CRYSTAL STRUCTURE OF TRIBUTYLTIN N-PHTHALOYLLEUCINATE

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Abstract

The two symmetry-independent molecules of tributyltin N-phthaloylleucinate are linked by long tinamide bridges [Sn-O = 2.897(7), 2.960(7) Å] into a linear chain that propagates along the a-c diagonal of the monoclinic cell. The tin atoms are five-coordinate in trans-C₃SnO₂ trigonal bipyramidal geometries.

Discussion

The tributyltin derivative of N-phthaloylglycine exists as two symmetry-independent hydrates, and the coordinated water molecules engage in hydrogen bonding interactions that consolidate the crystal structure [1]. The incorporation of an i-propyl group in the protected amino acid appears to render the tin atom less Lewis acidic in the resulting carboxylate; unlike trialkyltin carboxylates that generally are linked by strong carboxylate bridges [2,3], the molecules of tributyltin N-phthaloylleucinate are connected by a long tin-amide interaction into a linear chain. A similar long interaction is also found in tricyclohexyltin N-phthaloylglycinate [4].

Table 1. Crystal data for tributyltin *N*-phthaloylleucinate.

Programs used	C ₂₆ H ₄₁ NO ₄ Sn Monoclinic 550.29 P2 ₁ a = 10.609(1), b = 23.023(1), c = 11.951(1) Å; β = 101.73(1)° 4 25°C 1.279 g cm ⁻³ 0.921 mm ⁻¹ 1144 CAD-4 (λ = 0.71073 Å) 1.74 to 24.98°. 0 ≤ $h \le 12, -27 \le k \le 0, -14 \le l \le 13$ 5455 5166 ($R_{int} = 0.028$) 0.724 - 0.803 5166 / 246 / 579 4278 0.042 (0.110) 0.057 (0.121) 1.039 - 0.371 to 495 SHELXS-97 [6] SHELXL-97 [7]
Programs used Deposition number	CCDC [6] SHELXL-97 [7]

Experimental

N-Phthaloylleucinate was synthesized by melting phthalic anhydride with leucine according to the procedure for preparing N-phthaloylglycine [5]. The acid and bis(tributyltin) oxide (2:1 molar stoichiometry) were heated in a small volume of ethanol to afford the organotin carboxylate, which was isolated and recrystallized from ethanol. A specimen, 0.22 x 0.14 x 0.14 mm, was used in the diffraction measurements. The refinement of the crystal structure necessitated imposing several restraints as the butyl groups in molecule a were disordered. Each butyl group was refined as two butyl groups sharing a common C_{α} atom. Distances in all butyl groups and in the two i-propyl groups were restrained (C-C = 1.54±0.01; C-C = 2.52±0.02 Å, and the temperature factors in these groups were restrained to be nearly isotropic by an ISOR 0.02 command.

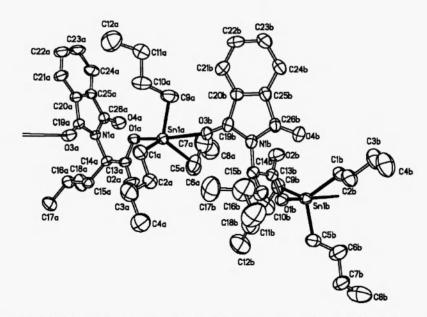


Figure 1. ORTEP [8] plot of tributyltin N-phthaloylleucinate. Selected bond distances and angles: Sn1a-C1a = 2.14(1), Sn1a-C5a = 2.13(1), Sn1a-C9a = 2.14(1), Sn1a-O1a = 2.068(6), Sn1a-O3b = 2.960(7), Sn1b-C1b = 2.14(1), Sn1b-C5b = 2.10(1), Sn1b-C9b = 2.12(1), Sn1b-O1b = 2.139(7), Sn1b- $O3a^1 = 2.897(7)$ Å; C1a-Sn1a-C5a = 119.0(5), C1a-Sn1a-C9a = 115.7(6), C1a-Sn1a-O1a = 102.7(4), C1a-Sn1a-O3b = 79.2(4), C5a-Sn1a-C9a = 115.3(6), C5a-Sn1a-O1a = 105.2(4), C5a-Sn1a-O3b = 77.2(4), C9a-Sn1a-O1a = 93.4(4), C9a-Sn1a-O3b = 81.9(4), O1a-Sn1a-O3b = 175.3(2), C1b-Sn1b-C5b = 114.3(6), C1b-Sn1b-C9b = 123.5(5), C1b-Sn1b-O1b = 100.7(4), C1b-Sn1b- $O3a^1 = 76.9(3)$, C5b-Sn1b-C9b = 115.7(7), C5b-Sn1b-O1b = 91.6(5), C5b-Sn1b- $O3a^1 = 89.0(4)$, C9b-Sn1b-O1b = 102.4(4), C9b-Sn1b- $O3a^1 = 79.4(4)$, O1b-Sn1b- $O3a^1 = 177.5(3)°. Translational code: <math>i = 1 + x$, y, 1 + z.

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