTAL STRUCTURE OF *catena*-[AQUABIS(3,5-DINITROBENZOATO)CALCIUM(II)]

Guang Yang<sup>1</sup> and Seik Weng Ng\*<sup>2</sup>

<sup>1</sup> Department of Chemistry, University of Puerto Rico, Rio Piedras, P. O. Box 23346, San Juan, PR 00931-3346, USA

<sup>2</sup> Institute of Postgraduate Studies, University of Malaya, 50603 Kuala Lumpur, Malaysia



**Figure 1.** *ORTEP* plot illustrating the pentagonal bipyramidal geometry of the calcium atom in the title compound. Selected bond distances and angles: Ca1-O1 2.381(2), Ca1-O2<sup>i</sup> 2.368(2), Ca1-O2<sup>ii</sup> 2.485(2), Ca1-O7 2.373(2), Ca1-O7<sup>ii</sup> 2.498(2), Ca1-O8<sup>i</sup> 2.377(2), Ca1-O1w 2.377(3) Å; O1-Ca1-O2<sup>i</sup> 105.5(1), O1-Ca1-O2<sup>ii</sup> 145.9(1), O1-Ca1-O7 81.4(1), O1-Ca1-O7<sup>ii</sup> 74.1(1), O1-Ca1-O8<sup>i</sup> 141.5(1), O1-Ca1-O1w 72.5(1), O2<sup>i</sup>-Ca1-O2<sup>ii</sup> 84.0(1), O2<sup>i</sup>-Ca1-O7 162.9(1), O2<sup>i</sup>-Ca1-O7<sup>ii</sup> 81.3(1), O2<sup>i</sup>-Ca1-O8<sup>i</sup> 82.2(1), O2<sup>i</sup>-Ca1-O1w 89.2(1), O2<sup>ii</sup>-Ca1-O7 81.9(1), O2<sup>ii</sup>-Ca1-O7<sup>ii</sup> 75.3(1), O2<sup>ii</sup>-Ca1-O8<sup>i</sup> 71.4(1), O2<sup>ii</sup>-Ca1-O1w 141.2(1), O7-Ca1-O7<sup>ii</sup> 85.8(1), O7-Ca1-O8<sup>i</sup> 102.2(1), O7-Ca1-O1w 107.8(1), O7<sup>ii</sup>-Ca1-O8<sup>i</sup> 144.0(1), O7<sup>ii</sup>-Ca1-O1w 141.3(1), O8<sup>i</sup>-Ca1-O1w 69.9(1)°. Symmetry/translational code:  $i = x, 1+y, z$ ;  $ii = 0.5-x, 1.5-y, 1-z$ .

### Comment

The calcium atom in the title compound is bonded to six 3,5-dinitrobenzoato groups through the carboxyl oxygen atoms; the compound adopts a linear chain structure that propagates by two-fold screw axial translations. The pentagonal bipyramidal geometry is completed by the water ligand, which forms two relatively weak hydrogen bonds within the chain. The 3,5-dinitrobenzoate anion bridges three calcium atoms in a  $\mu$ - $O,O,O$  manner. In an earlier study on the cadmium complex, the 3,5-dinitrobenzoato entity was found to chelate to the metal atom in the tetra(3,5-dinitrobenzoato)cadmate dianion, whose negative charge is balanced by two sodium atoms [1].

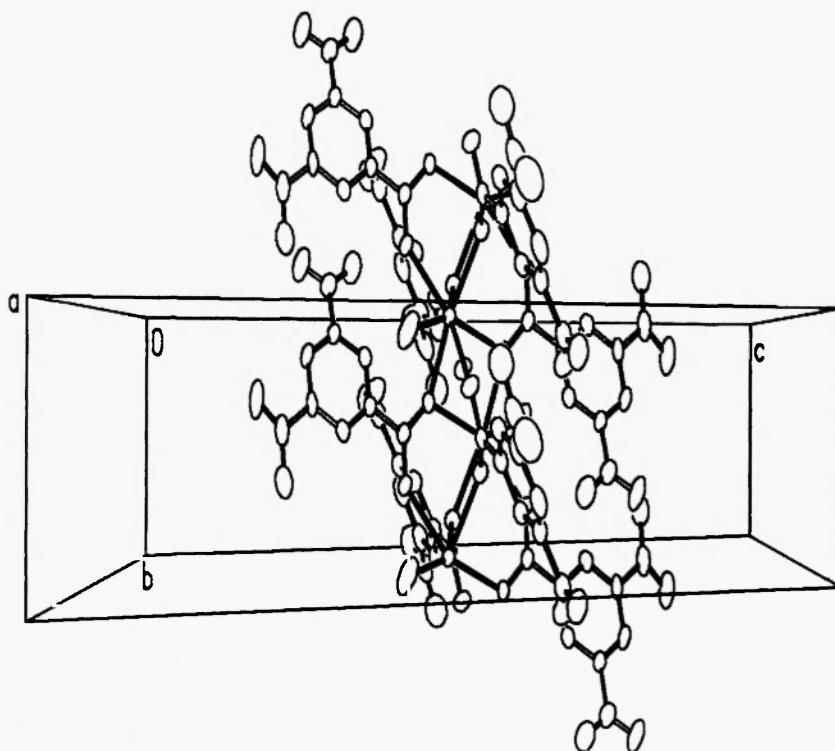
### Experimental

Ammonium hydroxide was added dropwise to a mixture of calcium nitrate tetrahydrate (1 mmol, 0.24 g), 3,5-dinitrobenzoic acid (2 mmol, 0.42 g) and water (10 cm<sup>3</sup>). When all had dissolved, the solution was filtered; slow evaporation of the solution gave colorless needles after two weeks. Calculated (Found) for C<sub>14</sub>H<sub>8</sub>CaN<sub>4</sub>O<sub>13</sub>: C 35.04 (35.01), H 1.80 (1.68), N 11.76% (11.66%). IR (KBr disc, cm<sup>-1</sup>): 3659s, 3565m.

3136w, 3110m, 3097w, 1620vs, 1584s, 1555s, 1530s, 1460m, 1388s, 1346vs, 1193w, 1089w, 1075w, 927w, 913w, 799m, 725s, 716s, 538w.

**Table 1.** Crystal data for calcium bis(3,5-dinitrobenzoate) hydrate

Empirical formula	C <sub>14</sub> H <sub>8</sub> CaN <sub>4</sub> O <sub>13</sub>	Formula weight	480.32
Space group	C <sub>2/c</sub>	Diffractometer	Siemens CCD
Unit cell dimensions	<i>a</i> 25.768(4) <i>b</i> 5.4660(8) <i>c</i> 28.304(4) Å β 114.271(3) °	Volume, Å <sup>3</sup>	3634(1)
μ (Mo-Kα), mm <sup>-1</sup>	0.431	<i>Z</i>	8
θ range, °	1.8 – 23.3	<i>D</i> <sub>calc</sub> , g cm <sup>-3</sup>	1.756
Reflections collected	7531	<i>F</i> (000)	1952
Reflections with <i>F</i> > 2σ( <i>I</i> )	1753	Temperature, °C	25
<i>R</i> (all data)	0.077	Independent reflections	2604 ( <i>R</i> <sub>int</sub> 0.053)
<i>wR</i> (all data)	0.093	No. parameters refined	297
<i>w</i>	[σ <sup>2</sup> + (0.0483 <i>P</i> ) <sup>2</sup> ] <sup>-1</sup>	Final <i>R</i> [ <i>F</i> > 2σ( <i>I</i> )]	0.040
Diff. hole and peak, eÅ <sup>-3</sup>	-0.25 – 0.26	<i>wR</i> [ <i>F</i> > 2σ( <i>I</i> )]	0.081
Programs	SHELX-97, ORTEP [2,3]	Goodness-of-fit on <i>F</i> <sup>2</sup>	0.95
		CCDC deposition no.	166889



**Figure 2.** Plot of the polymeric structure projected on the *a*–*c* plane.

#### Acknowledgments

We thank Dr. Peter Baran and Dr. Raphael G. Raptis the University of Puerto Rico for the use of the laboratory facilities, and the University of Malaya (F0758/2001A) for generously supporting this work. YG is grateful to Zhongshan University (China) and the Lingnan Foundation for a fellowship award.

#### References

- 1 H.-G. Zhu, G. Yang, X.-M. Chen and S.W. Ng, *Main Group Met. Chem.*, **24** (2001) 449.
- 2 G.M. Sheldrick, *SHELX-97*. Programs for the crystal structure analysis (Release 97-2). University of Göttingen (1997).
- 3 C.K. Johnson, *ORTEP-II*. Report ORNL-5138, Oak Ridge National Laboratory, Oak Ridge, Tennessee, USA (1976).